

Orbitally Matched Edge-Doping in Graphene Nanoribbons

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1.	Figure S1. STM topographical image of O-doped cGNR polymer intermediates and cGNR products.	S3
2.	Figure S2. Length GNR distribution analysis.	S3
3.	Figure S3. nc-AFM of Sulfur Vacancies in S-doped cGNR.	S4
4.	Figure S4. <i>dI/dV</i> point spectra taken at varying points along O-doped cGNR backbone.	S4
5.	Figure S5. DFT PDOS of 9 <i>H</i> -carbazole, dibenzofuran, and dibenzothiophene on four-layer Au(111).	S5
6.	Figure S6. DFT PDOS of O-, S- and N-doped cGNR in vacuum.	S5
8.	Materials and General Methods	S6
9.	Synthetic Procedures	S6–8
10.	Figure S7. Single crystal X-ray structure diagram of 1a	S9
11.	Table S1–S5. Crystal data and structure refinement for 1a	S9–21
12.	Figure S8. Single crystal X-ray structure diagram of 1b	S22
13.	Table S6–S10. Crystal data and structure refinement for 1b	S22–30
14.	Figure S9. Single crystal X-ray structure diagram of 1c	S30
15.	Table S11–S15. Crystal data and structure refinement for 1c	S30–38
16.	Figure S10. ¹ H NMR (600 MHz, CDCl ₃) spectrum of 4	S39
17.	Figure S11. ¹³ C NMR (151 MHz, CDCl ₃) spectrum of 4	S40
18.	Figure S12. ¹ H NMR (600 MHz, CDCl ₃) spectrum of 3a	S41
19.	Figure S13. ¹³ C NMR (600 MHz, CDCl ₃) spectrum of 3a	S42
20.	Figure S14. ¹ H NMR (500 MHz, CDCl ₃) spectrum of 3b-TMS	S43
21.	Figure S15. ¹³ C NMR (151 MHz, CDCl ₃) spectrum of 3b-TMS	S44
22.	Figure S16. ¹ H NMR (500 MHz, CDCl ₃) spectrum of 3c	S45

23.	Figure S17. ^{13}C NMR (151 MHz, CDCl_3) spectrum of 3c	S46
24.	Figure S18. ^1H NMR (600 MHz, CDCl_3) spectrum of 1a	S47
25.	Figure S19. ^{13}C NMR (151 MHz, CDCl_3) spectrum of 1a	S48
26.	Figure S20. ^1H NMR (500 MHz, CDCl_3) spectrum of 1b	S49
27.	Figure S21. ^{13}C NMR (151 MHz, CDCl_3) spectrum of 1b	S50
28.	Figure S22. ^1H NMR (600 MHz, CDCl_3) spectrum of 1c	S51
29.	Figure S23. ^{13}C NMR (151 MHz, CDCl_3) spectrum of 1c	S52
30.	References	S53

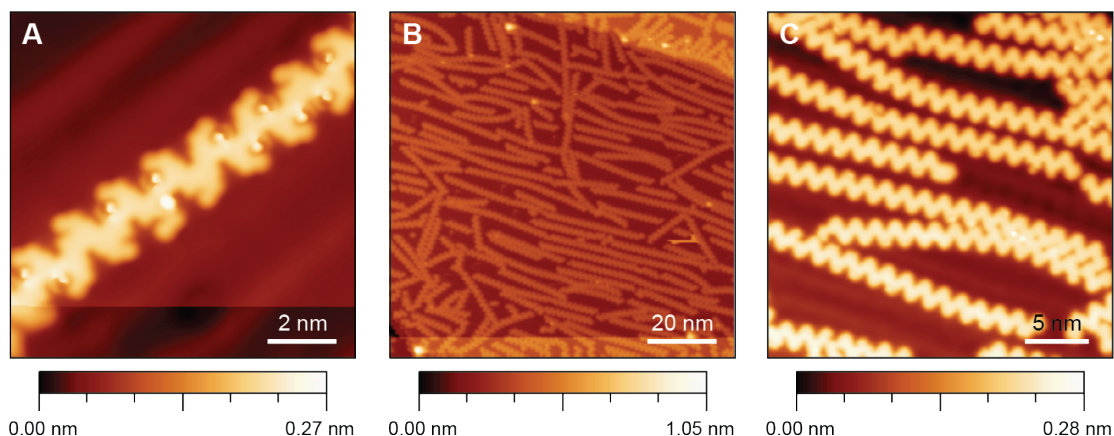


Figure S1. A) STM topographical image of O-doped cGNR polymer intermediate. Uncyclized wings protruding from polymer backbone appear as brighter protrusions. This polymer intermediate is representative of the intermediate structures also observed for the S-doped and N-doped precursor polymers; other examples of precursor polymer intermediates from this series of GNRs can be found in the literature.⁴ B) Representative large area STM topographical image of O-doped cGNR and C) magnified image of O-doped cGNR illustrating the virtually quantitative yield of fully cyclodehydrogenated GNRs. ($V_{\text{bias}} = 50\text{mV}$, $I_t = 20\text{ pA}$)

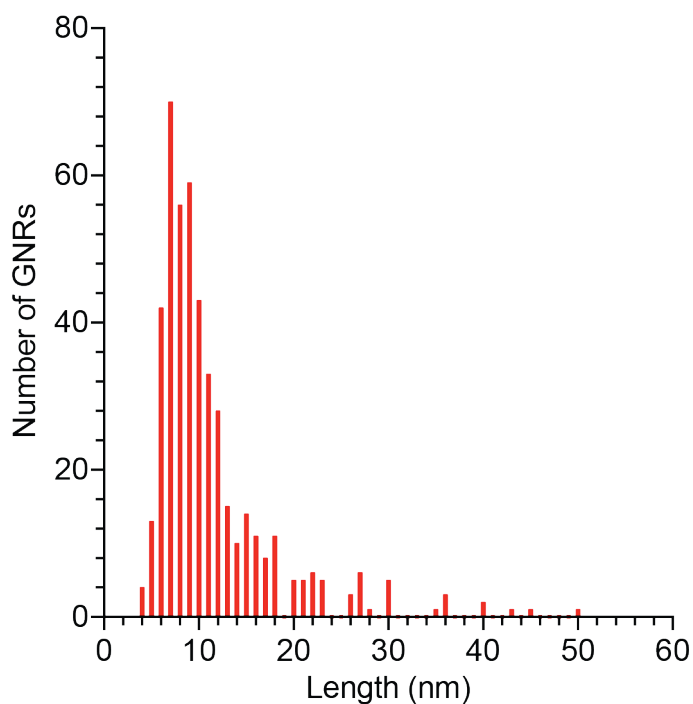


Figure S2. Statistical analysis on GNR length distribution. Data set combined from samples of O-, S-, and N-doped cGNRs; distribution is representative of all samples.

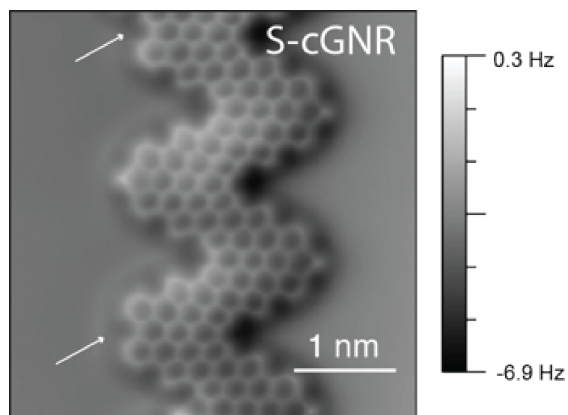


Figure S3. nc-AFM frequency shift image of S-cGNR; areas that have lost S-dopant atom indicated with white arrow.

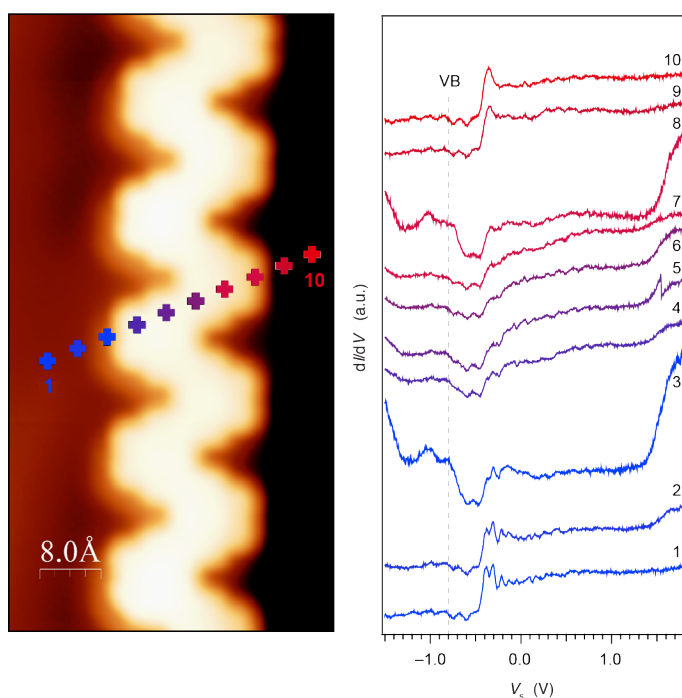


Figure S4. STM topographic image (left) and corresponding dI/dV point spectra (right) of O-doped cGNR recorded at various positions above the GNR. Spectra recorded at positions marked on the GNR (left), with the first spectrum shown in blue, moving across the ribbon to the tenth spectrum shown in red. Spectra are offset vertically (set point: $V_{\text{bias}} = 0.05$ V, $I_t = 20$ pA).

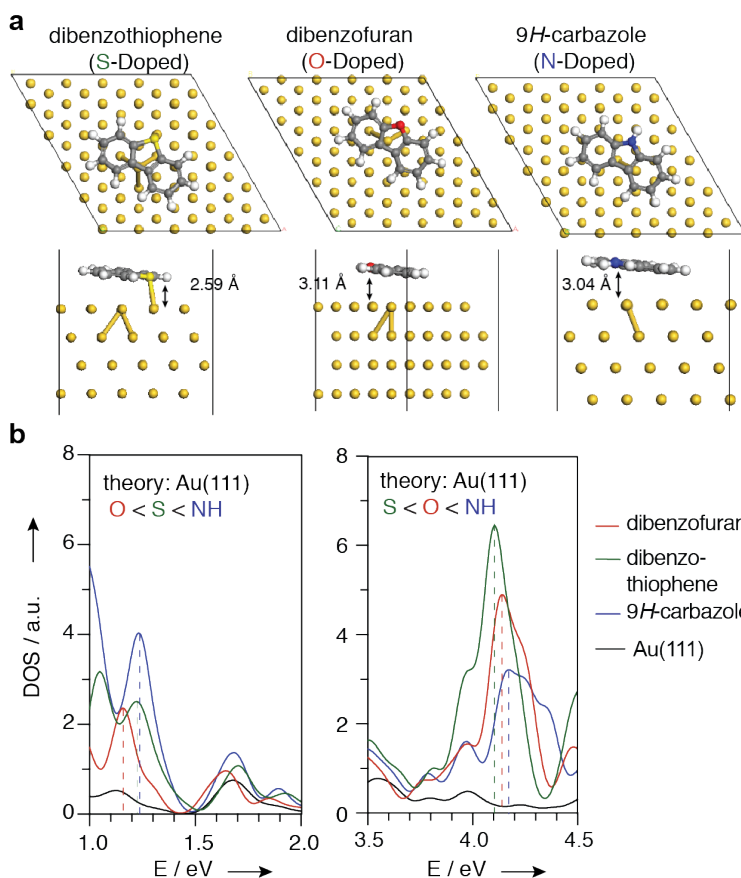


Figure S5. a) Schematic of structure of dibenzothiophene (left), dibenzofuran (middle), and 9H-carbazole (right) molecules on four-layer Au(111) substrate (upper panels; Color coding: C (gray), N (blue), O (red), S (green), H (white)); calculated heteroatom dopant height above the Au(111) surface (lower panels). b) Total density of states (DOS) from DFT-LDA calculations of dibenzofuran (red), dibenzothiophene (green), 9H-carbazole (red), on Au(111) surface (with the DOS of Au(111) surface only shown in black).

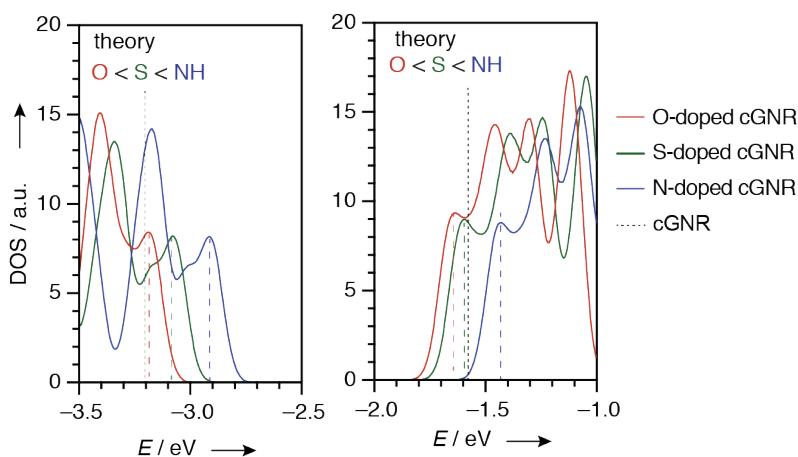


Figure S6. Density of states from DFT-LDA calculations of free-standing O-doped cGNR (red), S-doped cGNR (green), and N-doped cGNR (blue); dashed lines indicate valence and conduction band positions. Origin of energy scale is set at the vacuum level.

Materials and General Methods. Unless otherwise stated, all manipulations of air- and/or moisture-sensitive compounds were carried out in oven-dried glassware under and atmosphere of N₂. All solvents and reagents were purchased from Alfa Aesar, Spectrum Chemicals, Acros Organics, TCI America, Matrix Scientific, and Sigma-Aldrich and were used as received unless otherwise noted. Organic solvents were dried by passing through a column of alumina. Flash column chromatography was performed on SiliCycle silica gel (particle size 40–63 μm). Thin layer chromatography was performed using SiliCycle silica gel 60 Å F-254 precoated plates (0.25 mm thick) and visualized by UV absorption. All ¹H and {¹H} ¹³C NMR spectra were recorded on Bruker AV-600, AV-500, and AVQ-400 spectrometers and are referenced to residual solvent peaks (CDCl₃ ¹H NMR δ = 7.26 ppm, ¹³C NMR δ = 77.16 ppm). High resolution mass spectrometry (EI) was performed on an Autospec Permier (Waters) sector spectrometer in positive ionization mode. ESI mass spectrometry was performed on a Finnigan LTQFT (Thermo) spectrometer. X-ray crystallography was performed on a MicroSTAR-H APEX II, using a microfocus rotating anode (Cu-K_α radiation), Kappa Geometry with DX (Enraf-Nonius build) goniostat, a Bruker APEX II detector, Helios multilayer mirrors as the radiation monochromator, and Oxford Cryostream 700 held at 100 K (**1a**) or on an APEX II QUAZAR, using a Microfocus Sealed Source (Incoatec; Cu-K_α radiation), Kappa Geometry with DX (Bruker-AXS build) goniostat, a Bruker APEX II detector, QUAZAR multilayer mirrors as the radiation monochromator, and Oxford Cryostream 700 held at 100 K (**1b,c**). Crystallographic data were resolved with SHELXT, refined with SHELXL-2014, and visualized with ORTEP-32. 1-iododibenzofuran^{1,2}, 4-bromo-9*H*-carbazole^{1,3}, and 1-bromodibenzothiophene¹ were synthesized following literature procedures.

Synthetic Procedures

tert-butyl 4-bromo-9*H*-carbazole-9-carboxylate (**4**): A 150 mL Schlenk flask was charged with 4-bromo-9*H*-carbazole (3.15 g, 12.8 mmol), di-*tert*-butyl dicarbonate (5.59 g, 25.6 mmol) and 4-dimethylaminopyridine (1.56 g, 12.8 mmol) in acetonitrile (51 mL), and the reaction mixture was stirred at 25 °C for 2 h. The reaction mixture was extracted with CH₂Cl₂ (100 mL). The combined organic layers were dried over MgSO₄ and concentrated on a rotary evaporator. Column chromatography (hexane/EtOAc 1:0–50:1) yielded **4** (3.49 g, 77%) as a colorless solid. ¹H NMR (600 MHz, 25 °C, CDCl₃) δ = 8.82 (ddd, *J* = 7.9, 1.4, 0.7 Hz, 1H), 8.38–8.32 (m, 2H), 7.55–7.51 (m, 2H), 7.41 (ddd, *J* = 8.1, 7.3, 1.1 Hz, 1H), 7.30 (t, *J* = 8.1 Hz, 1H), 1.77 (s, 10H) ppm; ¹³C NMR (151 MHz, 25 °C, CDCl₃) δ = 150.7, 139.9, 138.7, 127.6, 127.5, 127.3, 125.1, 124.2, 122.8, 122.4, 116.0, 115.8, 115.1, 84.5, 28.3 ppm; HRMS (EI) *m/z*: [C₁₇H₁₆BrNO]⁺, calcd. For [C₁₇H₁₆BrNO] 345.0364; found 345.0368.

tert-butyl 4-ethynyl-9*H*-carbazole-9-carboxylate (**3a**): A 100 mL sealable Schlenk flask was charged with **4** (1.00 g, 2.9 mmol) in toluene (20 mL) and NEt₃ (20 mL). The reaction mixture was degassed and Pd(PPh₃)₄ (0.67 g, 0.58 mmol) and CuI (55 mg, 0.29 mmol) were added. TMSA (4 mL) was added, and the flask was sealed and stirred at 90 °C for 18 h. The reaction mixture was cooled to 25 °C, and extracted with Et₂O (100 mL). The combined organic layers were dried over MgSO₄, and concentrated on a rotary evaporator. Column chromatography (hexane/CH₂Cl₂ 1:0–20:1) yielded an inseparable mixture of partially deprotected **3a**. A 250 mL round bottom flask was charged with the crude intermediate in MeOH (77 mL) and THF (77 mL). K₂CO₃ (3 g) was added and the reaction mixture was stirred for 2 h at 25 °C. The reaction was extracted with Et₂O (100 mL), the combined organic layers were dried over MgSO₄ and concentrated on a rotary evaporator. Column chromatography (hexane/C₆H₆ 1:0–10:1) yielded **3a** (28 mg, 20%) as a colorless oil. ¹H NMR (600 MHz, 25 °C, CDCl₃) δ = 8.80–8.75 (m, 1H), 8.39 (d, *J* = 8.4 Hz, 1H), 8.33 (d, *J* = 8.4 Hz, 1H), 7.55–7.49 (m, 2H), 7.45–7.35 (m, 2H), 3.58 (s, 1H), 1.77 (s, 10H) ppm; ¹³C NMR (151 MHz, 25 °C, CDCl₃) δ = 150.7, 138.5, 138.4, 128.1, 127.4, 126.1, 125.6, 125.0, 122.8, 121.7, 116.7, 115.6, 114.6, 84.0, 82.3, 82.0, 28.2 ppm; HRMS (EI) *m/z*: [C₁₉H₁₇NO₂]⁺, calcd. For C₁₉H₁₇NO₂ 291.1259; found 291.1255.

(dibenzofuran-1-ylethynyl)trimethylsilane (**3b-TMS**): A 5 mL sealable Schlenk flask was charged with 1-iododibenzofuran (13 mg, 0.04 mmol) in THF (1 mL) and NEt₃ (1 mL). The reaction mixture degassed, and Pd(PPh₃)₄ (7.6 mg, 0.01 mmol) and CuI (0.5 mg, 0.01 mmol) were added. TMSA (0.1 mL) was added, and the flask was sealed and stirred at 80 °C for 16 h. The reaction mixture was cooled to 25 °C, and concentrated on a rotary evaporator. The reaction mixture was extracted with CH₂Cl₂ (50 mL), and the combined organic layers were washed with H₂O, dried over MgSO₄, and concentrated on a rotary evaporator. Column chromatography (hexane) yielded **3b-TMS** (8.4 mg, 79%) as a colorless oil. ¹H NMR (600 MHz, 25 °C, CDCl₃) δ = 8.43 (d, *J* = 7.7 Hz, 1H), 7.56 (dd, *J* = 20.8, 8.2, 2H), 7.52–7.47 (m, 1H), 7.45 (d, *J* = 7.3 Hz, 1H), 7.38 (t, *J* = 7.9 Hz, 2H), 0.38 (s, 9H) ppm; ¹³C NMR (151 MHz, 25 °C, CDCl₃) δ = 156.3,

155.7, 127.5, 126.9, 126.6, 124.8, 123.8, 122.7, 122.2, 166.7, 112.0, 111.3, 102.8, 99.5, -0.06 ppm; HRMS (ESI-TOF) m/z: $[\text{C}_{17}\text{H}_{16}\text{OSi}]^+$, calcd. for $[\text{C}_{17}\text{H}_{16}\text{OSi}]$ 264.0975; found 264.0970.

1-ethynyldibenzothiophene (**3c**): A 10 mL sealable Schlenk flask was charged with 1-bromodibenzothiophene (70.0 mg, 0.27 mmol) in toluene (2.6 mL) and NEt_3 (2.6 mL). The reaction mixture was degassed, and $\text{Pd}(\text{PPh}_3)_4$ (62.4 mg, 0.05 mmol) and CuI (5.1 mg, 0.03 mmol) were added. TMSA (1 mL) was added, and the flask was sealed and stirred at 90 °C for 18 h. The reaction mixture was cooled to 25 °C, and extracted with Et_2O (100 mL). The combined organic layers were dried over MgSO_4 , and the reaction mixture was concentrated on a rotary evaporator. Column chromatography (hexane) yielded an inseparable mixture of partially deprotected **3c**. A 25 mL round bottom flask was charged with the crude intermediate in THF (8 mL) and MeOH (8 mL). K_2CO_3 (1 g) was added, and the reaction mixture was stirred at 25 °C for 2 h. The reaction mixture was extracted with CH_2Cl_2 , and the combined organic layers were washed with H_2O , dried over MgSO_4 , and concentrated on a rotatory evaporator. Column chromatography (hexane) yielded **3c** (41.8 mg, 75%) as a colorless oil. ^1H NMR (600 MHz, 25 °C, CDCl_3) δ = 9.21–9.15 (m, 1 H), 7.90–7.84 (m, 2H), 7.65 (dd, J = 7.4, 0.8 Hz, 1H), 7.50 (m, 2H), 7.41 (t, J = 7.7 Hz, 1H), 3.65 (s, 1H) ppm; ^{13}C NMR (151 MHz, 25 °C, CDCl_3) δ = 140.2, 139.82, 135.4, 134.7, 131.1, 127.2, 125.8, 124.9, 124.4, 123.7, 122.6, 117.3, 83.4, 82.9 ppm; HRMS (EI) m/z: $[\text{C}_{14}\text{H}_8\text{S}]^+$, calcd. for $[\text{C}_{14}\text{H}_8\text{S}]$ 208.0349; found 208.0347.

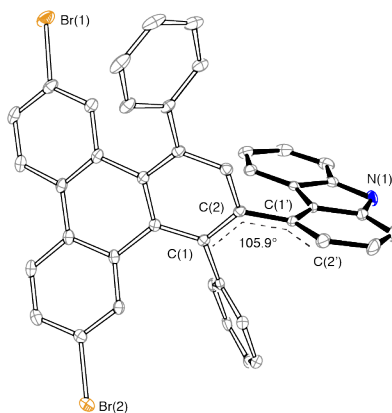


Figure S7. ORTEP representation of the molecular structures of the **1a**. Thermal ellipsoids are drawn at the 50% probability level. Color coding: C (gray), Br (orange), N (blue). Hydrogen atoms and solvent molecules are omitted for clarity.

Table S1. Crystal data and structure refinement for **1a**.

CCDC no.	1567587	
Empirical formula	C ₄₃ H ₂₆ Br ₂ Cl ₃ N	
Formula weight	822.82	
Temperature	100(2) K	
Wavelength	0.71073 Å	
Crystal system	Triclinic	
Space group	P-1	
Unit cell dimensions	<i>a</i> = 9.1436(6) Å	α = 93.985(2)°
	<i>b</i> = 13.3703(8) Å	β = 92.921(3)°
	<i>c</i> = 14.1905(8) Å	γ = 94.285(3)°
Volume	1723.04(18) Å ³	
Z	2	
Density (calculated)	1.586 Mg/ m ³	
Absorption coefficient	2.619 mm ⁻¹	
F(000)	824	
Crystal size	0.050 x 0.050 x 0.030 mm ³	
Theta range for data collection	1.441° to 25.422°	
Index ranges	-11 ≤ <i>h</i> ≤ 11, -16 ≤ <i>k</i> ≤ 16, -17 ≤ <i>l</i> ≤ 17	
Reflections collected	46988	
Independent reflections	6333 [R(int) = 0.0345]	
Completeness to theta = 25.000°	100.0 %	
Absorption correction	Semi-empirical from equivalents	
Max. and min. transmission	0.928 and 0.850	
Refinement method	Full-matrix least-squares on F ²	
Data / restraints / parameters	6333 / 0 / 468	
Goodness-of-fit on F ²	1.038	
Final R indices [I > 2σ(I)]	R1 = 0.0272, wR2 = 0.0658	
R indices (all data)	R1 = 0.0329, wR2 = 0.0683	
Extinction coefficient	n/a	
Largest diff. peak and hole	0.598 and -0.490 eÅ ⁻³	

Table S2. Atomic coordinates ($\times 10^4$) and equivalent isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for **1a**. U(eq) is defined as one third of the trace of the orthogonalized U^{ij} tensor.

	x	y	z	U(eq)
C(1)	6041(2)	7114(2)	2019(1)	16(1)
C(2)	5589(2)	6125(2)	1672(1)	16(1)
C(3)	4255(2)	5480(2)	1728(1)	17(1)
C(4)	2916(2)	5585(2)	2146(2)	20(1)
C(5)	1839(3)	4790(2)	2043(2)	24(1)
C(6)	2091(3)	3888(2)	1536(2)	27(1)
C(7)	3409(3)	3755(2)	1136(2)	26(1)
C(8)	4485(3)	4551(2)	1238(1)	21(1)
C(9)	6574(2)	5552(2)	1168(1)	21(1)
C(10)	7989(3)	5938(2)	1015(2)	28(1)
C(11)	8409(2)	6908(2)	1368(2)	28(1)
C(12)	7456(2)	7495(2)	1866(2)	22(1)
C(13)	5023(2)	7696(1)	2591(1)	14(1)
C(14)	3838(2)	8141(1)	2183(1)	14(1)
C(15)	2794(2)	8556(1)	2775(1)	14(1)
C(16)	1678(2)	9214(2)	2437(1)	15(1)
C(17)	1894(2)	9757(2)	1633(1)	16(1)
C(18)	907(2)	10419(2)	1363(1)	17(1)
C(19)	-345(2)	10567(2)	1852(2)	22(1)
C(20)	-539(2)	10070(2)	2657(2)	21(1)
C(21)	468(2)	9406(2)	2984(2)	17(1)
C(22)	314(2)	8944(2)	3877(2)	17(1)
C(23)	-986(2)	8960(2)	4367(2)	21(1)
C(24)	-1143(2)	8512(2)	5200(2)	24(1)
C(25)	19(2)	8004(2)	5548(2)	21(1)
C(26)	1318(2)	7978(2)	5109(2)	19(1)
C(27)	1511(2)	8471(2)	4272(1)	16(1)
C(28)	2858(2)	8386(1)	3752(1)	15(1)
C(29)	4181(2)	8073(2)	4167(1)	16(1)
C(30)	5230(2)	7741(2)	3576(1)	16(1)
C(31)	3663(2)	8080(2)	1123(1)	15(1)

C(32)	4694(2)	8565(2)	584(2)	18(1)
C(33)	4508(2)	8505(2)	-395(2)	21(1)
C(34)	3310(2)	7944(2)	-855(2)	23(1)
C(35)	2283(2)	7450(2)	-327(2)	21(1)
C(36)	2461(2)	7516(2)	654(2)	17(1)
C(37)	4588(2)	8140(2)	5204(1)	18(1)
C(38)	4471(2)	9029(2)	5764(2)	22(1)
C(39)	4993(3)	9118(2)	6706(2)	30(1)
C(40)	5627(3)	8317(2)	7097(2)	32(1)
C(41)	5719(3)	7429(2)	6559(2)	27(1)
C(42)	5214(2)	7342(2)	5615(2)	21(1)
C(43)	8278(5)	4930(3)	4091(3)	29(1)
C(43A)	7747(9)	4876(5)	4278(5)	29(1)
N(1)	5888(2)	4610(1)	914(1)	24(1)
Cl(1)	8876(3)	6144(2)	3813(2)	46(1)
Cl(2)	7064(2)	4358(1)	3179(1)	51(1)
Cl(3)	7440(6)	4957(3)	5207(3)	62(1)
Cl(1A)	9030(5)	5897(3)	4100(3)	47(1)
Cl(2A)	6102(3)	5007(2)	3615(2)	54(1)
Cl(3A)	7438(9)	4799(5)	5424(5)	45(1)
Br(1)	1302(1)	11194(1)	316(1)	21(1)
Br(2)	-246(1)	7252(1)	6626(1)	31(1)

Table S3. Bond lengths [Å] and angles [°] for **1a**.

C(1)-C(12)	1.387(3)
C(1)-C(2)	1.403(3)
C(1)-C(13)	1.493(3)
C(2)-C(9)	1.415(3)
C(2)-C(3)	1.449(3)
C(3)-C(4)	1.399(3)
C(3)-C(8)	1.416(3)
C(4)-C(5)	1.388(3)
C(4)-H(4)	0.9500
C(5)-C(6)	1.401(3)
C(5)-H(5)	0.9500
C(6)-C(7)	1.376(4)
C(6)-H(6)	0.9500
C(7)-C(8)	1.388(3)
C(7)-H(7)	0.9500
C(8)-N(1)	1.383(3)
C(9)-N(1)	1.379(3)
C(9)-C(10)	1.390(3)
C(10)-C(11)	1.377(4)
C(10)-H(10)	0.9500
C(11)-C(12)	1.401(3)
C(11)-H(11)	0.9500
C(12)-H(12)	0.9500
C(13)-C(14)	1.394(3)
C(13)-C(30)	1.397(3)
C(14)-C(15)	1.420(3)
C(14)-C(31)	1.501(3)
C(15)-C(28)	1.420(3)
C(15)-C(16)	1.480(3)
C(16)-C(17)	1.408(3)
C(16)-C(21)	1.412(3)
C(17)-C(18)	1.371(3)
C(17)-H(17)	0.9500
C(18)-C(19)	1.388(3)

C(18)-Br(1)	1.904(2)
C(19)-C(20)	1.372(3)
C(19)-H(19)	0.9500
C(20)-C(21)	1.410(3)
C(20)-H(20)	0.9500
C(21)-C(22)	1.457(3)
C(22)-C(23)	1.408(3)
C(22)-C(27)	1.414(3)
C(23)-C(24)	1.372(3)
C(23)-H(23)	0.9500
C(24)-C(25)	1.391(3)
C(24)-H(24)	0.9500
C(25)-C(26)	1.371(3)
C(25)-Br(2)	1.904(2)
C(26)-C(27)	1.411(3)
C(26)-H(26)	0.9500
C(27)-C(28)	1.475(3)
C(28)-C(29)	1.422(3)
C(29)-C(30)	1.384(3)
C(29)-C(37)	1.495(3)
C(30)-H(30)	0.9500
C(31)-C(32)	1.394(3)
C(31)-C(36)	1.397(3)
C(32)-C(33)	1.388(3)
C(32)-H(32)	0.9500
C(33)-C(34)	1.387(3)
C(33)-H(33)	0.9500
C(34)-C(35)	1.389(3)
C(34)-H(34)	0.9500
C(35)-C(36)	1.389(3)
C(35)-H(35)	0.9500
C(36)-H(36)	0.9500
C(37)-C(42)	1.396(3)
C(37)-C(38)	1.396(3)
C(38)-C(39)	1.391(3)
C(38)-H(38)	0.9500

C(39)-C(40)	1.390(4)
C(39)-H(39)	0.9500
C(40)-C(41)	1.378(4)
C(40)-H(40)	0.9500
C(41)-C(42)	1.389(3)
C(41)-H(41)	0.9500
C(42)-H(42)	0.9500
C(43)-Cl(1)	1.751(4)
C(43)-Cl(2)	1.756(4)
C(43)-Cl(3)	1.793(6)
C(43)-H(43)	1.0000
C(43A)-Cl(3A)	1.673(10)
C(43A)-Cl(2A)	1.760(7)
C(43A)-Cl(1A)	1.773(8)
C(43A)-H(43A)	1.0000
N(1)-H(1)	0.8800
C(12)-C(1)-C(2)	118.58(19)
C(12)-C(1)-C(13)	122.27(19)
C(2)-C(1)-C(13)	119.04(18)
C(1)-C(2)-C(9)	119.4(2)
C(1)-C(2)-C(3)	133.78(19)
C(9)-C(2)-C(3)	106.79(18)
C(4)-C(3)-C(8)	118.7(2)
C(4)-C(3)-C(2)	134.83(19)
C(8)-C(3)-C(2)	106.45(18)
C(5)-C(4)-C(3)	119.2(2)
C(5)-C(4)-H(4)	120.4
C(3)-C(4)-H(4)	120.4
C(4)-C(5)-C(6)	120.6(2)
C(4)-C(5)-H(5)	119.7
C(6)-C(5)-H(5)	119.7
C(7)-C(6)-C(5)	121.5(2)
C(7)-C(6)-H(6)	119.3
C(5)-C(6)-H(6)	119.3
C(6)-C(7)-C(8)	117.9(2)
C(6)-C(7)-H(7)	121.0

C(8)-C(7)-H(7)	121.0
N(1)-C(8)-C(7)	129.4(2)
N(1)-C(8)-C(3)	108.52(19)
C(7)-C(8)-C(3)	122.1(2)
N(1)-C(9)-C(10)	129.8(2)
N(1)-C(9)-C(2)	108.47(19)
C(10)-C(9)-C(2)	121.7(2)
C(11)-C(10)-C(9)	117.8(2)
C(11)-C(10)-H(10)	121.1
C(9)-C(10)-H(10)	121.1
C(10)-C(11)-C(12)	121.8(2)
C(10)-C(11)-H(11)	119.1
C(12)-C(11)-H(11)	119.1
C(1)-C(12)-C(11)	120.8(2)
C(1)-C(12)-H(12)	119.6
C(11)-C(12)-H(12)	119.6
C(14)-C(13)-C(30)	119.35(19)
C(14)-C(13)-C(1)	122.65(18)
C(30)-C(13)-C(1)	117.90(18)
C(13)-C(14)-C(15)	119.36(18)
C(13)-C(14)-C(31)	117.51(17)
C(15)-C(14)-C(31)	122.92(18)
C(28)-C(15)-C(14)	119.38(18)
C(28)-C(15)-C(16)	117.31(18)
C(14)-C(15)-C(16)	123.21(18)
C(17)-C(16)-C(21)	118.66(19)
C(17)-C(16)-C(15)	120.73(18)
C(21)-C(16)-C(15)	120.15(19)
C(18)-C(17)-C(16)	120.42(19)
C(18)-C(17)-H(17)	119.8
C(16)-C(17)-H(17)	119.8
C(17)-C(18)-C(19)	121.8(2)
C(17)-C(18)-Br(1)	118.97(16)
C(19)-C(18)-Br(1)	119.16(16)
C(20)-C(19)-C(18)	118.2(2)
C(20)-C(19)-H(19)	120.9

C(18)-C(19)-H(19)	120.9
C(19)-C(20)-C(21)	122.2(2)
C(19)-C(20)-H(20)	118.9
C(21)-C(20)-H(20)	118.9
C(20)-C(21)-C(16)	118.4(2)
C(20)-C(21)-C(22)	121.80(19)
C(16)-C(21)-C(22)	119.75(19)
C(23)-C(22)-C(27)	118.80(19)
C(23)-C(22)-C(21)	122.07(19)
C(27)-C(22)-C(21)	119.12(18)
C(24)-C(23)-C(22)	122.2(2)
C(24)-C(23)-H(23)	118.9
C(22)-C(23)-H(23)	118.9
C(23)-C(24)-C(25)	117.9(2)
C(23)-C(24)-H(24)	121.0
C(25)-C(24)-H(24)	121.0
C(26)-C(25)-C(24)	122.2(2)
C(26)-C(25)-Br(2)	118.79(17)
C(24)-C(25)-Br(2)	118.85(16)
C(25)-C(26)-C(27)	120.2(2)
C(25)-C(26)-H(26)	119.9
C(27)-C(26)-H(26)	119.9
C(26)-C(27)-C(22)	118.47(19)
C(26)-C(27)-C(28)	120.75(19)
C(22)-C(27)-C(28)	120.32(18)
C(15)-C(28)-C(29)	118.59(18)
C(15)-C(28)-C(27)	118.14(18)
C(29)-C(28)-C(27)	123.10(18)
C(30)-C(29)-C(28)	118.54(18)
C(30)-C(29)-C(37)	116.33(18)
C(28)-C(29)-C(37)	124.99(18)
C(29)-C(30)-C(13)	122.13(19)
C(29)-C(30)-H(30)	118.9
C(13)-C(30)-H(30)	118.9
C(32)-C(31)-C(36)	118.55(19)
C(32)-C(31)-C(14)	121.37(18)

C(36)-C(31)-C(14)	120.07(18)
C(33)-C(32)-C(31)	120.5(2)
C(33)-C(32)-H(32)	119.7
C(31)-C(32)-H(32)	119.7
C(34)-C(33)-C(32)	120.5(2)
C(34)-C(33)-H(33)	119.8
C(32)-C(33)-H(33)	119.8
C(33)-C(34)-C(35)	119.5(2)
C(33)-C(34)-H(34)	120.2
C(35)-C(34)-H(34)	120.2
C(34)-C(35)-C(36)	120.1(2)
C(34)-C(35)-H(35)	120.0
C(36)-C(35)-H(35)	120.0
C(35)-C(36)-C(31)	120.8(2)
C(35)-C(36)-H(36)	119.6
C(31)-C(36)-H(36)	119.6
C(42)-C(37)-C(38)	118.7(2)
C(42)-C(37)-C(29)	120.25(19)
C(38)-C(37)-C(29)	120.80(19)
C(39)-C(38)-C(37)	120.3(2)
C(39)-C(38)-H(38)	119.8
C(37)-C(38)-H(38)	119.8
C(40)-C(39)-C(38)	120.0(2)
C(40)-C(39)-H(39)	120.0
C(38)-C(39)-H(39)	120.0
C(41)-C(40)-C(39)	120.2(2)
C(41)-C(40)-H(40)	119.9
C(39)-C(40)-H(40)	119.9
C(40)-C(41)-C(42)	119.9(2)
C(40)-C(41)-H(41)	120.0
C(42)-C(41)-H(41)	120.0
C(41)-C(42)-C(37)	120.8(2)
C(41)-C(42)-H(42)	119.6
C(37)-C(42)-H(42)	119.6
Cl(1)-C(43)-Cl(2)	109.6(2)
Cl(1)-C(43)-Cl(3)	111.1(3)

Cl(2)-C(43)-Cl(3)	111.2(3)
Cl(1)-C(43)-H(43)	108.3
Cl(2)-C(43)-H(43)	108.3
Cl(3)-C(43)-H(43)	108.3
Cl(3A)-C(43A)-Cl(2A)	110.8(5)
Cl(3A)-C(43A)-Cl(1A)	111.8(5)
Cl(2A)-C(43A)-Cl(1A)	109.1(4)
Cl(3A)-C(43A)-H(43A)	108.3
Cl(2A)-C(43A)-H(43A)	108.3
Cl(1A)-C(43A)-H(43A)	108.3
C(9)-N(1)-C(8)	109.75(17)
C(9)-N(1)-H(1)	125.1
C(8)-N(1)-H(1)	125.1

Table S4. Anisotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for **1a**. The anisotropic displacement factor exponent takes the form: $-2p^2[h^2a^{*2}U^{11} + \dots + 2hk a^* b^* U^{12}]$

	U ¹¹	U ²²	U ³³	U ²³	U ¹³	U ¹²
C(1)	15(1)	23(1)	11(1)	5(1)	2(1)	6(1)
C(2)	18(1)	22(1)	10(1)	5(1)	3(1)	10(1)
C(3)	24(1)	16(1)	10(1)	3(1)	1(1)	7(1)
C(4)	26(1)	18(1)	15(1)	2(1)	3(1)	4(1)
C(5)	28(1)	24(1)	20(1)	5(1)	4(1)	-1(1)
C(6)	40(1)	21(1)	19(1)	4(1)	-3(1)	-6(1)
C(7)	44(2)	17(1)	16(1)	0(1)	-3(1)	6(1)
C(8)	31(1)	21(1)	13(1)	4(1)	1(1)	10(1)
C(9)	25(1)	29(1)	12(1)	3(1)	2(1)	13(1)
C(10)	24(1)	44(2)	20(1)	3(1)	8(1)	20(1)
C(11)	14(1)	48(2)	25(1)	7(1)	7(1)	6(1)
C(12)	18(1)	30(1)	20(1)	5(1)	1(1)	3(1)
C(13)	15(1)	13(1)	16(1)	3(1)	3(1)	-2(1)
C(14)	14(1)	12(1)	15(1)	1(1)	3(1)	-3(1)
C(15)	14(1)	10(1)	17(1)	-1(1)	2(1)	-2(1)
C(16)	15(1)	12(1)	18(1)	-5(1)	0(1)	1(1)
C(17)	15(1)	15(1)	17(1)	-3(1)	1(1)	1(1)
C(18)	20(1)	13(1)	17(1)	-2(1)	-2(1)	0(1)
C(19)	20(1)	21(1)	25(1)	-2(1)	-3(1)	7(1)
C(20)	16(1)	21(1)	26(1)	-4(1)	4(1)	5(1)
C(21)	15(1)	15(1)	20(1)	-5(1)	1(1)	1(1)
C(22)	16(1)	14(1)	19(1)	-4(1)	4(1)	-1(1)
C(23)	18(1)	20(1)	26(1)	-4(1)	5(1)	3(1)
C(24)	20(1)	24(1)	26(1)	-5(1)	12(1)	-2(1)
C(25)	27(1)	18(1)	19(1)	-2(1)	9(1)	-6(1)
C(26)	20(1)	16(1)	19(1)	-1(1)	5(1)	-2(1)
C(27)	17(1)	13(1)	17(1)	-5(1)	4(1)	-3(1)
C(28)	18(1)	10(1)	17(1)	-1(1)	4(1)	-1(1)
C(29)	17(1)	14(1)	15(1)	1(1)	2(1)	-2(1)
C(30)	15(1)	15(1)	18(1)	3(1)	1(1)	1(1)
C(31)	15(1)	14(1)	16(1)	1(1)	2(1)	7(1)

C(32)	17(1)	18(1)	19(1)	2(1)	3(1)	4(1)
C(33)	21(1)	25(1)	20(1)	5(1)	7(1)	7(1)
C(34)	28(1)	30(1)	12(1)	0(1)	2(1)	18(1)
C(35)	18(1)	24(1)	22(1)	-4(1)	-4(1)	13(1)
C(36)	14(1)	17(1)	21(1)	1(1)	3(1)	6(1)
C(37)	14(1)	23(1)	16(1)	1(1)	5(1)	-5(1)
C(38)	22(1)	22(1)	21(1)	0(1)	8(1)	-6(1)
C(39)	33(1)	35(1)	19(1)	-10(1)	10(1)	-14(1)
C(40)	27(1)	51(2)	14(1)	4(1)	3(1)	-14(1)
C(41)	23(1)	39(1)	19(1)	11(1)	2(1)	-5(1)
C(42)	19(1)	26(1)	17(1)	4(1)	3(1)	-3(1)
N(1)	32(1)	24(1)	20(1)	-2(1)	4(1)	17(1)
Cl(1)	49(1)	23(1)	64(2)	-3(1)	-4(1)	-3(1)
Cl(2)	59(1)	42(1)	47(1)	17(1)	-31(1)	-22(1)
Cl(3)	68(1)	47(2)	72(3)	-2(2)	9(2)	13(1)
Cl(1A)	43(2)	56(2)	41(2)	15(1)	-1(1)	-10(2)
Cl(2A)	61(2)	55(1)	43(1)	28(1)	-30(1)	-23(1)
Cl(3A)	54(2)	32(2)	48(2)	10(2)	-18(2)	-6(1)
Br(1)	24(1)	17(1)	22(1)	3(1)	-3(1)	2(1)
Br(2)	38(1)	28(1)	28(1)	5(1)	16(1)	-4(1)

Table S5. Hydrogen coordinates ($\times 10^4$) and isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for **1a**.

	x	y	z	U(eq)
H(4)	2745	6193	2495	24
H(5)	924	4858	2319	29
H(6)	1336	3356	1467	32
H(7)	3578	3138	801	31
H(10)	8644	5545	680	34
H(11)	9372	7186	1271	34
H(12)	7781	8161	2101	27
H(17)	2731	9663	1276	19
H(19)	-1048	10999	1636	26
H(20)	-1380	10177	3005	25
H(23)	-1781	9293	4111	25
H(24)	-2019	8548	5530	28
H(26)	2090	7626	5369	22
H(30)	6120	7536	3848	19
H(32)	5530	8940	890	21
H(33)	5207	8850	-754	26
H(34)	3193	7899	-1526	27
H(35)	1458	7066	-636	25
H(36)	1756	7174	1011	20
H(38)	4032	9576	5499	26
H(39)	4917	9726	7082	36
H(40)	5997	8383	7738	38
H(41)	6129	6877	6832	32
H(42)	5295	6731	5243	25
H(43)	9153	4526	4139	35
H(43A)	8157	4240	4039	35
H(1)	6283	4121	595	29

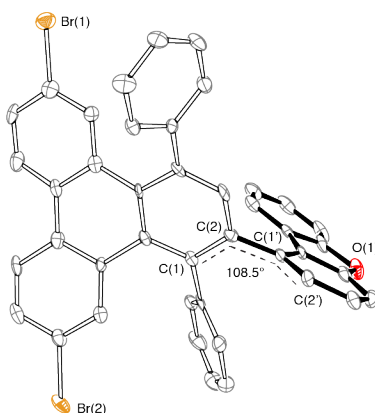


Figure S8. ORTEP representation of the molecular structures of the **1b**. Thermal ellipsoids are drawn at the 50% probability level. Color coding: C (gray), Br (orange), O (red). Hydrogen atoms and solvent molecules are omitted for clarity.

Table S6. Crystal data and structure refinement for **1b**.

CCDC no.	1567589	
Empirical formula	C ₄₂ H ₂₄ Br ₂ O	
Formula weight	704.43	
Temperature	100(2) K	
Wavelength	0.71073 Å	
Crystal system	Orthorhombic	
Space group	P b c a	
Unit cell dimensions	<i>a</i> = 7.8289(5) Å	$\alpha = 90^\circ$
	<i>b</i> = 24.8954(18) Å	$\beta = 90^\circ$
	<i>c</i> = 31.810(2) Å	$\gamma = 90^\circ$
Volume	6200.0(7) Å ³	
Z	8	
Density (calculated)	1.509 Mg/m ³	
Absorption coefficient	2.649 mm ⁻¹	
F(000)	2832	
Crystal size	0.100 x 0.010 x 0.005 mm ³	
Theta range for data collection	1.280° to 25.393°	
Index ranges	-9 ≤ <i>h</i> ≤ 9, -29 ≤ <i>k</i> ≤ 30, -38 ≤ <i>l</i> ≤ 38	
Reflections collected	59809	
Independent reflections	5706 [R(int) = 0.1869]	
Completeness to theta = 25.000°	100.0 %	
Absorption correction	Semi-empirical from equivalents	
Max. and min. transmission	0.7896 and 0.7196	
Refinement method	Full-matrix least-squares on F ²	
Data / restraints / parameters	5706 / 0 / 406	
Goodness-of-fit on F ²	1.007	
Final R indices [I > 2σ(I)]	R1 = 0.0655, wR2 = 0.1404	
R indices (all data)	R1 = 0.1384, wR2 = 0.1738	
Extinction coefficient	n/a	
Largest diff. peak and hole	1.323 and -0.986 eÅ ⁻³	

Table S7. Atomic coordinates ($\times 10^4$) and equivalent isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for **1b**. U(eq) is defined as one third of the trace of the orthogonalized U_{ij} tensor.

	x	y	z	U(eq)
C(1)	144(9)	2431(3)	7040(2)	26(2)
C(2)	-1398(9)	2619(3)	7185(2)	31(2)
C(3)	-1678(9)	3155(3)	7147(2)	33(2)
C(4)	-500(8)	3497(3)	6950(2)	26(2)
C(5)	1009(8)	3294(2)	6783(2)	21(2)
C(6)	1369(8)	2750(3)	6843(2)	22(2)
C(7)	2789(8)	2409(3)	6744(2)	24(2)
C(8)	4431(8)	2475(3)	6581(2)	27(2)
C(9)	5482(10)	2031(3)	6543(2)	35(2)
C(10)	4923(11)	1521(3)	6666(2)	40(2)
C(11)	3297(11)	1447(3)	6830(2)	40(2)
C(12)	2288(9)	1894(3)	6874(2)	28(2)
C(13)	2235(8)	3642(2)	6545(2)	20(1)
C(14)	2398(8)	3580(2)	6107(2)	19(1)
C(15)	3617(7)	3900(2)	5889(2)	19(2)
C(16)	3634(7)	3944(2)	5421(2)	17(1)
C(17)	2216(8)	3806(2)	5173(2)	23(2)
C(18)	2224(9)	3885(3)	4744(2)	26(2)
C(19)	3619(9)	4115(3)	4544(2)	33(2)
C(20)	4993(9)	4263(3)	4781(2)	29(2)
C(21)	5044(8)	4188(3)	5219(2)	24(2)
C(22)	6501(8)	4371(2)	5466(2)	23(2)
C(23)	8061(9)	4514(3)	5281(2)	34(2)
C(24)	9389(9)	4719(3)	5496(2)	31(2)
C(25)	9205(9)	4774(3)	5934(2)	30(2)
C(26)	7734(8)	4605(2)	6132(2)	22(2)
C(27)	6331(8)	4414(2)	5911(2)	22(2)
C(28)	4785(8)	4222(2)	6117(2)	16(1)
C(29)	4448(8)	4320(3)	6550(2)	21(2)
C(30)	3203(8)	4024(3)	6750(2)	23(2)
C(31)	1323(8)	3162(3)	5893(2)	22(2)
C(32)	-452(9)	3200(3)	5893(2)	31(2)
C(33)	-1438(10)	2816(3)	5678(2)	38(2)
C(34)	-644(11)	2404(3)	5469(2)	45(2)
C(35)	1118(11)	2352(3)	5477(2)	38(2)
C(36)	2085(9)	2736(3)	5688(2)	26(2)
C(37)	5213(7)	4782(3)	6785(2)	20(2)
C(38)	5941(8)	4715(3)	7184(2)	28(2)
C(39)	6642(9)	5149(3)	7388(2)	34(2)
C(40)	6637(9)	5653(3)	7215(2)	36(2)
C(41)	5875(9)	5728(3)	6818(2)	33(2)
C(42)	5174(8)	5295(2)	6611(2)	23(2)
O(1)	680(6)	1901(2)	7054(2)	35(1)
Br(1)	296(1)	3678(1)	4427(1)	37(1)
Br(2)	10993(1)	5077(1)	6253(1)	36(1)

Table S8. Bond lengths [Å] and angles [°] for **1b**.

C(1)-C(2)	1.375(9)
C(1)-O(1)	1.387(8)
C(1)-C(6)	1.394(9)
C(2)-C(3)	1.359(9)
C(2)-H(2)	0.9500
C(3)-C(4)	1.402(9)
C(3)-H(3)	0.9500
C(4)-C(5)	1.391(9)
C(4)-H(4)	0.9500
C(5)-C(6)	1.396(9)
C(5)-C(13)	1.497(9)
C(6)-C(7)	1.434(9)
C(7)-C(8)	1.396(9)
C(7)-C(12)	1.404(9)
C(8)-C(9)	1.384(9)
C(8)-H(8)	0.9500
C(9)-C(10)	1.400(10)
C(9)-H(9)	0.9500
C(10)-C(11)	1.388(11)
C(10)-H(10)	0.9500
C(11)-C(12)	1.370(9)
C(11)-H(11)	0.9500
C(12)-O(1)	1.382(8)
C(13)-C(30)	1.379(9)
C(13)-C(14)	1.408(8)
C(14)-C(15)	1.424(8)
C(14)-C(31)	1.502(9)
C(15)-C(28)	1.417(8)
C(15)-C(16)	1.490(8)
C(16)-C(17)	1.405(8)
C(16)-C(21)	1.415(8)
C(17)-C(18)	1.379(9)
C(17)-H(17)	0.9500
C(18)-C(19)	1.387(9)
C(18)-Br(1)	1.887(7)
C(19)-C(20)	1.363(10)
C(19)-H(19)	0.9500
C(20)-C(21)	1.406(9)
C(20)-H(20)	0.9500
C(21)-C(22)	1.460(9)
C(22)-C(23)	1.402(9)
C(22)-C(27)	1.425(9)
C(23)-C(24)	1.344(10)
C(23)-H(23)	0.9500
C(24)-C(25)	1.406(10)
C(24)-H(24)	0.9500
C(25)-C(26)	1.379(9)
C(25)-Br(2)	1.886(7)
C(26)-C(27)	1.388(9)
C(26)-H(26)	0.9500
C(27)-C(28)	1.456(8)
C(28)-C(29)	1.423(8)
C(29)-C(30)	1.378(9)
C(29)-C(37)	1.497(9)

C(30)-H(30)	0.9500
C(31)-C(36)	1.380(9)
C(31)-C(32)	1.393(9)
C(32)-C(33)	1.407(10)
C(32)-H(32)	0.9500
C(33)-C(34)	1.372(10)
C(33)-H(33)	0.9500
C(34)-C(35)	1.386(11)
C(34)-H(34)	0.9500
C(35)-C(36)	1.392(9)
C(35)-H(35)	0.9500
C(36)-H(36)	0.9500
C(37)-C(42)	1.394(9)
C(37)-C(38)	1.402(9)
C(38)-C(39)	1.374(9)
C(38)-H(38)	0.9500
C(39)-C(40)	1.371(10)
C(39)-H(39)	0.9500
C(40)-C(41)	1.407(10)
C(40)-H(40)	0.9500
C(41)-C(42)	1.379(9)
C(41)-H(41)	0.9500
C(42)-H(42)	0.9500

C(2)-C(1)-O(1)	125.3(6)
C(2)-C(1)-C(6)	124.2(6)
O(1)-C(1)-C(6)	110.4(6)
C(3)-C(2)-C(1)	116.4(6)
C(3)-C(2)-H(2)	121.8
C(1)-C(2)-H(2)	121.8
C(2)-C(3)-C(4)	122.1(7)
C(2)-C(3)-H(3)	119.0
C(4)-C(3)-H(3)	119.0
C(5)-C(4)-C(3)	120.6(7)
C(5)-C(4)-H(4)	119.7
C(3)-C(4)-H(4)	119.7
C(4)-C(5)-C(6)	118.2(6)
C(4)-C(5)-C(13)	121.9(6)
C(6)-C(5)-C(13)	120.0(6)
C(1)-C(6)-C(5)	118.3(6)
C(1)-C(6)-C(7)	107.1(6)
C(5)-C(6)-C(7)	134.6(6)
C(8)-C(7)-C(12)	118.4(6)
C(8)-C(7)-C(6)	136.4(6)
C(12)-C(7)-C(6)	105.1(6)
C(9)-C(8)-C(7)	119.1(7)
C(9)-C(8)-H(8)	120.5
C(7)-C(8)-H(8)	120.5
C(8)-C(9)-C(10)	121.0(7)
C(8)-C(9)-H(9)	119.5
C(10)-C(9)-H(9)	119.5
C(11)-C(10)-C(9)	120.8(7)
C(11)-C(10)-H(10)	119.6
C(9)-C(10)-H(10)	119.6
C(12)-C(11)-C(10)	117.4(7)
C(12)-C(11)-H(11)	121.3

C(10)-C(11)-H(11)	121.3
C(11)-C(12)-O(1)	125.3(7)
C(11)-C(12)-C(7)	123.3(7)
O(1)-C(12)-C(7)	111.4(6)
C(30)-C(13)-C(14)	119.5(6)
C(30)-C(13)-C(5)	120.8(6)
C(14)-C(13)-C(5)	119.7(5)
C(13)-C(14)-C(15)	118.9(6)
C(13)-C(14)-C(31)	118.3(5)
C(15)-C(14)-C(31)	122.8(6)
C(28)-C(15)-C(14)	119.9(6)
C(28)-C(15)-C(16)	117.6(5)
C(14)-C(15)-C(16)	122.3(5)
C(17)-C(16)-C(21)	117.7(6)
C(17)-C(16)-C(15)	122.4(5)
C(21)-C(16)-C(15)	119.5(5)
C(18)-C(17)-C(16)	121.2(6)
C(18)-C(17)-H(17)	119.4
C(16)-C(17)-H(17)	119.4
C(17)-C(18)-C(19)	121.1(6)
C(17)-C(18)-Br(1)	119.1(5)
C(19)-C(18)-Br(1)	119.8(5)
C(20)-C(19)-C(18)	118.7(7)
C(20)-C(19)-H(19)	120.7
C(18)-C(19)-H(19)	120.7
C(19)-C(20)-C(21)	122.2(6)
C(19)-C(20)-H(20)	118.9
C(21)-C(20)-H(20)	118.9
C(20)-C(21)-C(16)	119.1(6)
C(20)-C(21)-C(22)	121.1(6)
C(16)-C(21)-C(22)	119.8(6)
C(23)-C(22)-C(27)	118.6(6)
C(23)-C(22)-C(21)	122.2(6)
C(27)-C(22)-C(21)	119.1(6)
C(24)-C(23)-C(22)	123.8(7)
C(24)-C(23)-H(23)	118.1
C(22)-C(23)-H(23)	118.1
C(23)-C(24)-C(25)	117.4(6)
C(23)-C(24)-H(24)	121.3
C(25)-C(24)-H(24)	121.3
C(26)-C(25)-C(24)	120.6(6)
C(26)-C(25)-Br(2)	119.7(5)
C(24)-C(25)-Br(2)	119.7(5)
C(25)-C(26)-C(27)	122.2(7)
C(25)-C(26)-H(26)	118.9
C(27)-C(26)-H(26)	118.9
C(26)-C(27)-C(22)	117.0(6)
C(26)-C(27)-C(28)	122.9(6)
C(22)-C(27)-C(28)	119.9(6)
C(15)-C(28)-C(29)	118.3(5)
C(15)-C(28)-C(27)	119.5(6)
C(29)-C(28)-C(27)	122.2(6)
C(30)-C(29)-C(28)	119.1(6)
C(30)-C(29)-C(37)	117.6(6)
C(28)-C(29)-C(37)	122.8(6)
C(29)-C(30)-C(13)	122.7(6)

C(29)-C(30)-H(30)	118.7
C(13)-C(30)-H(30)	118.7
C(36)-C(31)-C(32)	119.0(6)
C(36)-C(31)-C(14)	120.3(6)
C(32)-C(31)-C(14)	120.7(6)
C(31)-C(32)-C(33)	120.0(7)
C(31)-C(32)-H(32)	120.0
C(33)-C(32)-H(32)	120.0
C(34)-C(33)-C(32)	119.7(7)
C(34)-C(33)-H(33)	120.2
C(32)-C(33)-H(33)	120.2
C(33)-C(34)-C(35)	120.8(7)
C(33)-C(34)-H(34)	119.6
C(35)-C(34)-H(34)	119.6
C(34)-C(35)-C(36)	119.1(7)
C(34)-C(35)-H(35)	120.5
C(36)-C(35)-H(35)	120.5
C(31)-C(36)-C(35)	121.3(7)
C(31)-C(36)-H(36)	119.3
C(35)-C(36)-H(36)	119.3
C(42)-C(37)-C(38)	118.6(6)
C(42)-C(37)-C(29)	119.8(6)
C(38)-C(37)-C(29)	121.6(6)
C(39)-C(38)-C(37)	119.7(7)
C(39)-C(38)-H(38)	120.2
C(37)-C(38)-H(38)	120.2
C(40)-C(39)-C(38)	122.1(7)
C(40)-C(39)-H(39)	119.0
C(38)-C(39)-H(39)	119.0
C(39)-C(40)-C(41)	118.8(7)
C(39)-C(40)-H(40)	120.6
C(41)-C(40)-H(40)	120.6
C(42)-C(41)-C(40)	119.7(7)
C(42)-C(41)-H(41)	120.2
C(40)-C(41)-H(41)	120.2
C(41)-C(42)-C(37)	121.2(7)
C(41)-C(42)-H(42)	119.4
C(37)-C(42)-H(42)	119.4
C(12)-O(1)-C(1)	105.9(5)

Table S9. Anisotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for **1b**. The anisotropic displacement factor exponent takes the form: $-2p^2[h^2 a^{*2} U^{11} + \dots + 2 h k a^* b^* U^{12}]$

	U ¹¹	U ²²	U ³³	U ²³	U ¹³	U ¹²
C(1)	42(4)	24(4)	12(4)	-5(3)	-3(3)	-8(3)
C(2)	36(4)	39(5)	18(4)	-4(3)	7(3)	-14(3)
C(3)	38(4)	48(5)	14(4)	-2(3)	3(3)	-2(4)
C(4)	29(4)	34(4)	15(4)	0(3)	4(3)	-4(3)
C(5)	27(4)	25(4)	11(3)	2(3)	-3(3)	-6(3)
C(6)	30(4)	29(4)	8(3)	1(3)	-1(3)	-9(3)
C(7)	34(4)	31(4)	7(3)	-4(3)	-3(3)	0(3)
C(8)	38(4)	28(4)	15(4)	-2(3)	-2(3)	6(3)
C(9)	43(4)	44(5)	19(4)	-6(3)	-1(3)	9(4)
C(10)	63(6)	39(5)	18(4)	-6(3)	-8(4)	13(4)
C(11)	75(6)	29(5)	16(4)	2(3)	-2(4)	7(4)
C(12)	48(5)	28(4)	10(4)	-2(3)	3(3)	-3(3)
C(13)	22(3)	24(4)	15(4)	-3(3)	-1(3)	3(3)
C(14)	27(4)	14(3)	15(4)	0(3)	-1(3)	2(3)
C(15)	23(4)	18(3)	14(4)	4(3)	1(3)	7(3)
C(16)	20(3)	20(4)	12(3)	-2(3)	7(3)	-1(3)
C(17)	32(4)	17(4)	21(4)	-1(3)	3(3)	1(3)
C(18)	38(4)	25(4)	17(4)	-3(3)	1(3)	-2(3)
C(19)	53(5)	28(4)	17(4)	5(3)	3(4)	4(4)
C(20)	41(5)	34(4)	12(4)	4(3)	11(3)	-1(3)
C(21)	31(4)	22(4)	19(4)	-4(3)	9(3)	1(3)
C(22)	35(4)	19(4)	15(4)	2(3)	5(3)	1(3)
C(23)	42(5)	33(4)	26(5)	-1(3)	13(4)	-5(4)
C(24)	29(4)	35(4)	30(5)	6(3)	16(3)	-7(3)
C(25)	34(4)	21(4)	33(5)	-3(3)	2(4)	0(3)
C(26)	20(3)	23(4)	23(4)	-1(3)	2(3)	2(3)
C(27)	29(4)	16(4)	22(4)	0(3)	4(3)	5(3)
C(28)	20(3)	21(3)	8(3)	5(3)	5(3)	1(3)
C(29)	26(4)	27(4)	9(3)	0(3)	-3(3)	7(3)
C(30)	32(4)	33(4)	4(3)	-3(3)	5(3)	3(3)
C(31)	39(4)	25(4)	3(3)	0(3)	3(3)	-2(3)
C(32)	33(4)	43(5)	16(4)	-4(3)	4(3)	-10(3)
C(33)	39(5)	48(5)	28(5)	2(4)	-5(4)	-17(4)
C(34)	71(7)	33(5)	32(5)	1(4)	-18(4)	-24(4)
C(35)	66(6)	28(4)	21(4)	-4(3)	-12(4)	-1(4)
C(36)	34(4)	29(4)	15(4)	3(3)	-3(3)	-7(3)
C(37)	15(3)	30(4)	14(3)	1(3)	5(3)	-1(3)
C(38)	30(4)	32(4)	20(4)	4(3)	1(3)	0(3)
C(39)	41(4)	47(5)	14(4)	-5(4)	-4(3)	-3(4)
C(40)	45(5)	40(5)	23(4)	-14(4)	-3(4)	-13(4)
C(41)	39(4)	25(4)	36(5)	-15(3)	-3(4)	4(3)
C(42)	28(4)	22(4)	17(4)	0(3)	3(3)	4(3)
O(1)	58(4)	27(3)	19(3)	-4(2)	7(2)	-8(2)
Br(1)	52(1)	44(1)	14(1)	-3(1)	-9(1)	1(1)
Br(2)	24(1)	39(1)	43(1)	3(1)	-4(1)	-1(1)

Table S10. Hydrogen coordinates ($\times 10^4$) and isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for **1b**.

	x	y	z	U(eq)
H(2)	-2224	2386	7307	37
H(3)	-2704	3303	7256	40
H(4)	-734	3871	6932	31
H(8)	4821	2820	6498	32
H(9)	6600	2073	6431	42
H(10)	5665	1221	6636	48
H(11)	2898	1101	6908	48
H(17)	1234	3655	5303	28
H(19)	3616	4168	4249	39
H(20)	5949	4421	4645	35
H(23)	8187	4463	4987	40
H(24)	10412	4823	5357	38
H(26)	7679	4620	6430	26
H(30)	3004	4085	7040	28
H(32)	-997	3486	6038	37
H(33)	-2649	2842	5677	46
H(34)	-1309	2151	5317	54
H(35)	1659	2058	5341	46
H(36)	3295	2705	5690	31
H(38)	5950	4371	7314	33
H(39)	7146	5098	7656	41
H(40)	7140	5947	7360	43
H(41)	5843	6076	6695	40
H(42)	4656	5348	6344	27

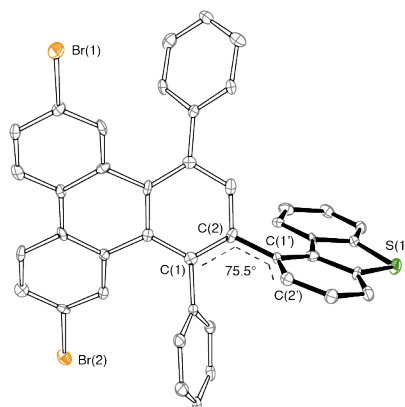


Figure S9. ORTEP representation of the molecular structures of the **1c**. Thermal ellipsoids are drawn at the 50% probability level. Color coding: C (gray), Br (orange), S (green). Hydrogen atoms are omitted for clarity.

Table S11. Crystal data and structure refinement for **1c**.

CCDC no.	1567588	
Empirical formula	C ₄₉ H ₃₂ Br ₂ S	
Formula weight	812.62	
Temperature	100(2) K	
Wavelength	0.71073 Å	
Crystal system	Triclinic	
Space group	P-1	
Unit cell dimensions	<i>a</i> = 7.0949(4) Å	α = 104.009(3)°
	<i>b</i> = 16.1951(9) Å	β = 92.325(3)°
	<i>c</i> = 16.9526(9) Å	γ = 93.650(3)°
Volume	1882.97(18) Å ³	
Z	2	
Density (calculated)	1.433 Mg/ m ³	
Absorption coefficient	2.243 mm ⁻¹	
F(000)	824	
Crystal size	0.050 x 0.040 x 0.020 mm ³	
Theta range for data collection	1.240° to 25.393°	
Index ranges	-8 ≤ <i>h</i> ≤ 8, -19 ≤ <i>k</i> ≤ 19, -20 ≤ <i>l</i> ≤ 20	
Reflections collected	11841	
Independent reflections	11841 [R(int) = 0.0503]	
Completeness to theta = 25.000°	98.8 %	
Absorption correction	Semi-empirical from equivalents	
Max. and min. transmission	0.928 and 0.824	
Refinement method	Full-matrix least-squares on F ²	
Data / restraints / parameters	11841 / 0 / 436	
Goodness-of-fit on F ²	1.027	
Final R indices [I > 2σ(I)]	R1 = 0.0520, wR2 = 0.1274	
R indices (all data)	R1 = 0.0706, wR2 = 0.1360	
Extinction coefficient	n/a	
Largest diff. peak and hole	1.103 and -0.480 eÅ ⁻³	

Table S12. Atomic coordinates ($\times 10^4$) and equivalent isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for **1c**. U(eq) is defined as one third of the trace of the orthogonalized U^{ij} tensor.

	x	y	z	U(eq)
C(1)	6384(8)	6741(4)	2586(4)	21(1)
C(2)	5713(7)	7549(3)	2734(3)	16(1)
C(3)	6559(8)	8189(3)	3396(3)	16(1)
C(4)	6439(8)	9111(4)	3464(3)	17(1)
C(5)	5972(7)	9423(3)	2782(4)	19(1)
C(6)	5929(8)	10285(3)	2845(4)	17(1)
C(7)	6375(8)	10879(4)	3583(3)	18(1)
C(8)	6888(7)	10584(4)	4242(4)	19(1)
C(9)	6975(8)	9710(3)	4198(3)	14(1)
C(10)	7634(7)	9408(3)	4900(3)	16(1)
C(11)	7878(8)	9974(4)	5696(4)	19(1)
C(12)	8413(8)	9716(4)	6365(4)	20(1)
C(13)	8742(8)	8850(4)	6260(3)	17(1)
C(14)	8535(7)	8276(3)	5509(3)	16(1)
C(15)	7993(7)	8551(3)	4801(3)	14(1)
C(16)	7710(8)	7947(3)	4001(3)	15(1)
C(17)	8428(7)	7123(3)	3796(3)	16(1)
C(18)	7863(8)	6543(3)	3072(3)	15(1)
C(19)	4043(8)	7635(3)	2209(3)	15(1)
C(20)	3845(9)	7215(4)	1395(3)	23(1)
C(21)	2158(9)	7207(4)	936(4)	33(2)
C(22)	662(9)	7637(4)	1292(4)	30(2)
C(23)	851(8)	8074(4)	2091(4)	24(2)
C(24)	2525(8)	8078(3)	2553(3)	16(1)
C(25)	9960(8)	6878(3)	4327(3)	18(1)
C(26)	11793(8)	7249(4)	4370(3)	20(1)
C(27)	13236(8)	7032(4)	4855(4)	25(2)
C(28)	12796(9)	6444(3)	5313(3)	21(1)
C(29)	10974(8)	6061(3)	5259(3)	20(1)
C(30)	9580(8)	6258(3)	4755(3)	16(1)
C(31)	8803(8)	5751(3)	2734(3)	18(1)
C(32)	10540(8)	5852(4)	2380(3)	18(1)
C(33)	11433(8)	5179(4)	1954(4)	23(1)
C(34)	10647(8)	4344(4)	1847(3)	21(1)
C(35)	8952(8)	4233(3)	2184(3)	17(1)
C(36)	5889(8)	3718(4)	2667(3)	19(1)
C(37)	4312(8)	3288(4)	2860(3)	19(1)
C(38)	3041(8)	3743(4)	3327(4)	24(1)
C(39)	3353(8)	4630(4)	3630(4)	24(2)
C(40)	4941(8)	5060(4)	3425(3)	18(1)
C(41)	6249(8)	4610(3)	2937(3)	15(1)
C(42)	7978(8)	4926(4)	2642(3)	18(1)
C(43)	9919(14)	9017(6)	250(6)	67(3)
C(44)	8237(11)	9104(5)	620(5)	50(2)
C(45)	6619(13)	8691(6)	299(6)	68(3)
C(46)	6484(13)	8111(6)	-463(5)	63(2)
C(47)	8197(13)	8025(6)	-850(6)	68(3)
C(48)	9851(12)	8425(5)	-540(5)	52(2)
C(49)	11666(13)	9444(6)	621(6)	80(3)
S(1)	7724(2)	3239(1)	2096(1)	23(1)
Br(1)	5307(1)	10699(1)	1921(1)	29(1)
Br(2)	9478(1)	8463(1)	7189(1)	26(1)

Table S13. Bond lengths [Å] and angles [°] for **1c**.

C(1)-C(2)	1.389(7)
C(1)-C(18)	1.407(8)
C(1)-H(1)	0.9500
C(2)-C(3)	1.413(8)
C(2)-C(19)	1.488(7)
C(3)-C(16)	1.429(8)
C(3)-C(4)	1.477(7)
C(4)-C(9)	1.403(8)
C(4)-C(5)	1.405(8)
C(5)-C(6)	1.376(7)
C(5)-H(5)	0.9500
C(6)-C(7)	1.392(8)
C(6)-Br(1)	1.895(6)
C(7)-C(8)	1.363(8)
C(7)-H(7)	0.9500
C(8)-C(9)	1.405(8)
C(8)-H(8)	0.9500
C(9)-C(10)	1.463(8)
C(10)-C(15)	1.399(7)
C(10)-C(11)	1.432(8)
C(11)-C(12)	1.349(8)
C(11)-H(11)	0.9500
C(12)-C(13)	1.406(8)
C(12)-H(12)	0.9500
C(13)-C(14)	1.379(8)
C(13)-Br(2)	1.898(6)
C(14)-C(15)	1.424(8)
C(14)-H(14)	0.9500
C(15)-C(16)	1.465(7)
C(16)-C(17)	1.427(7)
C(17)-C(18)	1.380(8)
C(17)-C(25)	1.516(8)
C(18)-C(31)	1.484(7)
C(19)-C(20)	1.379(8)
C(19)-C(24)	1.400(7)
C(20)-C(21)	1.398(8)
C(20)-H(20)	0.9500
C(21)-C(22)	1.385(9)
C(21)-H(21)	0.9500
C(22)-C(23)	1.364(8)
C(22)-H(22)	0.9500
C(23)-C(24)	1.395(8)
C(23)-H(23)	0.9500
C(24)-H(24)	0.9500
C(25)-C(26)	1.389(8)
C(25)-C(30)	1.394(8)
C(26)-C(27)	1.402(8)
C(26)-H(26)	0.9500
C(27)-C(28)	1.395(8)
C(27)-H(27)	0.9500
C(28)-C(29)	1.387(8)
C(28)-H(28)	0.9500
C(29)-C(30)	1.384(8)
C(29)-H(29)	0.9500

C(30)-H(30)	0.9500
C(31)-C(42)	1.394(8)
C(31)-C(32)	1.410(8)
C(32)-C(33)	1.362(8)
C(32)-H(32)	0.9500
C(33)-C(34)	1.397(8)
C(33)-H(33)	0.9500
C(34)-C(35)	1.372(8)
C(34)-H(34)	0.9500
C(35)-C(42)	1.436(7)
C(35)-S(1)	1.751(6)
C(36)-C(37)	1.373(8)
C(36)-C(41)	1.409(7)
C(36)-S(1)	1.761(6)
C(37)-C(38)	1.357(8)
C(37)-H(37)	0.9500
C(38)-C(39)	1.404(8)
C(38)-H(38)	0.9500
C(39)-C(40)	1.387(8)
C(39)-H(39)	0.9500
C(40)-C(41)	1.390(7)
C(40)-H(40)	0.9500
C(41)-C(42)	1.454(8)
C(43)-C(44)	1.370(11)
C(43)-C(49)	1.424(12)
C(43)-C(48)	1.443(11)
C(44)-C(45)	1.318(11)
C(44)-H(44)	0.9500
C(45)-C(46)	1.397(12)
C(45)-H(45)	0.9500
C(46)-C(47)	1.403(11)
C(46)-H(46)	0.9500
C(47)-C(48)	1.327(11)
C(47)-H(47)	0.9500
C(48)-H(48)	0.9500
C(49)-H(49A)	0.9800
C(49)-H(49B)	0.9800
C(49)-H(49C)	0.9800
C(2)-C(1)-C(18)	122.3(5)
C(2)-C(1)-H(1)	118.8
C(18)-C(1)-H(1)	118.8
C(1)-C(2)-C(3)	118.3(5)
C(1)-C(2)-C(19)	115.7(5)
C(3)-C(2)-C(19)	125.8(5)
C(2)-C(3)-C(16)	119.1(5)
C(2)-C(3)-C(4)	123.2(5)
C(16)-C(3)-C(4)	117.6(5)
C(9)-C(4)-C(5)	117.6(5)
C(9)-C(4)-C(3)	120.6(5)
C(5)-C(4)-C(3)	121.5(5)
C(6)-C(5)-C(4)	121.1(5)
C(6)-C(5)-H(5)	119.4
C(4)-C(5)-H(5)	119.4
C(5)-C(6)-C(7)	121.3(5)
C(5)-C(6)-Br(1)	120.8(4)
C(7)-C(6)-Br(1)	117.9(4)

C(8)-C(7)-C(6)	118.0(5)
C(8)-C(7)-H(7)	121.0
C(6)-C(7)-H(7)	121.0
C(7)-C(8)-C(9)	122.3(5)
C(7)-C(8)-H(8)	118.8
C(9)-C(8)-H(8)	118.8
C(4)-C(9)-C(8)	119.5(5)
C(4)-C(9)-C(10)	118.9(5)
C(8)-C(9)-C(10)	121.6(5)
C(15)-C(10)-C(11)	119.2(6)
C(15)-C(10)-C(9)	119.9(5)
C(11)-C(10)-C(9)	120.9(5)
C(12)-C(11)-C(10)	122.9(5)
C(12)-C(11)-H(11)	118.5
C(10)-C(11)-H(11)	118.5
C(11)-C(12)-C(13)	117.4(5)
C(11)-C(12)-H(12)	121.3
C(13)-C(12)-H(12)	121.3
C(14)-C(13)-C(12)	122.2(5)
C(14)-C(13)-Br(2)	119.3(4)
C(12)-C(13)-Br(2)	118.5(4)
C(13)-C(14)-C(15)	120.4(5)
C(13)-C(14)-H(14)	119.8
C(15)-C(14)-H(14)	119.8
C(10)-C(15)-C(14)	117.9(5)
C(10)-C(15)-C(16)	120.8(5)
C(14)-C(15)-C(16)	121.2(5)
C(17)-C(16)-C(3)	117.8(5)
C(17)-C(16)-C(15)	124.8(5)
C(3)-C(16)-C(15)	117.3(5)
C(18)-C(17)-C(16)	121.1(5)
C(18)-C(17)-C(25)	117.2(5)
C(16)-C(17)-C(25)	121.5(5)
C(17)-C(18)-C(1)	118.2(5)
C(17)-C(18)-C(31)	124.5(5)
C(1)-C(18)-C(31)	117.1(5)
C(20)-C(19)-C(24)	117.9(5)
C(20)-C(19)-C(2)	121.7(5)
C(24)-C(19)-C(2)	120.1(5)
C(19)-C(20)-C(21)	121.1(6)
C(19)-C(20)-H(20)	119.5
C(21)-C(20)-H(20)	119.5
C(22)-C(21)-C(20)	120.1(6)
C(22)-C(21)-H(21)	119.9
C(20)-C(21)-H(21)	119.9
C(23)-C(22)-C(21)	119.6(6)
C(23)-C(22)-H(22)	120.2
C(21)-C(22)-H(22)	120.2
C(22)-C(23)-C(24)	120.5(6)
C(22)-C(23)-H(23)	119.8
C(24)-C(23)-H(23)	119.8
C(23)-C(24)-C(19)	120.9(5)
C(23)-C(24)-H(24)	119.6
C(19)-C(24)-H(24)	119.6
C(26)-C(25)-C(30)	118.8(5)
C(26)-C(25)-C(17)	120.1(5)

C(30)-C(25)-C(17)	121.0(5)
C(25)-C(26)-C(27)	121.2(6)
C(25)-C(26)-H(26)	119.4
C(27)-C(26)-H(26)	119.4
C(28)-C(27)-C(26)	118.8(6)
C(28)-C(27)-H(27)	120.6
C(26)-C(27)-H(27)	120.6
C(29)-C(28)-C(27)	120.0(6)
C(29)-C(28)-H(28)	120.0
C(27)-C(28)-H(28)	120.0
C(30)-C(29)-C(28)	120.6(6)
C(30)-C(29)-H(29)	119.7
C(28)-C(29)-H(29)	119.7
C(29)-C(30)-C(25)	120.4(6)
C(29)-C(30)-H(30)	119.8
C(25)-C(30)-H(30)	119.8
C(42)-C(31)-C(32)	118.2(5)
C(42)-C(31)-C(18)	124.7(5)
C(32)-C(31)-C(18)	116.4(5)
C(33)-C(32)-C(31)	122.7(6)
C(33)-C(32)-H(32)	118.7
C(31)-C(32)-H(32)	118.7
C(32)-C(33)-C(34)	120.7(6)
C(32)-C(33)-H(33)	119.6
C(34)-C(33)-H(33)	119.6
C(35)-C(34)-C(33)	117.5(5)
C(35)-C(34)-H(34)	121.3
C(33)-C(34)-H(34)	121.3
C(34)-C(35)-C(42)	123.3(5)
C(34)-C(35)-S(1)	124.1(4)
C(42)-C(35)-S(1)	112.5(4)
C(37)-C(36)-C(41)	122.8(5)
C(37)-C(36)-S(1)	125.2(4)
C(41)-C(36)-S(1)	112.0(4)
C(38)-C(37)-C(36)	118.5(5)
C(38)-C(37)-H(37)	120.8
C(36)-C(37)-H(37)	120.8
C(37)-C(38)-C(39)	121.1(6)
C(37)-C(38)-H(38)	119.5
C(39)-C(38)-H(38)	119.5
C(40)-C(39)-C(38)	119.9(6)
C(40)-C(39)-H(39)	120.0
C(38)-C(39)-H(39)	120.0
C(39)-C(40)-C(41)	120.1(5)
C(39)-C(40)-H(40)	120.0
C(41)-C(40)-H(40)	120.0
C(40)-C(41)-C(36)	117.6(5)
C(40)-C(41)-C(42)	129.4(5)
C(36)-C(41)-C(42)	113.1(5)
C(31)-C(42)-C(35)	117.6(5)
C(31)-C(42)-C(41)	131.6(5)
C(35)-C(42)-C(41)	110.8(5)
C(44)-C(43)-C(49)	123.4(8)
C(44)-C(43)-C(48)	115.9(8)
C(49)-C(43)-C(48)	120.7(9)
C(45)-C(44)-C(43)	124.3(9)

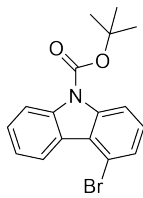
C(45)-C(44)-H(44)	117.9
C(43)-C(44)-H(44)	117.9
C(44)-C(45)-C(46)	121.9(9)
C(44)-C(45)-H(45)	119.0
C(46)-C(45)-H(45)	119.0
C(45)-C(46)-C(47)	114.2(9)
C(45)-C(46)-H(46)	122.9
C(47)-C(46)-H(46)	122.9
C(48)-C(47)-C(46)	125.3(9)
C(48)-C(47)-H(47)	117.3
C(46)-C(47)-H(47)	117.3
C(47)-C(48)-C(43)	118.4(8)
C(47)-C(48)-H(48)	120.8
C(43)-C(48)-H(48)	120.8
C(43)-C(49)-H(49A)	109.5
C(43)-C(49)-H(49B)	109.5
H(49A)-C(49)-H(49B)	109.5
C(43)-C(49)-H(49C)	109.5
H(49A)-C(49)-H(49C)	109.5
H(49B)-C(49)-H(49C)	109.5
C(35)-S(1)-C(36)	91.6(3)

Table S14. Anisotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for **1c**. The anisotropic displacement factor exponent takes the form: $-2p^2[h^2 a^{*2} U^{11} + \dots + 2 h k a^* b^* U^{12}]$

	U ¹¹	U ²²	U ³³	U ²³	U ¹³	U ¹²
C(1)	20(3)	14(3)	25(4)	-1(3)	7(3)	-1(3)
C(2)	14(3)	17(3)	18(3)	8(3)	2(3)	-2(2)
C(3)	15(3)	8(3)	23(3)	0(2)	7(2)	0(2)
C(4)	11(3)	19(3)	23(3)	8(3)	3(3)	5(2)
C(5)	13(3)	17(3)	20(3)	-7(3)	1(3)	-5(2)
C(6)	16(3)	14(3)	23(3)	9(3)	4(3)	0(2)
C(7)	19(3)	13(3)	19(3)	-1(3)	6(3)	1(3)
C(8)	10(3)	21(3)	23(3)	-2(3)	4(3)	-6(3)
C(9)	13(3)	15(3)	12(3)	-2(2)	6(2)	1(2)
C(10)	9(3)	14(3)	20(3)	-3(3)	1(2)	-6(2)
C(11)	19(3)	11(3)	24(3)	-2(2)	9(3)	2(3)
C(12)	18(3)	25(4)	17(3)	5(3)	7(3)	0(3)
C(13)	9(3)	25(3)	15(3)	4(3)	-3(2)	-2(3)
C(14)	11(3)	12(3)	22(3)	1(2)	4(3)	-1(2)
C(15)	10(3)	12(3)	20(3)	3(2)	0(2)	-3(2)
C(16)	13(3)	13(3)	18(3)	3(2)	5(2)	0(2)
C(17)	11(3)	18(3)	19(3)	6(3)	4(2)	-3(2)
C(18)	14(3)	9(3)	24(3)	6(2)	9(2)	-1(2)
C(19)	19(3)	8(3)	19(3)	7(2)	1(2)	-2(2)
C(20)	29(4)	15(3)	20(3)	-4(3)	-4(3)	3(3)
C(21)	42(4)	37(4)	18(3)	1(3)	-10(3)	7(3)
C(22)	27(4)	32(4)	24(4)	-4(3)	-10(3)	6(3)
C(23)	21(3)	17(3)	37(4)	10(3)	6(3)	1(3)
C(24)	17(3)	10(3)	19(3)	0(2)	4(2)	-7(2)
C(25)	23(3)	12(3)	16(3)	-1(2)	-3(3)	4(3)
C(26)	16(3)	21(3)	20(3)	-2(3)	3(3)	2(3)
C(27)	19(3)	19(3)	32(4)	1(3)	-2(3)	-2(3)
C(28)	27(4)	13(3)	19(3)	-4(3)	-3(3)	2(3)
C(29)	34(4)	10(3)	15(3)	1(2)	3(3)	-2(3)
C(30)	17(3)	14(3)	17(3)	3(2)	2(3)	-1(2)
C(31)	22(3)	14(3)	14(3)	-3(2)	-4(3)	5(3)
C(32)	19(3)	15(3)	20(3)	6(3)	0(3)	-3(2)
C(33)	21(3)	26(4)	20(3)	2(3)	3(3)	6(3)
C(34)	28(4)	17(3)	15(3)	1(3)	0(3)	1(3)
C(35)	20(3)	10(3)	21(3)	0(2)	1(3)	5(2)
C(36)	21(3)	17(3)	16(3)	0(2)	-3(3)	4(3)
C(37)	26(3)	13(3)	17(3)	4(3)	-2(3)	-2(3)
C(38)	18(3)	24(4)	27(4)	4(3)	-4(3)	-10(3)
C(39)	18(3)	36(4)	20(3)	8(3)	1(3)	1(3)
C(40)	15(3)	20(3)	17(3)	4(3)	-3(3)	3(3)
C(41)	18(3)	8(3)	15(3)	-2(2)	-1(2)	-4(2)
C(42)	17(3)	19(3)	20(3)	9(3)	-2(2)	0(3)
S(1)	24(1)	13(1)	30(1)	2(1)	5(1)	1(1)
Br(1)	37(1)	21(1)	30(1)	10(1)	-2(1)	3(1)
Br(2)	29(1)	26(1)	20(1)	6(1)	0(1)	-6(1)

Table S15. Hydrogen coordinates ($\times 10^4$) and isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for **1c**.

	x	y	z	U(eq)
H(1)	5826	6308	2142	25
H(5)	5681	9032	2269	22
H(7)	6324	11473	3625	21
H(8)	7198	10984	4749	23
H(11)	7652	10557	5755	22
H(12)	8563	10104	6886	24
H(14)	8755	7694	5463	19
H(20)	4870	6926	1143	27
H(21)	2038	6906	379	40
H(22)	-488	7628	982	36
H(23)	-165	8378	2334	29
H(24)	2638	8385	3108	20
H(26)	12074	7659	4066	24
H(27)	14491	7280	4872	30
H(28)	13744	6306	5661	25
H(29)	10682	5660	5571	24
H(30)	8356	5969	4701	19
H(32)	11110	6413	2440	21
H(33)	12603	5279	1727	27
H(34)	11262	3871	1552	25
H(37)	4116	2685	2671	23
H(38)	1923	3456	3450	29
H(39)	2476	4936	3975	29
H(40)	5136	5662	3619	21
H(44)	8238	9486	1143	60
H(45)	5512	8791	594	81
H(46)	5332	7801	-698	76
H(47)	8170	7651	-1378	81
H(48)	10963	8324	-831	63
H(49A)	11424	9984	998	120
H(49B)	12477	9559	200	120
H(49C)	12298	9083	921	120



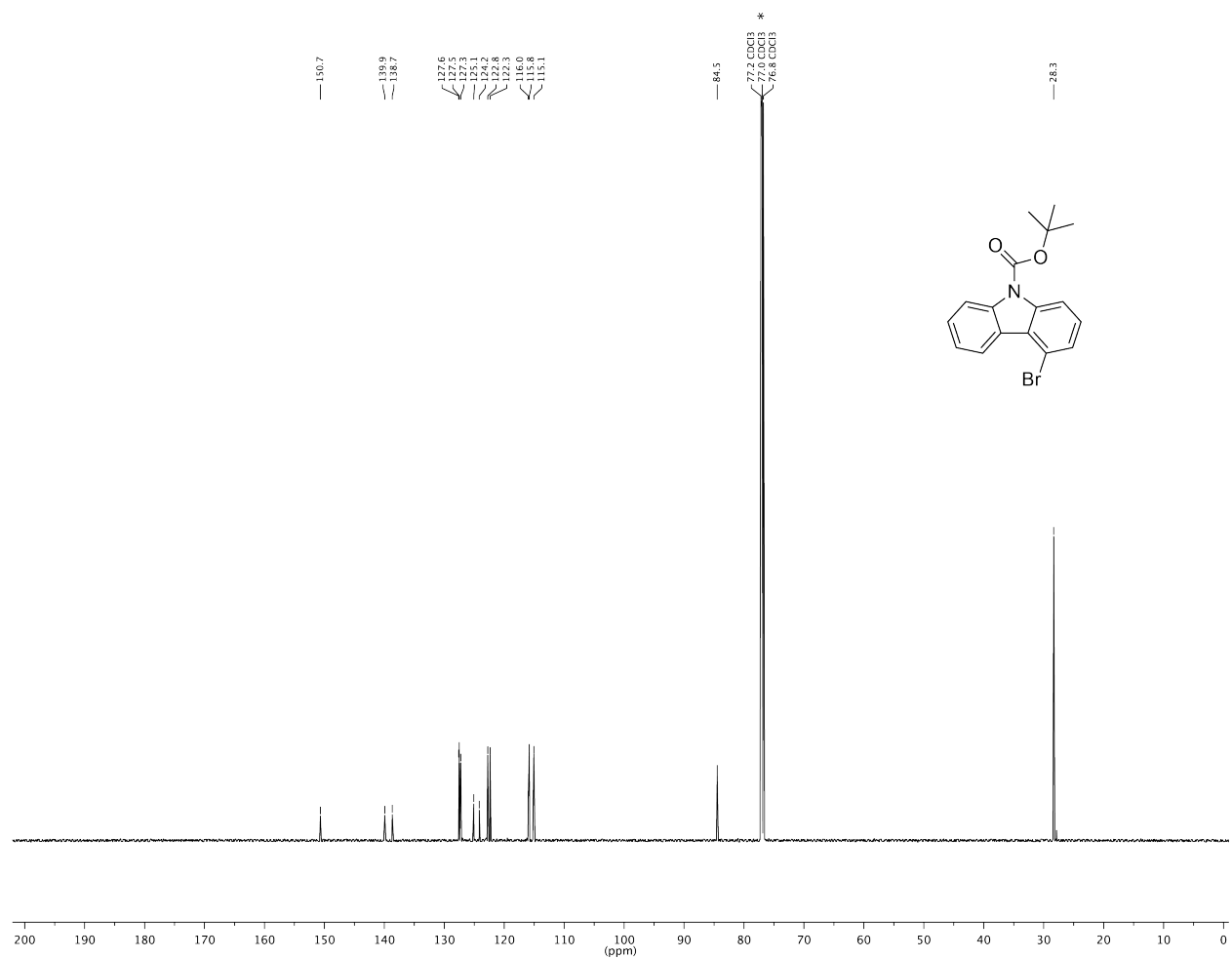


Figure S11. ^{13}C NMR (151 MHz, 25 °C) spectrum of **4** (* solvent residual peak).

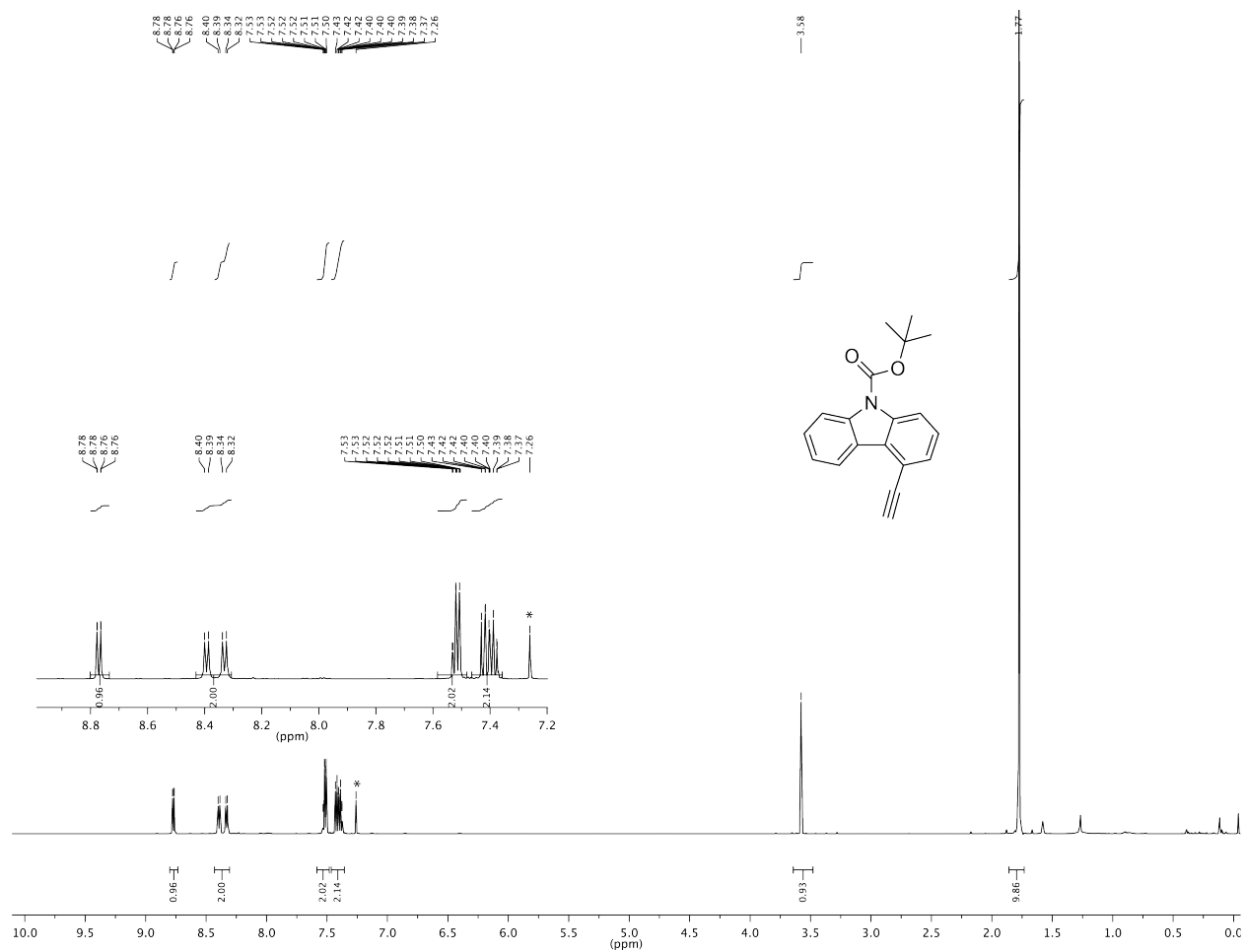


Figure S12. ¹H NMR (600 MHz, 25 °C) spectrum of **3a** (* solvent residual peak).

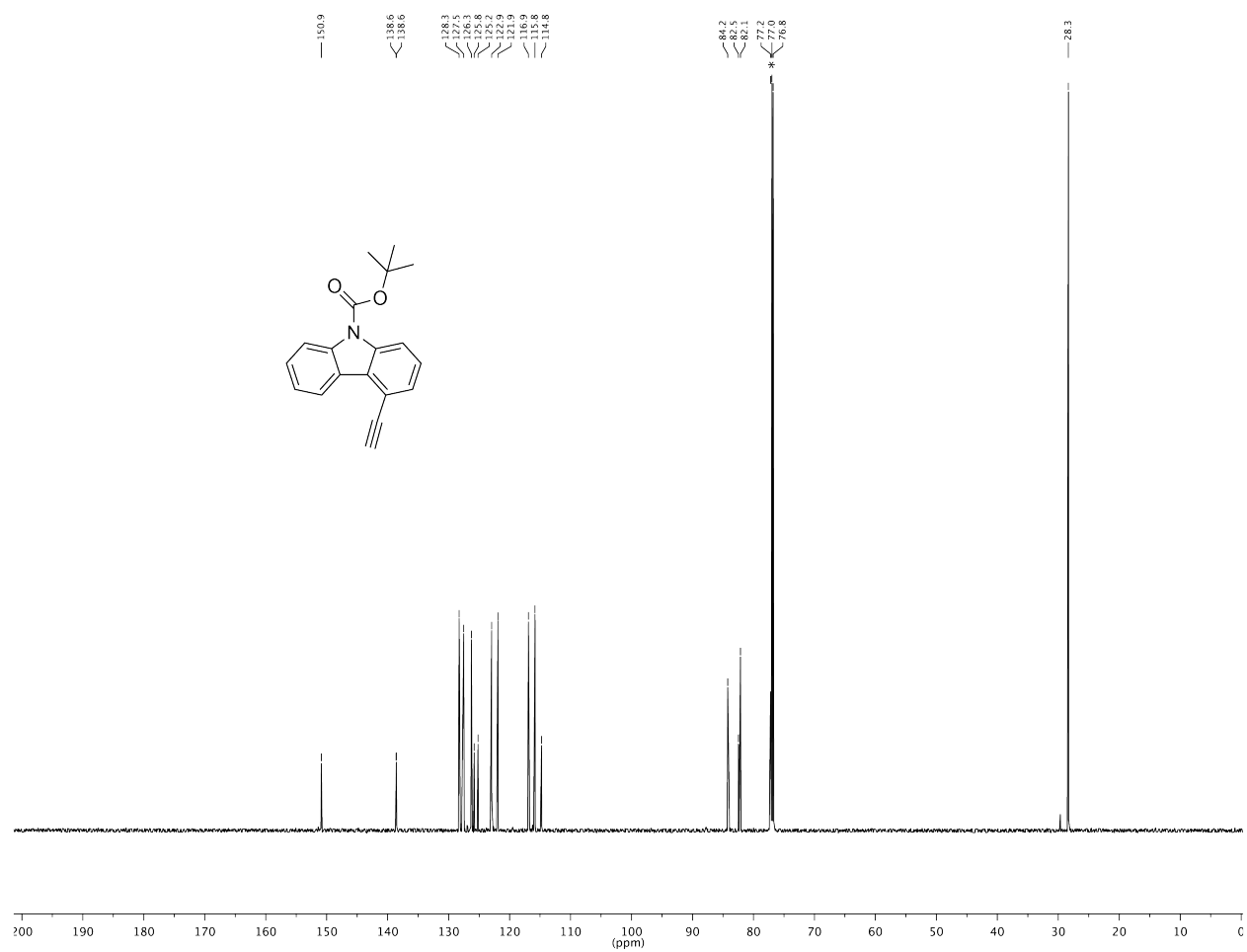


Figure S13. ¹³C NMR (151 MHz, 25 °C) spectrum of **3a** (* solvent residual peak).

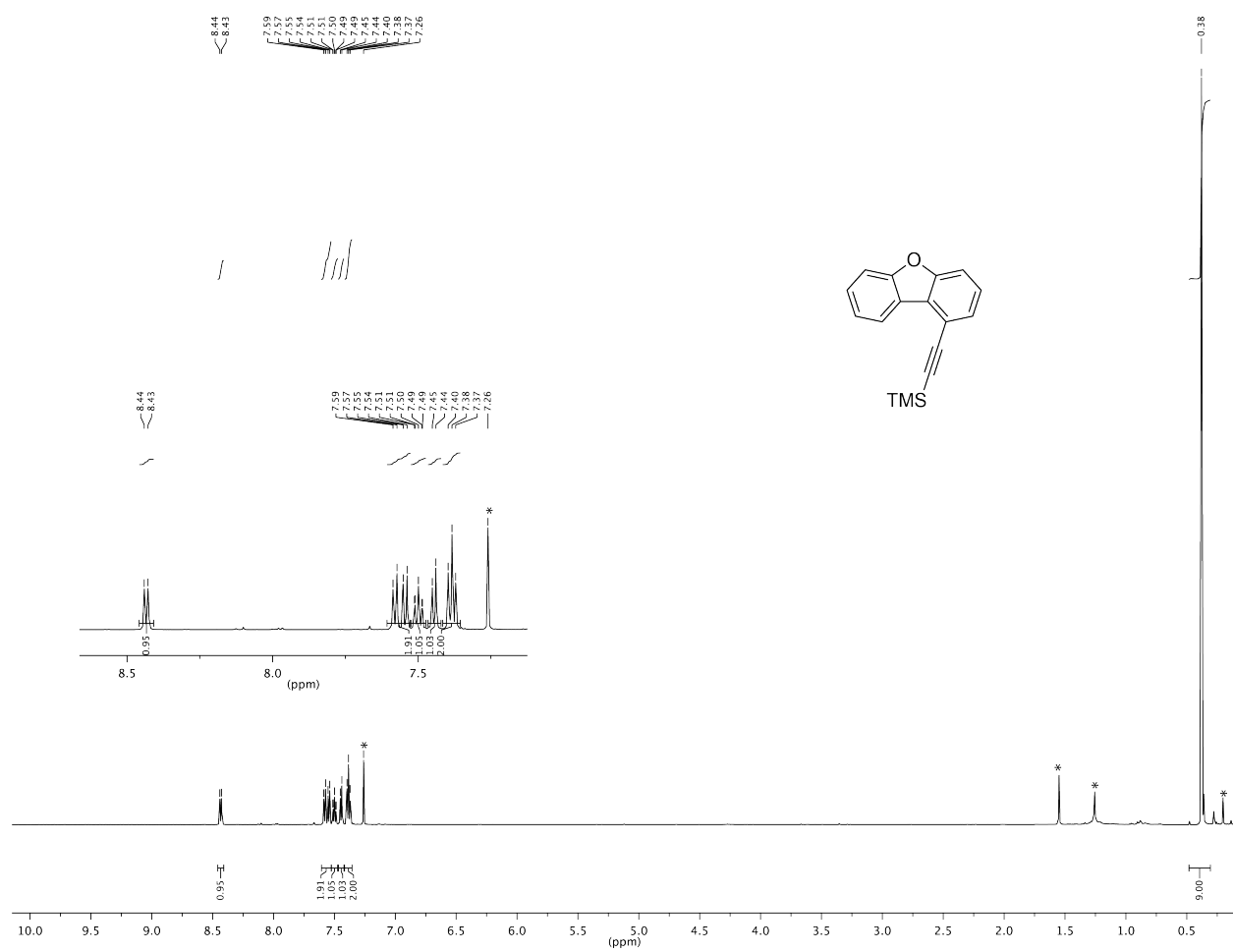


Figure S14. ^1H NMR (500 MHz, 25 °C) spectrum of **3b-TMS** (* solvent residual peak).

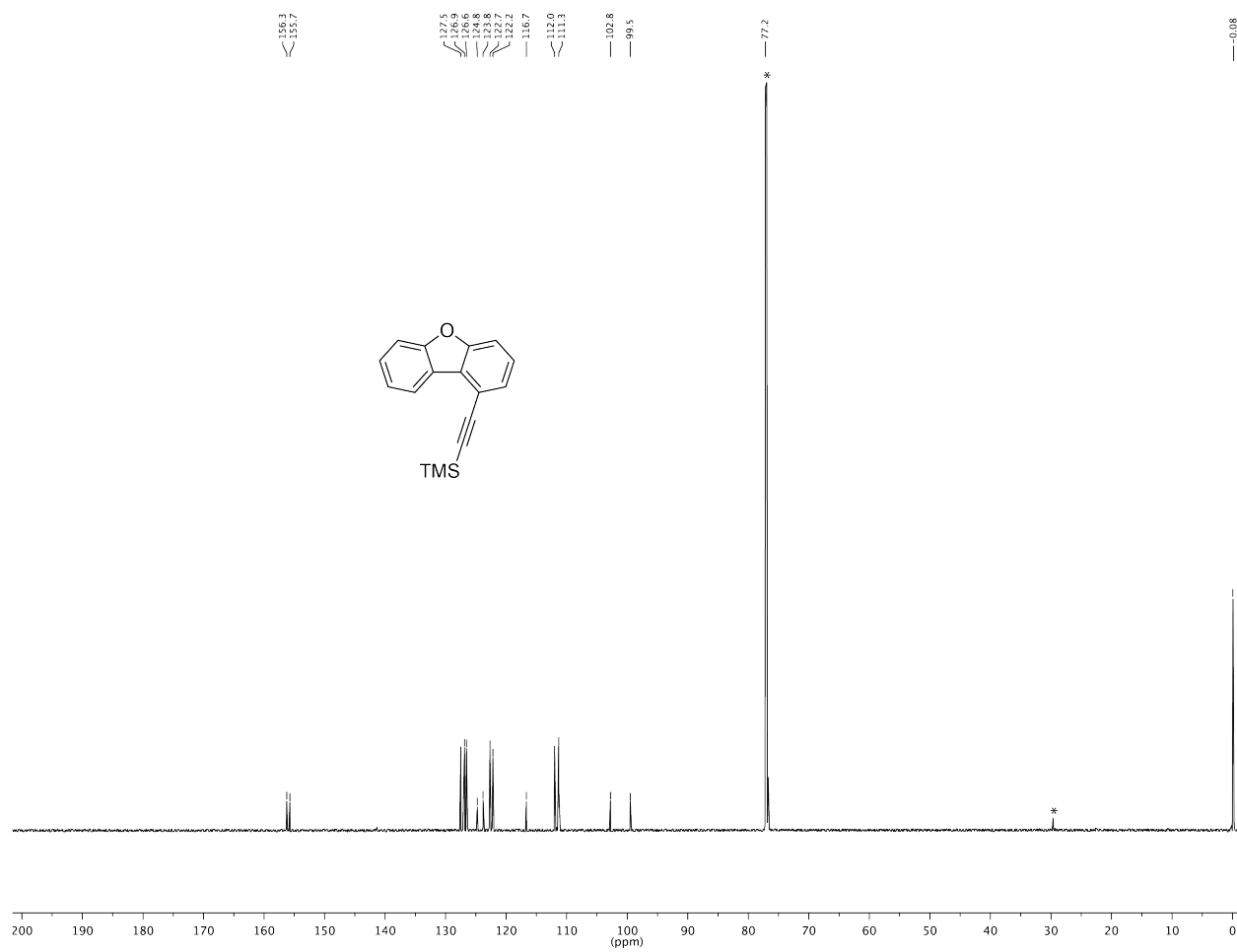


Figure S15. ¹³C NMR (151 MHz, 25 °C) spectrum of **3b-TMS** (* solvent residual peak).

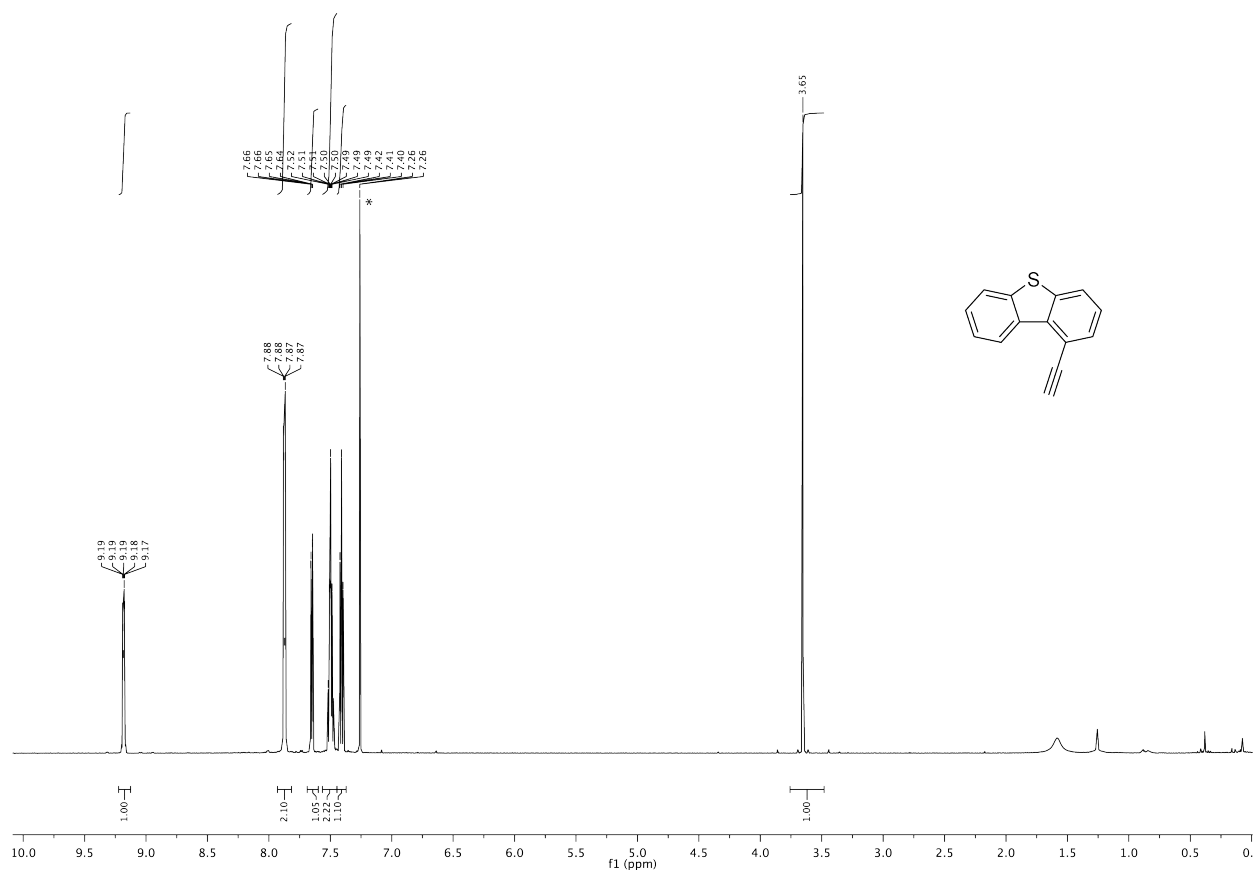


Figure S16. ¹H NMR (500 MHz, 25 °C) spectrum of **3c** (* solvent residual peak).

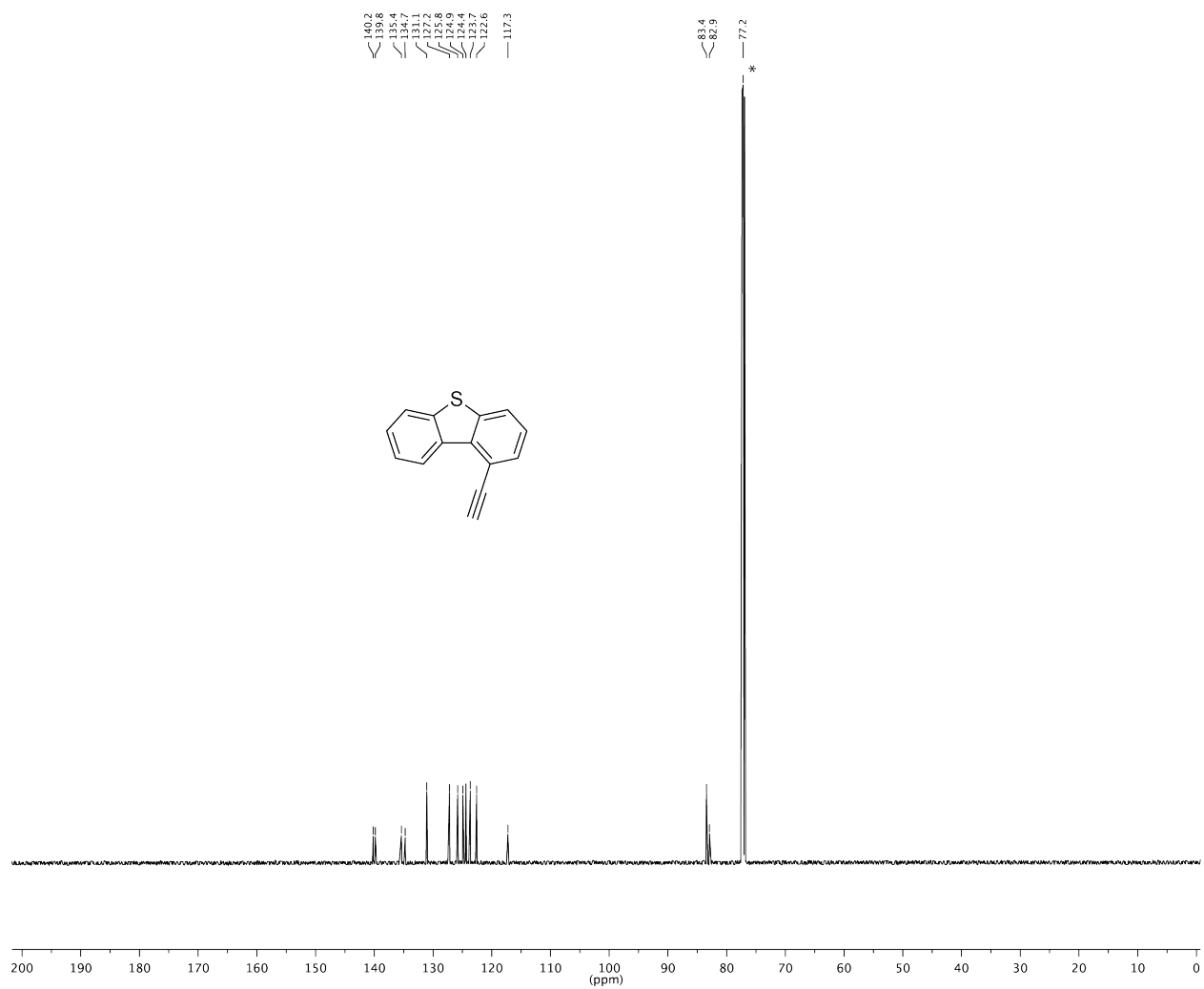


Figure S17. ¹³C NMR (151 MHz, 25 °C) spectrum of **3c** (* solvent residual peak).

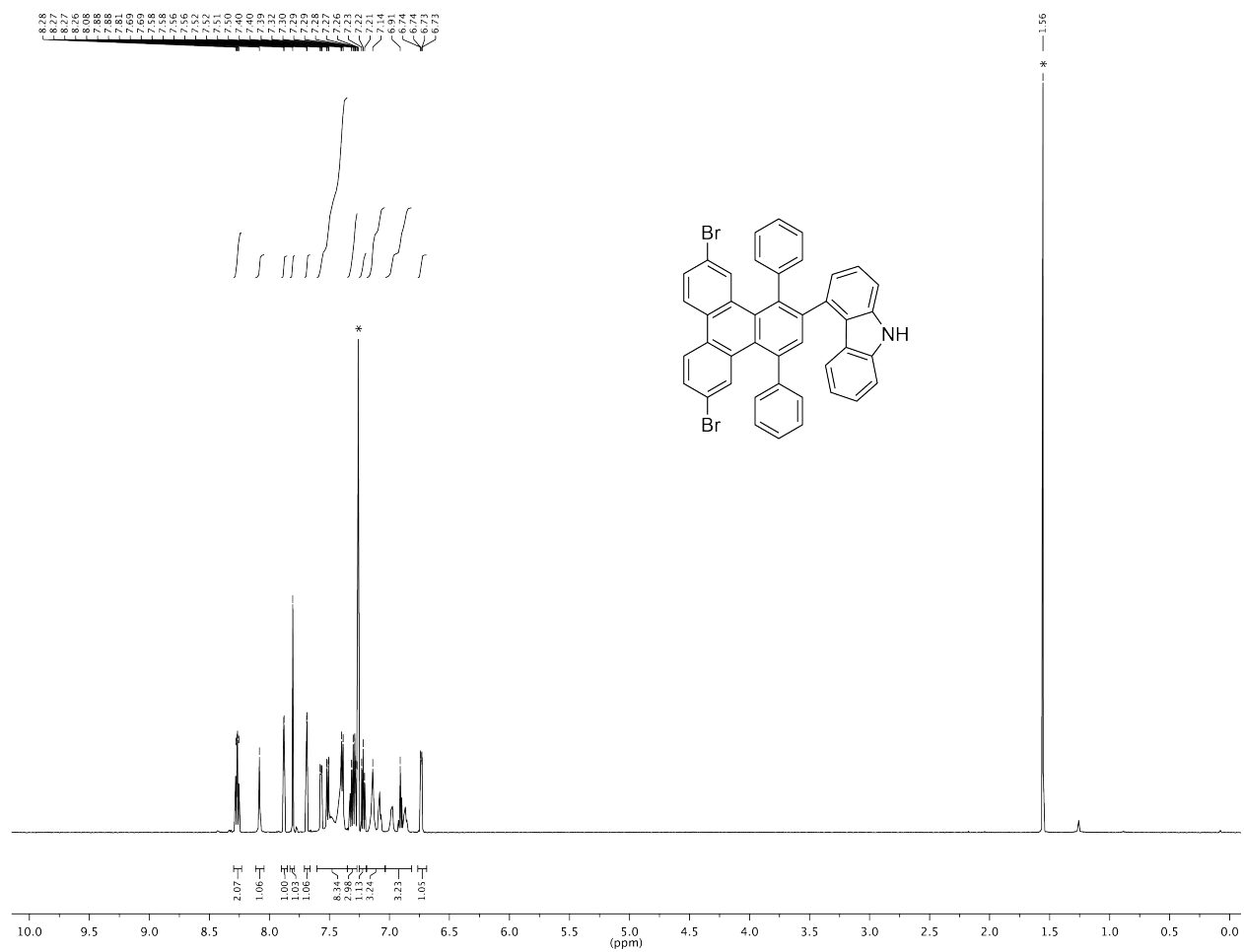


Figure S18. ^1H NMR (600 MHz, 25 °C) spectrum of **1a** (* solvent residual peak).

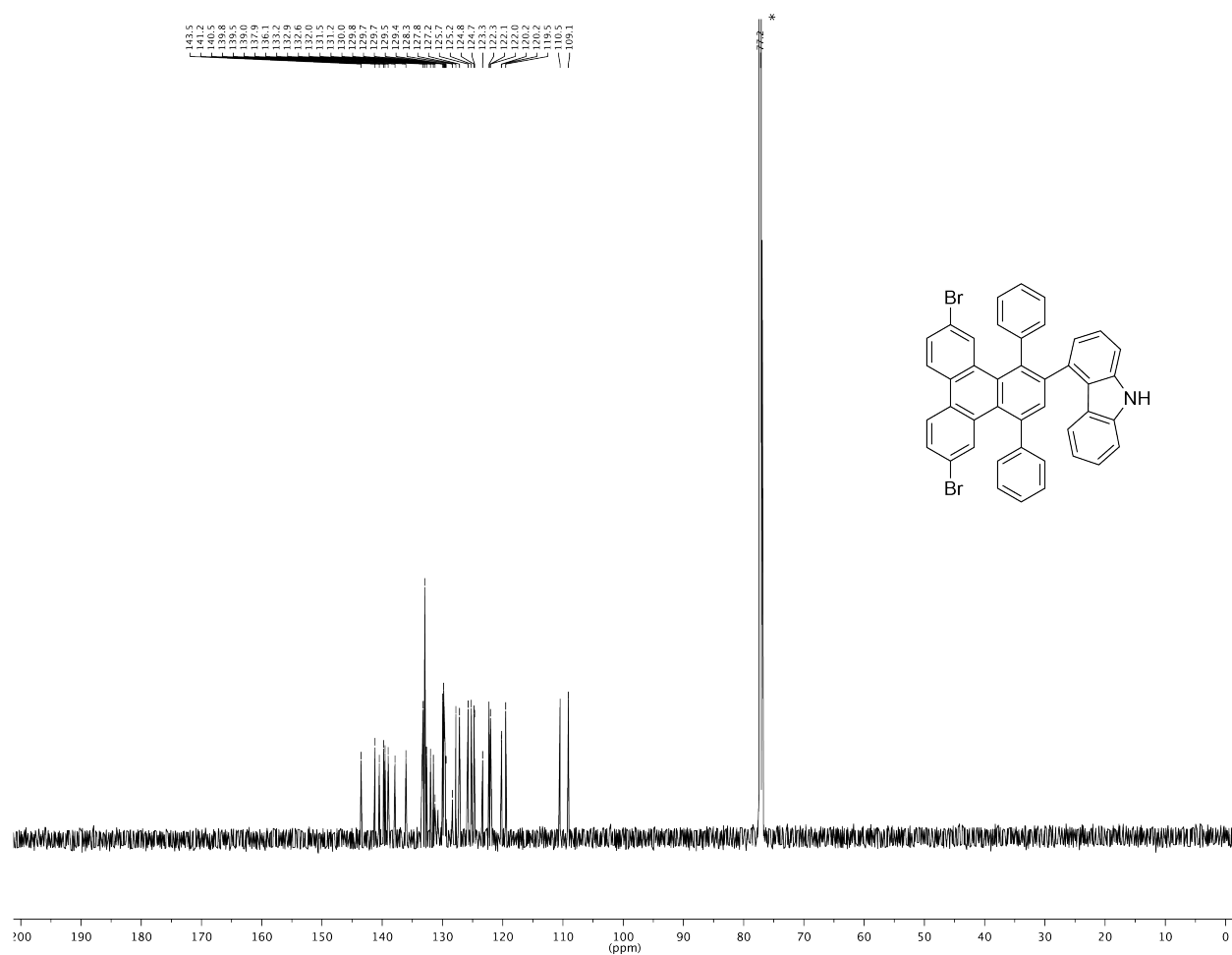
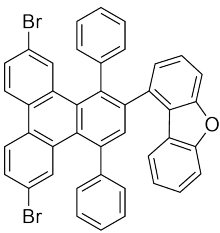


Figure S19. ¹³C NMR (151 MHz, 25 °C) spectrum of **1a** (* solvent residual peak).



S49

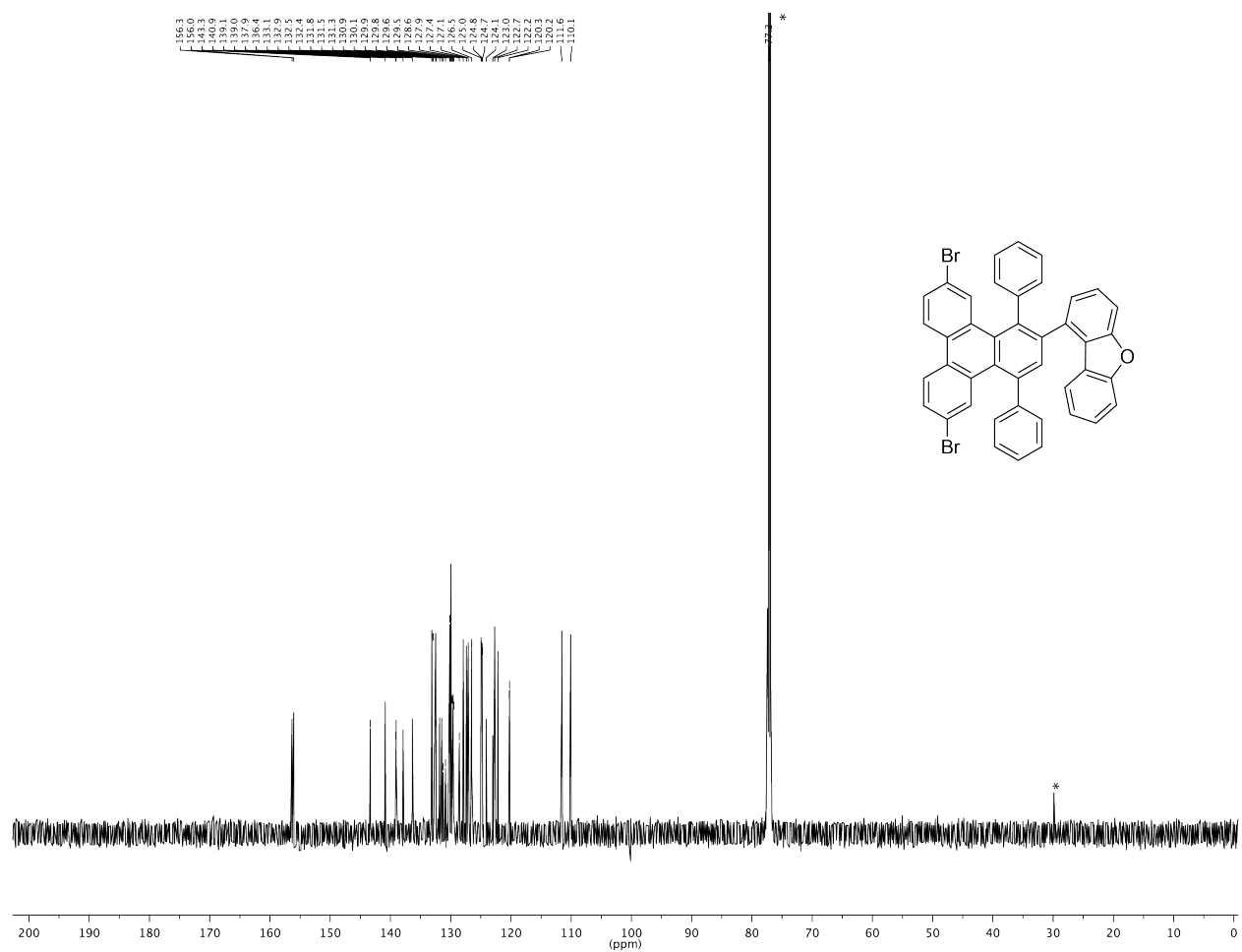


Figure S21. ¹³C NMR (151 MHz, 25 °C) spectrum of **1b** (* solvent residual peak).

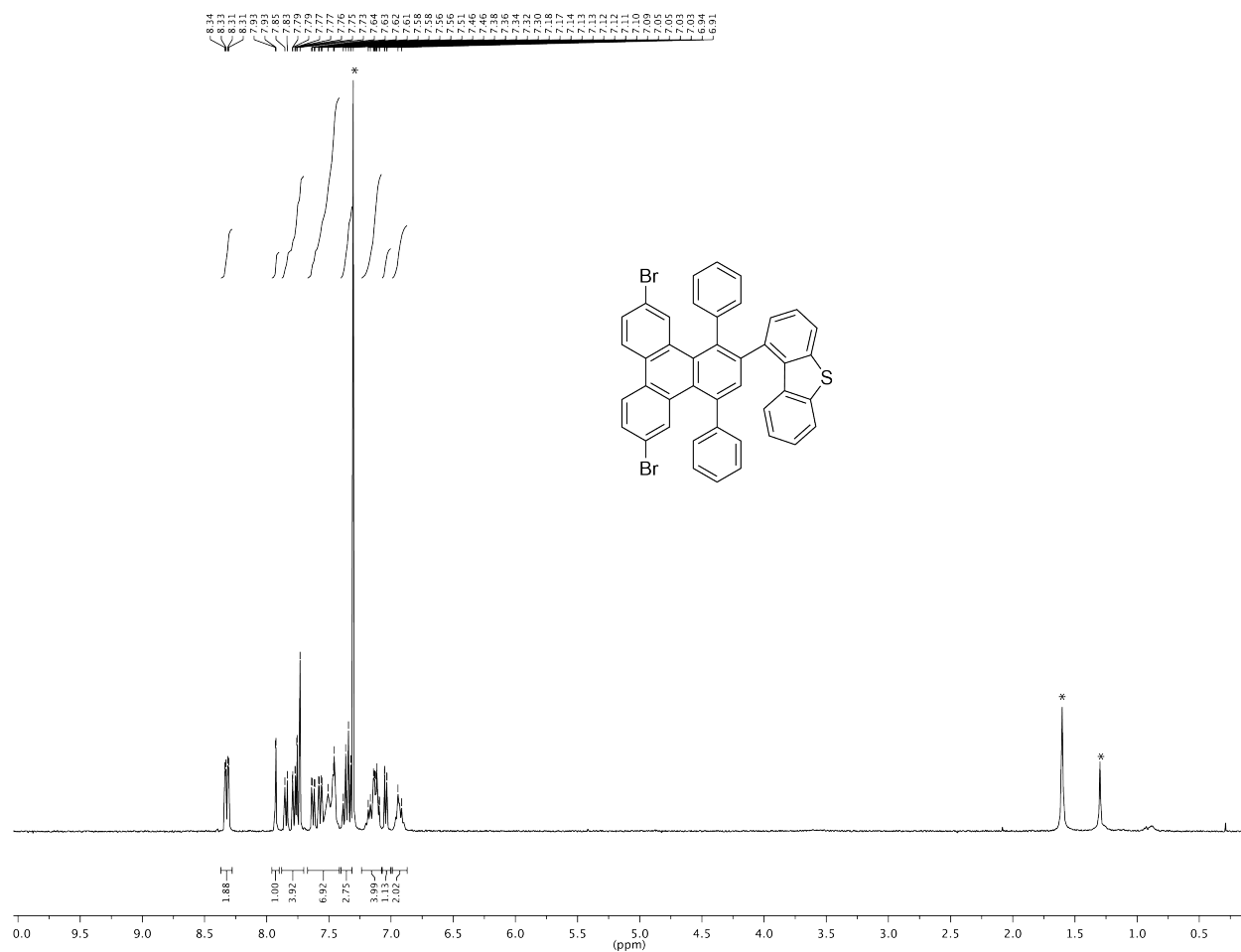


Figure S22. ^1H NMR (600 MHz, 25 °C) spectrum of **1c** (* solvent residual peak).

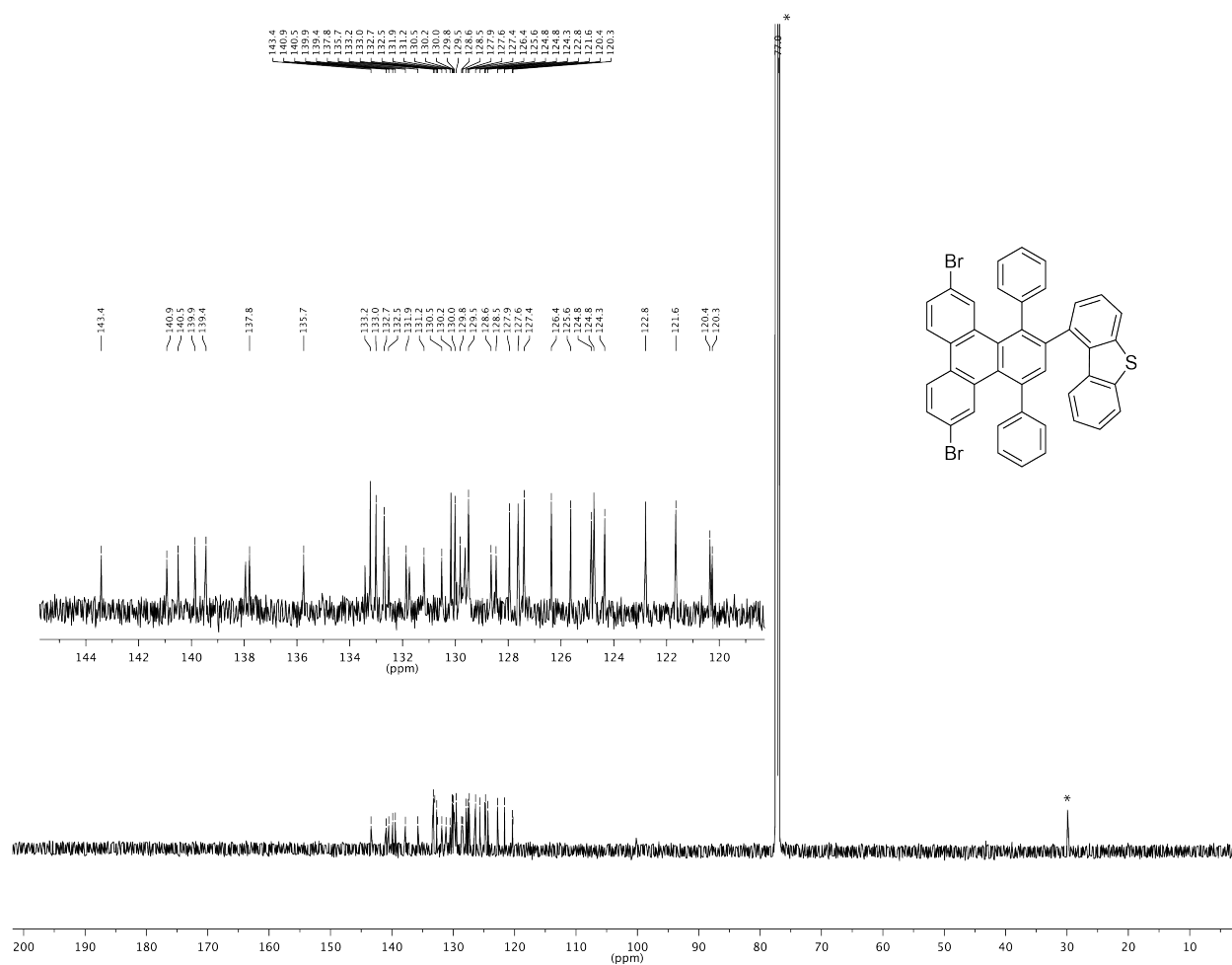


Figure S23. ^{13}C NMR (151 MHz, 25 °C) spectrum of **1c** (* solvent residual peak).

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