

Supporting Information

Multi-layer Ag-embedded Silica Nanostructure as SERS-based Chemical Sensor with Dual-Function Internal Standards

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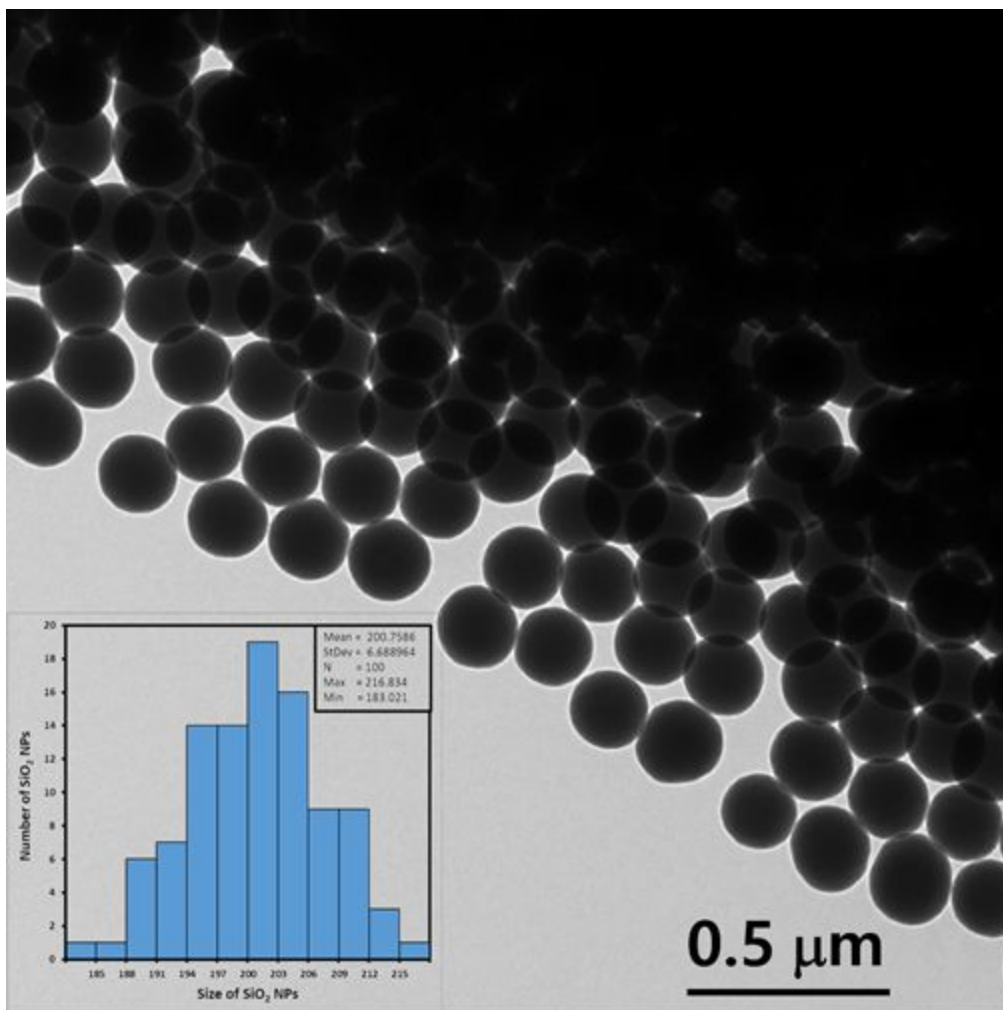


Figure S1. TEM image of SiO₂ NPs and histogram of size distribution.

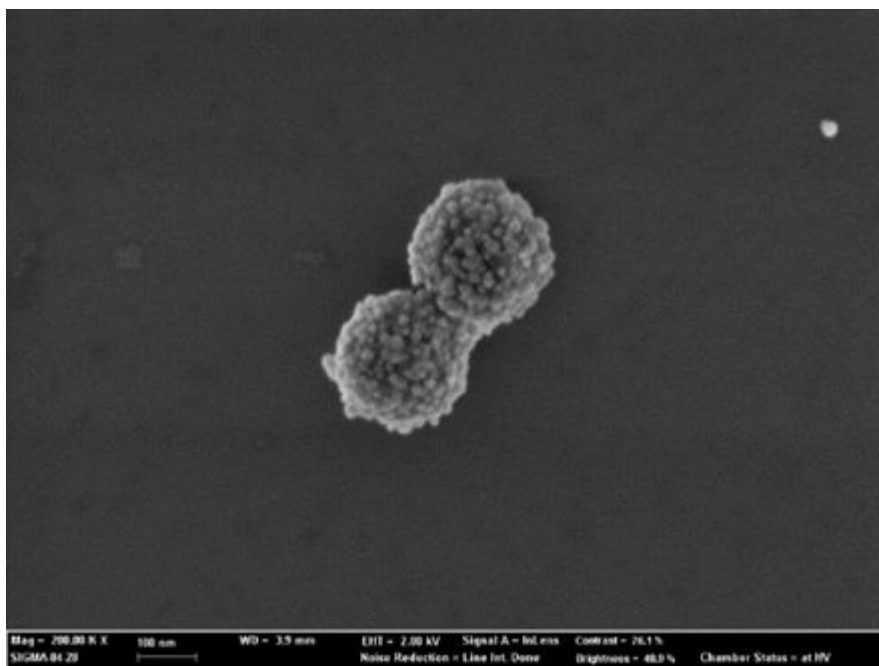


Figure S2. SEM image of ML₄-BBT dot colloids.

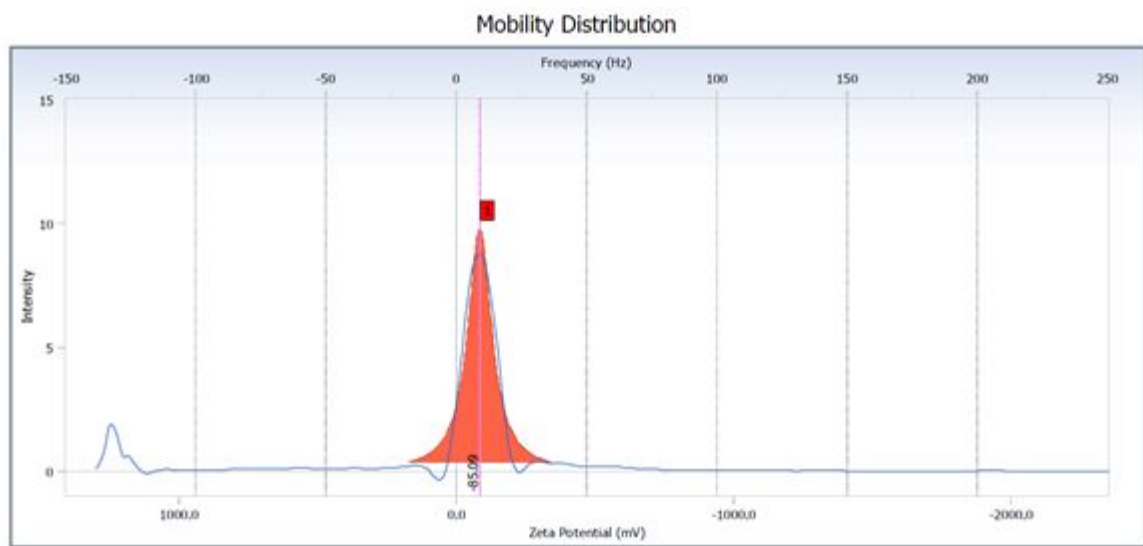


Figure S3. Zeta potential of ML₄-BBT dot colloids.



Figure S4. EDX analysis results of ML₄-BBT dots.

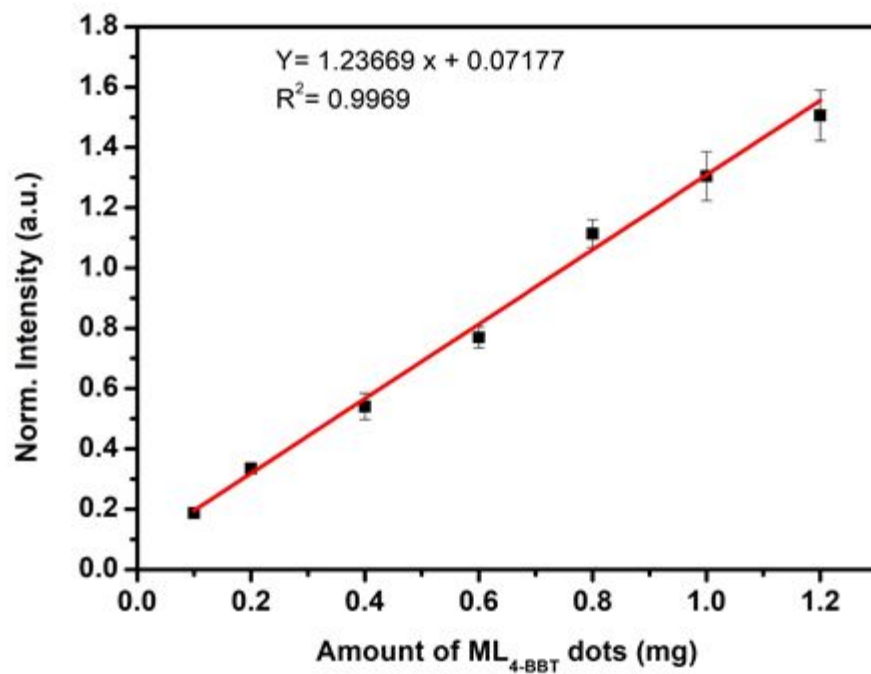


Figure S5. The internal standard SERS intensity according to amount of ML_{4-BBT} dots

To prove that the ML dots solution can remain its well dispersion during the assay, we mixed the ML dots solution with analytes (4-FBT and thiram) and took pictures of the samples after 5 hours. As shown in Figure S6 (a), the ML dots mixed with analyte were still dispersed well. Also, the UV-Visible spectrum of ML dots solution mixed with analytes (4-FBT, several concentration) was very similar to UV-Visible spectrum of ML dots. Moreover, the z- potential of ML dots mixed with 4-FBT showed also high negative value (-73.75 mV), indicating that the ML dots remain its well dispersion during the assay. The UV-Vis spectrum and z-potential graph were shown below (Figure S6 (b) and (c)).

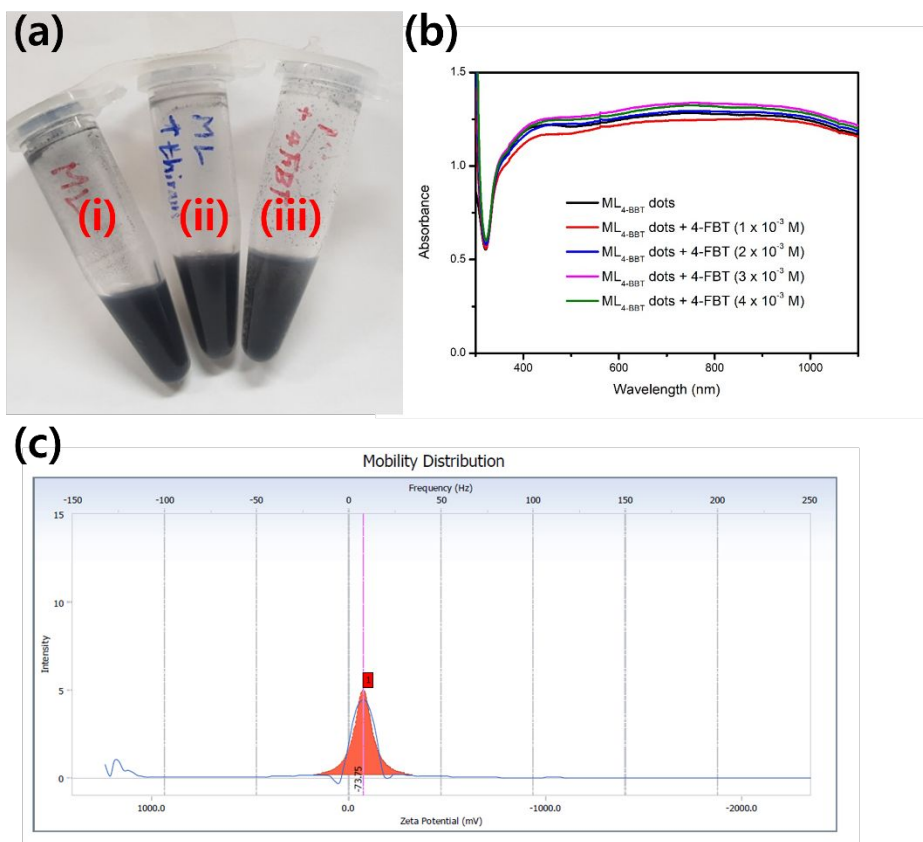


Figure S6. (a) The picture of ML_{4-BBT} dots after 5 hours of mixing with analytes. (i) ML_{4-BBT} dots, (ii) ML_{4-BBT} dots mixed with thiram and (iii) ML_{4-BBT} dots mixed with 4-FBT. (b) The UV-Vis spectrum of ML_{4-BBT} dots mixed with several concentration of analytes. (c) The z-potential of ML_{4-BBT} dots mixed with 4-FBT.