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Evaluation of potent isoquinoline-based thiosemicarbazone antiproliferatives against solid tumor models

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ABSTRACT: A potent class of isoquinoline-based α-N-heterocyclic carboxaldehyde thiosemicarbazone (HCT) compounds has been re-discovered; based on this scaffold we have synthesized three series of antiproliferative agents through iterative rounds of methylation and fluorination modifications. Synergy between isoquinoline fluorination and 4' amine methylation was identified, and incubation of the compounds with physiologically relevant levels of CuCl₂ was shown to further potentiate their activity. The lead compound, HCT-13, was highly potent against a panel of pancreatic, small cell lung carcinoma, and prostate cancer models, with IC₉₀ values in the low-to-mid nanomolar range. We show that the cytotoxicity of HCT-13 is copper-dependent, that it acts as a copper ionophore, induces production of reactive oxygen species (ROS), and promotes mitochondrial dysfunction and S-phase arrest. Lastly, DNA damage response/replication stress response (DDR/RSR) pathways, specifically Ataxia-Telangiectasia Mutated (ATM) and Rad3-related protein kinase (ATR), were identified as actionable adaptive resistance mechanisms following HCT-13 treatment. Taken together, HCT-13 is potent against solid tumor models and warrants *in vivo* evaluation against aggressive tumor models, either as a single agent or as part of a combination therapy.

INTRODUCTION

The diverse therapeutic potential of α -N-heterocyclic carboxaldehyde thiosemicarbazones (HCTs) have been investigated since the 1940s, with tuberculostatic activity being first observed *in vivo* as early as 1946. This class of compounds was subsequently shown to possess antitumor, antiviral, antibacterial, and antifungal activities, prompting decades of research and development. The particular, isoquinoline-based HCTs such as IQ-1 (HCT-1)³ were the subject of early interest due to their efficacy, particularly in terms of 50-day survival rates of tumor-bearing mice (Figure 1). The research groups of Sartorelli and French spent decades developing isoquinoline HCTs and investigating other HCT scaffolds, eventually turning their attention to pyridyl-based HCT analogs. Notably, in 1992 Sartorelli and coworkers developed 3-aminopyridine-2-carboxaldehyde thiosemicarbazone (3-AP, also known as Triapine), a pyridyl-based HCT. It has since undergone multiple clinical trials for the treatment of various cancers, and it is widely accepted to inhibit ribonucleotide reductase (RNR), a critical enzyme for rapidly proliferating cells such as bacteria and cancer cells. Two other HCT compounds, namely di-2-pyridylketone 4-cyclohexyl-4-methyl-3-thiosemicarbazone (DpC) hydrochloride and 4-(2-pyridinyl)-2-(6,7-dihydro-8(5H)-quinolinylidene)hydrazide-1-piperazinecarbothioic acid (COTI-2)^{15,16}, have been investigated in the clinic. However, despite the early promise, no HCT compounds have yet advanced beyond phase II clinical trials. However, despite the early promise, no HCT compounds have yet advanced beyond phase II clinical trials.

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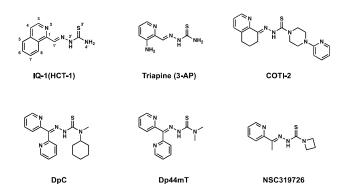


Figure 1. Structures of α-N-heterocyclic carboxaldehyde thiosemicarbazone (HCT) compounds of clinical and research significance.

While the mechanisms of action of HCTs are multi-modal and have not yet been fully defined ^{17–25}, their biological activities generally stem from the ability to chelate transition metals through their heterocyclic nitrogen, Schiff base nitrogen, and thiosemicarbazone sulfur. The resulting HCT-metal complexes can undergo redox cycles, a property that generates cytotoxic reactive oxygen species (ROS) through Fenton and/or Haber-Weiss processes. ²⁶ HCTs are particularly adept at binding copper ²⁷, which can be either detrimental or beneficial to the compound's biological activity. For instance, physiological concentrations of copper in human plasma (11-18 µM)^{28,29} strongly interfere with the RNR-inhibitory activity of 3-AP²⁷, while the cytotoxicities of Dp44mT^{30,31} and NSC-319726³² against glioblastoma and other cancer models are potentiated by copper. Binding of this transition metal is intriguing from an anticancer therapy standpoint, as cancers rely upon higher intracellular levels of copper, relative to healthy cells, to promote angiogenesis, tumor growth, and metastasis. ^{33,34} Indeed, several therapeutic strategies have employed small molecules to disrupt copper homeostasis in cancers, either through chelation-mediated copper sequestration, or by increasing intracellular copper to cytotoxic levels through ionophoric modalities. ^{32,35}

The antiproliferative effects of HCTs coupled with their ability to bind copper make them a compelling scaffold from which to develop copper-mediated therapeutics. We have identified isoquinoline-based HCTs as a viable scaffold for this purpose, as previously reported analogs demonstrated both *in vitro* and *in vivo* efficacy. Additionally, studies investigating the isoquinoline HCT chemical space have not emerged for years, as focus shifted away from this scaffold following the report of 3-AP.^{3,5,6} Therefore, we aimed to develop novel isoquinoline-based HCTs which leverage copper against malignancies. Herein, the design and synthesis of such a series of molecules is reported, with identification of a lead compound (HCT-13) which inhibits proliferation of pancreatic ductal adenocarcinoma (PDAC), small cell lung carcinoma (SCLC), and prostate cancer (PC) models at nanomolar concentrations in the presence of physiologically relevant levels of copper. We further demonstrate that HCT-13 increases intracellular copper levels, generates reactive oxygen species (ROS), and interferes with the mitochondrial electron transport chain. Additionally, we have identified actionable adaptive resistance mechanisms which could be exploited in combination therapies.

RESULTS

Synthesis of isoquinoline-based 2-N-heterocyclic carboxaldehyde thiosemicarbazone (HCT) compounds

Among the previously reported isoquinoline HCTs, the 5-, 7-, and 8-fluoro analogs of **HCT-1** were of particular interest to our group. These compounds varied in terms of potency and toxicity⁵, demonstrating that the effects of fluorination were dependent upon the position of isoquinoline substitution. The benefits that fluorination endows upon small-molecule drugs are well documented^{36,37}, and are also reflected in the marketplace, since approximately 20% of all pharmaceuticals are fluorinated. ³⁸ Therefore, it was important to investigate how the previously unreported 4- and 6-fluoro analogs of **HCT-1** would perform in antiproliferative assays against cancer models. Additionally, we were intrigued as to whether fluorination at these positions would synergize with 4' amine alkylation, as Kowol et al. discovered that 4' amine alkylation potentiated the cytotoxicity of 3-AP analogs.³⁹

Scheme 1. Synthesis HCT compounds 1-15 from simple isoquinolines^{a,b}

^a(a) allyl chloroformate, MeMgBr, THF; (b) Pd(PPh₃)₄, morpholine; DDQ, CH₂Cl₂; (c) SeO₂, 1,4-dioxane, 60 °C; (d) appropriate thiosemicarbazide, HCl, EtOH, reflux or microwave 50 °C; (e) KNO₃, H₂SO₄; (f) Fe, HCl, MeOH, reflux; (g) Boc₂O, DMAP, TEA, THF; (h) Boc₂O, DMAP, TEA, THF; NaHCO₃, MeOH, reflux or K₂CO₃, MeOH, reflux; (i) NaH, THF; MeI. ^bHCTs 1-3, 6-8, and 11-13 synthesized through Route A; HCTs 5 and 10 synthesized through Route B; HCTs 4, 9, 14, and 15 synthesized through Route C. See Supporting Information for full synthetic details.

Table 1. Substitution patterns of HCTs 1-15

	HCT-1	HCT-2	нст-з	HCT-4	HCT-5	НСТ-6	нст-7	нст-8	НСТ-9	HCT-10	HCT-11	HCT-12	HCT-13	HCT-14	HCT-15
R1	Н	F	Н	н	Н	н	F	Н	Н	н	н	F	н	н	F
R ²	Н	Н	F	н	Н	н	н	F	н	н	н	н	F	F	Н
R ³	Н	Н	Н	Н	Н	Me	Me	Me	Me	Me	Me	Me	Me	Me	Me
R ⁴	Н	Н	Н	н	Н	Н	Н	Н	Н	Н	Me	Me	Me	Me	Me
5-Pos	Н	н	н	NHMe	NH ₂	н	н	н	NH2	NHMe	н	н	н	NHMe	NHMe

A total of 15 isoquinoline-based HCTs - four known compounds (HCT-1, HCT-4, HCT-5, HCT-6) and 11 novel compounds - were synthesized and tested for antiproliferative potency against cancer models (Scheme 1, Table 1). The synthetic approach began with methylation of the appropriate isoquinoline 1 to generate 2 (Scheme 1). Depending upon the desired 5-position substituent, 2 was then subjected to either Route A (5-hydrido), Route B (5-amino), or Route C (5-methylamino). Syntheses of HCTs 1-3, HCTs 6-8, and HCTs 11-13 were carried out by Route A, wherein the methyl substituent of 2 was oxidized using selenium dioxide (SeO2) to furnish the carboxaldehyde 3. Condensation with the appropriate thiosemicarbazide under acidic conditions yielded the desired HCT. HCT-5 and HCT-10 were synthesized via Route B, which began with nitration of 2 followed by an iron-mediated reduction to the amine, which was subsequently Boc-protected and oxidized to produce carboxaldehyde 4. This intermediate was then simultaneously Boc-deprotected and condensed with the appropriate thiosemicarbazide under acidic conditions to furnish the target HCT. Syntheses of HCT-4, HCT-9, HCT-14, and HCT-15 via Route C proceeded from 2 with installation of a nitro group, subsequent conversion to the mono-Boc-methylamine, and SeO₂-mediated oxidation to furnish 5. Concurrent Boc-deprotection and thiosemicarbazide condensation were again achieved under acidic conditions to provide the desired HCT compound. While characterizing the HCTs, we occasionally observed the presence of a minor Z-isomeric product, particularly for HCTs 11-15. This isomer arose from an intramolecular hydrogen bond between the 2' amine of the thiosemicarbazone and the heterocyclic isoquinoline nitrogen, forming a stable 6-membered hydrogen bonded species. 40 The E and Z isomers were inseparable by reversed-phase HPLC purification and were used as a mixture in vitro, as previous studies reported no significant difference in potency.⁴⁰

Fluorination and dimethylation display synergistic effects further potentiated by copper supplementation

IC₉₀ values against MIAPACA2 cells for all compounds were first determined in normal cell culture conditions (DMEM media and 10% FBS) (**Figure 2**). Compounds were separated into three series – 4' primary amines, 4' secondary amines, and 4' tertiary amines – to reflect the relative degrees of 4' amine methylation. A series of non-methylated 4' primary amine compounds was first synthesized, with known compounds **HCT-1**, **HCT-4**, and **HCT-5** included to gauge whether fluorination of the isoquinoline proved

beneficial for biological activity. 41,42 Within the 4' primary amine series, fluorination at the isoquinoline 4-position (**HCT-2**) did not show an increase in potency relative to unsubstituted analog **HCT-1**. However, fluorination at the 6-position (**HCT-3**) showed a 3-fold increase in potency, demonstrating that the fluorine position impacts the potency of these compounds.

In the 4' secondary amine series, synergistic effect was observed when combining isoquinoline fluorination with 4' amine methylation. The 4-fluorine substituted **HCT-7** and the 6-fluorine substituted **HCT-8** were each significantly more potent than their non-fluorinated analog **HCT-6**, as well more potent than their 4' primary amine analogs (**HCT-2** and **HCT-3**, respectively). The trend of isoquinoline substitution and 4' secondary amine combining to enhance potency held for 5-methylamino substituted **HCT-9** and 5-amino substituted **HCT-10**. Taken together, these results suggested that combining isoquinoline substitution, particularly 4- or 6-fluorination, with 4' amine methylation produced synergistic antiproliferative effects when compared with either modification alone.

The effects of fluorine substitution became significantly more pronounced for the di-methylated 4'-tertiary amine compounds **HCT-12** and **HCT-13**, whose IC₉₀ values were in the nM range and were roughly 110- and 270-fold more potent, respectively, when compared to their non-fluorinated analog **HCT-11**. Fluorination at the 6-position (**HCT-13**) was found to be a superior modification when compared to fluorination at the 4-position (**HCT-12**), a trend which also held for the fluorine-substituted 5-methylamino compounds **HCT-14** and **HCT-15**.

Figure 2 | Cu(II) supplementation potentiates the activity of isoquinoline HCTs. MIAPACA2 cells were treated with the indicated HCT \pm 20 μ M Cu(II) for 72 h, then cell viability was measured with CellTiter-Glo to determine IC₉₀ values.

HCT compounds are known to be copper chelators, and a recent publication by Stockwell and coworkers demonstrated that the activity of a known 4' tertiary amine HCT (NSC-319726) was significantly potentiated by the addition of copper.³² To test whether our compounds were similarly potentiated, IC₉₀ (+Cu IC₉₀) values were determined against MIAPACA2 cells in media supplemented with physiologically relevant levels of copper (DMEM media + 10% FBS + 20 μM CuCl₂) (**Fig. 2**). While the activity of **HCT-5** was attenuated, all other compounds displayed a significant increase in potency under copper-supplemented conditions. For non-fluorinated isoquinoline compounds **HCT-1**, **HCT-6**, and **HCT-11**, the +Cu IC₉₀ values improved as the degree of methylation at the 4' amine increased (10-fold, 60-fold, and 400-fold increase in potency, respectively, versus non-copper-supplemented IC₉₀ values). Copper supplementation was similarly beneficial for fluorinated isoquinolines – all such compounds displayed significant improvements in antiproliferative potency in presence of copper, and fluorine substitution led to greater potency when compared with corresponding non-fluorinated analogs. Compounds bearing 4' tertiary amines were the most active, achieving +Cu IC₉₀ values as low as 21.6 nM (**HCT-13**). The results demonstrated that physiologically relevant levels of copper potentiated the activity of

isoquinoline HCTs, and that 4' amine methylation synergized with fluorine substitution. Due to its potency and straightforward synthesis, we chose **HCT-13** as a lead compound for in-depth mechanism of action studies.

HCT-13 highly potent against a panel of solid tumor models and induces copper-dependent cytotoxicity

Serum copper levels are elevated (>20 µM) in individuals with solid tumor types such as pancreatic ductal adenocarcinoma (PDAC), small cell lung carcinoma (SCLC), and prostate cancer (PC). ^{43–48} These cancers rely upon elevated copper levels to sustain growth, making this transition metal a viable target for therapeutic modulation. ³⁴ In a panel of PDAC, SCLC, and PC cancer models cultured in media supplemented with physiologically relevant levels of copper (20 µM CuCl₂), **HCT-13** was a highly potent growth inhibitor, with +Cu IC₉₀ values ranging from 1 nM to 200 nM (**Figure 3a**). The MIAPACA2 cell line, a well-characterized PDAC model that was highly sensitive to **HCT-13** treatment, was used to further investigate the mechanism of action of the lead compound.

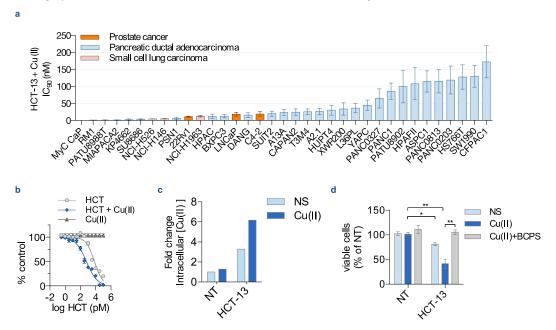


Fig. 3 | **Copper potentiates HCT-13 toxicity against a panel of cancer models.** (a) IC₉₀ in a panel of human and mouse prostate cancer (PC), small cell lung carcinoma (SCLC) and pancreatic ductal adenocarcinoma (PDAC) models treated with HCT-13 + Cu(II) (20 μM) for 72h measured by CellTiterGlo. (b) Proliferation rate of MIAPACA2 PDAC cells measured by CellTiterGlo following HCT-13 treatment for 72h \pm Cu(II) (20 μM), and with Cu(II) alone. (c) Intracellular concentrations of copper measured by inductively coupled plasma mass spectrometry (ICP-MS) in MIAPACA2 PDAC cells treated with HCT-13 (25nM) for 24h \pm Cu(II) (20 μM). (d) Inhibition of proliferation of MIAPACA2 cells treated with HCT-13 (25nM) + Cu(II) (20 μM) for 24h \pm bathocuproine disulfonate (BCPS, 300 μM) measured by trypan blue exclusion.

(mean \pm SD, n = 2, one-way ANOVA corrected for multiple comparisons by Bonferroni adjustment, * P < 0.05; ** P < 0.01; *** P < 0.001).

MIAPACA2 proliferation was measured in response to **HCT-13** in the presence and absence of 20 μM CuCl₂ (Cu(II)), as well as in response to 20 μM Cu(II) alone (**Figure 3b**). The potency of **HCT-13** improved by greater than 5-fold under Cu(II) supplemented conditions, with its IC₉₀ decreasing from 110 nM to 21 nM. Importantly, Cu(II) supplementation alone did not affect proliferation at all. To probe whether **HCT-13** was acting as an ionophore, intracellular copper levels were measured using inductively coupled plasma mass spectrometry (ICP-MS). In the presence of **HCT-13**, intracellular copper levels increased both with and without Cu(II) supplementation (**Figure 3c**). Additionally, treatment with bathocuproine disulfonate (BCPS), a membrane impermeable Cu(II) chelator, abrogated the cytotoxicity of **HCT-13** in the presence of Cu(II), suggesting that the growth inhibitory effect of our lead compound is largely dependent upon the availability of copper (**Figure 3d**). Collectively, this data suggests that **HCT-13** is a Cu(II) ionophore which increases intracellular copper concentration, and whose cytotoxicity is copper-dependent.

HCT-13 induces oxidative stress and ROS production

While **HCT-13** in the presence of Cu(II) exhibited nanomolar potency against a panel of solid cancer models, the underlying reasons remained unknown as the canonical mechanism of HCT cytotoxicity is poorly defined in the literature. Cu(II)-supplemented MIAPACA2 cells treated with **HCT-13** showed induction of AMPK phosphorylation (T172) at 24 h, demonstrating suppression of mitochondrial oxidative phosphorylation (**Figure 4a**). Further, **HCT-13** treatment increased heme oxygenase-1 (HO-1) levels in MIAPACA2 cells, indicative of ROS induction (**Figure 4a**). Based on the immunoblot results, we set out to determine whether **HCT-13** treatment was leading to ROS generation. We found that treatment with our lead compound induced ROS generation detectable by CM-H₂DCFDA staining (**Figure 4b**). Interestingly, ROS generation resulting from **HCT-13** treatment of MIAPACA2 cells was also observed in the mitochondria, as measured by mitochondria-specific dye MitoSOX (**Figure 4c**).

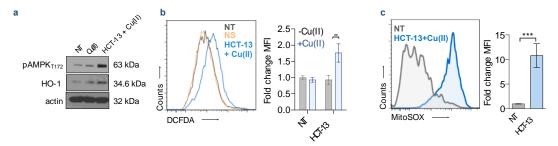


Figure 4 | **Induction of ROS by HCT-13.** (a) Representative immunoblots of MIAPACA2 PDAC cells treated as indicated for 24h. (b) Reactive oxygen species (ROS) measurement using CM-H2DCFDA staining after HCT-13 (25nM) + Cu(II) (20 μ M) treatment for 24h. (c) Mitochondrial ROS detection using MitoSOX staining in MIAPACA2 PDAC cells treated with HCT-13 (25 nM) + Cu(II) (20 μ M) for 24h.

(mean \pm SD, n = 2, Student t-test, ***P < 0.001).

HCT-13 induces mitochondrial dysfunction and displays mitochondria-dependent cytotoxicity

Considering the canonical role AMPK plays in energy homeostasis, and the implication that AMPK activation may signal mitochondrial dysfunction⁴⁹, we compared the metabolic status of MIAPACA2 cells treated with **HCT-13** for 24 hr with and without 20 µM Cu(II) using a Seahorse Bioscience XFe24 analyzer. In the presence of Cu(II), **HCT-13** significantly reduced both the basal respiration and maximum respiratory capacities of MIAPACA2 cells, indicating mitochondrial electron transport chain (mtETC) impairment (**Figure 5a**). *In vitro* mitochondrial complex activity following **HCT-13** treatment was dissected by an electron flow assay in isolated mitochondria, which showed decreased activity of complexes I and II (**Figure 5b**).

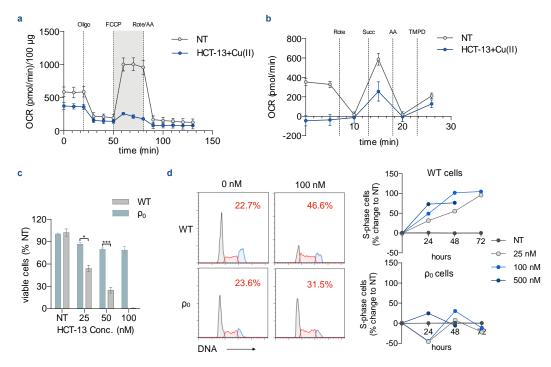


Figure 5 | HCT-13 alters cellular energetics through inhibition of electron transport chain and has selective mitochondrial toxicity. (a) Mito Stress Test of MIAPACA2 PDAC cells treated with HCT-13 (25 nM) + Cu(II) (20 μ M) for 24h. (b) Electron flow assay in isolated mitochondria treated with HCT-13 (100 nM) + Cu(II) (20 μ M) for 1h. (c) Viability of 143 BTK parental (wild type, WT) and ρ_0 cells after 48h of the indicated HCT-13 concentration + Cu(II) (20 μ M) treatment, assessed with Trypan Blue Staining. (d) 48h cell cycle histogram and plots of S-phase arrest plots in 143 BTK WT and 143 BTK ρ_0 cells at 24, 48 and 72h following treatment with indicated concentrations of HCT-13 + Cu(II) (20 μ M).

(mean \pm SD, n = 2, one-way ANOVA corrected for multiple comparisons by Bonferroni adjustment, * P < 0.05; ** P < 0.01; *** P < 0.001).

These findings suggested that **HCT-13** inhibited mtETC activity but did not indicate whether the lead compound's cytotoxicity stemmed from effects independent of mitochondrial function. Another HCT compound, Dp44mT, was reported to increase AMPK expression and induce ROS, although its cytotoxicity was not attributed to the functionality of the mitochondria. To determine whether the cytotoxicity of **HCT-13** was mitochondria-dependent, its effects upon 143 BTK ρ_0 , an mtDNA-deficient fibroblast cell line, were examined. Both 143 BTK ρ_0 and parental (wild type, WT) cells were treated with **HCT-13** + 20 μ M Cu(II) for 48 h, after which cell viability was determined through trypan blue staining (**Figure 5c**). Compared to WT, the ρ_0 cells were significantly less sensitive to the treatment. The cytotoxicity of **HCT-13** was partially abolished by supplementation with uridine (rU) but not by pyruvate, suggesting disruption to the supply of pyrimidine nucleotides in addition to impaired mitochondria (**Supplementary Figure S2a**). Proper mitochondrial function is necessary for the action of dihydroorotate dehydrogenase, an enzyme critical for the *de novo* production of pyrimidine nucleotides, and one for which **HCT-13** did not demonstrate affinity (**Supplementary Figure S2b**). Additionally, cell cycle analysis revealed marked S-phase arrest in 143 BTK WT cells but not in 143 BTK ρ_0 (**Figure 5d**). Taken together, these results indicate that the cytotoxic effects of **HCT-13** are mitochondria-dependent and suggest that **HCT-13** may be indirectly targeting DHODH, and thus *de novo* pyrimidine nucleotide production, through induction of mitochondrial dysfunction.

Small molecule kinase inhibitor screen with HCT-13 identifies the DNA-damage pathway as a resistance mechanism

To identify potential resistance mechanisms and synergistic interactions with **HCT-13**, we performed an unbiased pharmacological inhibition screen using a chemical genomics platform consisting of 430 kinase inhibitors (Selleckchem, Cat. L1200). MIAPACA2 cells were treated with the 430-member library, covering a 7-point concentration range spanning between 6.5 nM and 5 μM, with and without 25 nM **HCT-13** in presence of 20 μM CuCl₂. After 72 h of incubation, ATP content was measured using CellTiter-Glo (**Figure 6a**). A composite synergy score was calculated for each combination, defined as the sum of the Bliss Additivity Score (%

proliferation inhibition observed - % proliferation inhibition expected). A positive synergy score indicates greater-than-additive interaction, and a negative score indicates less-than-additive interaction i.e. antagonism.

The ten highest scoring compounds were kinase inhibitors contained in the DNA damage response/replication stress response (DDR/RSR) module, with the ATM/ATR/CHK pathway featuring as the most prominent codependency (**Figure 6b, 6c**). All six ATM, ATR and CHEK1 inhibitors included in the library scored positively in the screen, indicating synergy with our lead compound. Upon **HCT-13** + Cu(II) treatment, we consistently observed phosphorylation of the downstream targets of ATR and ATM, CHEK1 and CHEK2, respectively, indicating activation of this pathway as an adaptive resistance mechanism. This phosphorylation was accompanied by induction of DNA damage marker pH2AX and by cleavage of apoptotic marker caspase 3 (**Figure 6d**). The synergistic interaction between **HCT-13** and ATR inhibition was further validated using cell death assays measured by Annexin V/PI and Trypan Blue Viability Staining in PDAC (MIAPACA2, CFPAC-1) and PC (C4-2) cell lines (**Figure 6E, 6F, and Supplementary Figure S3b, S3c**).

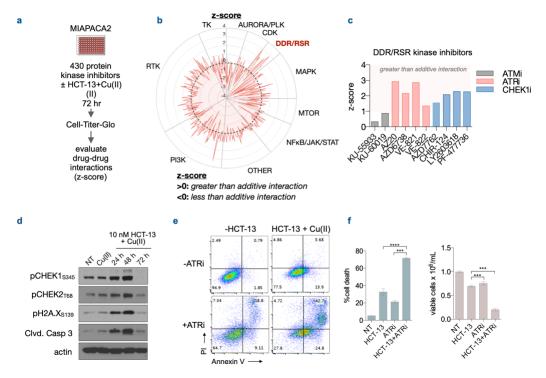


Figure 6 | A chemical genomics screen identifies the replication stress response pathway as an actionable co-dependency of HCT-13-treated cells. (a) Experimental design of a synthetic lethality screen using a library of protein kinase inhibitors against HCT-13-treated cells in the presence of Cu(II) (20 μ M). (b) Radar plot of screen results. (c) z-score values for kinase inhibitors within the DNA damage response/replication stress response (DDR/RSR) pathway module. (d) Representative immunoblot of replication stress and cell death biomarkers in MIAPACA2 PDAC cells treated with HCT-13 (10 nM) + Cu(II) (20 μ M). (e) Annexin V/PI staining in MIAPACA2 cells to validate the synergistic interaction of HCT-13 (25nM) + Cu(II) (20 μ M) with ATRi (250nM VE-822) treated for 72h. (f) Trypan Blue Viability staining in MIAPACA2 cells to validate the synergistic interaction of HCT-13 (25nM) with ATRi (250nM VE-822) treated for 72h in presence of Cu(II) (20 μ M).

(mean \pm SD, n = 2, one-way ANOVA corrected for multiple comparisons by Bonferroni adjustment, * P < 0.05; ** P < 0.01; *** P < 0.001; **** P < 0.0001).

DISCUSSION

We primarily focused on two modifications of the HCT-1 (IQ-1) scaffold during our synthetic planning: fluorination of the isoquinoline ring, and sequential methylation of the 4' amine. Though the 5-, 7-, and 8-fluoro-substituted analogs of HCT-1 were previously reported, and while it was apparent that fluorine position could influence a compound's toxicity and produce differential antiproliferative effects upon various cell lines, no trends had previously emerged regarding the effects of fluorine position.⁵ Additionally, multiple groups have shown that 4' amine methylation potentiates the activity HCTs such as pyridine 2-carboxyaldehyde thiosemicarbazones and 2-acetylpyridine thiosemicarbazones. ^{24,39,40} We therefore sought to synthesize the novel 4- and 6-fluoro analogs of HCT-1 and investigate what effects sequential 4' amine methylation had upon antiproliferative activity. Following our isoquinoline HCT synthetic campaign, analysis of the antiproliferative data revealed several trends. Without copper supplementation, fluorination at either the 4- or 6-position of the isoquinoline ring led to an increase in potency for five out of six compounds, when compared with their corresponding non-fluorinated analogs (Figure 2). In some cases, the change was dramatic – for example, the IC₉₀ of HCT-13 was nearly 270-fold lower than its non-fluorinated analog HCT-11. Secondly, 4' amine methylation in the absence of isoquinoline substitution or copper supplementation was detrimental to activity, as demonstrated by the decrease in potencies for HCTs 6 and 11 when compared with 4' primary amine HCT-1 (Figure 2). However, combining 4' amine methylation and isoquinoline substitution in a single compound, as in HCTs 7-10, 12 and 13, produced synergistic antiproliferative effects when compared with either their 4' primary amine or unsubstituted isoquinoline analogs. HCT-13 again exemplified this trend, with potency nearly 270-fold greater than its non-substituted isoquinoline analog HCT-11 and nearly 50-fold greater than its non-methylated analog HCT-3 (Figure 2). The underlying mechanisms responsible for this synergy remain under study by our group. Finally, the antiproliferative activities of all but one of our isoquinoline HCTs were potentiated by supplementation with physiologically relevant levels of copper, with HCT-5 being the exception. The observed trends culminated with identification of our lead compound as HCT-13, which was uncomplicated in its synthesis and potent both in the absence and presence of copper supplementation.

The potency of HCT-13 is highlighted by its nanomolar IC90 values against a panel of PDAC, SCLC, and PC cancer models in the presence of physiologically relevant levels of copper (Figure 3a). The use of copper-chelating small molecules in anticancer therapy is an established strategy which is executed either through sequestration of copper from tumor tissue, or through increasing intracellular copper to cytotoxic levels.⁵¹ HCT-13 behaved as an ionophore and increased intracellular levels of copper, both in the presence and absence of copper supplementation. This property is essential for the cytotoxicity of HCT-13, as sequestration of copper via BCPS-chelation negated our lead compound's growth inhibitory effects (Figure 3d). We further demonstrated that HCT-13 leverages copper to effect its cytotoxicity in a mitochondria-dependent manner. Specifically, the data indicate that HCT-13 induces mitochondrial dysfunction and mitochondria-dependent S-phase arrest, and generates ROS and oxidative stress in different cancer models. Strikingly, mitochondria-deficient 143 BTK ρ₀ cells were significantly less sensitive to **HCT-13** in the presence of copper compared to their parental 143 BTK WT counterpart, providing further evidence of mitochondria-dependent cytotoxicity (Fig 5c). It is possible that the observed S-phase arrest results from disruption of the de novo pathway (DNP) for pyrimidine nucleotide biosynthesis, which supplies cells with the pyrimidine nucleotides necessary for replication. The lone oxidation step of the pyrimidine DNP is carried out by dihydroorotate dehydrogenase (DHODH), an enzyme located in the inner mitochondrial membrane, which utilizes ubiquinone as a redox partner. Without a properly functioning mitochondrial ETC, DHODH does not have access to the levels of ubiquinone necessary for the oxidative enzyme to adequately turn-over, leading to shortages in pyrimidine nucleotides and corresponding S-phase arrest. 52-57

In general, cancer cells exhibit higher levels of ROS and higher baseline oxidative stress than healthy cells, which may imbue our lead compound with selectivity towards them. The ROS and mitochondrial dysfunction produced in MIAPACA2 cells by HCT-13 lead to an increase in DNA damage marker pH2AX, which may explain why our compound synergized with inhibitors of ATR (Ataxia-Telangiectasia Mutated (ATM) and Rad3-related protein kinase), the most upstream kinase in the DNA-damage response/replication stress response (DDR/RSR) pathway. Synergy with DDR/RSR inhibitors may increase the therapeutic window of our lead compound, should it be administered in combination therapy. The observed mechanisms of action of HCT-13 suggest that

it may also synergize with radiotherapy, as therapeutic ionizing radiation increases ROS, thereby increasing oxidative stress and DNA damage in the targeted area(s). Therefore, **HCT-13** could also function as a radiosensitizer by further increasing the load of ROS, oxidative stress, and DNA damage when administered in combination with radiation therapy. Taken together, the potency of **HCT-13** as a single-agent therapeutic against aggressive solid tumor models, its mechanism of action, and the observed synergy with ATR inhibitors warrant further testing *in vivo*.

CONCLUSION

A class of isoquinoline-based HCTs has been expanded upon to produce a set of novel antiproliferative compounds. The synergistic effects of combining 4' amine methylation with isoquinoline substitution were demonstrated, and **HCT-13** was identified as a highly potent antiproliferative which is active against a panel of PDAC, SCLC, and PC cancer models. The presence of physiologically-relevant levels of Cu(II) greatly potentiated our lead compound's activity, and subsequent investigation into **HCT-13**'s mechanism of action revealed that it acts as a copper ionophore and requires copper to effect its cytotoxicity. Furthermore, **HCT-13** induces ROS production, oxidative stress, S-phase arrest, and mitochondrial dysfunction which may contribute to indirect inhibition of DHODH. Lastly, a high-throughput phenotypic screen of protein kinase inhibitors was used to identify actionable adaptive resistance mechanisms of **HCT-13**-treated cells, and the DDR/RSR pathways were revealed as actionable vulnerabilities. Specifically, we show that ATR inhibition synergizes with **HCT-13** in presence of physiologically-relevant levels of Cu(II). Taken together, this study demonstrates the potential of **HCT-13** for use in anti-cancer therapy, either as a single agent or as part of a larger combination therapy.

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Materials and Methods

General

All chemicals, reagents and solvents were obtained from commercial sources and were used without further purification. Unless otherwise noted, reactions were carried out in oven-dried glassware under an atmosphere of argon using commercially available anhydrous solvents. Tetrahydrofuran (THF) was distilled from sodium under an argon atmosphere. Dichloromethane was distilled from calcium hydride. Solvents used for extractions and chromatography were not anhydrous. Analytical TLC was carried out on precoated silica gel (Merck silica gel 60, F254) and visualized with UV light. Column chromatography was performed with silica (Fisher, 230-400 mesh). ¹H NMR, ¹³C NMR, and ¹⁹F NMR spectra were measured in CDCl₃ or DMSO-d₆ on Bruker AV spectrometers at 400 or 500 MHz. Chemical shifts were reported in parts per million (δ) relative to residual solvent signals. The signals observed were described as follows: s (singlet), d (doublet), t (triplet), q (quartet), dd (doublet of doublets), dt (doublet of triplets), ddd (doublet of doublet of doublets), tt (triplet of triplets), tdd (triplet of doublet of doublets), m (multiplet), br s (broad singlet). Mass spectra were obtained on a Waters LCT Premier with ACQUITY UPLC mass spectrometer under electrospray ionization (ESI) or Thermo Fisher Scientific Exactive Plus with direct analysis in real time (DART) ionization. Purity of all compounds used in biological assays was determined on a Hewlett Packard 1090 HPLC system using an Aquasil C18 column (250 mm × 2 mm, 5 µm, Keystone Scientific) with an acetonitrile/water solvent system containing 0.1% TFA with detection performed at 254 nm (minute/%acetonitrile: 0/0, 8/0, 35/95, 43/95, 45/0, 55/0). HPLC purification was performed on a Hewlett Packard 1090 HPLC system with Hypersil Gold column (250 mm × 10 mm, 5 μm, Thermo Scientific) with and acetonitrile/water solvent system containing 0.05% formic acid and 10 mM ammonium formate. All microwave-assisted reactions were carried out in a CEM Discover 908005 Microwave synthesizer system.

Cell culture and culture conditions

Pancreatic adenocarcinoma cell lines: PATU8988T, MIAPACA2, SU8686, PSN1, HPAC, BXPC3, DANG, SUIT2, A13A, CAPAN2, T3M4, A2.1, HUPT4, XWR200, L36PL, YAPC, PANC0327, PANC1, PATU8902, HPAF11, ASPC1, PANC0813, PANC0203, HS766T, SW1990, and CFPAC1; prostate cancer cell lines: 22Rv1, LNCaP, RM1 and C4-2; and small cell lung carcinoma cell lines: NCI-H526, NCI-H146, and NCI-H1963 were obtained from American Type Culture Collection (ATCC). 143 BTK WT and 143 BTK ρ_0 , BJ WT and BJ ρ_0 cells were gifts from Prof. Michael Teitell in UCLA. Murine Prostate cancer cell line MyC CaP was a kind gift from Prof. DLJ Thorek at WUSTL. Murine Pancreatic cancer cells KP4662 was a kind gift from Prof. Robert Vonderheide at UPenn. With a few exceptions, cell lines were cultured in DMEM (Corning) or RPMI (Corning) containing 10% fetal bovine serum (FBS, Omega Scientific) and were grown at 37 °C, 20% O_2 and 5% CO_2 . All cultured cells were incubated in antibiotic free media and were regularly tested for mycoplasma contamination using MycoAlert kit (Lonza) following the manufacturer's instructions, except that the reagents were diluted 1:4 from their recommended amount.

Proliferation assay

Cells were plated in 384-well plates (500 cells/well for adherent cell lines in 30 μ l volume). Drugs were serially diluted to the desired concentrations and an equivalent volume of DMSO was added to vehicle control. Following 72 h incubation, ATP content was measured using CellTiter-Glo reagent according to manufacturer's instructions (Promega, CellTiter-Glo Luminescent Cell Viability Assay), and analyzed by SpectraMax luminometer (Molecular Devices). IC₅₀ and IC₉₀ values, concentrations required to inhibit

proliferation by 50% and 90% respectively compared to DMSO treated cells, were calculated using Prism 6.0 h (Graphpad Software).

Western blot

Cells were lysed using RIPA buffer supplemented with protease (ThermoFisher, 78,430) and phosphatase (ThermoFisher, 78,420) inhibitors, scraped, sonicated, and centrifuged (20,000 × *g* at 4 °C). Protein concentrations in the supernatant were determined using the Micro BCA Protein Assay kit (Thermo), and equal amounts of protein were resolved on pre-made Bis-Tris polyacrylamide gels (Life Technologies). Primary antibodies: pAMPK_{T172} (Cell signaling, #2535, 1:1000), HO-1 (Cell signaling, #5061S, 1:1000), pS345 CHEK1(Cell signaling, #2348L, 1:1000), pT68 CHEK2 (Cell signaling, #2197 S, 1:1000), pS139 H2A.X (Millipore, 05-636, 1:1000), clvd. Casp3 (Cell signaling, #9662, 1:1000), and anti-actin (Cell Signaling Technology, 9470, 1:10,000). Primary antibodies were stored in 5% BSA (Sigma-Aldrich) and 0.1% NaN₃ in TBST solution. Anti-rabbit IgG HRP-linked (Cell Signaling Technology, 7074, 1:2500) and anti-mouse IgG HRP-linked (Cell Signaling Technology, 7076, 1:2500) were used as secondary antibodies. Chemiluminescent substrates (ThermoFisher Scientific, 34,077 and 34,095) and autoradiography film (Denville) were used for detection.

Viability/Apoptosis assay

Viable cells were measured by Trypan blue staining using vi-cell counter (Beckman Coulter, CA, USA). Apoptosis and cell death were assayed using Annexin V-FITC and PI according to manufacturer's instructions (FITC Annexin V Apoptosis Detection Kit, BD Sciences, #556570).

Cell cycle

Cell cycle was assessed using Propidium iodide staining at indicated timepoints. Cells were pulsed with EdU 1 h before collection at different time points. Cells were fixed 4% paraformaldehyde, permeabilized with perm/wash reagent (Invitrogen), stained with Azide-AF647 (using click-chemistry, Invitrogen; Click-iT EdU Flow cytometry kit, #C10634) and FxCycle-Violet (Invitrogen), and then analyzed by flow cytometry (a detailed description is available in the Supplementary Information).

ROS Measurements

Cellular ROS measurement was assayed with CM-H2DCFDA staining after treatment according to manufacturer's instructions (Reactive Oxygen Species (ROS) Detection Reagents, Invitrogen, #D399). The cells were then incubated with 5 μ M of CM-H2DCFDA for 30 min, spun down at 450 x g for 4 mins, and the supernatant was replaced with fresh media containing lethal compounds and/or Cu(II). Then, the cells were incubated for 30 mins, spun down, and the supernatant was replaced with PBS. The samples were analyzed using flow cytometry.

Mitochondrial ROS was measured using MitoSOX staining according to manufacturer's instructions (MitoSOX, Invitrogen, #M36008). Cells were treated with HCT-13, washed and treated with MitoSOX. Cells were then incubated for 30 minutes at 37°C. After incubation, media is aspirated and cells are washed with PBS and analyzed by flow cytometry.

Mito Stress Test and Electron Flow Assay

Oxygen consumption rate (OCR) was measured using a XF24 Analyzer (Agilent) and normalized per μg protein. For cellular OCR, cells were incubated in unbuffered DMEM containing 25 mM glucose, 1 mM pyruvate and 2 mM glutamine. OCR was measured before (total respiration) and after the sequential injection of 1 μ M oligomycin (complex V inhibitor), 0.75 μ M FCCP (uncoupler), and 1 μ M of rotenone and myxothiazol (complex I and III inhibitors, respectively), as described previously. Mitochondrial respiration was calculated by subtracting the non-mitochondrial respiration left after rotenone and myxothiazol injection. Oligomycin-sensitive respiration represents ATP-linked respiration (coupled respiration).

To measure electron transport chain complex activity from cells, cells were incubated in MAS buffer with 10 mM pyruvate (complex I substrate), 2 mM malate, 4 μ M FCCP, 4 mM ADP, and 1 nM of XF Plasma Membrane Permeabilizer (PMP) reagent (Agilent). OCR was measured before and after the sequential injection of 2 μ M rotenone, 10 mM succinate (complex II substrate), 4 μ M antimycin A (complex III inhibitor), and a mix of 10 mM ascorbate and 100 μ M TMPD (complex IV substrates), as described previously (2). Antimycin A-sensitive respiration represents the complex III respiration.

To measure OCR directly from mitochondria, mitochondria were isolated from fresh mouse liver by dual centrifugation at 800g and 8000g and seeded by centrifugation (2). Mitochondria were incubated with 1 mM pyruvate (complex I), 2 mM malate, 4 μ M FCCP in MAS buffer, as well as the "corresponding drugs" for 30 min at 37°C. OCR was measured before and after the sequential injections described in the previous paragraph.

Intracellular Cu(II) measurement

Cells were plated in 6-well plates and cultured for one day. Vehicle of **HCT13** were added to the cells the following day and incubated for 24 hours. The plates were then washed 2 times with PBS containing 1 mM EDTA and 2 times with PBS alone. The concentration of Cu(II) was measured using Inductive Coupled Plasma Mass Spectrometry (ICP-MS) using standard procedure.

DHODH activity

Recombinant protein was incubated in an aqueous solution (total volume, 1.0 mL) containing 500 μ M DHO (Sigma, D1728), 200 mM K₂CO₃-HCl (pH 8.0), 0.2% triton x-100, and 100 μ M coenzyme Q10 (Sigma, C9538) at 37 °C for 0, 15, 30, 45, or 60 min. An aliquot (100 μ L) of the mixture of enzyme reaction mixture or cell/tissue lysate was mixed with 100 μ L of 0, 0.5, or 1.0 μ M orotic acid, 50 μ L of H2O, 250 μ L of 4.0 mM 4-TFMBAO (Sigma, 422231), 250 μ L of 8.0 mM K₃[Fe(CN)₆] (Sigma, 244023), and 250 μ L of 80 mM K₂CO₃ (Sigma, P5833) and then heated at 80 °C for 4.0 min. The reaction was stopped by cooling in an ice-water bath and the absorbance was measured with a spectrofluorometer (FP-6300 Jasco, Tokyo, Japan): excitation and emission wavelengths were 340 nm and 460 nm, respectively.

FACS analyses

All flow cytometry data were acquired on a five-laser LSRII cytometer (BD), and analyzed using the FlowJo software (Tree Star).

Supplementary Figures

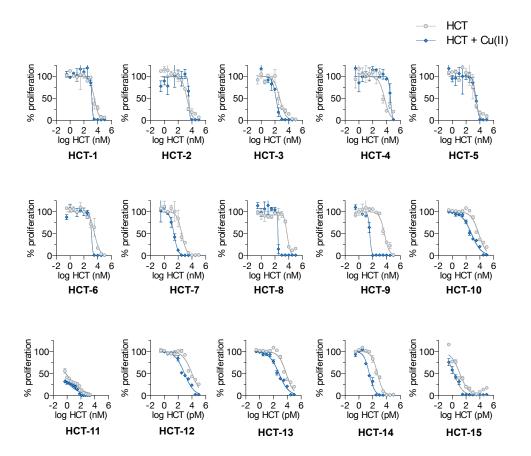


Fig. S1 | Summary of HCT compound dose response curves. Viability curves measured with CellTiterGlo in MIAPACA2 cells treated with each HCT \pm Cu(II) (20 μ M) for 72h.

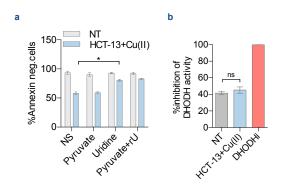
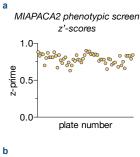


Fig. S2 | Cell proliferation inhibition induced by HCT-13 is partially rescued by uridine supplementation. (a) Rescue of HCT-13 (25 nM)-induced cell death by Uridine (rU) (200 μ M), Pyruvate (1 mM), or both following 48h of treatment (b) Measurement of DHODH activity using recombinant DHODH assay following treatment with indicated perturbations for 2 min. DHODH inhibitor used: NITD-982 - 1 μ M; HCT-13 - 100 nM, 1 μ M, (100 nM data shown); Cu(II) - 20 μ M. (mean \pm SD, n = 2, Student t-test, * P < 0.05).



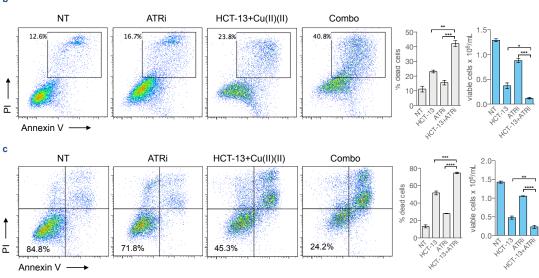


Fig. S3 | Identification of resistance mechanisms to HCT-13 using a synthetic lethality screen (a) Assay quality, as measured by Z-factor (Z') scores (b and c) Annexin V/PI staining and Trypan Blue staining in CFPAC-1 PDAC cells and C4-2 PC cells to validate the synergistic interaction of HCT-13 with ATRi (250 nM VE-822) treated for 72h in presence of Cu(II) (20 μ M).

(mean \pm SD, n = 2, one-way ANOVA corrected for multiple comparisons by Bonferroni adjustment, * P < 0.05; ** P < 0.01; *** P < 0.001; **** P < 0.0001).

Synthesis of compounds S1-S22 and HCT1-HCT16

Isoquinoline-1-carboxaldehyde (S1). To a solution of 1-methylisoquinoline (1.0 g, 6.98 mmol) in 1,4-dioxane (10 mL) was added selenium dioxide (0.930 g, 8.38 mmol) and the mixture was refluxed for 4 h. The mixture was filtered, then concentrated *in vacuo*. The crude residue was purified by column chromatography (25% DCM:Hexanes) to give the product **S1** as a taupe powder (0.840 g, 69% yield). ¹H NMR (500 MHz, DMSO- d_6) δ 10.28 (s, 1H), 9.15 (ddd, J = 7.7, 1.9, 0.8 Hz, 1H), 8.82 (d, J = 5.5 Hz, 1H), 8.21 (dd, J = 5.6, 0.9 Hz, 1H), 8.17–8.12 (m, 1H), 7.93–7.84 (m, 2H). ¹³C NMR (125 MHz, DMSO- d_6) δ 195.64, 149.38, 142.47, 136.49, 131.00, 130.30, 127.45, 125.77, 125.41, 124.73. DART-MS: m/z calcd. for $C_{10}H_8NO$ (M+H) ⁺ 158.06004, found 158.05977.

(*E*)-2-(isoquinolin-1-ylmethylene)hydrazine-1-carbothioamide (HCT1): Synthesized from S1 as previously reported. HNMR (500 MHz, DMSO- d_6) δ 11.74 (s, 1H), 9.19 (d, J = 8.5 Hz, 1H), 8.60–8.54 (m, 2H) 8.49 (br s, 1H), 8.02 (d, J = 8.1 Hz, 1H), 7.86 (d, J = 5.6 Hz, 1H), 7.84–7.78 (m, 2H), 7.75 (ddd, J = 8.3, 6.8, 1.4 Hz, 1H). CNMR (125 MHz, DMSO- d_6) δ 178.41, 150.78, 145.99, 142.13, 136.24, 130.47, 129.08, 127.22, 126.94, 125.58, 121.77. DART-MS: m/z calcd. for C₁₁H₁₀N₄S (M+H)⁺ 231.06989, found 231.06938.

(*E*)-2-(isoquinolin-1-ylmethylene)-*N*-methylhydrazine-1-carbothioamide (HCT6): To a solution of S1 (0.060 g, 0.382 mmol) in ethanol (3 mL) was added 4-methyl-3-thiosemicarbazide (0.040g, 0.382 mmol) and HCl (0.318 mL, 12 M in H₂O). The mixture was refluxed for 4 h. The solid that formed was collected by filtration, washed with water, and recrystallized from EtOH to yield HCT6 as a yellow powder (0.058 g, 62% yield). ¹H NMR (500 MHz, DMSO- d_6) δ 11.78 (br s, 1H), 9.11 (br s, 1H), 8.61 (s, 1H), 8.56 (d, J = 5.6 Hz, 1H), 8.31 (br s, 1H), 8.02 (d, J = 8.1 Hz, 1H), 7.89–7.79 (m, 2H), 7.76 (t, J = 7.7 Hz, 1H), 3.07 (s, 3H). ¹³C NMR (125 MHz, DMSO- d_6) δ 178.36, 151.06, 144.76, 142.15, 136.25, 130.42, 128.86, 127.21, 126.82, 125.64, 121.48, 31.34. DART-MS: m/z calcd. for C₁₂H₁₃N₄S (M+H)⁺ 245.08554, found 245.08505.

(*E*)-2-(Isoquinolin-1-ylmethylene)-*N*,*N*-dimethylhydrazine-1-carbothioamide and (*Z*)-2-(Isoquinolin-1-ylmethylene)-*N*,*N*-dimethylhydrazine-1-carbothioamide (HCT11). To a solution of S1 (0.060 g, 0.382 mmol) in ethanol (3 mL) was added 4,4-dimethyl-3-thiosemicarbazide (0.046g, 0.382 mmol) and HCl (0.318 mL, 12 M in H₂O). The mixture was refluxed for 4 h. The solid that formed was collected by filtration, washed with water, and recrystallized from EtOH to yield HCT11 as a yellow powder (0.056 g, 57% yield) (mixture of E and Z isomers). ¹H NMR (500 MHz, DMSO-*d*6) δ 15.99 (s, 0.33H), 11.26 (br s, 1H), 9.77 (dd, J = 8.8, 5.1 Hz, 1H), 8.81 (d, J = 8.6 Hz, 0.33H), 8.70 (d, J = 1.7 Hz, 1H), 8.69 (d, J = 1.2 Hz, 0.33H), 8.63 (s, 0.33H), 8.55 (d, J = 5.5 Hz, 1H), 8.12 (d, J = 8.2 Hz, 0.33H), 8.01–7.96 (m, 1.33H), 7.92 (ddd, J = 8.1, 7.0, 1.1 Hz, 0.33H), 7.88–7.76 (m, 2.33H), 7.72 (ddd, J = 8.4, 6.8, 1.4 Hz, 1H), 3.43 (s, 1.98H), 3.35 (s, 6H). ¹³C NMR (125 MHz, DMSO-*d*6) δ 180.88, 180.81, 151.94, 150.58, 147.81, 142.50, 140.45, 136.86, 136.82, 131.80, 130.77, 129.48, 129.16, 128.21, 128.17 (2C), 127.68, 126.83, 125.87, 124.60, 122.53, 121.93, 42.08 (4C). DART-MS: m/z calcd. for C₁₃H₁₅N₄S (M+H)⁺ 259.10119, found 259.10080.

$$O_2N$$
 Me

1-Methyl-5-nitroisoquinoline (**S2**). To a solution of 1-methylisoquinoline (28.80 g, 201.2 mmol) in sulfuric acid (92.4 mL) at 0 °C was added KNO₃ (20.4 g, 201.2 mmol) in sulfuric acid (78.0 mL). The mixture was heated at 60 °C for 2 h and then poured slowly over crushed ice. The solution was made alkaline with NH₄OH; the resulting tan precipitate was filtered, washed with water, and dried to afford **S2** as a tan solid (20.00 g, 53%). ¹H NMR (500 MHz, CDCl₃) δ 8.61 (d, J = 6.2 Hz, 1H), 8.47−8.50 (m, 2H), 8.28 (d, J = 6.3 Hz, 1H), 7.71 (t, J = 8.1 Hz, 1H), 3.05 (s, 3H). ¹³C NMR (125 MHz, CDCl₃) δ 159.53, 145.38, 132.53, 128.65, 128.23, 127.79, 125.58, 114.26 (2C), 23.38. DART-MS: m/z calcd. for $C_{10}H_9N_2O_2$ (M+H) [†] 189.06585, found 189.06544.

1-Methylisoquinolin-5-amine (S3). To a solution of **S2** (20.00 g, 106.28 mmol) in MeOH (530 mL) and iron powder (44.40 g, 795.05 mmol) was added concentrated HCl (1 mL, 12 M in H_2O). The mixture was refluxed for 2 h and then a solution of sodium hydroxide (6 mL, 2 M in H_2O) was added. The mixture was filtered, then concentrated *in vacuo*, and resuspended in EtOAc (200 mL) and water (200 mL). The organic layer was separated and the aqueous layer was extracted with EtOAc (3 × 200 mL). The organic

layers were combined and dried over Na₂SO₄, filtered, and then concentrated *in vacuo*. The crude residue was purified by flash column chromatography (gradient, 10–30% EtOAc:hexanes). The isoquinoline **S3** was obtained as a brown solid (15.0 g, 90%). ¹H NMR (500 MHz, CDCl₃) δ 8.36 (d, J = 6.1 Hz, 1H), 7.55 (dt, J = 8.4, 1.0 Hz, 1H), 7.45 (d, J = 5.7 Hz, 1H), 7.39 (dd, J = 8.5, 7.4 Hz, 1H), 6.95 (dd, J = 7.5, 0.9 Hz, 1H), 4.18 (br s, 2H), 2.93 (s, 3H). ¹³C NMR (125 MHz, CDCl₃) δ 165.39, 159.15, 141.94, 128.35, 127.51, 126.16, 116.19, 113.09, 112.73, 23.06. DART-MS: m/z calcd. for C₁₀H₁₁N₂ (M+H)⁺ 159.09167, found 159.09136.

tert-Butyl (tert-butoxycarbonyl)(1-methylisoquinolin-5-yl)carbamate (S4). To a solution of S3 (360.0 mg, 2.28 mmol) in THF (10 mL) was added Boc₂O (1.68 g, 6.83 mmol), DMAP (28.0 mg, 0.23 mmol), and TEA (0.69 g, 3.65 mmol) and the mixture was stirred at 22 °C overnight. The reaction was quenched with water (10 mL) and the organic layers were separated. The aqueous layer was extracted with EtOAc (3 × 10 mL). The organic layers were combined and dried over Na₂SO₄, filtered, and then concentrated *in vacuo*. The crude residue was purified by flash column chromatography (gradient, 10–30% EtOAc:hexanes). The isoquinoline S4 was obtained as a brown solid (420.0 mg, 51%). ¹H NMR (500 MHz, CDCl₃) δ 8.43 (d, J = 6.0 Hz, 1H), 8.11 (d, J = 8.5 Hz, 1H), 7.58 (t, J = 7.9 Hz, 1H), 7.51 (dd, J = 7.3, 1.1 Hz, 1H), 7.46 (d, J = 5.9 Hz, 1H), 2.99 (s, 3H), 1.31 (s, 18H). ¹³C NMR (125 MHz, CDCl₃) δ 159.16, 151.59, 142.73, 135.74, 133.82, 129.64, 128.20, 126.55, 125.86, 113.74, 83.19 (2C), 27.90 (6C), 22.90. DART-MS: m/z calcd. for C₂₀H₂₇N₂O₄ (M+H)⁺ 359.19653, found 359.19540.

tert-Butyl (1-methylisoquinolin-5-yl)carbamate (S5). To a solution of S3 (10.00 g, 63.21 mmol) in THF (250 mL) was added Boc₂O (34.38 g, 158.0 mmol), DMAP (772.2 mg, 6.32 mmol), and TEA (15.96 g, 158.0 mmol) and the mixture was stirred at 22 °C overnight. After completion of the reaction as judged by TLC, NaHCO₃ (15.93 g, 189.6 mmol) and MeOH (100 mL) were added to the reaction mixture and it was refluxed overnight. After completion of the reaction (monitored by TLC), the mixture was concentrated *in vacuo* and then resuspended in EtOAc (200 mL) and water (200 mL). The organic layer was separated and the aqueous layer was extracted with EtOAc (3 × 200 mL). The organic layers were combined and dried over Na₂SO₄, filtered, and then concentrated *in vacuo*. The crude residue was purified by flash column chromatography (gradient, 10–30% EtOAc:hexanes). The isoquinoline S5 was obtained as a brown oil (4.73 g, 29%). ¹H NMR (500 MHz, CDCl₃) δ 8.37 (d, J = 6.1 Hz, 1H), 7.56 (dt, J = 8.3, 1.0 Hz, 1H), 7.46 (d, J = 6.7 Hz, 1H), 7.40 (dd, J = 8.5, 7.4 Hz, 1H), 6.95 (dd, J = 7.5, 0.9 Hz, 1H), 2.94 (s, 4H), 1.56 (s, 9H). ¹³C NMR (125 MHz, CDCl₃) δ 165.39, 159.15, 141.94, 128.35, 127.51, 126.16, 116.19, 113.09, 112.73, 76.91, 29.86 (3C), 23.06, one low-field carbon were either not observed or is overlapping

with another low- field carbon. DART-MS: m/z calcd. for $C_{15}H_{19}N_2O_2\left(M+H\right)^+259.14410$, found 259.14349.

tert-Butyl methyl(1-methylisoquinolin-5-yl)carbamate (S6). To a solution of S5 (1.99 g, 7.68 mmol) in THF (50 mL) was added NaH 60% in mineral oil (399.6 mg, 9.99 mmol). After effervescence ceased, the resulting solution was refluxed for 30 min. To the reaction mixture was then added MeI (622 μL, 9.99 mmol) in THF (2 mL) and the solution was subsequently refluxed overnight. The mixture was concentrated *in vacuo* and the crude residue was purified by flash column chromatography (gradient, 10–30% EtOAc:hexanes). The isoquinoline S6 was obtained as an amber oil (4.73 g, 29%). ¹H NMR (500 MHz, CDCl₃) δ 8.42 (d, J = 6.0 Hz, 1H), 8.08 (d, J = 8.4 Hz, 1H), 7.46–7.61 (m, 3H), 3.31 (s, 3H), 3.01 (s, 3H), 1.23 (s, 9H). ¹³C NMR (125 MHz, CDCl₃) δ 159.01, 155.21, 140.21, 133.54, 128.35, 126.95, 124.96, 121.52, 114.65, 76.15, 29.71, 28.06 (3C), 22.51, one low-field carbon were either not observed or is overlapping with another low-field carbon. DART-MS: m/z calcd. for $C_{16}H_{21}N_2O_2$ (M+H)⁺ 273.15975, found 273.15891.

tert-Butyl (1-formylisoquinolin-5-yl)(methyl)carbamate (S7). To a solution of S6 (1.50 g, 5.51 mmol) in 1,4-dioxane (60 mL) was added SeO₂ (1.22 g, 11.0 mmol). The mixture was stirred at 60 °C overnight then cooled to 22 °C. The mixture was concentrated *in vacuo* and the crude residue was purified by flash column chromatography (gradient, 5–25% EtOAc:hexanes). The isoquinoline S7 was obtained as a white solid (711.9 mg, 45%). ¹H NMR (500 MHz, CDCl₃) δ 10.39, 9.28 (d, J = 8.6 Hz, 1H), 8.80 (d, J = 5.7 Hz, 1H), 7.88 (d, J = 5.1 Hz, 1H), 7.74 (d, J = 8.0 Hz, 1H), 7.61 (m, 1H), 3.33 (s, 3H), 1.22 (s, 9H). ¹³C NMR (125 MHz, CDCl₃) δ 195.44, 155.05, 150.18, 142.85, 139.95, 134.55, 129.90, 128.96, 127.08, 124.99, 120.65, 80.72, 37.85, 28.08 (3C). DART-MS: m/z calcd. for C₁₆H₁₉N₂O₃ (M+H)⁺ 287.13902, found 287.13812.

$$Boc_2N$$
 O

tert-Butyl (tert-butoxycarbonyl)(1-formylisoquinolin-5-yl)carbamate (S8). To a solution of S4 (200.0 mg, 0.558 mmol) in 1,4-dioxane (5.5 mL) was added SeO₂ (123.8 mg, 1.12 mmol). The mixture was

stirred at 60 °C overnight then cooled to 22 °C. The mixture was concentrated *in vacuo* and the crude residue was purified by flash column chromatography (gradient, 5–25% EtOAc:hexanes). The isoquinoline **S8** was obtained as a white solid (63.2 mg, 30%). 1 H NMR (500 MHz, CDCl₃) δ 10.40 (s, 1H), 9.34 (dt, J = 8.7, 1.0 Hz, 1H), 8.81 (d, J = 5.7 Hz, 1H), 7.89 (dd, J = 5.7, 1.0 Hz, 1H), 7.76 (dd, J = 8.7, 7.4 Hz, 1H), 7.61 (dd, J = 7.3, 1.1 Hz, 1H), 1.32 (s, 18H). 13 C NMR (125 MHz, CDCl₃) δ 195.59, 151.36, 150.23, 143.31, 134.94, 130.49, 129.74, 126.88, 126.00, 119.90, 83.60 (2C), 27.91 (6C), two low-field carbon were either not observed or is overlapping with another low-field carbon . DART-MS: m/z calcd. for $C_{20}H_{25}N_2O_5$ (M+H) $^+$ 373.17580, found 373.17496.

(*E*)-2-((5-(Methylamino)isoquinolin-1-yl)methylene)hydrazine-1-carbothioamide (HCT4). To a solution of S7 (100.0 mg, 0.3492 mmol) in EtOH (1.75 mL) was added thiosemicarbazide (31.8 mg, 0.3492 mmol) and HCl (350 μL, 6 M in H₂O). The mixture was stirred and refluxed for 1.5 h and then cooled to 22 °C. The hydrochloride salt that formed was neutralized with 1.4 mL of a saturated aqueous NaHCO₃ solution. The precipitate of the desired compound was collected by filtration, washed with water, EtOH and then dried to yield the isoquinoline HCT4 as a black solid (622.4 mg, 97%). ¹H NMR (500 MHz, DMSO- d_6) δ 12.32 (s, 1H), 9.07 (br s, 1H), 8.92 (s, 1H), 8.90 (s, 1H), 8.57 (d, J = 6.6 Hz, 1H), 8.52 (d, J = 6.7 Hz, 1H), 7.85 (t, J = 8.2 Hz, 1H), 7.56 (d, J = 8.3 Hz, 1H), 7.30 (br s, 1H), 7.01 (d, J = 8.0 Hz, 1H), 2.91 (s, 3H). ¹³C NMR (125 MHz, DMSO- d_6) δ 179.40, 146.25, 146.00, 133.09, 130.09, 128.72, 126.75 (2C), 119.38, 111.17, 110.66, 30.39. DART-MS: m/z calcd. for C₁₂H₁₄N₅S (M+H)⁺ 260.09644, found 260.09501.

(*E*)-*N*-Methyl-2-((5-(methylamino)isoquinolin-1-yl)methylene)hydrazine-1-carbothioamide and (*Z*)-*N*-Methyl-2-((5-(methylamino)isoquinolin-1-yl)methylene)hydrazine-1-carbothioamide (HCT9). To a solution of S7 (51.4 mg, 0.18 mmol) in EtOH (0.88 mL) was added 4-methyl-3-thiosemicarbazide (18.9 mg, 0.18 mmol) and HCl (0.18 mL, 6 M in H₂O). The mixture was stirred and refluxed for 1.5 h then cooled to 22 °C. The hydrochloride salt that formed was neutralized with saturated aqueous NaHCO₃ solution (0.88 mL). The precipitate of the desired compound was collected by filtration, washed with water, EtOH and then dried to yield the isoquinoline HCT9 as a black solid (49.0 mg, 94%) (mixture of *E* and *Z* isomers). ¹H NMR (500 MHz, DMSO- d_6) δ 14.74 (s, 0.15H), 12.22 (s, 1H), 9.39 (br s, 1H), 8.93 (q, *J* = 4.7 Hz, 0.15H), 8.78 (s, 1H), 8.54 (d, *J* = 5.9 Hz, 0.15H), 8.50 (d, *J* = 6.5 Hz, 1H), 8.38 (s, 1H), 8.17 (s, 0.15H), 8.12 (d, *J* = 5.9 Hz, 0.15H), 7.82 (d, *J* = 8.6 Hz, 0.15H), 7.76 (t, *J* = 8.1 Hz, 1H), 7.69 (s, 1H), 7.56 (t, *J* = 8.1 Hz, 0.15H), 7.08 (br s, 1H), 6.92 (d, *J* = 7.7 Hz, 1.15H), 6.72 (d, *J* = 7.8 Hz, 0.15H), 3.07 (d, *J* = 4.6 Hz, 3H), 3.02 (d, *J* = 4.6 Hz, 0.45 H), 2.88 (s, 3H), 2.86 (s, 0.45H). ¹³C NMR (125 MHz, DMSO- d_6) δ 178.84, 178.38, 150.11, 147.37, 145.83 (2C), 145.56, 138.94, 132.45, 130.58, 129.19,

128.23, 128.22, 126.87 (2C), 126.80, 118.47, 117.04, 111.57, 110.13, 109.63, 106.57, 31.59, 31.42, 30.42 (2C). DART- MS: m/z calcd. for $C_{13}H_{16}N_{5}S$ (M+H)⁺ 274.11209, found 274.11104.

(*E*)-2-((5-Aminoisoquinolin-1-yl)methylene)hydrazine-1-carbothioamide (HCT5). To a solution of S8 (30.0 mg, 0.081 mmol) in EtOH (0.39 mL) was added thiosemicarbazide (7.3 mg, 0.802 mmol) and HCl (80.6 μL, 6 M in H₂O). The mixture was stirred and refluxed for 1.5 h then cooled to 22 °C. The hydrochloride salt that formed was neutralized with saturated aqueous NaHCO₃ solution (0.39 mL). The precipitate of the desired compound was collected by filtration, washed with water, EtOH and then dried to yield the isoquinoline **HCT5** as a green solid (19.6 mg, 99%). ¹H NMR (500 MHz, DMSO- d_6) δ 11.66 (br s, 1H), 8.57 (s, 1H), 8.42 (d, J = 5.8 Hz, 1H), 8.31 (br s, 1H), 8.25 (d, J = 8.5 Hz, 1H), 7.98 (d, J = 5.8 Hz, 1H), 7.60 (br s, 1H), 7.42 (t, J = 8.1 Hz, 1H), 6.89 (d, J = 7.1 Hz, 1H), 6.02 (s, 2H). ¹³C NMR (125 MHz, DMSO- d_6) δ 178.46, 150.36, 145.86, 144.62, 140.01, 129.74, 126.78. 125.83. 116.50, 113.12, 110.74. DART-MS: m/z calcd. for C₁₁H₁₁N₅S (M+H)⁺ 246.08079, found 246.08020.

(*E*)-2-((5-Aminoisoquinolin-1-yl)methylene)-*N*-methylhydrazine-1-carbothioamide and (*Z*)- 2-((5-Aminoisoquinolin-1-yl)methylene)-*N*-methylhydrazine-1-carbothioamide (HCT10). To a solution of S8 (27.1 mg, 0.0728 mmol) in EtOH (0.73 mL) was added 4-methyl-3-thiosemicarbazide (7.7 mg, 0.0732 mmol) and HCl (72.8 μL, 6 M in H₂O). The mixture was stirred and refluxed for 1.5 h then cooled to 22 °C. The hydrochloride salt that formed was neutralized with saturated aqueous NaHCO₃ solution (0.73 mL). The precipitate of the desired compound was collected by filtration, washed with water, EtOH and then dried to yield the isoquinoline HCT10 as a yellow solid (5.1 mg, 27%) (mixture of *E* and *Z* isomers). ¹H NMR (500 MHz, DMSO- d_6) δ 14.80 (s, 0.08H), 11.66 (br s, 1H), 8.95 (d, J = 4.9 Hz, 0.08H), 8.62 (s, 1H), 8.52 (d, J = 5.9 Hz, 0.8H), 8.42 (d, J = 5.8 Hz, 1H), 8.25 (d, J = 3.3 Hz, 1H), 8.18 (s, 0.08H), 8.11–8.15 (m, 1.08H), 7.99 (d, J = 5.9 Hz, 1H), 7.83 (d, J = 8.4 Hz, 0.08H), 7.48 (t, J = 7.9 Hz, 0.08H), 7.43 (t, J = 8.0 Hz, 1H), 6.97 (d, J = 7.6 Hz, 0.08H), 6.91 (dd, J = 7.6, 0.9 Hz, 1H), 6.21 (s, 0.16H), 6.04 (s, 2H), 3.05–3.07 (m, 3.24H). ¹³C NMR (125 MHz, DMSO- d_6) δ 178.85, 178.52, 150.83, 150.19, 145.53, 145.20, 145.10, 140.49, 138.40, 130.28, 130.10, 129.16, 128.41, 127.36, 126.27, 126.16, 117.81, 116.96, 113.24, 111.66, 111.18, 110.61, 31.74, 31.59. DART-MS: m/z calcd. for C₁₂H₁₄N₅S (M+H)⁺ 260.09644, found 260.09563.

4-Fluoro-1-methylisoquinoline (S9). To a solution of 4-fluoroisoquinoline (1.50 g, 10.19 mmol) in THF (102 mL) was added allyl chloroformate (2.17 mL, 20.38 mmol). MeMgBr (10.19 mL, 2 M in diethyl ether) was then added dropwise to the reaction mixture at 0 °C with stirring. The reaction mixture was gradually warmed to 22 °C over a period of 2 h. The mixture was quenched with saturated aqueous NH₄Cl (10 mL) and water (100 mL) was added. The organic layer was separated and the aqueous layer was extracted with EtOAc (3 × 100 mL). The organic layers were combined and dried over MgSO₄, filtered, and then concentrated in vacuo. The crude residue in EtOAc was filtered through a silica plug, concentrated in vacuo and the residue was subjected to the next reaction without further purification. To a solution of the crude residue and Pd(PPh₃)₄ (70.1 mg, 0.061 mmol) in DCM (60 mL) at 0 °C was added morpholine (523.1 uL, 6.07 mmol). The reaction mixture was stirred and slowly warmed to 22 °C over a period of 3 h. The mixture was cooled to 0 °C and DDQ (1.38 g, 6.07 mmol) was added in portions. After the reaction mixture stirred at 0 °C for 30 min, the reaction was slowly poured into a solution of saturated NaHCO₃ solution (60 mL) and extracted with DCM (3 × 60). The combined extracts are washed with brine, dried over Na₂SO₄, and concentrated in vacuo. The crude residue was purified by flash column chromatography (gradient, 5-25% EtOAc:hexanes). The isoquinoline S9 was obtained as a brown oil (241.9 mg, 15% over three steps). ¹H NMR $(500 \text{ MHz}, \text{CDCl}_3) \delta 8.23 \text{ (d, } J = 1.7 \text{ Hz}, 1\text{H}), 8.06-8.09 \text{ (m, } 1.00 \text{ m})$ 2H), 7.73-7.76 (m, 1H), 7.64-7.67 (m, 1H), 2.90 (s, 3H). 13 C NMR (125 MHz, CDCl₃) δ 154.48 (d, $^{1}J_{\text{C-F}}$ = 257.5 Hz), 154.24 (d, ${}^{3}J_{\text{C-F}}$ = 4.9 Hz), 130.14 (d, ${}^{4}J_{\text{C-F}}$ = 1.6 Hz), 128.35 (d, ${}^{4}J_{\text{C-F}}$ = 2.4 Hz), 127.83, 126.64 (d, ${}^{2}J_{\text{C-F}}$ = 15.3 Hz), 126.57 (d, ${}^{2}J_{\text{C-F}}$ = 22.3 Hz), 125.60 (d, ${}^{4}J_{\text{C-F}}$ = 2.1 Hz), 120.09 (d, ${}^{3}J_{\text{C-F}}$ = 4.5 Hz), 22.10. ¹⁹F NMR (376 MHz, CDCl₃) δ –143.11, extraneous peak found at –139.82. DART-MS: m/z calcd. for $C_{10}H_9FN (M+H)^+ 162.07135$, found 162.07092.

4-Fluoroisoquinoline-1-carboxaldehyde (S10). To a solution of **S9** (40.0 mg, 0.248 mmol) in 1,4-dioxane (2.5 mL) was added SeO₂ (55.1 mg, 0.496 mmol). The mixture was stirred at 60 °C overnight then cooled to 22 °C. The mixture was concentrated *in vacuo* and the crude residue was purified by flash column chromatography (gradient, 5–25% EtOAc:hexanes). The isoquinoline **S10** was obtained as a white solid (27.3 mg, 63%). ¹H NMR (500 MHz, DMSO-*d*₆) δ 10.32 (s, 1H), 9.37–9.41 (m, 1H), 8.59 (d, J = 1.5 Hz, 1H), 8.16–8.20 (m, 1H), 7.82-7.87 (m, 2H). ¹³C NMR (125 MHz, DMSO-*d*₆) δ 194.23, 157.16 (d, $^{1}J_{\text{C-F}}$ = 270.3 Hz), 146.44 (d, $^{3}J_{\text{C-F}}$ = 5.6 Hz), 131.07 (d, $^{4}J_{\text{C-F}}$ = 2.2 Hz), 130.95 128.48 (d, $^{2}J_{\text{C-F}}$ = 24.6 Hz), 128.14 (d, $^{4}J_{\text{C-F}}$ = 4.1 Hz), 126.85 (d, $^{2}J_{\text{C-F}}$ = 14.4 Hz), 125.64 (d, $^{4}J_{\text{C-F}}$ = 1.8 Hz), 119.85 (d, $^{3}J_{\text{C-F}}$ = 4.7 Hz). ¹⁹F NMR (376 MHz, CDCl₃) δ –129.02. DART-MS: m/z calcd. for C₁₀H₆FNO (M+H)⁺ 176.05062, found 176.05012.

(*E*)-2-((4-Fluoroisoquinolin-1-yl)methylene)hydrazine-1-carbothioamide (HCT2). To a solution of S10 (6.0 mg, 0.0343 mmol) in EtOH (0.5 mL) was added thiosemicarbazide (3.3 mg, 0.0343 mmol) and HCl (34 μL, 0.206 mmol, 6 M in H₂O). The mixture was stirred and refluxed for 1.5 h then cooled to 22 °C. The hydrochloride salt that formed was then neutralized with saturated aqueous NaHCO₃ solution (0.5 mL). The precipitate of the desired compound was collected by filtration, washed with water, EtOH and then dried to yield the isoquinoline HCT2 as a pale-yellow solid (3.0 mg, 35%). ¹H NMR (500 MHz, DMSO- d_6) δ 11.70 (s, 1H), 9.28 (d, J = 8.5 Hz, 1H), 8.56 (d, J = 1.5 Hz, 1H), 8.53 (s, 1H), 8.48 (s, 1H), 8.13 (d, J = 8.2 Hz, 1H), 7.94 (ddd, J = 8.2, 7.0, 0.9 Hz, 1H), 7.85 (m, 2H). ¹³C NMR (125 MHz, DMSO- d_6) δ 178.84, 154.73 (d, $^1J_{C-F}$ = 262.2 Hz), 148.03 (d, $^3J_{C-F}$ = 5.2 Hz), 145.84, 131.75, 130.69, 128.10 (d, $^2J_{C-F}$ = 23.3 Hz), 127.75, 127.35 (d, $^4J_{C-F}$ = 3.0 Hz), 126.51 (d, $^2J_{C-F}$ = 14.9 Hz), 119.79 (d, $^3J_{C-F}$ = 4.6 Hz). ¹⁹F NMR (376 MHz, DMSO- d_6) δ –137.31. DART-MS: m/z calcd. for C₁₁H₉FN₄S (M+H)⁺ 249.06047, found 249.05042.

(*E*)-2-((4-Fluoroisoquinolin-1-yl)methylene)-*N*-methylhydrazine-1-carbothioamide (HCT7). To a solution of **S10** (6.0 mg, 0.0343 mmol) in EtOH (0.5 mL) was added 4-methyl-3-thiosemicarbazide (3.6 mg, 0.0343 mmol) and HCl (34 μL, 0.206 mmol, 6 M in H₂O). The mixture was stirred and refluxed for 1.5 h then cooled to 22 °C. The hydrochloride salt that formed was then neutralized with saturated aqueous NaHCO₃ solution (0.5 mL). The precipitate of the desired compound was collected by filtration, washed with water, EtOH and then dried to yield the isoquinoline HCT7 as a pale-yellow solid (2.6 mg, 29%). ¹H NMR (500 MHz, DMSO- d_6) δ 11.76 (s, 1H), 9.19 (d, J = 8.6 Hz, 1H), 8.56 (d, J = 1.4 Hz, 1H), 8.56 (s, 1H), 8.34 (d, J = 4.4 Hz, 1H), 8.14 (d, J = 8.3 Hz, 1H), 7.94–7.97 (m, 1H), 7.85–7.89 (m, 1H), 3.06 (d, J = 4.6 Hz, 3H). ¹³C NMR (125 MHz, DMSO- d_6) δ 178.84, 154.73 (d, $^1J_{C-F}$ = 262.2 Hz), 148.03 (d, $^3J_{C-F}$ = 5.2 Hz), 145.84, 131.75, 130.69, 128.10 (d, $^2J_{C-F}$ = 23.3 Hz), 127.75, 127.35 (d, $^4J_{C-F}$ = 3.0 Hz), 126.51 (d, $^2J_{C-F}$ = 14.9 Hz), 119.79 (d, $^3J_{C-F}$ = 4.6 Hz), 31.86. ¹⁹F NMR (376 MHz, DMSO- d_6) δ –137.53, extraneous peak found at –134.32. DART-MS: m/z calcd. for C₁₂H₁₂FN₄S (M+H)⁺ 263.07612, found 263.07520.

(E)-2-((4-Fluoroisoquinolin-1-yl)methylene)-N,N-dimethylhydrazine-1-carbothioamide and (Z)-2-((4-Fluoroisoquinolin-1-yl)methylene)-N,N-dimethylhydrazine-1-carbothioamide (HCT12). To a solution of S10 (17.8 mg, 0.102 mmol) in MeOH (1.0 mL) was added 4,4-dimethyl-3-thiosemicarbazide (12.0 mg, 0.102 mmol) and HCl $(101 \mu L, 0.610 \text{ mmol}, 6 \text{ M in H}_2\text{O})$. The mixture was microwaved at 300 W and 50 °C for 1.0 h. The hydrochloride salt that formed was then neutralized with saturated aqueous NaHCO₃ solution (1.0 mL). The precipitate of the desired compound was collected by filtration, washed with water, EtOH and then dried to yield the isoquinoline HCT12 as a pale-yellow solid (16.0 mg, 57%) (mixture of E and Z isomers). ¹H NMR (500 MHz, DMSO- d_6) δ 15.52 (s, 0.15H), 11.28 (s, 1H), 9.87 (d, J = 8.7 Hz, 1H), 8.87 (d, J = 9.0 Hz, 0.15 H), 8.77 (d, J = 1.9 Hz, 0.15 H), 8.69 (s, 1H), 8.57 (d, J = 1.6 Hz, 1.15H), 8.23 (d, J = 8.2 Hz, 0.15H), 8.15 (d, J = 8.2 Hz, 1H), 8.02–8.05 (m, 0.15H), 7.92–7.97 (m, 1.15H), 7.86 (ddd, J = 8.3, 6.9, 1.3 Hz, 1H), 3.42 (s, 0.90H), 3.36 (s, 6H). ¹³C NMR (125 MHz, DMSO d_6) δ 180.79, 180.73, 154.60 (d, ${}^{1}J_{\text{C-F}}$ = 261.7 Hz), 154.13 (d, ${}^{1}J_{\text{C-F}}$ = 261.8 Hz), 148.75 (d, ${}^{3}J_{\text{C-F}}$ = 5.1 Hz), $147.76 \text{ (d, }^{3}J_{C-F} = 5.7 \text{ Hz)}, 147.05, 132.52, 131.59 \text{ (d, }^{4}J_{C-F} = 5.1 \text{ Hz)}, 131.10, 130.49, 130.30, 128.60 \text{ (d, }^{4}J_{C-F} = 5.1 \text{ Hz)}$ $^{4}J_{\text{C-F}} = 3.3 \text{ Hz}$), 128.41 (d, $^{4}J_{\text{C-F}} = 1.0 \text{ Hz}$), 127.90 (d, $^{2}J_{\text{C-F}} = 23.3 \text{ Hz}$), 127.19 (d, $^{4}J_{\text{C-F}} = 2.6 \text{ Hz}$), 126.88 (d, $^{2}J_{C-F} = 14.8 \text{ Hz}$), 126.71 (d, $^{2}J_{C-F} = 14.7 \text{ Hz}$), 126.35 (d, $^{2}J_{C-F} = 25.2 \text{ Hz}$), 124.98, 120.23 (d, $^{3}J_{C-F} = 4.3 \text{ Hz}$), 119.79 (d, ${}^{3}J_{C-F}$ = 4.7 Hz), 42.04 (4C). ${}^{19}F$ NMR (376 MHz, DMSO- d_{6}) δ –134.93, –138.02. DART-MS: m/z calcd. for C₁₃H₁₄FN₄S (M+H)⁺277.09177, found 277.09096.

6-Fluoro-1-methylisoquinoline (S11). To a solution of 6-fluoroisoquinoline (1.00 g, 6.80 mmol) in THF (120 mL) was added allyl chloroformate (1.64 mL, 13.59 mmol). MeMgBr (6.98 mL, 13.59 mmol, 2 M in diethyl ether) was then added dropwise to the reaction mixture at 0 °C while stirring and the mixture was gradually warmed to 22 °C over a period of 2 h. The reaction was quenched with saturated aqueous NH₄Cl (12 mL) and water (120 mL) was added. The organic layer was separated and the aqueous layer was extracted with EtOAc (3 × 120 mL). The organic layers were combined and dried over MgSO₄, filtered, and then concentrated in vacuo. The crude residue in EtOAc was filtered through a silica plug, concentrated in vacuo and the crude residue was subjected to the next reaction without further purification. To a solution of the crude residue and Pd(PPh₃)₄ (293.4 mg, 0.254 mmol) in DCM (50 mL) at 0 °C was added morpholine (437.9 uL, 5.08 mmol). The reaction was stirred and slowly warmed to 22 °C over a period of 3 h. The mixture was cooled to 0 °C and DDQ (1.15 g, 5.08 mmol) was added portionwise. After the reaction mixture stirred at 0 °C for 30 min, the reaction was slowly poured into a solution of saturated NaHCO₃ solution (50 mL) and extracted with DCM (3 × 50). The combined extracts are washed with brine, dried over Na₂SO₄, and concentrated in vacuo and the crude residue was purified by flash column chromatography (gradient, 5–25% EtOAc; hexanes). The isoquinoline S11 was obtained as a brown oil (583.9 mg, 53% over three steps). ¹H NMR (500 MHz, DMSO- d_6) δ 8.37 (d, J = 5.8 Hz,

1H), 8.13 (dd, J = 9.2, 5.5 Hz, 1H), 7.46 (d, J = 5.8 Hz, 1H), 7.40 (dd, J = 9.3, 2.6 Hz, 1H), 7.34 (td, J = 8.8, 2.6 Hz, 1H), 2.95 (s, 3H). ¹³C NMR (125 MHz, DMSO- d_6) δ 162.94 (d, ¹ $J_{\text{C-F}}$ = 252.2 Hz), 158.53 (d, ⁵ $J_{\text{C-F}}$ = 1.0 Hz), 142.77, 137.58 (d, ³ $J_{\text{C-F}}$ = 10.4 Hz), 128.79 (d, ³ $J_{\text{C-F}}$ = 9.6 Hz), 124.72 (d, ⁴ $J_{\text{C-F}}$ = 1.0 Hz), 119.01 (d, ⁴ $J_{\text{C-F}}$ = 5.0 Hz), 117.31 (d, ² $J_{\text{C-F}}$ = 25.0 Hz), 110.44 (d, ² $J_{\text{C-F}}$ = 20.6 Hz), 22.53. ¹⁹F NMR (376 MHz, CDCl₃) δ –108.23. DART-MS: m/z calcd. for C₁₀H₉FN (M+H)⁺ 162.07135, found 162.07096.

6-Fluoroisoquinoline-1-carboxaldehyde (S12). To a solution of **S11** (500.0 mg, 3.10 mmol) in 1,4-dioxane (19.0 mL) was added SeO₂ (688.4 mg, 6.20 mmol). The mixture was stirred at 60 °C overnight then cooled to 22 °C. The mixture was concentrated *in vacuo* and the crude residue was purified by flash column chromatography (gradient, 5–25% EtOAc:hexanes). The isoquinoline **S12** was obtained as a white solid (200.9 mg, 37%). ¹H NMR (500 MHz, DMSO-*d*₆) δ 10.35 (s, 1H), 9.39 (ddd, J = 10.1, 5.6, 0.9 Hz, 1H), 8.75 (dd, J = 5.6, 0.4 Hz, 1H), 7.85 (d, J = 5.5 Hz, 1H), 7.50–7.54 (m, 2H). ¹³C NMR (125 MHz, DMSO-*d*₆) δ 195.51, 163.18 (d, ${}^{1}J_{\text{C-F}}$ = 255.1 Hz), 149.71 (d, ${}^{4}J_{\text{C-F}}$ = 1.8 Hz), 143.34 (d, ${}^{5}J_{\text{C-F}}$ = 1.0 Hz), 138.70 (d, ${}^{3}J_{\text{C-F}}$ = 10.4 Hz), 129.24 (d, ${}^{3}J_{\text{C-F}}$ = 9.2 Hz), 124.90 (d, ${}^{4}J_{\text{C-F}}$ = 5.4 Hz), 123.49 (d, ${}^{5}J_{\text{C-F}}$ = 1.0 Hz), 120.53 (d, ${}^{2}J_{\text{C-F}}$ = 24.8 Hz), 110.23 (d, ${}^{2}J_{\text{C-F}}$ = 20.9 Hz). ¹⁹F NMR (376 MHz, CDCl₃) δ –105.46. DART- MS: m/z calcd. for C₁₀H₇FNO (M+H)⁺ 176.05062, found 176.05015.

(*E*)-2-((6-Fluoroisoquinolin-1-yl)methylene)hydrazine-1-carbothioamide (HCT3). To a solution of S12 (10.2 mg, 0.0582 mmol) in EtOH (0.5 mL) was added thiosemicarbazide (5.3 mg, 0.0582 mmol) and HCl (58 μL, 0.349 mmol, 6 M in H₂O). The mixture was stirred and refluxed for 1.5 h then cooled to 22 °C. The hydrochloride salt that formed was then neutralized with saturated aqueous NaHCO₃ solution (0.5 mL). The precipitate of the desired compound was collected by filtration, washed with water, EtOH and then dried to yield the isoquinoline **HCT3** as a pale-yellow solid (13.4 mg, 93%). ¹H NMR (500 MHz, DMSO- d_6) δ 11.74 (s, 1H), 9.30 (dd, J = 9.4, 5.8 Hz, 1H), 8.55 (d, J = 5.6 Hz, 1H), 8.51 (s, 2H), 7.80–7.85 (m, 3H), 7.57 (td, J = 9.0, 2.8 Hz, 1H). ¹³C NMR (125 MHz, DMSO- d_6) δ 178.88, 162.70 (d, $^1J_{C-F}$ = 250.4 Hz), 151.35, 146.33, 143.50, 138.62 (d, $^3J_{C-F}$ = 10.7 Hz), 131.52 (d, $^3J_{C-F}$ = 9.5 Hz), 123.24, 121.79 (d, $^4J_{C-F}$ = 5.0 Hz), 119.36 (d, $^2J_{C-F}$ = 24.5 Hz), 110.86 (d, $^2J_{C-F}$ = 20.7 Hz). ¹⁹F NMR (376 MHz, DMSO- d_6) δ – 107.79, extraneous peak found at –106.49. DART-MS: m/z calcd. for C₁₁H₁₀FN₄S (M+H)⁺ 249.06047, found 249.05984.

(*E*)-2-((6-Fluoroisoquinolin-1-yl)methylene)-*N*-methylhydrazine-1-carbothioamide (HCT8). To a solution of S12 (8.8 mg, 0.0502 mmol) in EtOH (0.5 mL) was added 4-methyl-3-thiosemicarbazide (5.3 mg, 0.0502 mmol) and HCl (50 μL, 0.300 mmol, 6 M in H₂O). The mixture was stirred and refluxed for 1.5 h then cooled to 22 °C. The hydrochloride salt that formed was then neutralized with saturated aqueous NaHCO₃ solution (0.5 mL). The precipitate of the desired compound was collected by filtration, washed with water, EtOH and then dried to yield the isoquinoline HCT8 as a pale-yellow solid (10.8 mg, 82%). ¹H NMR (500 MHz, DMSO- d_6) δ 11.80 (s, 1H), 9.20 (dd, J = 9.4, 5.7 Hz, 1H), 8.55 (d, J = 5.6 Hz, 1H), 8.54 (s, 1H), 8.35 (d, J = 4.7 Hz, 1H), 7.83 (dd, J = 9.2, 3.9 Hz, 2H) 7.60 (td, J = 9.0, 2.7 Hz, 1H), 3.06 (d, J = 4.5 Hz, 3H). ¹³C NMR (125 MHz, DMSO- d_6) δ 178.56, 162.71 (d, $^1J_{\text{C-F}}$ = 250.4 Hz), 151.55, 145.22, 143.53, 138.62 (d, $^3J_{\text{C-F}}$ = 10.6 Hz), 131.28 (d, $^3J_{\text{C-F}}$ = 9.5 Hz), 123.29, 121.67 (d, $^4J_{\text{C-F}}$ = 5.1 Hz), 119.23 (d, $^2J_{\text{C-F}}$ = 24.8 Hz), 110.89 (d, $^2J_{\text{C-F}}$ = 20.8 Hz), 31.85. ¹⁹F NMR (376 MHz, DMSO- d_6) δ -106.55, extraneous peak found at -107.74. DART-MS: m/z calcd. for C₁₂H₁₂FN₄S (M+H)⁺ 263.07612, found 263.07538.

(E)-2-((6-Fluoroisoquinolin-1-yl)methylene)-N,N-dimethylhydrazine-1-carbothioamide and (Z)-2-((6-Fluoroisoquinolin-1-vl)methylene)-N,N-dimethylhydrazine-1-carbothioamide (HCT13). To a solution of S12 (8.6 mg, 0.0491 mmol) in EtOH (0.5 mL) was added 4,4-dimethyl-3-thiosemicarbazide (5.9 mg, 0.0491 mmol) and HCl (49 μL, 0.294 mmol, 6 M in H₂O). The mixture was stirred and refluxed for 1.5 h then cooled to 22 °C. The hydrochloride salt that formed was then neutralized with saturated aqueous NaHCO₃ solution (0.5 mL), the precipitate of the desired compound was collected by filtration, washed with water, EtOH and then dried to yield the isoquinoline HCT13 as a pale-yellow solid (7.4 mg, 55%). H NMR (500 MHz, DMSO- d_6) δ 15.90 (s, 0.21H), 11.30 (s, 1H), 9.87 (dd, J = 9.5, 5.9 Hz, 1H). 8.91 (dd, J = 9.4, 5.4 Hz, 0.21H), 8.66 (m, 1.21H), 8.59 (s, 0.21H), 8.55 (d, J = 5.6 Hz, 1H), 7.97 (d, J =5.6 Hz, 0.21H), 7.91 (dd, J = 9.6, 2.7 Hz, 0.21H), 7.79–7.82 (m, 2H), 7.73 (td, J = 9.1, 2.7 Hz, 0.21H), 7.62 (ddd, J = 9.6, 8.6, 2.8 Hz, 1H), 3.40 (s, 1.26H), 3.33 (s, 6H). ¹³C NMR (125 MHz, DMSO- d_6) δ 180.78 (2C), 163.19 (d, ${}^{1}J_{C-F} = 251.7 \text{ Hz}$), 162.66 (d, ${}^{1}J_{C-F} = 250.6 \text{ Hz}$), 151.99 (d, ${}^{5}J_{C-F} = 1.2 \text{ Hz}$), 150.63 $(d, {}^{5}J_{C-F} = 0.9 \text{ Hz}), 147.57, 143.42, 141.42, 138.83 (d, {}^{3}J_{C-F} = 15.5 \text{ Hz}), 138.76 (d, {}^{4}J_{C-F} = 10.7 \text{ Hz}), 132.08$ $(d, {}^{3}J_{C-F} = 9.3 \text{ Hz}), 131.62, 128.64 (d, {}^{3}J_{C-F} = 9.9 \text{ Hz}), 124.14, 123.10, 122.12 (d, {}^{4}J_{C-F} = 5.2 \text{ Hz}), 121.56 (d, {}^{4}J_{C-F} = 1.2 \text{ Hz}), 121.56 (d, {}^{4}J_{C-F} =$ $^{4}J_{\text{C-F}} = 5.1 \text{ Hz}$, 119.54 (d, $^{2}J_{\text{C-F}} = 25.6 \text{ Hz}$), 119.14 (d, $^{2}J_{\text{C-F}} = 24.4 \text{ Hz}$), 111.48 (d, $^{2}J_{\text{C-F}} = 20.8 \text{ Hz}$), 110.90 $(d, {}^{2}J_{C-F} = 20.7 \text{ Hz}), 42.04 (4C). {}^{19}\text{F NMR} (376 \text{ MHz}, \text{DMSO-}d_{6}) \delta -106.34, -107.95. DART-MS: m/z$ calcd. for $C_{13}H_{14}FN_4S (M+H)^+277.09177$, found

$$O_2N$$
 Me

4-Fluoro-1-methyl-5-nitroisoquinoline (S13). To a solution of **S11** (0.376 g, 2.333 mmol) in sulfuric acid (0.4 mL) at 0 °C was added KNO₃ (0.234 g, 2.333 mmol) in sulfuric acid (0.6 mL). The mixture was heated at 60 °C for 2 h and then poured slowly over crushed ice. The solution was made alkaline with NH₄OH; the resulting tan precipitate was filtered, washed with water, and dried to afford **S13** as a tan solid (0.210 g, 44%). ¹H NMR (500 MHz, CDCl₃) δ 8.42 (d, J = 2.9 Hz, 1H), 8.36 (d, J = 8.3 Hz, 1H), 7.98 (d, J = 7.4 Hz, 1H), 7.77 (t, J = 7.8 Hz, 1H), 3.03 (s, 3H). ¹³C NMR (125 MHz, CDCl₃) δ 155.10 (d, ${}^4J_{C-F}$ = 5.2 Hz), 151.08 (d, ${}^1J_{C-F}$ = 262.1 Hz), 144.92, 130.04 (d, ${}^2J_{C-F}$ = 25.2 Hz), 129.62, 128.88, 127.24, 125.53, 118.43 (d, ${}^3J_{C-F}$ = 12.1 Hz), 22.66. ¹⁹F NMR (376 MHz, CDCl₃) δ -133.19. DART-MS: m/z calcd. for C₁₀H₈FN₂O₂ (M+H)⁺ 207.05643, found 207.05705.

4-Fluoro-1-methylisoquinolin-5-amine (S14). To a solution of **S13** (0.210 g, 1.02 mmol) in MeOH (50 mL) iron powder (0.171 g, 3.06 mmol and HCl (1 mL, 12 M in H₂O). The mixture was refluxed for 2 h and then a solution of sodium hydroxide (2 mL, 6 M in H₂O) was added. The mixture was filtered and extracted with diethyl ether (200 mL). The organic layer was dried over Na₂SO₄, filtered, and then concentrated *in vacuo*. The crude residue was purified by flash column chromatography (gradient, 10–30% EtOAc:hexanes). The isoquinoline **S14** was obtained as a brown solid (0.173 g, 96%). ¹H NMR (500 MHz, CDCl₃) δ 8.09 (d, J = 5.1 Hz, 1H), 7.46–7.39 (m, 2H), 6.88 (dd, J = 6.9, 1.8 Hz, 1H), 4.83 (br s, 2H), 2.87 (d, J = 1.3, 3H). ¹³C NMR (125 MHz, CDCl₃) δ 155.92 (d, ${}^{1}J_{\text{C-F}}$ = 253.3 Hz), 154.74 (d, ${}^{4}J_{\text{C-F}}$ = 4.9 Hz), 142.23 (d, ${}^{4}J_{\text{C-F}}$ = 3.0 Hz), 129.19, 115.79 (d, ${}^{3}J_{\text{C-F}}$ = 8.8 Hz), 114.77, 114.76, 113.81 (2C), 22.61. ¹⁹F NMR (376 MHz, CDCl₃) δ -136.45. DART-MS: m/z calcd. for C₁₀H₁₀FN₂ (M+H)⁺ 177.08225, found 177.08220.

tert-butyl (4-fluoro-1-methylisoquinolin-5-yl)carbamate (S15). To a solution of S14 (1.14 g, 6.49 mmol) in THF (15 mL) was added DMAP (79.3 mg, 0.65 mmol) then Boc₂O (3.54 g, 16.23 mmol) and the mixture was stirred at 22 °C overnight. After completion of the reaction as judged by TLC, K₂CO₃ (2.69 g, 19.47 mmol) and MeOH (10 mL) were added to the reaction mixture and then refluxed overnight.

The mixture was then concentrated *in vacuo* and resuspended in EtOAc (20 mL) and water (20 mL). The organic layer was separated and the aqueous layer was extracted with EtOAc (3 × 20 mL). The organic layers were combined and dried over Na₂SO₄, filtered, and then concentrated *in vacuo*. The crude residue was purified by flash column chromatography (gradient, 5–20% EtOAc:hexanes). The isoquinoline **S15** was obtained as a brown oil (0.572 g, 33%). ¹H NMR (500 MHz, CDCl₃) δ 8.50 (dd, J = 7.9, 1.0 Hz, 1H), 8.17 (d, J = 5.7 Hz, 1H), 8.05 (d, J = 17.8 Hz, 1H), 7.73 (ddd, J = 8.4, 3.0, 1.0 Hz, 1H), 7.60 (t, J = 8.2 Hz, 1H), 2.88 (d, J = 1.3 Hz, 3H), 1.55 (s, 9H). ¹³C NMR (125 MHz, CDCl₃) δ 158.37 (d, ¹J_{C-F} = 296.5 Hz), 153.19, 137.68, 131.10, 128.23 (d, ³J_{C-F} = 10.7 Hz), 124.83, 124.45, 121.14 (d, ²J_{C-F} = 22.3 Hz), 119.58, 82.72, 28.15 (3C), 17.84, one low-field carbon were either not observed or is overlapping with another low-field carbon. ¹⁹F NMR (376 MHz, CDCl₃) δ -136.85. DART-MS: m/z calcd. for C₁₅H₁₈FN₂O₂ (M+H)⁺ 277.13468, found 277.13425.

tert-Butyl methyl(4-fluoro-1-methylisoquinolin-5-yl)carbamate (S16). To a solution of S15 (0.524 g, 1.90 mmol) in THF (10 mL) was added NaH 60% in mineral oil (59.2 mg, 2.49 mmol). After effervescence ceased, the resulting solution was refluxed for 30 min. To the reaction mixture was then added MeI (0.350 g, 4.49 mmol) in THF (2 mL) and the solution refluxed overnight. The mixture was concentrated and passed through a silica plug (1:10-2:1 EtOAc:hexanes). The mixture was concentrated *in vacuo* and the crude residue was purified by flash column chromatography (gradient, 10–30% EtOAc:hexanes). The isoquinoline S16 was obtained as an amber oil containing a mixture of rotamers (0.456 g, 82%). ¹H NMR (400 MHz, CDCl₃) δ 8.25–8.22 (m, 1.5H), 8.11–8.02 (m, 1.5H), 7.71–7.61 (m, 2H), 7.55 (dd, J = 7.3, 1.3 Hz, 1H), 3.28 (s, 3H), 3.27 (s, 1.5H), 2.96 (s, 3H), 2.95 (s, 1.5H), 1.53 (s, 4.5H), 1.21 (s, 9H). ¹³C NMR (125 MHz, CDCl₃) δ 155.41, 154.92 (d, ${}^4J_{\text{C-F}}$ = 5.7 Hz), 154.90 (d, ${}^4J_{\text{C-F}}$ = 5.4 Hz), 154.63, 154.51, 153.54 (d, ${}^1J_{\text{C-F}}$ = 259.3 Hz), 137.94, 131.60, 130.52, 130.05, 129.74, 128.44, 128.19, 127.82 (d, ${}^2J_{\text{C-F}}$ = 27.6 Hz), 125.49, 125.08, 124.53, 124.34 (d, ${}^3J_{\text{C-F}}$ = 8.1 Hz), 80.79, 80.23, 38.52 (d, ${}^5J_{\text{C-F}}$ = 3.82 Hz), 37.81 (d, ${}^5J_{\text{C-F}}$ = 3.07 Hz), 28.38 (3C), 28.05 (3C), 22.46, 22.26, two low-field carbons were either not observed or is overlapping with another low-field carbon. ¹⁹F NMR (376 MHz, CDCl₃) δ -140.37, -141.22. DART-MS: m/z calcd. for C₁₆H₂₀FN₂O₂ (M+H) + 291.15033, found 291.14981.

tert-Butyl methyl(4-fluoro-1-formylisoquinolin-5-yl)carbamate (S17). To a solution of S16 (0.40 g, 1.38 mmol) in 1,4-dioxane (10 mL) was added SeO₂ (0.183 g, 1.65 mmol). The mixture was stirred at 60 °C overnight then cooled to 22 °C. The mixture was concentrated *in vacuo* and the crude residue was purified by flash column chromatography (gradient, 5–25% EtOAc:hexanes). The isoquinoline S17 was obtained as an off-white solid containing a mixture of rotamers (0.152 g, 36%). ¹H NMR (500 MHz, CDCl₃) δ 10.32 (d, J = 1.5 Hz, 1H), 10.29 (d, J = 1.6 Hz, 0.5H), 9.38 (tdd, J = 7.5, 2.7, 1.4 Hz, 1.5H),

8.58 (dd, J = 3.9, 1.1 Hz, 1H), 8.56 (dd, J = 4.0, 1.3 Hz, 0.5H), 7.80 (tt, J = 7.3, 1.4 Hz, 1.5H), 7.68 (dt, J = 7.5, 0.5Hz, 0.5H), 7.62 (dt, J = 7.4, 1.0 Hz, 1H), 3.31 (d, J = 1.1 Hz, 3H), 3.30 (d, J = 0.9 Hz, 1.5H), 1.54 (s, 4.5H), 1.21 (s, 9H). ¹³C NMR (125 MHz, CDCl₃) 194.12, 194.09, 156.39 (d, ${}^{1}J_{\text{C-F}}$ = 272.5 Hz), 156.35 (d, ${}^{1}J_{\text{C-F}}$ = 255.9 Hz), 155.21, 154.41, 146.67 (d, ${}^{3}J_{\text{C-F}}$ = 6.1 Hz), 137.64 (d, ${}^{4}J_{\text{C-F}}$ = 1.7 Hz), 131.73 (d, ${}^{4}J_{\text{C-F}}$ = 1.9 Hz), 131.12 (d, ${}^{4}J_{\text{C-F}}$ = 1.9 Hz), 131.05, 130.95 (d, ${}^{2}J_{\text{C-F}}$ = 21.5 Hz), 130.85, 130.57 (d, ${}^{2}J_{\text{C-F}}$ = 28.5 Hz), 130.25 (d, ${}^{2}J_{\text{C-F}}$ = 28.2 Hz), 129.83 (d, ${}^{4}J_{\text{C-F}}$ = 2.4 Hz), 129.49 (d, ${}^{4}J_{\text{C-F}}$ = 2.5 Hz), 125.05, 124.68 (d, ${}^{4}J_{\text{C-F}}$ = 1.7 Hz), 124.56 (${}^{3}J_{\text{C-F}}$, J = 7.4 Hz), 124.44 (d, ${}^{3}J_{\text{C-F}}$ = 7.1 Hz), 81.01, 80.49, 38.56 (d, ${}^{5}J_{\text{C-F}}$ = 3.3 Hz), 37.83 (d, ${}^{5}J_{\text{C-F}}$ = 2.6 Hz), 28.36 (3C), 28.03 (3C), one low-field carbon were either not observed or is overlapping with another low-field carbon. ¹⁹F NMR (376 MHz, CDCl₃) δ -133.9. DART-MS: m/z calcd. for $C_{16}H_{18}FN_2O_3$ (M+H) $^+$ 305.12960, found 305.12824.

(E)-2-((4-Fluoro-5-(methylamino)isoquinolin-1-yl)methylene)-N₂N-dimethylhydrazine-1carbothioamide and (Z)-2-((4-Fluoro-5-(methylamino)isoquinolin-1-yl)methylene)-N,Ndimethylhydrazine-1-carbothioamide (HCT15). To a solution of S17 (30.0 mg, 0.099 mmol) in MeOH (3.0 mL) was added 4,4-dimethyl-3-thiosemicarbazide (11.7 mg, 0.985 mmol) and HCl (98 µL, 0.59 mmol, 6 M in H₂O). The mixture was microwaved at 300 W and 50 °C for 1.0 h. The hydrochloride salt that formed was then neutralized with saturated aqueous NaHCO₃ solution (1.5 mL). The precipitate of the desired compound was collected by filtration, washed with water, EtOH and then dried to yield the isoguinoline HCT15 as a pale-yellow solid containing a mixture of E- and Z-isomers (12.2 mg, 41%). ¹H NMR (500 MHz, DMSO- d_6) δ 15.46 (s, 0.33H), 11.13 (br s, 1H), 8.91 (dd, J = 8.4, 2.9 Hz, 1H), 8.62 (s, 1H), 8.50 (d, J = 5.1 Hz, 0.33H), 8.39 (s, 0.33H), 8.32 (d, J = 4.8 Hz, 1H), 7.89 (dd, J = 8.5, 2.9 Hz, 0.33H), 7.65 (t, J = 8.2 Hz, 0.33H), 7.57 (t, J = 8.2 Hz, 1H), 6.82 (d, J = 8.0 Hz, 0.33H), 6.73 (d, J = 7.9Hz, 1H), 6.55 (dd, J = 11.9, 5.2 Hz, 0.33H), 6.39 (dd, J = 12.4, 5.0 Hz, 1H), 3.37 (s, 1.98H), 3.31 (s, 6H), 2.86-2.84 (m, 3.99H). ¹³C NMR (125 MHz, DMSO- d_6) δ 180.95, 180.72, 156.41 (d, J = 260.4 Hz), $147.99 \text{ (d, }^{4}J_{\text{C-F}} = 4.3 \text{ Hz)}, 147.41, 147.16, 144.92, 144.61 (d, {}^{4}J_{\text{C-F}} = 3.7 \text{ Hz)}, 131.90, 131.69, 130.83,$ 130.78, 129.29 (d, ${}^{4}J_{C-F}$ = 2.4 Hz), 127.41 (d, ${}^{2}J_{C-F}$ = 28.8 Hz), 125.43 (d, ${}^{2}J_{C-F}$ = 30.5 Hz), 115.98 (d, ${}^{2}J_{C-F}$ = 7.6 Hz), 113.89, 113.84, 110.26, 108.50, 107.78, 42.15 (4C), 30.95 (2C), one low-field carbon were either not observed or is overlapping with another low-field carbon. 19 F NMR (376 MHz, DMSO- d_6) δ -125.86, -129.02. DART- MS: m/z calcd. for $C_{14}H_{17}FN_5S$ (M+H)⁺ 306.11832, found 306.11716.

6-Fluoro-1-methyl-5-nitroisoquinoline (S18). To a solution of **S11** (0.584 g, 3.623 mmol) in sulfuric acid (0.8 mL) at 0° C was added KNO₃ (0.366 g, 3.623 mmol) in sulfuric acid (1.2 mL). The mixture was heated at 60° C for 2 h and then poured slowly over crushed ice. The solution was made alkaline with

NH₄OH; the resulting tan precipitate was filtered, washed with water, and dried to afford **S18** as a tan solid (0.264 g, 35%). 1 H NMR (500 MHz, CDCl₃) δ 8.58 (d, J = 6.1 Hz, 1H), 8.41 (dd, J = 9.4, 4.9 Hz, 1H), 7.70 (d, J = 6.0 Hz, 1H), 7.55 (t, J = 9.2 Hz, 1H), 3.07 (s, 3H). 13 C NMR (125 MHz, CDCl₃) δ 159.08, 155.06 (d, 1 J_{C-F} = 266.6 Hz), 144.50, 132.29 (d, 3 J_{C-F} = 10.0 Hz), 129.83 (2C), 124.19, 117.40 (d, 2 J_{C-F} = 23.5 Hz), 113.60, 22.41. 19 F NMR (376 MHz, CDCl₃) δ -113.01. DART-MS: m/z calcd. for C₁₀H₈FN₂O₂ (M+H) $^{+}$ 207.05643, found 207.05690.

6-Fluoro-1-methylisoquinolin-5-amine (S19). To a solution of **S18** (0.264 g, 1.28 mmol) in MeOH (60 mL) iron powder (0.214 g, 3.83 mmol) and HCl (1 mL, 12 M in H₂O). The mixture was refluxed for 2 h and then a solution of sodium hydroxide (2 mL, 6 M in H₂O) was added. The mixture was filtered and extracted with diethyl ether (200 mL). The organic layer was dried over Na₂SO₄, filtered, and then concentrated *in vacuo*. The crude residue was purified by flash column chromatography (gradient, 10–30% EtOAc:hexanes). The isoquinoline **S19** was obtained as a brown solid (145.8 mg, 82%). ¹H NMR (500 MHz, CDCl₃) δ 8.36 (d, J = 6.2 Hz, 1H), 7.62 (dd, J = 9.1, 4.8 Hz, 2H), 7.44 (d, J = 9.9 Hz, 1H), 4.27 (br s, 2H), 3.06 (s, 3H). ¹³C NMR (125 MHz, CDCl₃) δ 164.80, 159.77 (d, $^1J_{C-F}$ = 263.7 Hz), 150.34, 139.41 (d, $^3J_{C-F}$ = 10.8 Hz), 137.94, 133.78, 128.91, 122.74 (d, $^2J_{C-F}$ = 22.7 Hz), 118.04 (d, $^4J_{C-F}$ = 5.2 Hz), 27.69. ¹⁹F NMR (376 MHz, CDCl₃) δ -125.82. DART-MS: m/z calcd. for C₁₀H₁₀FN₂ (M+H)⁺ 177.08225, found 177.08291.

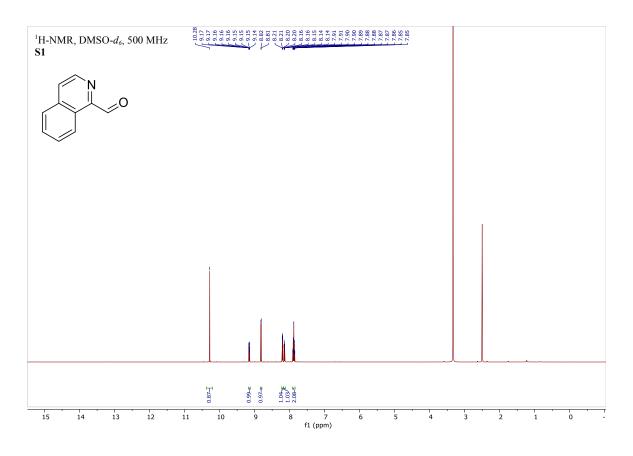
tert-Butyl (6-fluoro-1-methylisoquinolin-5-yl)carbamate (S20). To a solution of S19 (0.715 g, 4.06 mmol) in THF (15 mL) was added DMAP (49.5 mg, 0.41 mmol) then Boc₂O (2.21 g, 10.14 mmol) and the mixture was stirred at 22 °C overnight. After completion of the reaction as attested by TLC, K₂CO₃ (1.68 g, 12.17 mmol) and MeOH (10 mL) were added to the reaction mixture and was refluxed overnight. The mixture was then concentrated *in vacuo* and resuspended in EtOAc (20 mL) and water (20 mL). The organic layer was separated and the aqueous layer was extracted with EtOAc (3 × 20 mL). The organic layers were combined and dried over Na₂SO₄, filtered, and then concentrated *in vacuo*. The crude residue was purified by flash column chromatography (gradient, 5–20% EtOAc:hexanes). The isoquinoline S20 was obtained as a brown oil (0.303 g, 27%). ¹H NMR (500 MHz, CDCl₃) δ 8.40 (d, J = 6.0 Hz, 1H), 8.06 (dd, J = 9.3, 5.0 Hz, 1H), 7.65 (d, J = 6.0 Hz, 1H), 7.39 (t, J = 9.3 Hz, 1H), 6.59 (br s, 1H), 2.95 (s, 3H), 1.50 (s, 9H). ¹³C NMR (125 MHz, CDCl₃) δ 160.58 (d, ¹J_{C-F} = 260.2 Hz), 157.19, 153.19, 137.68, 131.09 (d, ⁴J_{C-F} = 4.9 Hz), 128.23 (d, ³J_{C-F} = 10.7 Hz), 124.83, 124.45, 121.14 (d, ²J_{C-F} = 22.1 Hz), 119.60, 82.72, 28.15 (3C), 17.84. ¹⁹F NMR (376 MHz, CDCl₃) δ -112.86. DART-MS: m/z calcd. for C₁₅H₁₈FN₂O₂ (M+H)⁺ 277.13468, found 277.13425.

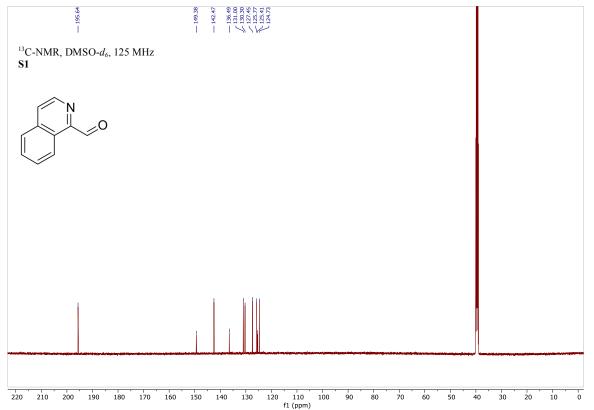
tert-Butyl methyl(6-fluoro-1-methylisoquinolin-5-yl)carbamate (S21). To a solution of S20 (0.150 g. 0.543 mmol) in THF (4 mL) was added NaH 60% in mineral oil (28.0 mg, 0.706 mmol). After effervescence ceased, the resulting solution was refluxed for 30 min. To the reaction mixture was added the MeI (0.10 g, 0.706 mmol) in THF (0.5 mL) and the solution refluxed overnight. The mixture was concentrated and passed through a silica plug (1:10-2:1 EtOAc:hexanes). The mixture was concentrated in vacuo and the crude residue was purified by flash column chromatography (gradient, 10–30%) EtOAc:hexanes). The isoquinoline S21 was obtained as a mixture of rotational isomers as an amber oil (0.120 g, 76%). H NMR (500 MHz, CDCl₃) δ 8.44 (d, J = 5.8 Hz, 1.27H), 8.16–8.02 (m, 1.27H), 7.51 (d, 1.27H)J = 6.0 Hz, 1H), 7.49 (d, J = 6.3 Hz, 0.27H), 7.39 (t, J = 9.3 Hz, 1.27H), 3.26 (s, 0.81H), 3.25 (s, 3H), 2.98 (s, 3H), 2.96 (s, 0.81H), 1.56 (s, 2.43H), 1.26 (s, 9H). ¹³C NMR (125 MHz, CDCl₃) δ 158.89, 158.88, 158.09 (d, ${}^{1}J_{C-F}$ = 254.3 Hz), 154.97, 154.78, 143.18 (2C), 135.55, 135.36 (d, ${}^{4}J_{C-F}$ = 3.7 Hz), 127.66 (d, $^{3}J_{\text{C-F}} = 9.7 \text{ Hz}$), 127.38 (d, $^{3}J_{\text{C-F}} = 9.6 \text{ Hz}$), 125.25, 125.06 (2C), 124.88, 124.78, 117.42 (d, $^{2}J_{\text{C-F}} = 24.0 \text{ Hz}$) Hz), 117.12 (d, ${}^{2}J_{C-F}$ = 24.1 Hz), 114.35 (d, ${}^{3}J_{C-F}$ = 5.8 Hz), 81.18, 80.61, 37.43, 36.39, 28.35 (3C), 27.99 (3C), 22.62, 22.56, one low-field carbon were either not observed or is overlapping with another low-field carbon. ¹⁹F NMR (376 MHz, CDCl₃) δ -114.54, -115.33. DART-MS: m/z calcd. for $C_{16}H_{20}FN_2O_2(M+H)^+$ 291.15033, found 291.15011.

tert-Butyl methyl(6-fluoro-1-formylisoquinolin-5-yl)carbamate (S22). To a solution of S21 (0.1000 g, 0.344 mmol) in 1,4-dioxane (2 mL) was added SeO₂ (38.2 mg, 0.344 mmol). The mixture was stirred at 60 °C overnight then cooled to 22 °C. The mixture was concentrated *in vacuo* and the crude residue was purified by flash column chromatography (gradient, 5–25% EtOAc:hexanes). The isoquinoline S22 was obtained as an off-white solid containing a mixture of rotamers (45.3 mg, 43%). ¹H NMR (500 MHz, CDCl₃) δ 10.37 (s, 1H), 10.35 (s, 0.3H), 9.35 (dd, J = 9.4, 5.1 Hz, 1.3H), 8.82 (d, J = 5.8 Hz, 1.3H), 7.92 (d, J = 5.7 Hz, 1H), 7.88 (d, J = 5.8 Hz, 0.3H), 7.56 (t, J = 9.4 Hz, 1.3H), 3.29 (s, 0.9H), 3.28 (s, 3H), 1.57 (s, 2.7H), 1.25 (s, 9H). ¹³C NMR (125 MHz, CDCl₃) δ 195.40 (2C), 158.41 (d, $^{1}J_{\text{C-F}}$ = 257.6 Hz), 154.77 (2C), 149.92, 143.81, 143.72 (2C), 136.73 (d, $^{4}J_{\text{C-F}}$ = 4.6 Hz), 136.59 (d, $^{4}J_{\text{C-F}}$ = 3.8 Hz), 128.17 (d, $^{3}J_{\text{C-F}}$ = 10.7 Hz), 127.91 (d, $^{3}J_{\text{C-F}}$ = 9.3 Hz), 124.72 (d, $^{3}J_{\text{C-F}}$ = 13.3 Hz), 124.00, 123.80, 120.68 (d, $^{2}J_{\text{C-F}}$ = 24.7 Hz), 120.44 (d, $^{2}J_{\text{C-F}}$ = 24.1 Hz), 120.31, 120.22 (d, $^{3}J_{\text{C-F}}$ = 6.3 Hz), 81.55, 81.00, 37.58, 36.53, 28.32 (3C), 27.97 (3C), two low-field carbon were either not observed or is overlapping with another low-field carbon. ¹⁹F NMR (376 MHz, CDCl₃) δ -112.18, -112.95. DART-MS: m/z calcd. for C₁₆H₁₈FN₂O₃ (M+H) ⁺ 305.1296, found 305.12819.

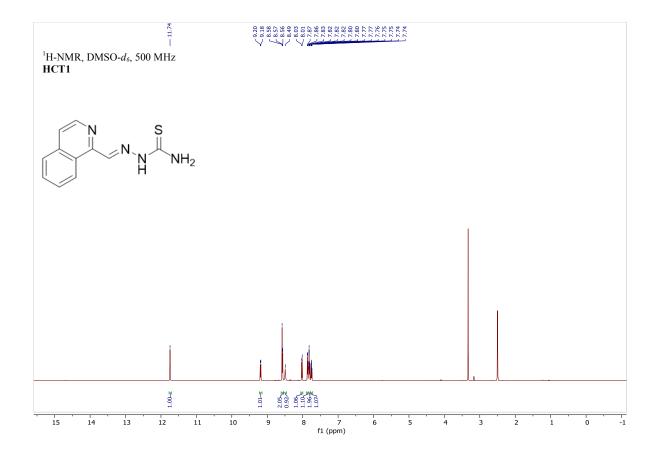
(*E*)-2-((6-Fluoro-5-(methylamino)isoquinolin-1-yl)methylene)-*N*,*N*-dimethylhydrazine-1-carbothioamide and (*Z*)-2-((6-Fluoro-5-(methylamino)isoquinolin-1-yl)methylene)-*N*,*N*-dimethylhydrazine-1-carbothioamide (HCT14). To a solution of S22 (10.0 mg, 0.033 mmol) in EtOH (0.5 mL) was added 4,4-dimethyl-3-thiosemicarbazide (3.9 mg, 0.033 mmol) and HCl (33 μL, 0.197 mmol, 6 M in H₂O). The mixture was stirred and refluxed for 1.5 h then cooled to 22 °C. The hydrochloride salt that formed was then neutralized with saturated aqueous NaHCO₃ solution (0.5 mL). the precipitate of the desired compound was collected by filtration, washed with water, EtOH and then dried to yield the isoquinoline HCT14 as a pale-yellow solid (6.7 mg, 67%). ¹H NMR (500 MHz, DMSO- d_6) δ 15.96 (s, 0.17H), 11.22 (Br s, 1H), 9.20 (s, 1H), 8.62–8.54 (m, 1.17H), 8.52 (s, 0.17H), 8.34 (d, J = 5.5 Hz, 1H), 8.20 (d, J = 6.2 Hz, 0.17H), 8.07 (dd, J = 9.3, 4.2 Hz, 0.17H), 7.87 (br s, 1H), 7.56 (dd, J = 13.6, 9.2 Hz, 0.17H), 7.33 (dd, J = 13.4, 9.5 Hz, 1H), 6.10 (br s, 0.17H), 5.69 (br s, 1H), 3.41 (s, 1.02H), 3.27 (s, 6H), 3.10 (t, J = 5.5 Hz, 0.51H), 3.05 (t, J = 5.2 Hz, 3H). A ¹³C NMR was not obtained. ¹⁹F NMR (376 MHz, CDCl₃) δ - 129.05, -129.53. DART-MS: m/z calcd. for C₁₄H₁₇FN₃S (M+H)⁺ 305.11832, found 305.11719.

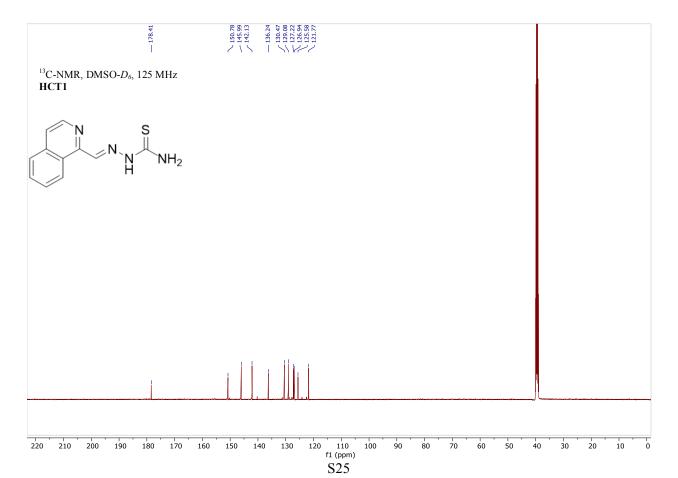
1 HNMR and 13 CNMR of compounds **S1-S22** and **HCT1-15**

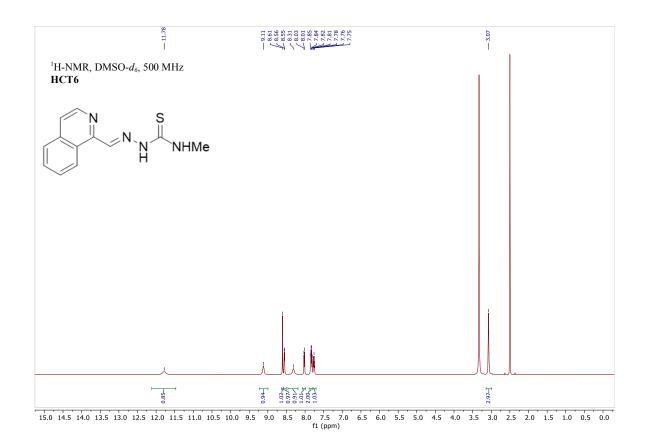


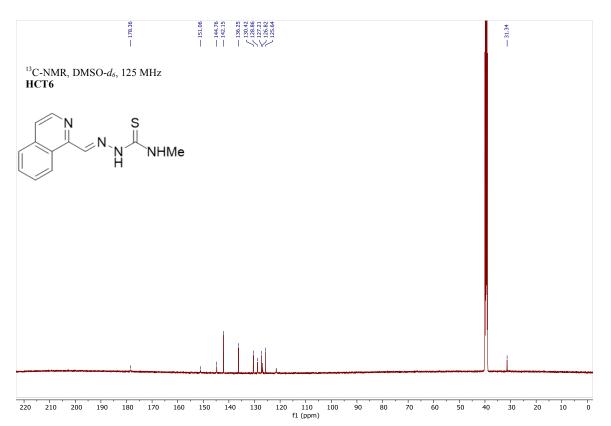


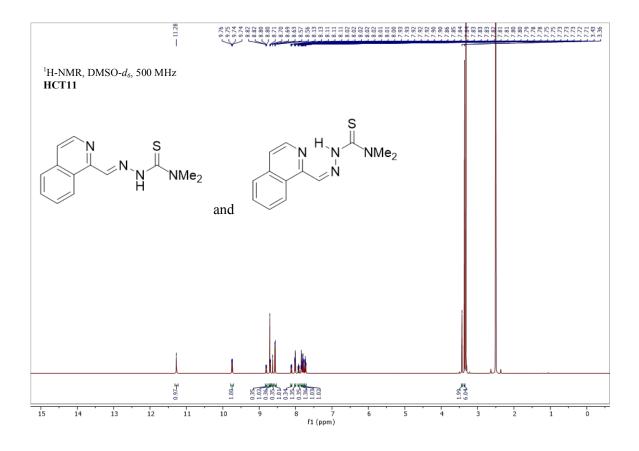
S24

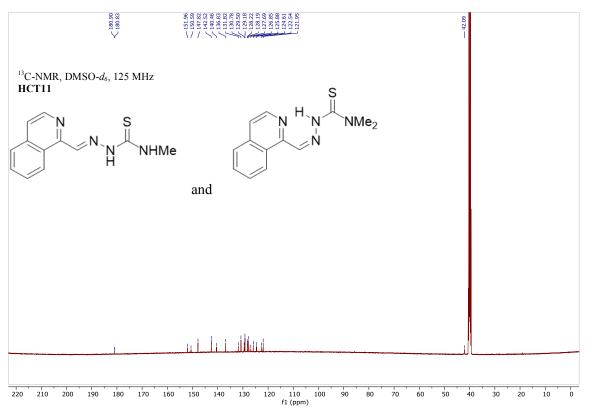


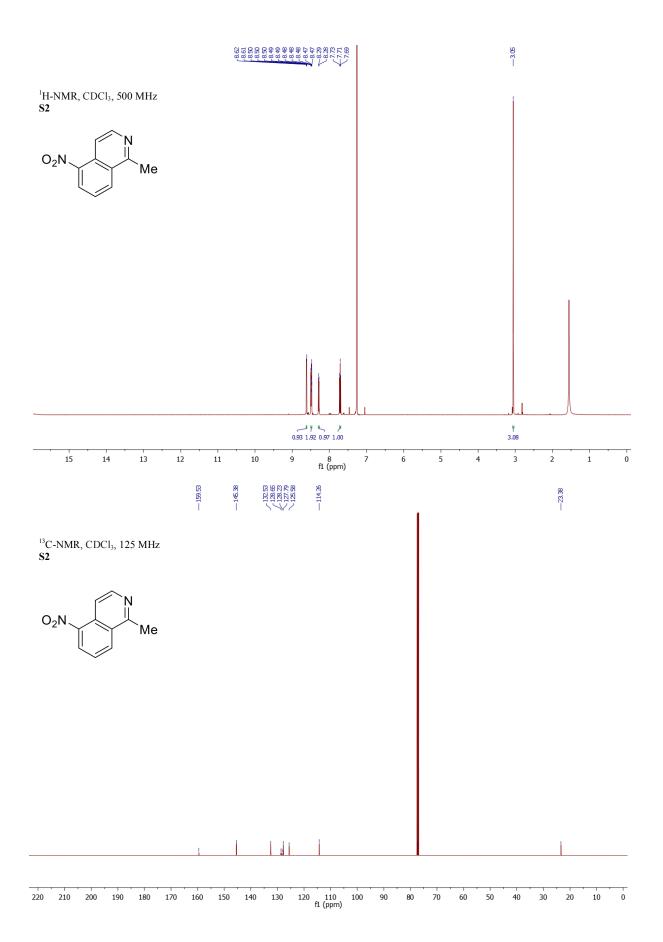


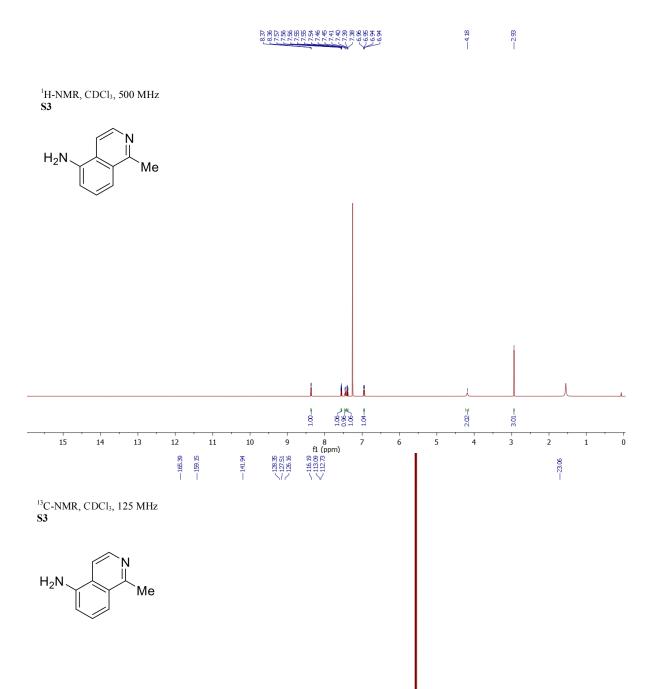








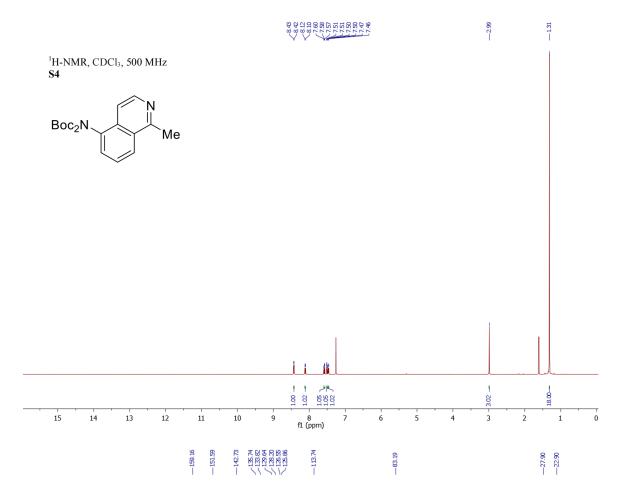




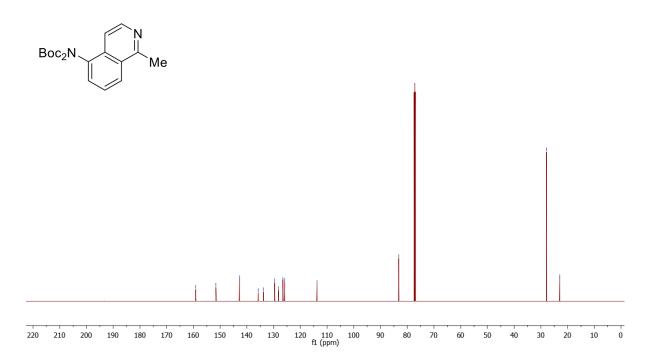
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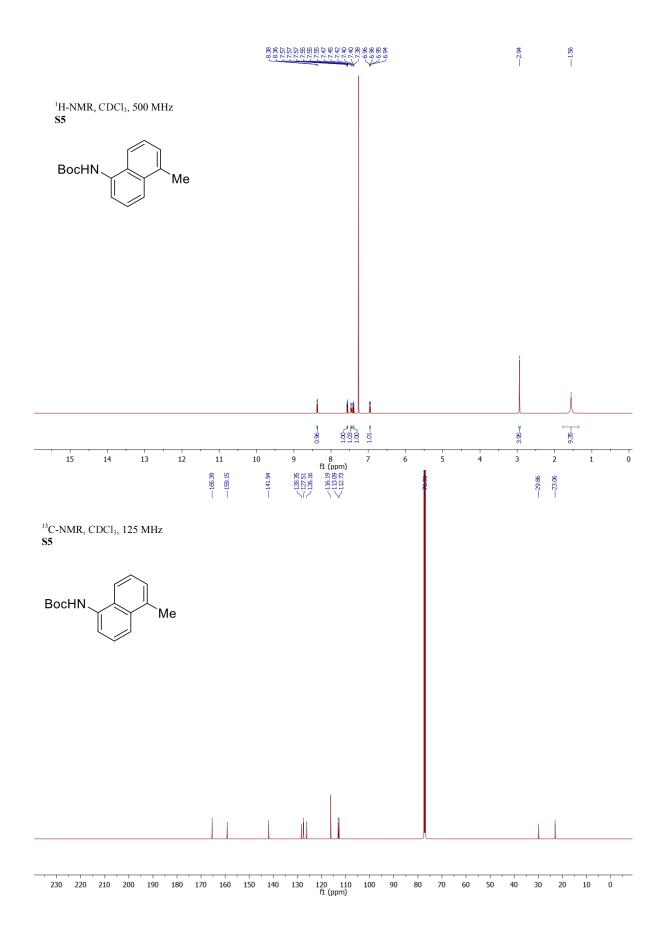
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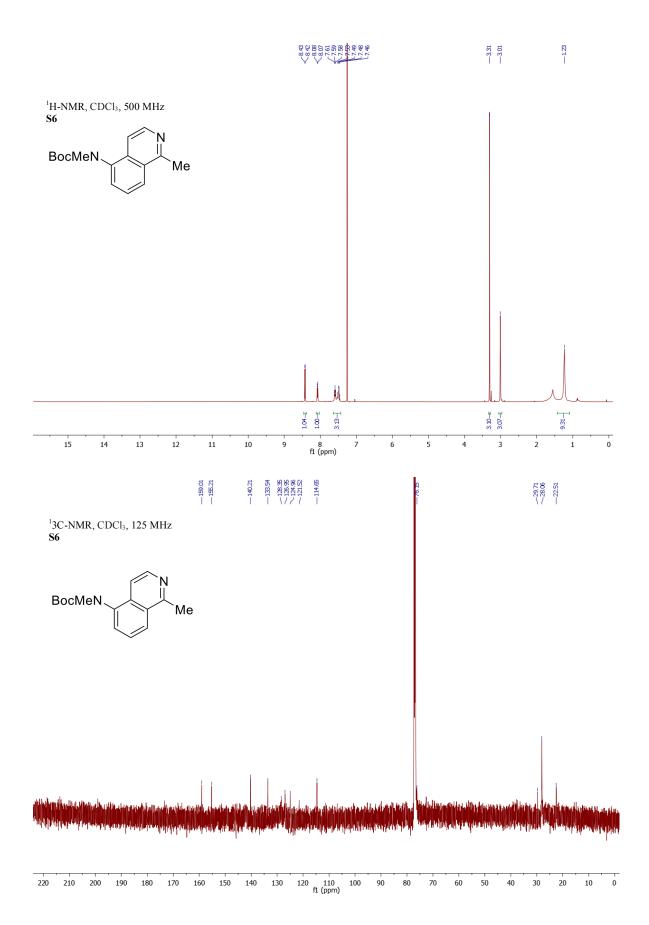
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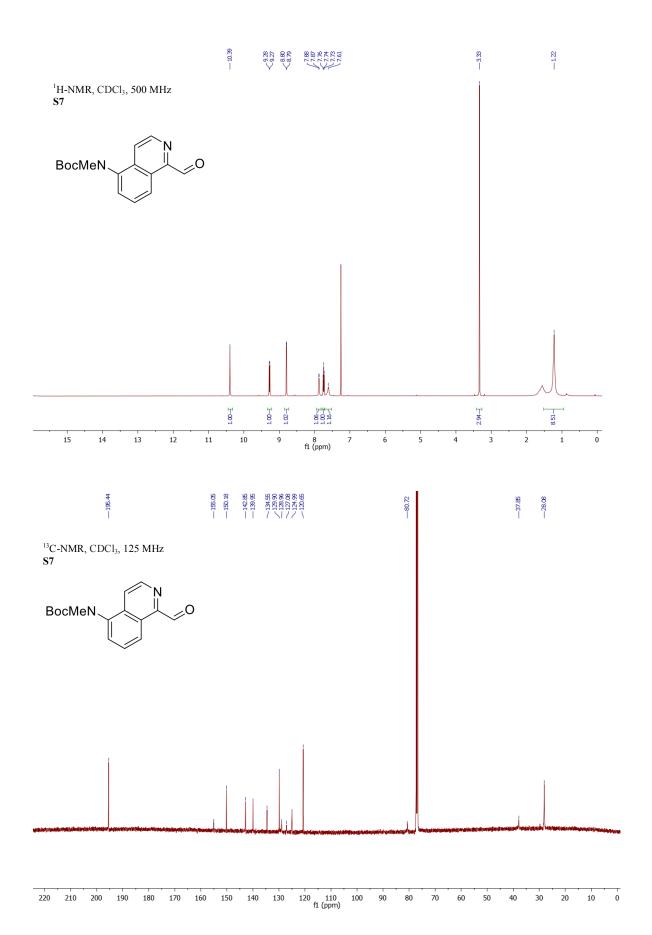


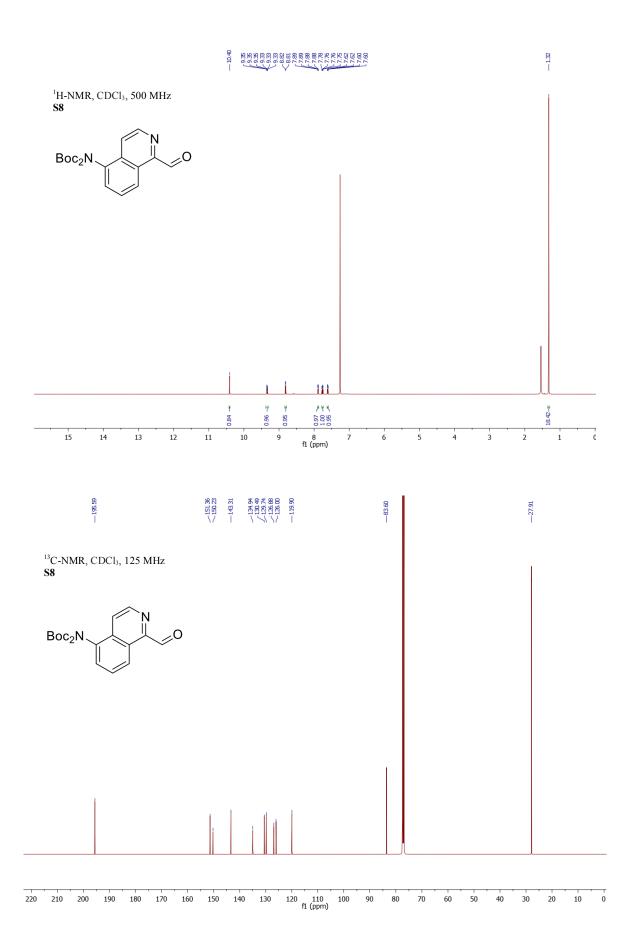
¹³C-NMR, CDCl₃, 125 MHz **S4**

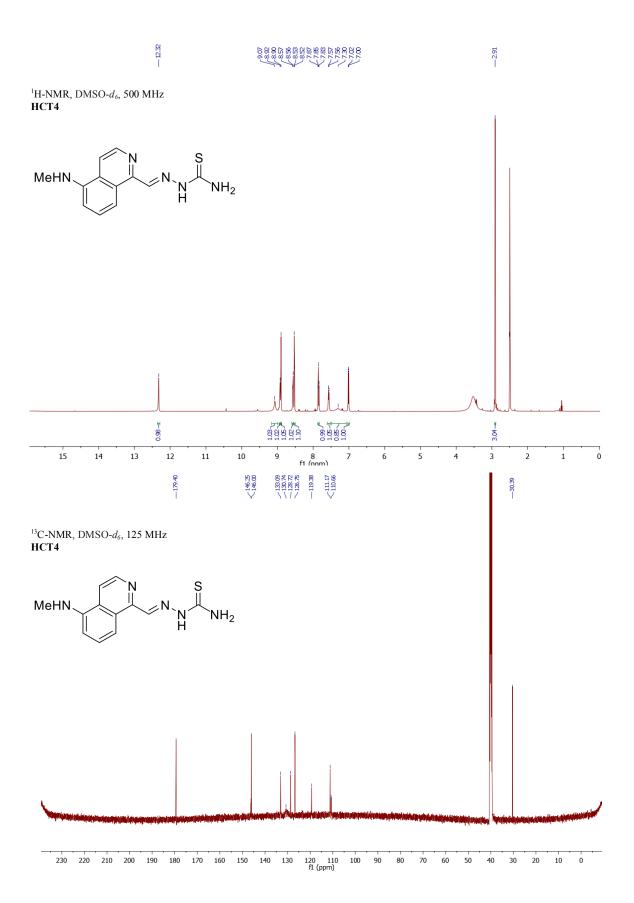


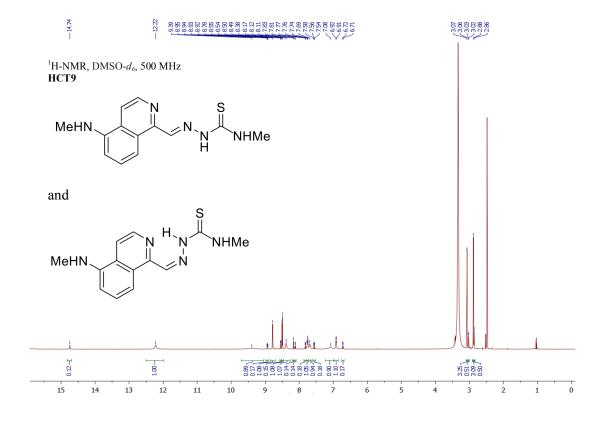


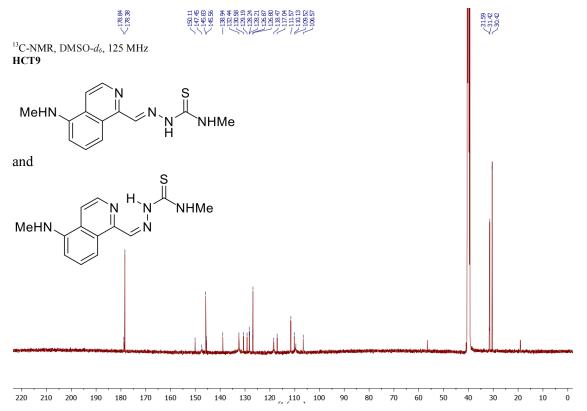


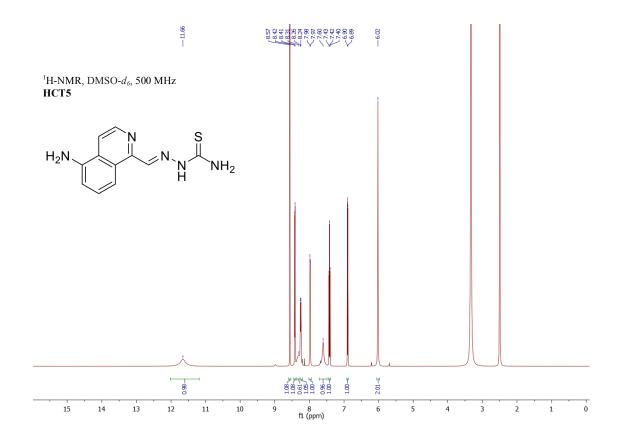


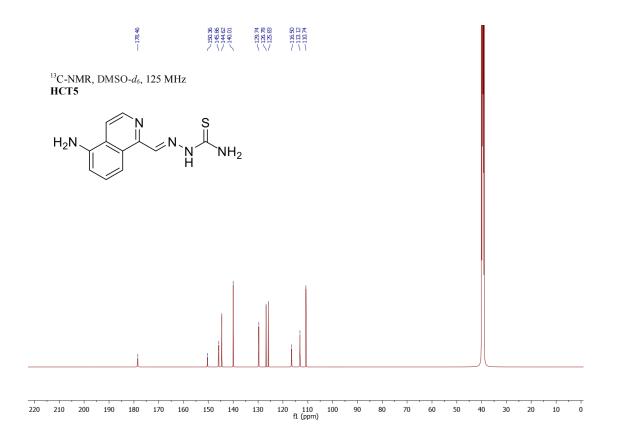


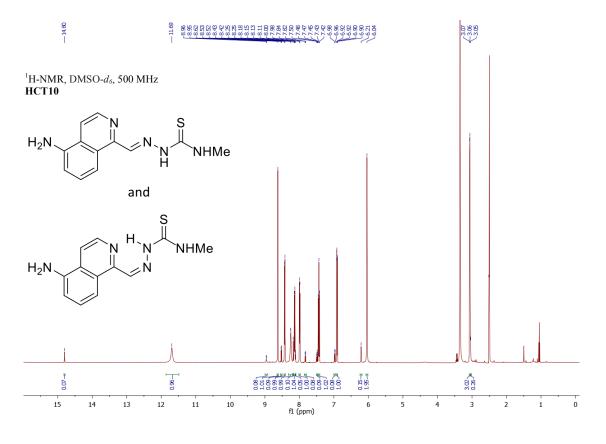


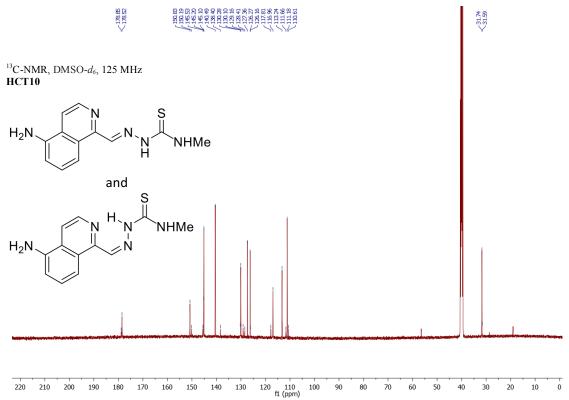


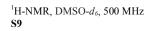






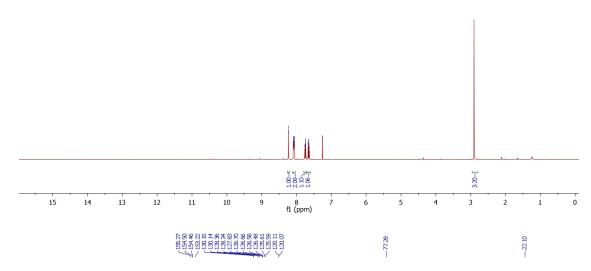




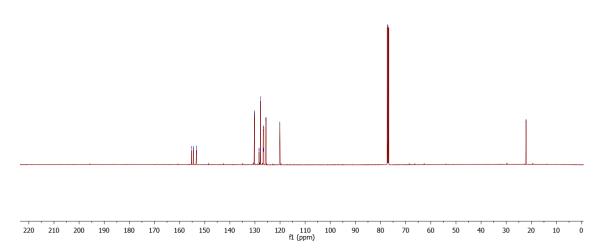






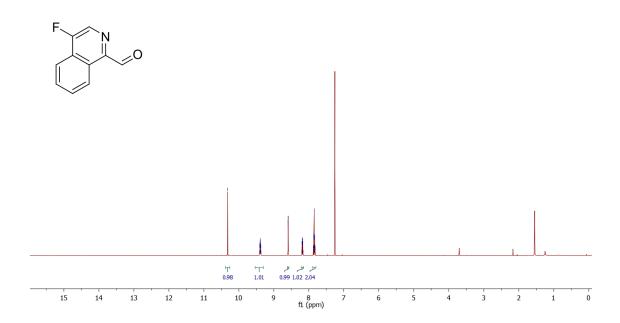


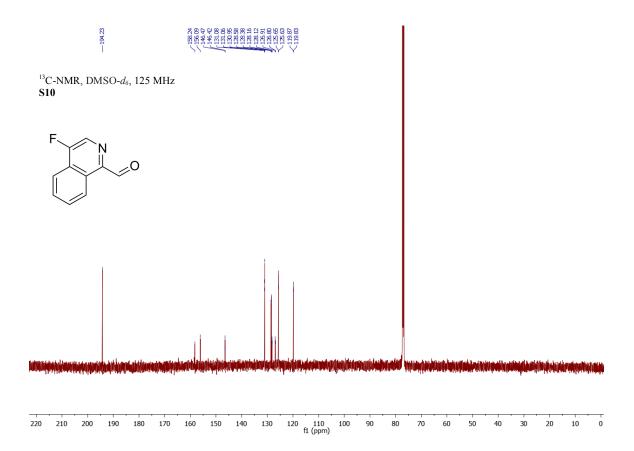
 13 C-NMR, DMSO- d_6 , 125 MHz **S9**

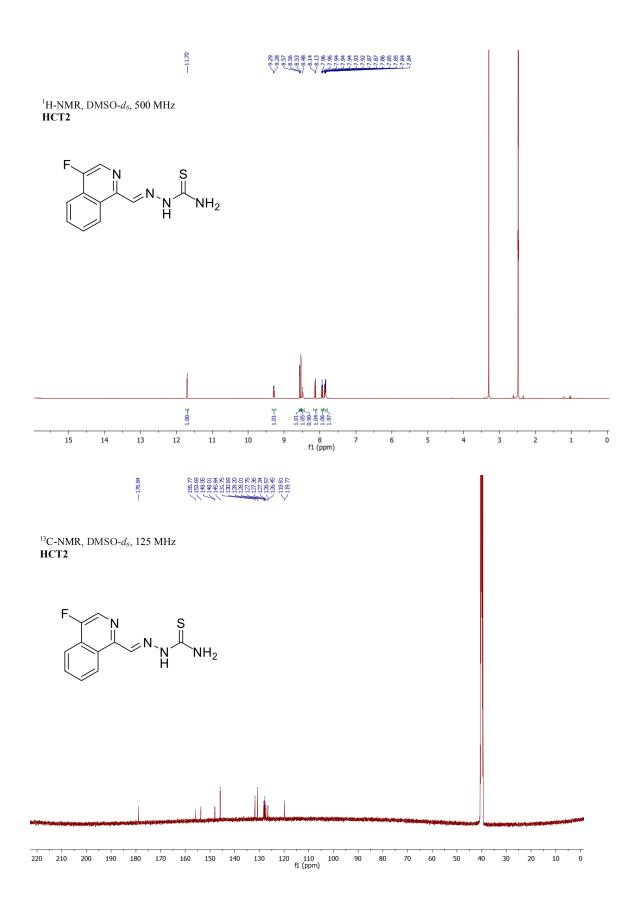


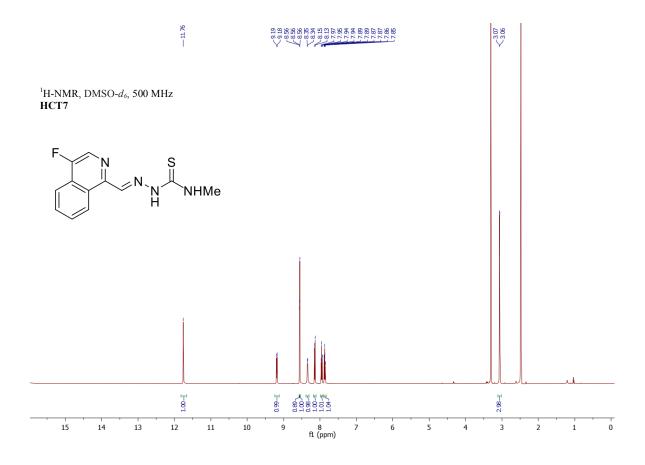


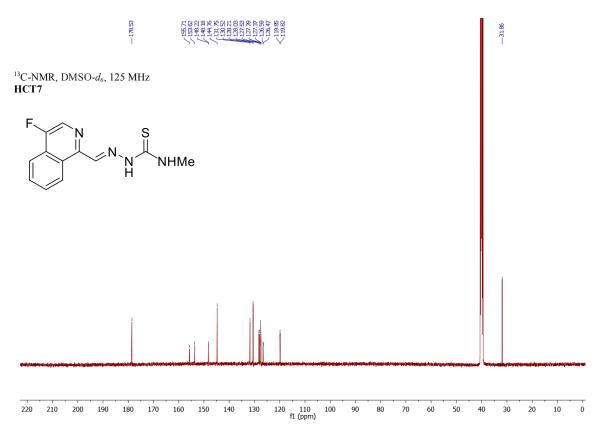
 1 H-NMR, DMSO- d_{6} , 500 MHz **S10**

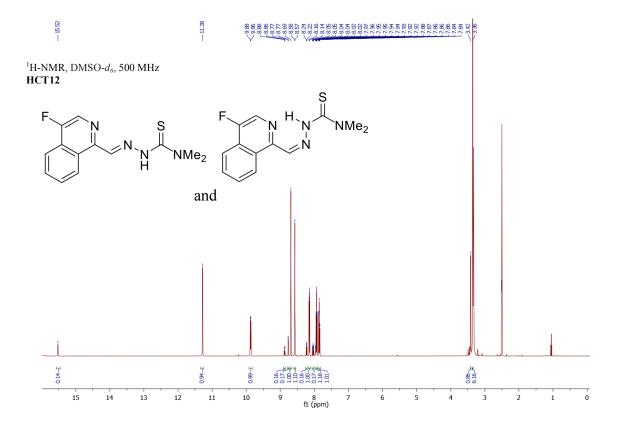


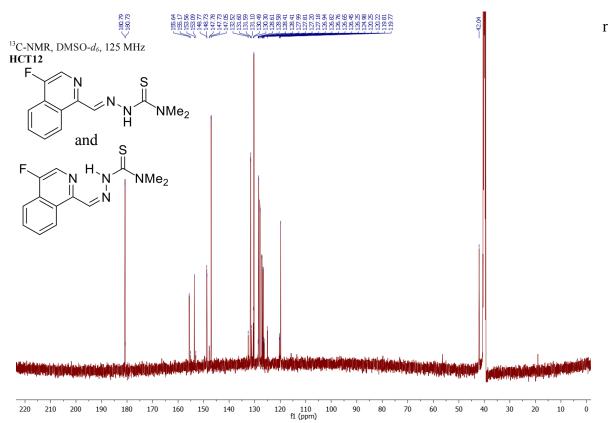


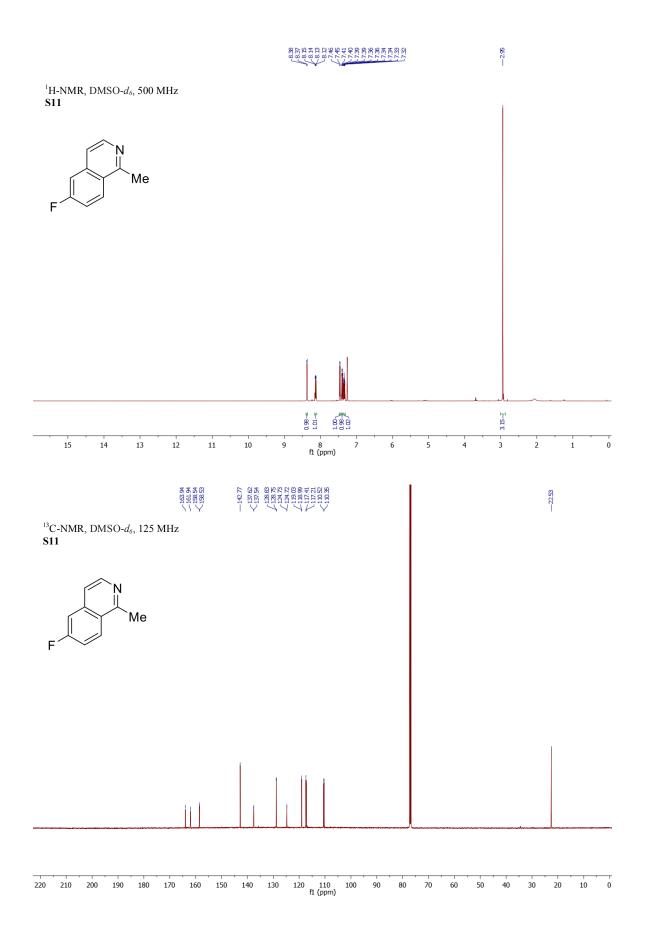




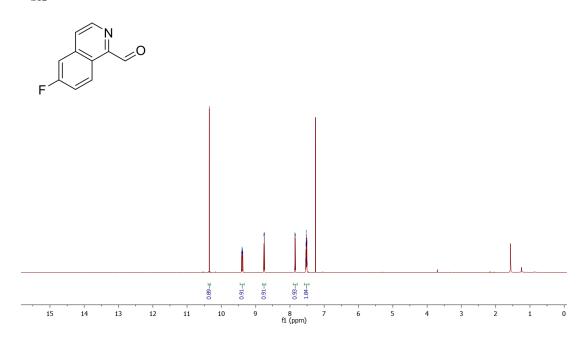




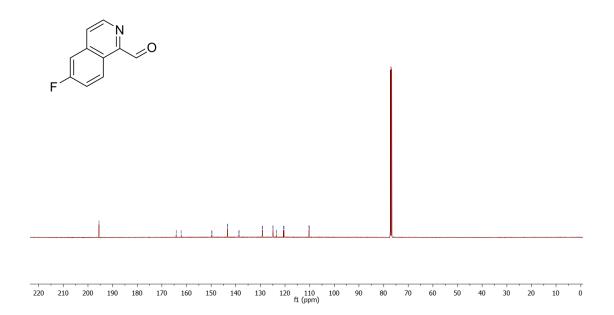


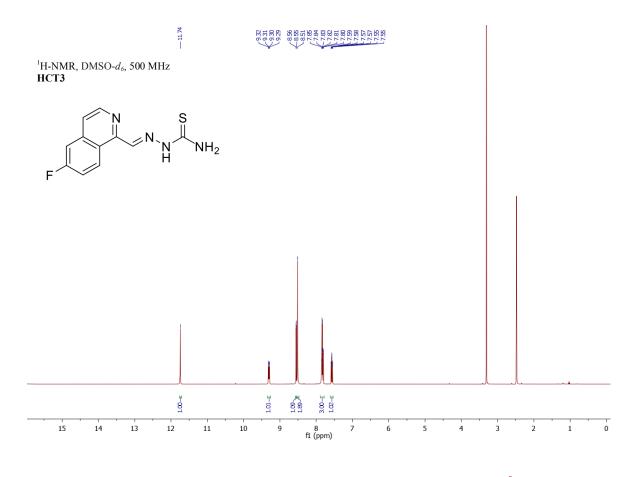


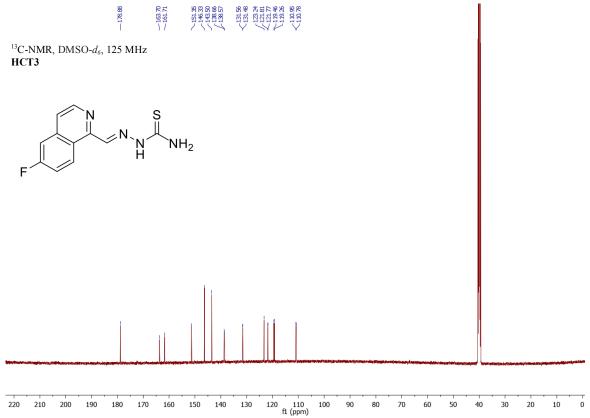
 1 H-NMR, DMSO- d_{6} , 500 MHz **S12**

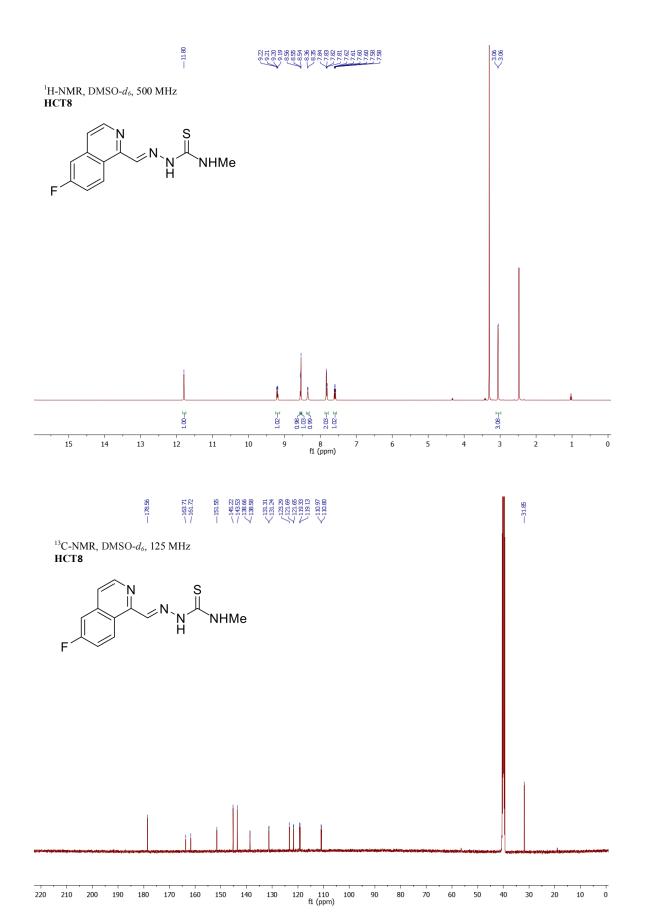


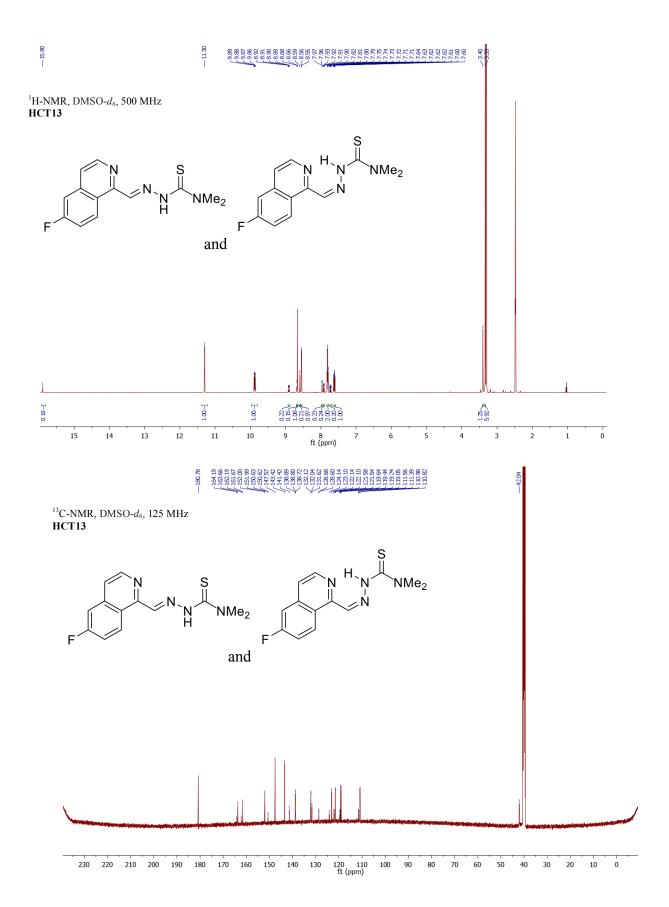
 13 C-NMR, DMSO- d_6 , 125 MHz **S12**

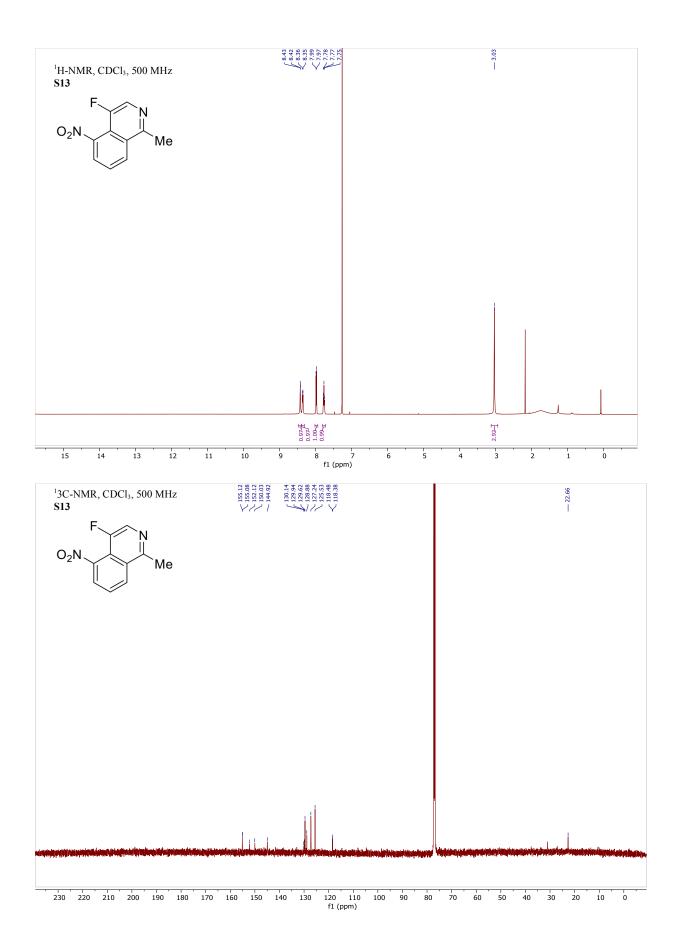


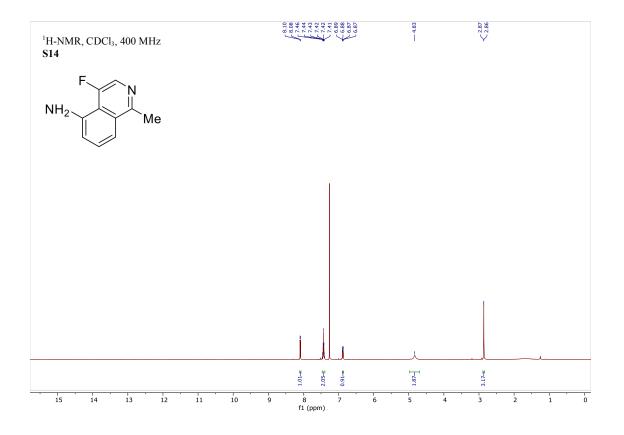


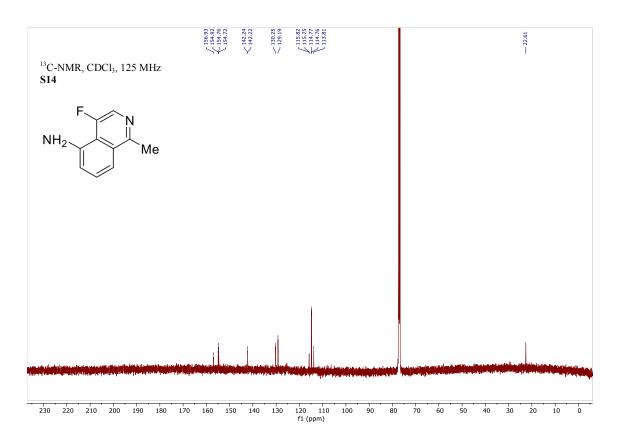


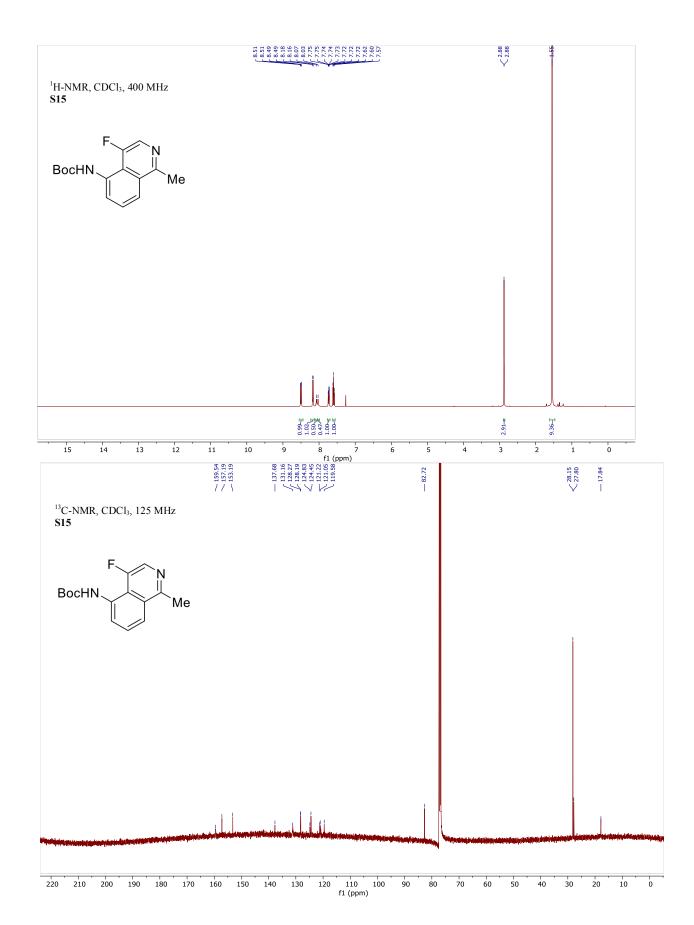


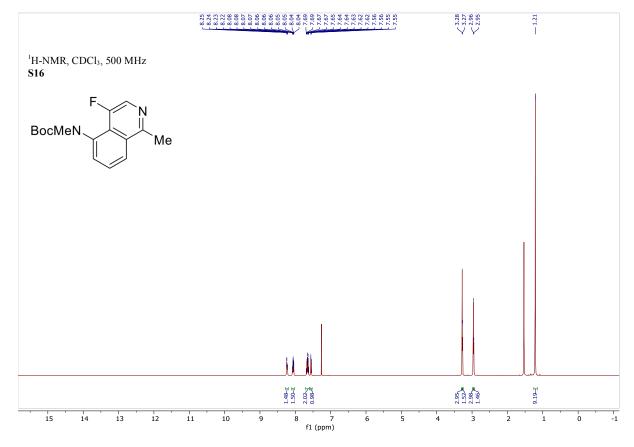


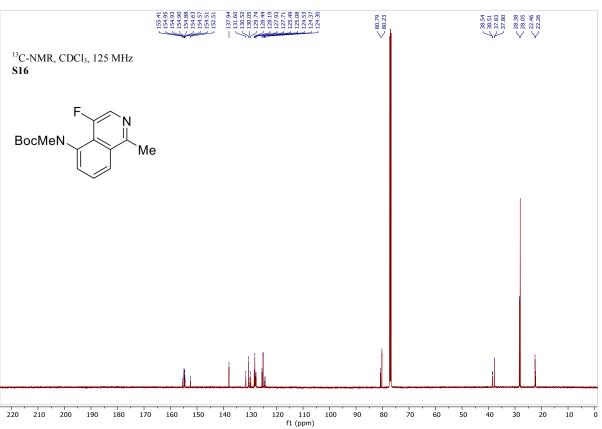


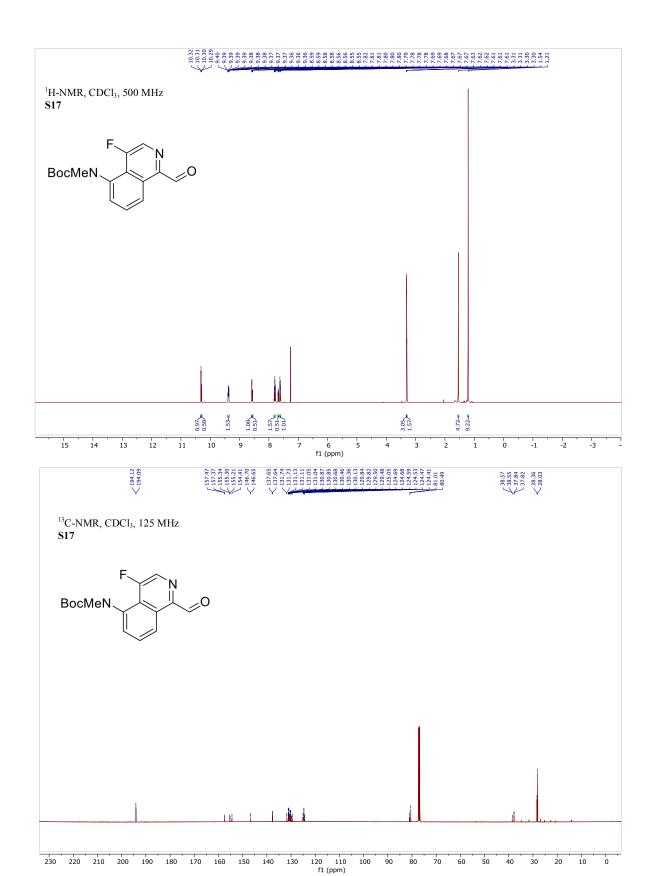


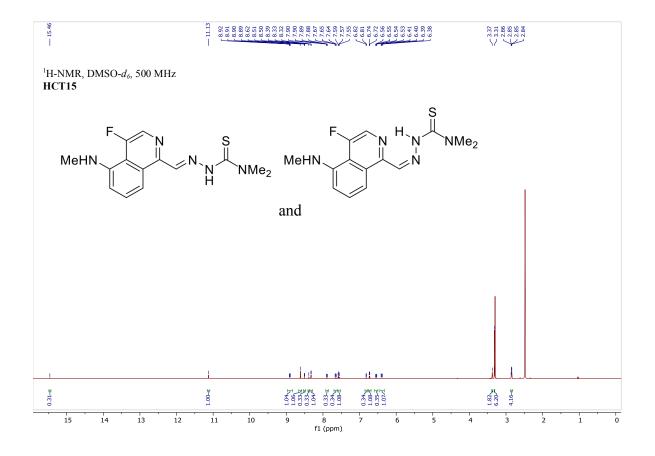


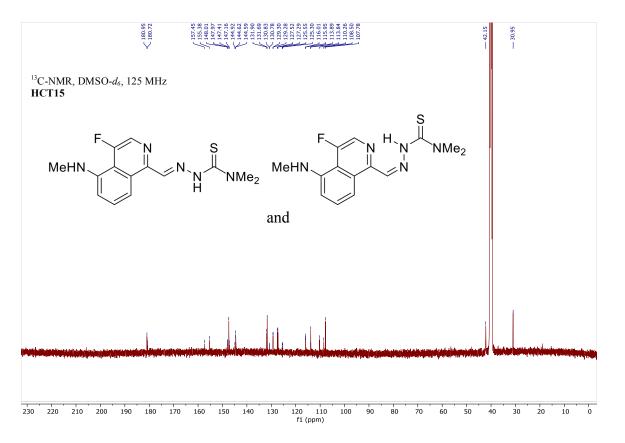


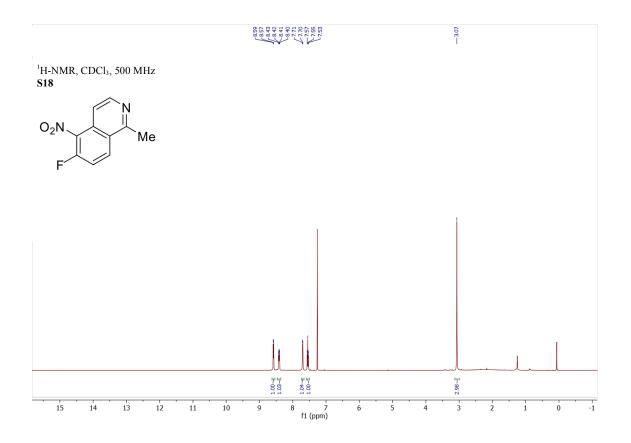


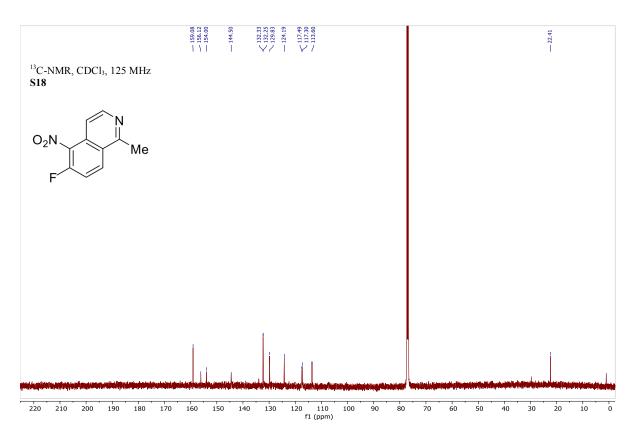


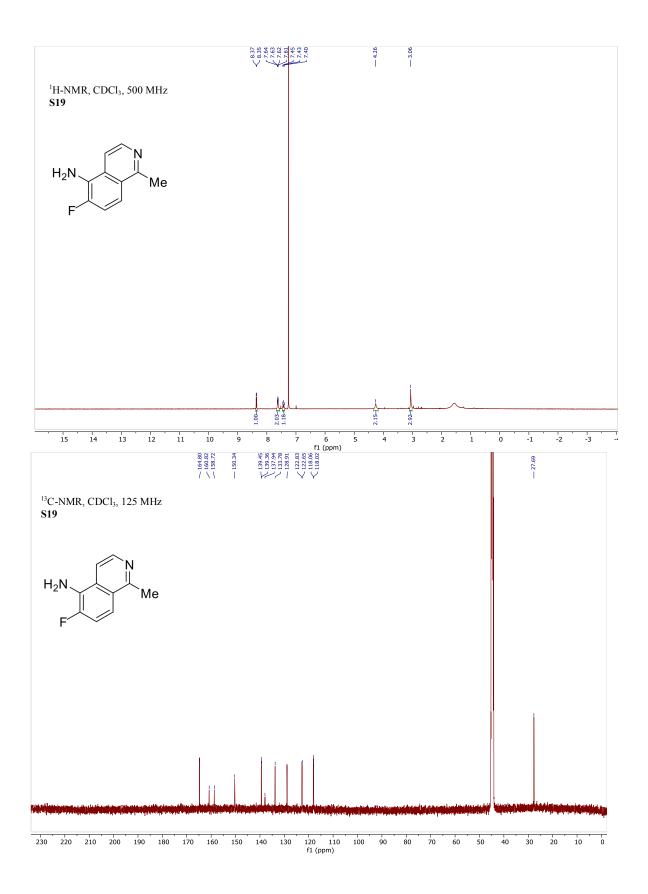


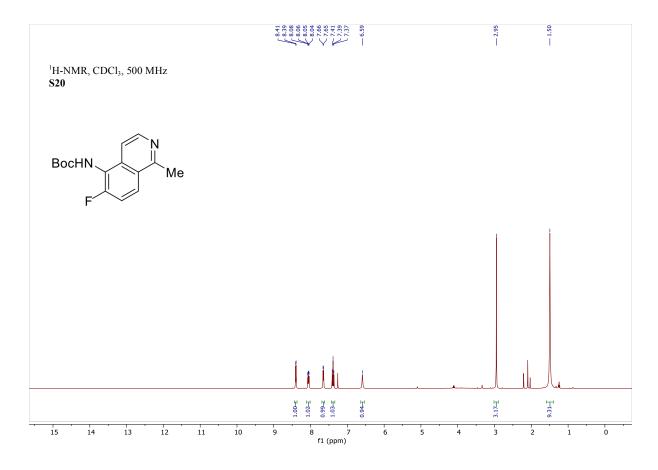


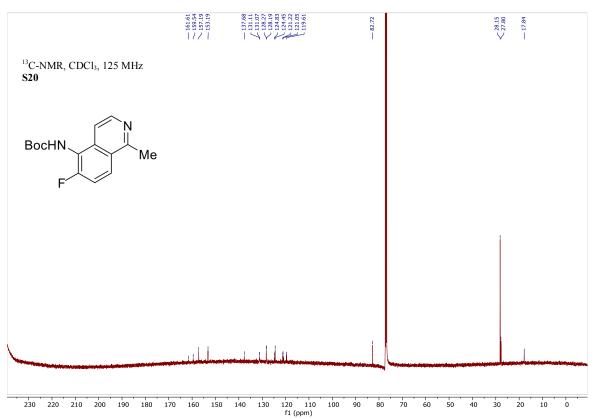


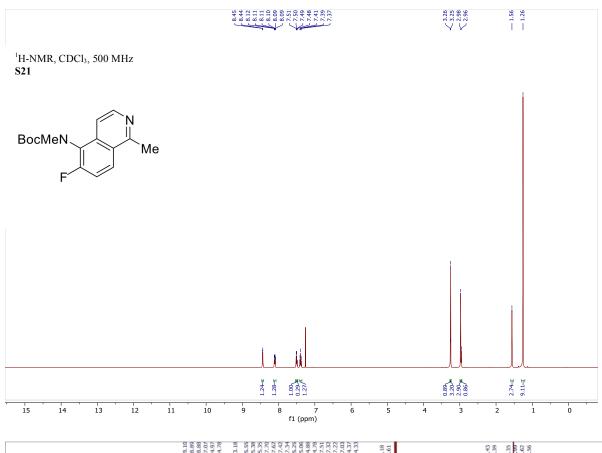


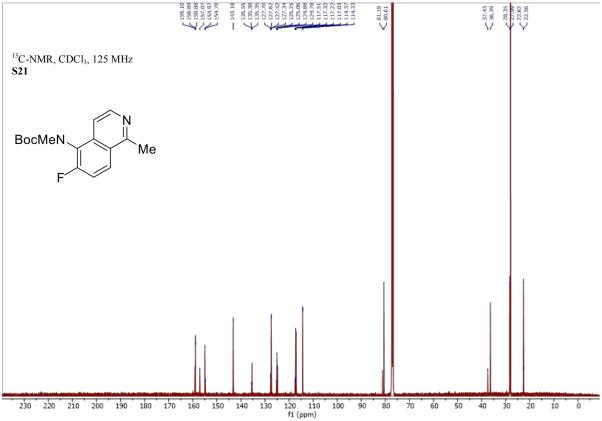


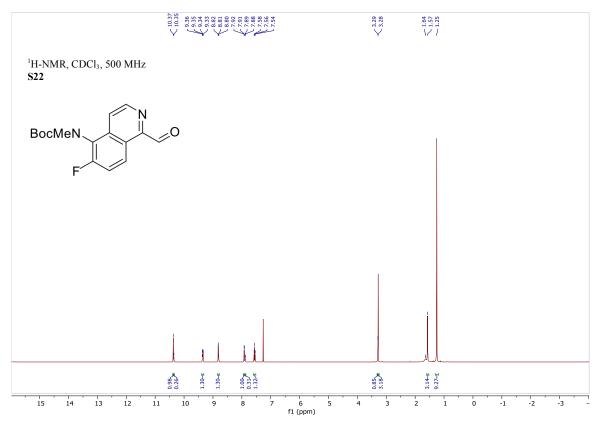


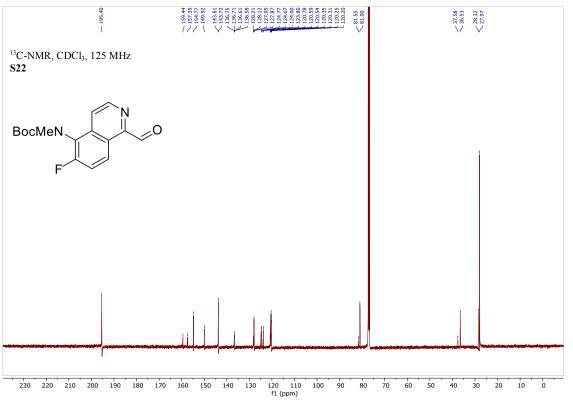


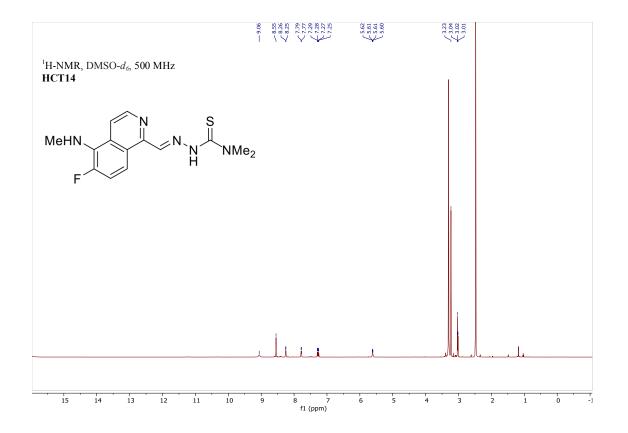


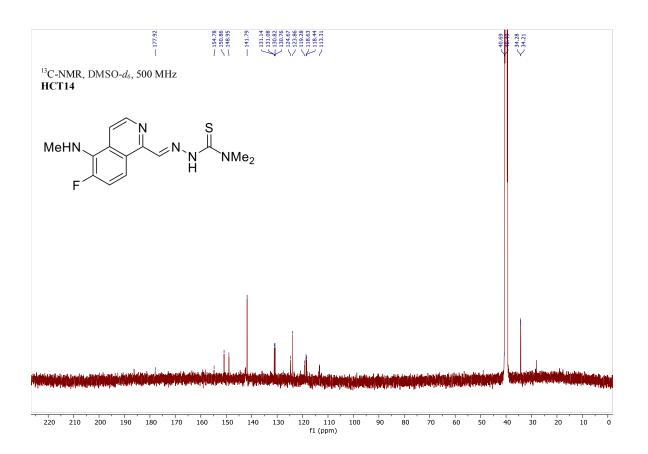


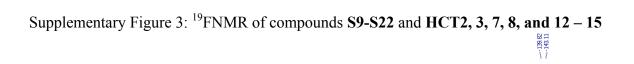






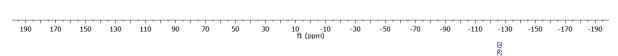




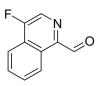






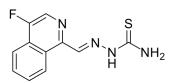


¹⁹F-NMR, CDCl₃, 376 MHz **S10**



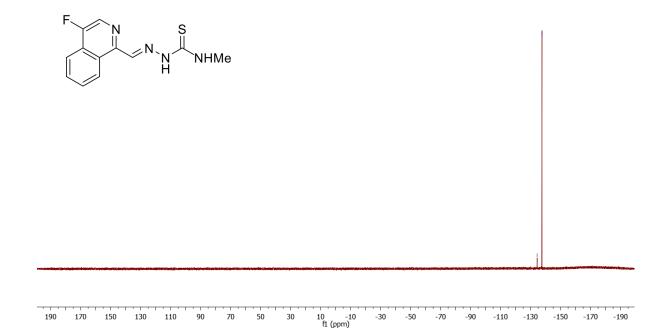


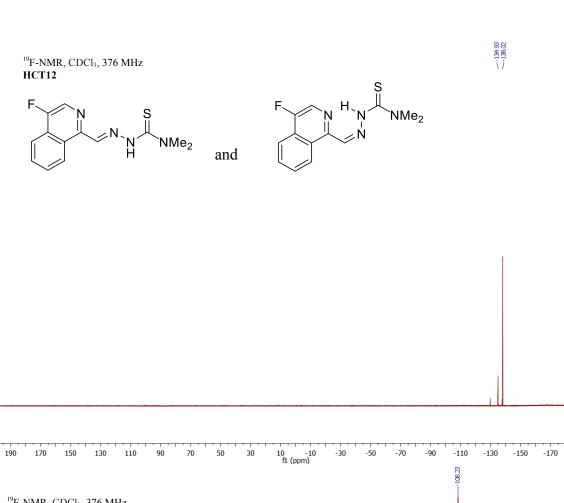
10 -10 f1 (ppm) ¹⁹F-NMR, CDCl₃, 376 MHz **HCT2**

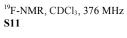


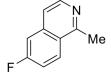


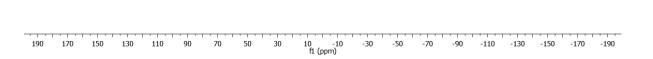
¹⁹F-NMR, CDCl₃, 376 MHz **HCT7**

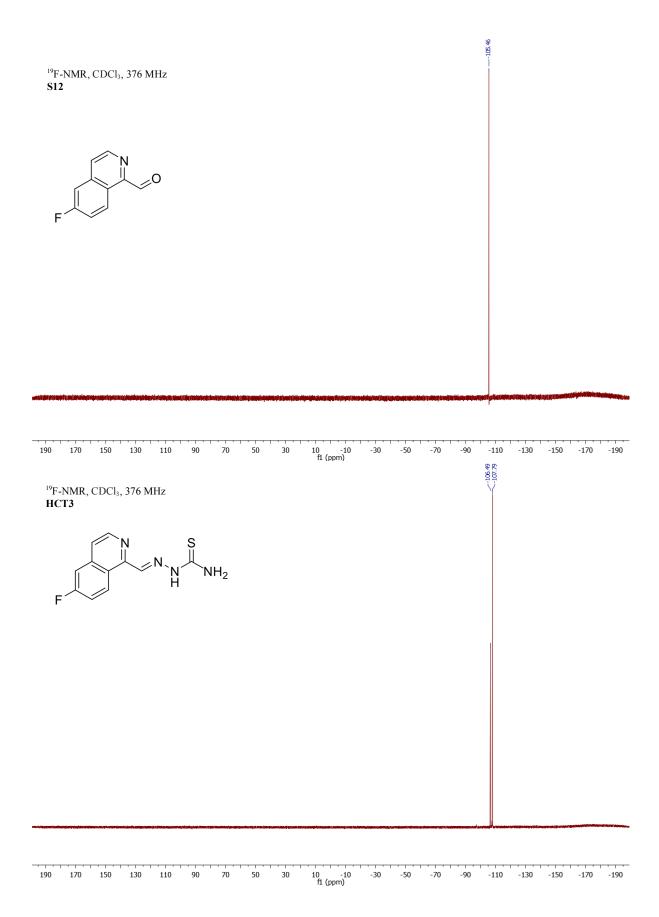


















-110 -130 -150 -170 170 150 110 -70 7-106.34 --107.97

¹⁹F-NMR, CDCl₃, 376 MHz HCT13

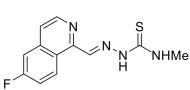
190

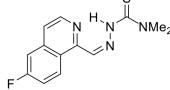
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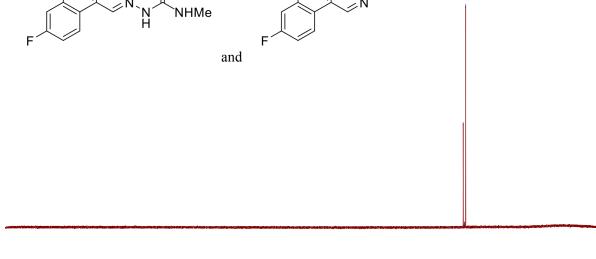
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130

110



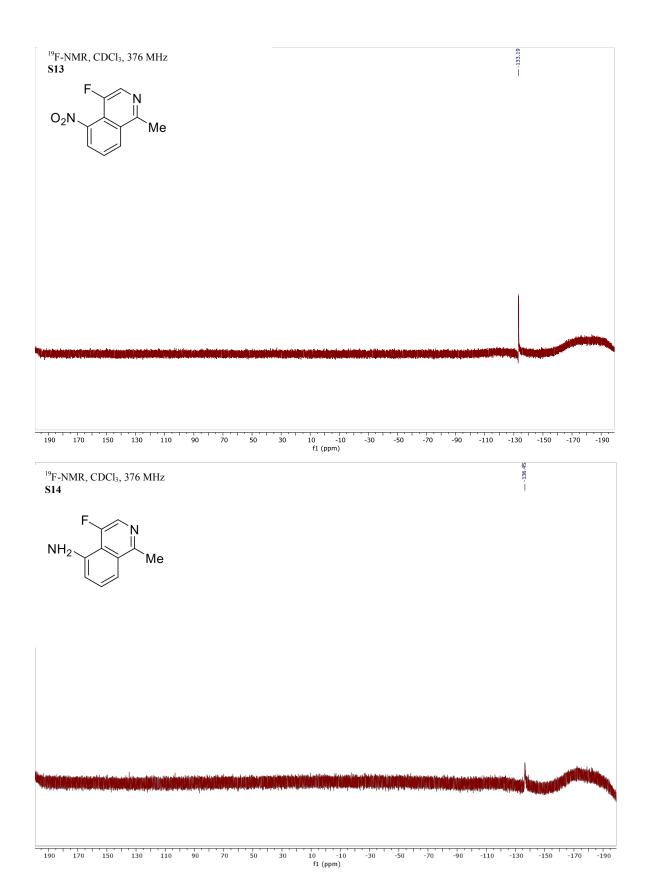


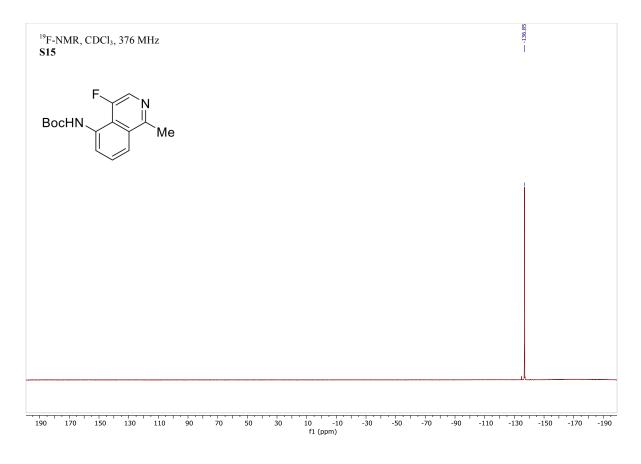


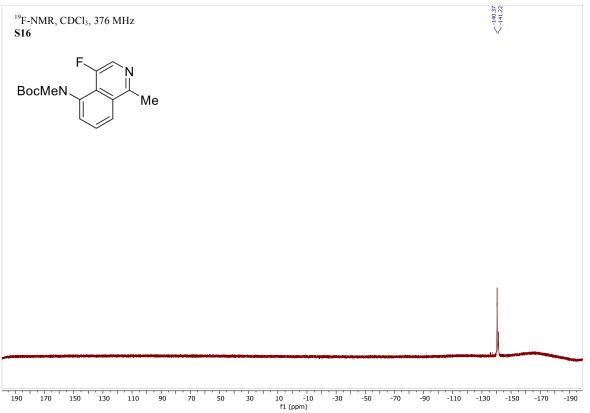
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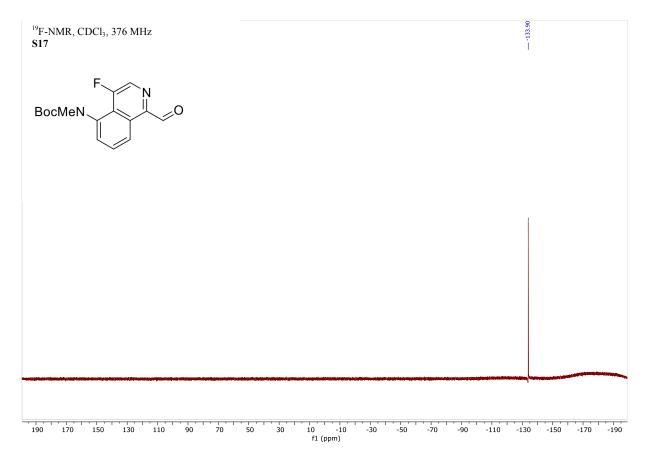
-50

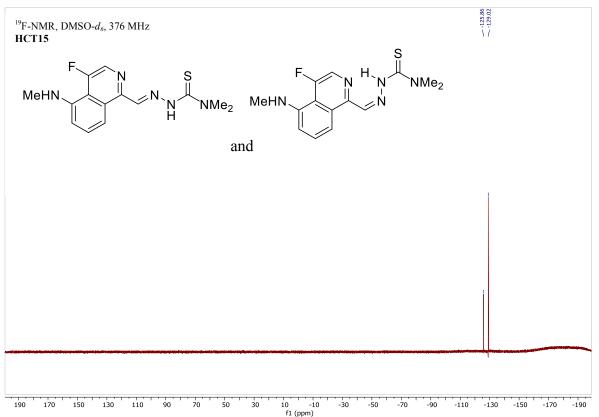
-110 -130 -150 -170

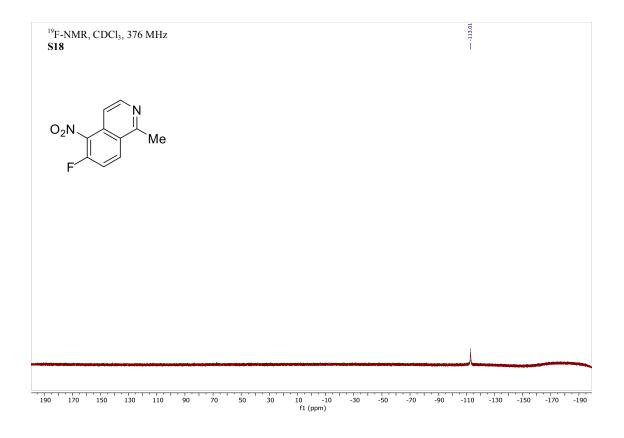


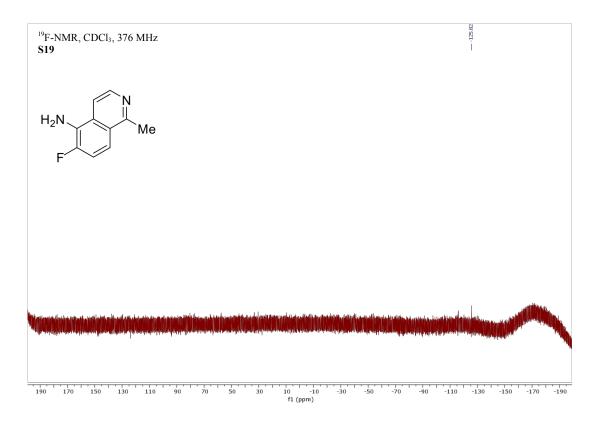


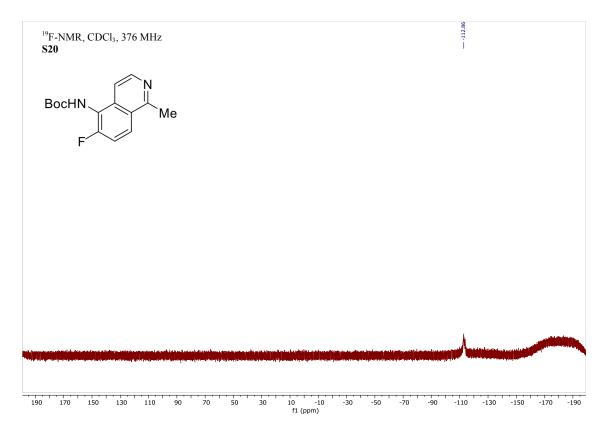


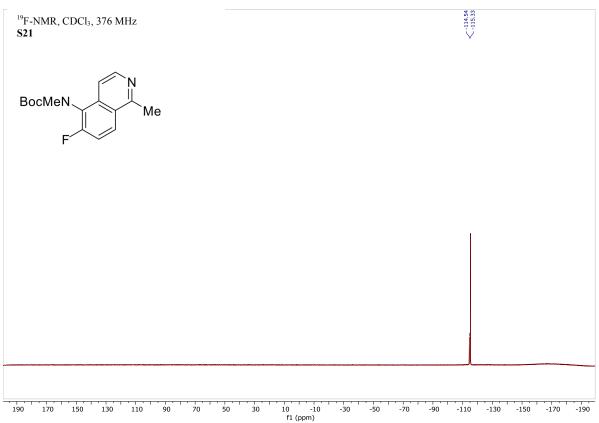


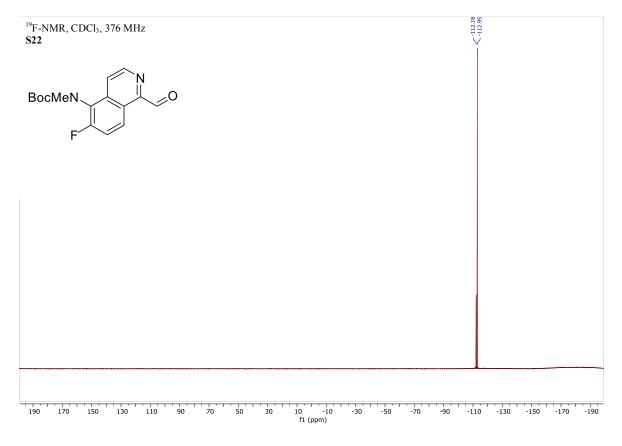


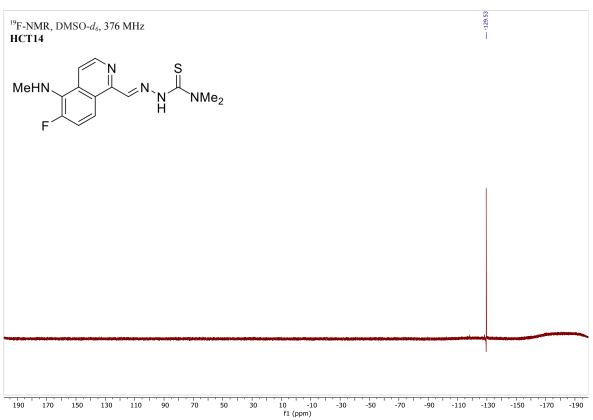












Supplementary Figure 4: HPLC analysis of HCT1-15

HPLC analysis of **HCT1**

Data File C:\HPCHEM\1\DATA\2018\052218\HCT-2.D

Sample Name: HCT-2

Injection Date : 5/22/2018 5:23:56 PM Sec. Line : 3
Sample Name : HCT-2 Location : Vial 2
Acq. Operator : JC Inj : 1
Inj Volume : 10 µl

Acc. Method : C:\HPCHEM\1\METHODS\JC-RP1.M
Last changed : 5/22/2018 3:51:27 PM by JC
Analysis Method : C:\HPCHEM\1\METHODS\JV095.M
Last changed : 12/12/2018 12:25:44 PM by Juno

Area Percent Report

Sorted By : Signal Multiplier : 1.0000 Dilution : 1.0000

Use Multiplier & Dilution Factor with ISTDs

Signal 1: DAD1 B, Sig=254,4 Ref=off

#	RetTime [min]		Width [min]	Area [mAU*s]	Height [mAU]	Area *
1	13.467	MM	0.2548	772.57324	50.52861	1.6953
2	14.451	MM	0.2133	4.34016e4	3390.72119	95.2392
3	14.799	MM	0.1675	979.93152	97.52905	2.1503
4	15.044	MM	0.1827	417.05139	32.53645	0.9152

Totals: 4.55711e4 3571.31531

\\PmsI-analysis2\HP LaserJet P3015 on Ne00:
Letter 8 1/2 x 11 in/Portrait

Close

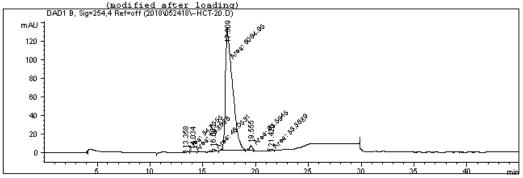
normal
Page 1
of 1

Data File C:\HPCHEM\1\DATA\2018\052418\--HCT-20.D

Sample Name: HCT-20

Injection Date : 5/24/2018 8:41:49 PM Sample Name : HCT-20 Acq. Operator : JC Seq. Line : 4 Location : Vial 3 Inj : 1 Inj Volume : 10 µl

Acc. Method : C:\HPCHEM\1\METHODS\JC-RP1.M
Last changed : 5/23/2018 1:25:21 PM by JC
Analysis Method : C:\HPCHEM\1\METHODS\JV095.M
Last changed : 12/12/2018 12:25:44 PM by Juno



Area Percent Report ______

Signal Multiplier 1.0000 Dilution : 1.0000 Use Multiplier & Dilution Factor with ISTDs

Signal 1: DAD1 B, Sig=254,4 Ref=off

Peak	RetTime	Type	Width	Area	Height	Area
#	[min]		[min]	[mAU*s]	[mAU]	*
1	13.358	MM	0.2960	34.75550	1.95683	0.5498
2	14.034	MM	0.4160	18.26581	7.31878e-1	0.2890
3	16.003	MM	0.3672	46.05307	2.09044	0.7286
4	17.309	MM	0.7684	6094.95312	132.20535	96.4241
5	19.555	MM	0.2949	93.59449	5.28906	1.4807
6	21.426	MM	0.3386	33.36292	1.64216	0.5278

\\Pmsl-analysis2\HP LaserJet P3015 on Ne00: Page 1 **▼** of 1 Close normal 🔻 Letter 8 1/2 x 11 in/Portrait

Data File C:\HPCHEM\1\DATA\2018\070218\HCT25(B1.D Sample Name: HCT25(B1)

______ Injection Date : 7/2/2018 1:46:02 PM Sample Name : HCT25(B1) Acq. Operator : JC Seq. Line : Location : Vial 61 Inj : 1 Inj Volume : 250 µl

Different Ini Volume from Secuence! Actual Inj Volume: 10 µl Acq. Method : C:\HPCHEM\1\METHODS\JC-ISOQS.M
Last changed : 7/2/2018 1:35:45 PM by JC

Analysis Method : C:\HPCHEM1\METHODS\JV095.M Last changed : 12/12/2018 12:25:44 PM by Juno (modified after loading)
DAD1 B, Sig=254,4 Ref=off (2018/070218/HCT25(B1.D) mAU -1000

800 600 400 200 10

Area Percent Report ______

Signal Multiplier 1.0000 1.0000 Dilution Use Multiplier & Dilution Factor with ISTDs

Signal 1: DAD1 B, Sig=254,4 Ref=off

Peak RetTime Type Width Height # [min] [min] [mAU*s] [mAU] [mAU] 1 16.265 MM 0.3742 2.64226e4 1176.90479 95.2581 2 16.969 MM 0.4744 1315.29187 33.62722 4.7419

2.77379e4 1210.53201

Results obtained with enhanced integrator!

\\Pmsl-analysis2\HP LaserJet P3015 on Ne00:

Letter 8 1/2 x 11 in/Portrait





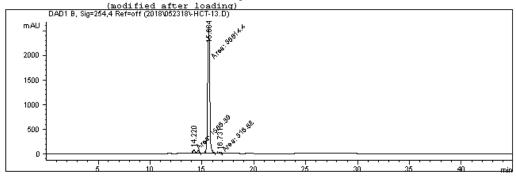
▼ of 1

Data File C:\HPCHEM\1\DATA\2018\052318\-HCT-13.D Sample Name: HCT-13

Seq. Line: 4 Location: Vial 3

Injection Date : 5/23/2018 3:43:29 PM Sample Name : HCT-13 Acq. Operator : JC Inj : Inj Volume : 10 μl

Acg. Method : C:\HPCHEM\1\METHODS\JC-RP1.M Last changed : 5/23/2018 1:25:21 PM by JC Analysis Method : C:\HPCHEM\1\METHODS\JV095.M Last changed : 12/12/2018 12:25:44 PM by Juno



Area Percent Report

Sorted By Signal Multiplier 1.0000 Dilution 1.0000

Use Multiplier & Dilution Factor with ISTDs

Signal 1: DAD1 B, Sig=254,4 Ref=off

#	RetTime [min]		[min]	Area [mAU*s]		Area *
1	14.220	им	0.4198	1566.38757	62.18721	4.0478
_	15.684 16.731			3.68144e4 316.82050		

3.86976e4 2628.15338

Results obtained with enhanced integrator!

\\Pmsl-analysis2\HP LaserJet P3015 on Ne00:

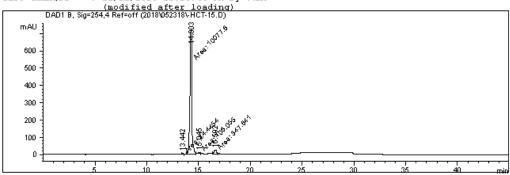
Letter 8 1/2 x 11 in/Portrait

Data File C:\HPCHEM\1\DATA\2018\052318\-HCT-15.D

Sample Name: HCT-15

Injection Date : 5/23/2018 5:14:51 PM Seq. Line : 6
Sample Name : HCT-15 Location : Vial 5
Acq. Operator : JC Inj : 1
Inj Volume : 10 µl

Acc. Method : C:\HPCHEM\1\METHODS\JC-RP1.M
Last changed : 5/23/2018 1:25:21 PM by JC
Analysis Method : C:\HPCHEM\1\METHODS\JV095.M
Last changed : 12/12/2018 12:25:44 PM by Juno



Area Percent Report

med Perceno Reporo

Sorted By : Signal Multiplier : 1.00000 Dilution : 1.00000 Use Multiplier 4 Dilution Factor with ISTDs

Signal 1: DAD1 B, Sig=254,4 Ref=off

				Area	Height	Area
				[mAU*s]		
1	13.442	MM	0.1510	24.44536	2.69893	0.2316
2	14.303	MM	0.2321	1.00778e4	723.61749	95.4686
3	15.045	MM	0.2343	106.05492	7.54276	1.0047
4	16.592	MM	0.2723	347.84100	21.28913	3.2951

Totals: 1.05562e4 755.14832

\\PmsI-analysis2\HP LaserJet P3015 on Ne00:
Letter 8 1/2 x 11 in/Portrait

Close | normal | Page | 1 | of 1

Data File C:\HPCHEM\1\DATA\2018\062718\IQ1'.D

Sample Name: IQ1'

Injection Date : 6/27/2018 5:51:21 PM Seq. Line : 8
Sample Name : IQ1' Location : Vial 51
Acq. Operator : JC Inj : 1
Inj Volume : 10 µl

(modified after loading)

DAD1 8. Sig=254.4 Ref=off (2018/062718\lQ1'.D)

mAU

400

200

100

5

10 15

20 25

30 35

40 min

Area Percent Report

Sorted By : Signal Multiplier : 1.0000 Dilution : 1.0000 Use Multiplier & Dilution Factor with ISTDs

Signal 1: DAD1 B, Sig=254,4 Ref=off

	RetTime			Area	Height	Area
	. — —		. — —	[mAU*s]		*
1	19.761	MM	0.1097	3552.81543	539.77997	95.1028
2	20.163	MM	0.1535	141.98978	10.95285	3.8008
3	21.714	MM	0.1266	40.95975	5.39103	1.0964

Totals: 3735.76495 556.12385

\\PmsI-analysis2\HP LaserJet P3015 on Ne00:
Letter 8 1/2 x 11 in/Portrait

Close | normal | Page | 1 | of 1

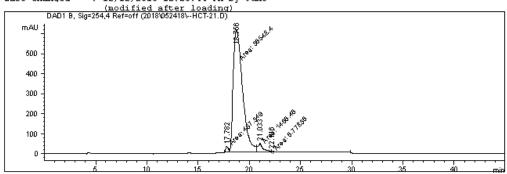
Data File C:\HPCHEM\1\DATA\2018\052418\--HCT-21.D

Sample Name: HCT-21

normal Page 1

Injection Date : 5/24/2018 9:27:30 PM Sample Name : HCT-21 Acq. Operator : JC Seq. Line : 5 Location : Vial 4 Inj : 1 Inj Volume : 10 μl

: C:\HPCHEM\1\METHODS\JC-RP1.M : 5/23/2018 1:25:21 PM by JC Acg. Method Last changed Analysis Method : C:\HPCHEM\1\METHODS\JV095.M Last changed : 12/12/2018 12:25:44 PM by Juno



Area Percent Report

Sorted By Signal Multiplier 1.0000 Dilution

Use Multiplier & Dilution Factor with ISTDs

Signal 1: DAD1 B, Sig=254,4 Ref=off

#	[min]		Width [min]	[mAU*s]		Area *
1	17.782	MF	0.2595	467.34909	30.01972	1.1543
2	18.756	MF	1.0520	3.85424e4	610.60626	95.1970
3	21.033	FM	0.6042	1468.46191	40.50690	3.6270
4	22.116	MM	0.0907	8.77236	1.61284	0.0217

4.04870e4

\\Pmsl-analysis2\HP LaserJet P3015 on Ne00: Close

682.74572

Letter 8 1/2 x 11 in/Portrait

Totals :

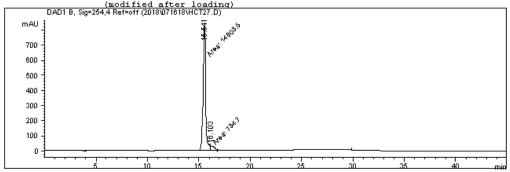
Data File C:\HPCHEM\1\DATA\2018\071618\HCT27.D

______ Seq. Line : Location: Vial 62 Inj: 1

Sample Name: HCT27

Injection Date : 7/16/2018 6:43:05 PM Sample Name : HCT27 Acq. Operator : JC Inj Volume : 10 μl

Acg. Method Last changed : C:\HPCHEM\1\METHODS\JC-RP2.M : 7/16/2018 5:08:35 PM by JC Analysis Method : C:\HPCHEM\l\METHODS\JV095 M Last changed : 12/12/2018 12:25:44 PM by Juno



_____ Area Percent Report

Signal Sorted By Multiplier 1.0000 Dilution 1.0000

Use Multiplier & Dilution Factor with ISTDs

Signal 1: DAD1 B, Sig=254,4 Ref=off

Peak	RetTime	Type	Width	Area	Height	Area
#	[min]		[min]	[mAU*s]	[mAU]	*
1	15.541	MM	0.2966	1.46035e4	820.66400	95.2100
2	16.103	MM	0.3201	734.70013	29.00597	4.7900

1.53382e4

Results obtained with enhanced integrator! ______

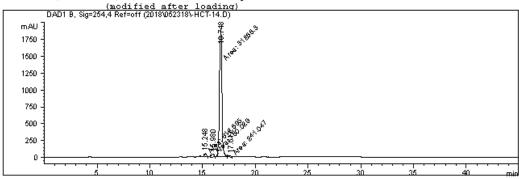
\\Pmsl-analysis2\HP LaserJet P3015 on Ne00: normal Page 1 **▼** of 1 Letter 8 1/2 x 11 in/Portrait

Data File C:\HPCHEM\1\DATA\2018\052318\-HCT-14.D

______ Injection Date : 5/23/2018 4:29:10 PM Sample Name : HCT-14 Acq. Operator : JC Seq. Line : Location : Vial 4

Inj: 1 Inj Volume : 10 µl

: C:\HPCHEM\1\METHODS\JC-RP1.M : 5/23/2018 1:25:21 PM by JC Acg. Method Last changed Analysis Method : C:\HPCHEM\1\METHODS\JV095.M Last changed : 12/12/2018 12:25:44 PM by Juno



Sample Name: HCT-14

______ Area Percent Report

Sorted By Signal Multiplier 1.0000 Dilution 1.0000

Use Multiplier & Dilution Factor with ISTDs

Signal 1: DAD1 B, Sig=254,4 Ref=off

Peak	RetTime	Type	Width	Area	Height	Area
#	[min]		[min]	[mAU*s]	[mAU]	*
1	15.248	MM	0.2206	612.69458	46.28159	1.8633
2	15.980	MM	0.3007	760.02936	42.12210	2.3114
3	16.748	MM	0.2747	3.12683e4	1897.07532	95.0923
4	17.617	MM	0.2869	241.04727	14.00450	0.7331

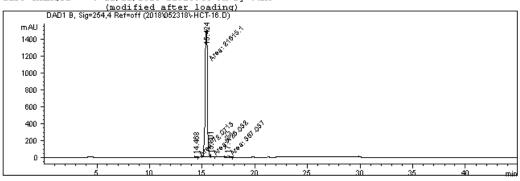
Totals : 3.28821e4 1999.48352

\\Pmsl-analysis2\HP LaserJet P3015 on Ne00: ▼ of 1 Close normal Page 1 Letter 8 1/2 x 11 in/Portrait

Data File C:\HPCHEM\1\DATA\2018\052318\-HCT-16.D

Injection Date : 5/23/2018 6:00:34 PM Sample Name : HCT-16 Acq. Operator : JC Seq. Line: 7 Location: Vial 6 Inj: 1 Inj Volume : 10 µl

Acc. Method : C:\HPCHEM\1\METHODS\JC-RP1.M
Last changed : 5/23/2018 1:25:21 PM by JC
Analysis Method : C:\HPCHEM\1\METHODS\JV095.M
Last changed : 12/12/2018 12:25:44 PM by Juno



Sample Name: HCT-16

______ Area Percent Report

Sorted By Signal Multiplier 1.0000 Dilution 1.0000

Use Multiplier & Dilution Factor with ISTDs

Signal 1: DAD1 B, Sig=254,4 Ref=off

Peak			Width		Height	Area
#				[mAU*s]		*
1	14.468	MM	0.2170	72.07128	5.53416	0.3217
2	15.424	MM	0.2410	2.18151e4	1508.36548	97.3878
3	15.801	MM	0.1281	126.03214	13.05827	0.5626
4	17.433	MM	0.3947	387.03723	16.34200	1.7278

2.24003e4 1543.29991 Totals :

\\Pmsl-analysis2\HP LaserJet P3015 on Ne00: ▼ of 1 Close normal 🔻 Page 1 Letter 8 1/2 x 11 in/Portrait

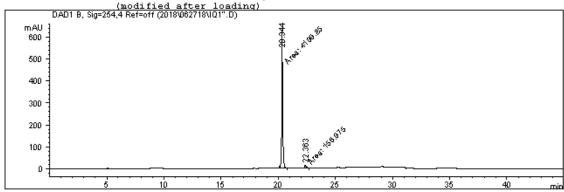
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Sample Name: IQ1''

______ Injection Date : 6/27/2018 6:37:07 PM Sample Name : IQ1'' Acq. Operator : JC Seq. Line : 9 Location : Vial 52 Inj : 1 Inj Volume : 10 μl

: C:\HPCHEM\1\METHODS\JC-RP2.M : 6/27/2018 4:51:44 PM by JC Aca. Method Last changed (modified after loading)
Analysis Method: C:\HPCHEM\1\METHODS\JV095.M

: 12/12/2018 12:25:44 PM by Juno Last changed



______ Area Percent Report

Sorted By Signal 1.0000 Multiplier 1.0000 Dilution

Use Multiplier & Dilution Factor with ISTDs

Signal 1: DAD1 B, Sig=254,4 Ref=off

Peak RetTime Type Width Area Height. Area [mAU*s] [mAU] # [min] [min] * 0.1105 4199.24951 633.14288 96.3965 0.1682 156.97470 15.55715 3.6035 1 20.344 MM 2 22.363 MM

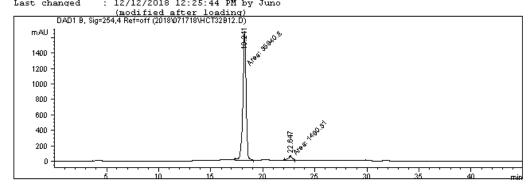
Totals : 4356.22421 648.70003

Results obtained with enhanced integrator!

Data File C:\HPCHEM\1\DATA\2018\071718\HCT32B12.D

______ Injection Date : 7/17/2018 10:33:19 PM Sample Name : HCT32b12 Seq. Line : 2 Location : Vial 61 Acq. Operator : JC Inj: 1 Inj Volume : 10 μl

Acc. Method : C:\HPCHEM\1\METHODS\JC-RP2.M
Last changed : 7/17/2018 9:46:20 PM by JC
Analysis Method : C:\HPCHEM\1\METHODS\JV095.M
Last changed : 12/12/2018 12:25:44 PM by Juno



Sample Name: HCT32b12

normal 🔻

Close

Page 1

▼ of 1

Area Percent Report

Sorted By Signal Multiplier 1.0000 1.0000 Dilution

Use Multiplier & Dilution Factor with ISTDs

Signal 1: DAD1 B, Sig=254,4 Ref=off

Peak RetTime Type Width Height [min] [min] [mAU*s] [mAU] 18.241 MM 0.3684 3.59402e4 1626.01050 96.0185 0.4370 1490.30554 56.83894 3.9815 2 22.647 MM

Totals : 3.74305e4 1682.84944

Results obtained with enhanced integrator!

\\Pmsl-analysis2\HP LaserJet P3015 on Ne00:

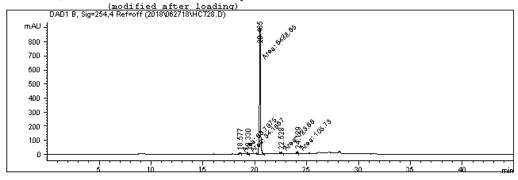
Letter 8 1/2 x 11 in/Portrait

Data File C:\HPCHEM\1\DATA\2018\062718\HCT28.D

Sample Name: HCT28

______ Inj Volume : 10 μl

Acg. Method : C:\HPCHEM\1\METHODS\JC-RP2.M Last changed : 6/27/2018 12:29:49 PM by JC Analysis Method : C:\HPCHEM\1\METHODS\JV095.M Last changed : 12/12/2018 12:25:44 PM by Juno



______ Area Percent Report

Sorted By Signal Multiplier 1.0000 Dilution 1.0000

Use Multiplier & Dilution Factor with ISTDs

Signal 1: DAD1 B, Sig=254,4 Ref=off

Peak #	RetTime [min]	Type	Width [min]	Area [mAU*s]	Height [mAU]	Area %
1						
1	18.577	MM	0.1399	63.79748	7.60011	0.9451
2	19.330	MM	0.0869	34.18670	6.55693	0.5065
3	20.485	MM	0.1189	6422.64941	900.17096	95.1469
4	22.528	MM	0.1418	123.87982	14.56061	1.8352
5	24.129	MM	0.1046	105.72972	16.84713	1.5663

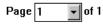
6750.24314 945.73574

\\Pmsl-analysis2\HP LaserJet P3015 on Ne00:

Letter 8 1/2 x 11 in/Portrait

Close





Data File C:\HPCHEM\1\DATA\2018\082818\HCT33.D

Sample Name: HCT33

_____ Injection Date : 8/28/2018 11:11:26 AM Sample Name : HCT33 Acq. Operator : JC Seq. Line : 3 Location : Vial 65 Inj : 1
Inj Volume : 250 µl

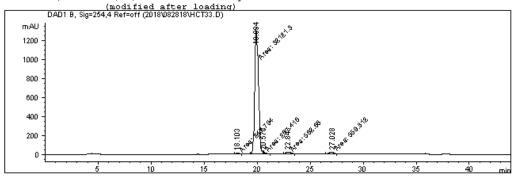
Different Ini Volume from Sequence ! Actual Inj Volume : 20 µl
Acq. Method : C:\HPCHEM\1\METHODS\JC-ISOQH.M

Last changed : 8/24/2018 4:40:23 PM by JC

Analysis Method : C:\HPCHEM\1\METHODS\JV095.M

Last changed : 12/12/2018 12:25:44 PM by Juno

(medical defense localing)



_____ Area Percent Report

Sorted By Signal 1.0000 Multiplier Dilution

Use Multiplier & Dilution Factor with ISTDs

Signal 1: DAD1 B, Sig=254,4 Ref=off

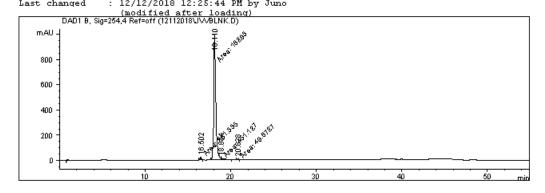
#	[min]			[mAU*s]		Area *
1	18.103	MM	0.3311	247.79393	12.47283	0.7331
2	19.894	MM	0.4048	3.21213e4	1322.54614	95.0371
3	20.578	MM	0.2189	287.41635	21.88802	0.8504
4	22.843	MM	0.5525	582.85968	17.58376	1.7245
5	27.028	MM	0.4419	559.31183	21.09399	1.6548

\\Pmsl-analysis2\HP LaserJet P3015 on Ne00: ▼ of 1 normal - Page 1 Letter 8 1/2 x 11 in/Portrait

Data File C:\HPCHEM\1\DATA\12112018\JVVBLNK.D

Injection Date : 12/11/2018 7:27:46 AM Sample Name : JVV blank Acq. Operator : Juno Seq. Line : Location : Vial 40 Inj : 1 Inj Volume : 250 µl Actual Inj Volume : 30 μ l

Different In: Volume from Sequence ! Actual Acq. Method : C:\HPCHEM\1\METHODS\JV095.M Last changed : 12/11/2018 6:56:31 AM by Juno (modified after loading)
Analysis Method : C:\HPCHEM\l\METHODS\JV095.M
Last changed : 12/12/2018 12:25:44 PM by Juno



Sample Name: JVV blank

Area Percent Report _____

Sorted By Signal Multiplier 1.0000 Dilution 1.0000

Use Multiplier & Dilution Factor with ISTDs

Signal 1: DAD1 B, Sig=254,4 Ref=off

Peak	RetTime	Type	Width	Area	Height	Area
#	[min]		[min]	[mAU*s]	[mAU]	*
1	16.502	MM	0.1977	281.33484	23.71385	1.6476
2	18.110	MM	0.2741	1.62930e4	990.64117	95.4195
3	18.834	MM	0.3645	451.12701	14.48537	2.6420
4	20.828	MM	0.2080	49.67268	3.98066	0.2909

\\Pmsl-analysis2\HP LaserJet P3015 on Ne00: Close normal 🔻 Page 1 ▼ of 1 Letter 8 1/2 x 11 in/Portrait

References

1. Agrawal, et al., J. Med. Chem. 1968, 11(4), 700-703.