

Supporting Information

A Long-life Rechargeable Zn Air Battery Based on Binary Metal Carbide Armored by Nitrogen-doped Carbon

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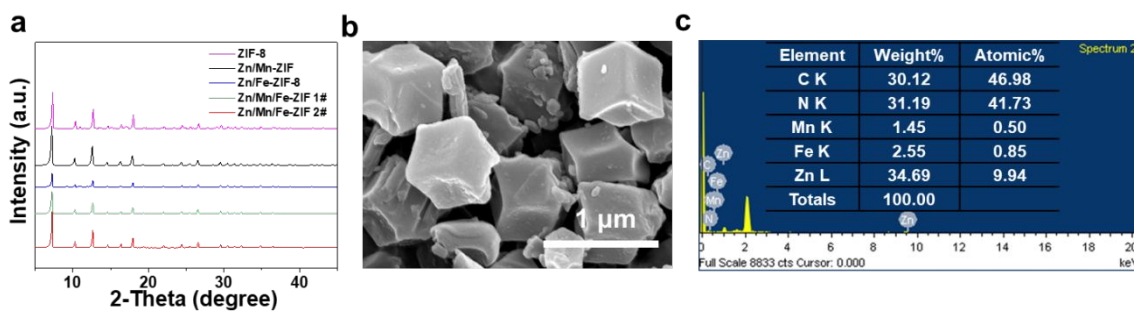


Figure S1. a) XRD patterns of ZIF-8, Zn/Mn-ZIF, Zn/Fe-ZIF, Zn/Mn/Fe-ZIF 1#, and Zn/Mn/Fe-ZIF 2#. b) SEM image of Zn/Fe-ZIF. c) EDS data for Zn/Mn/Fe-ZIF 2#.

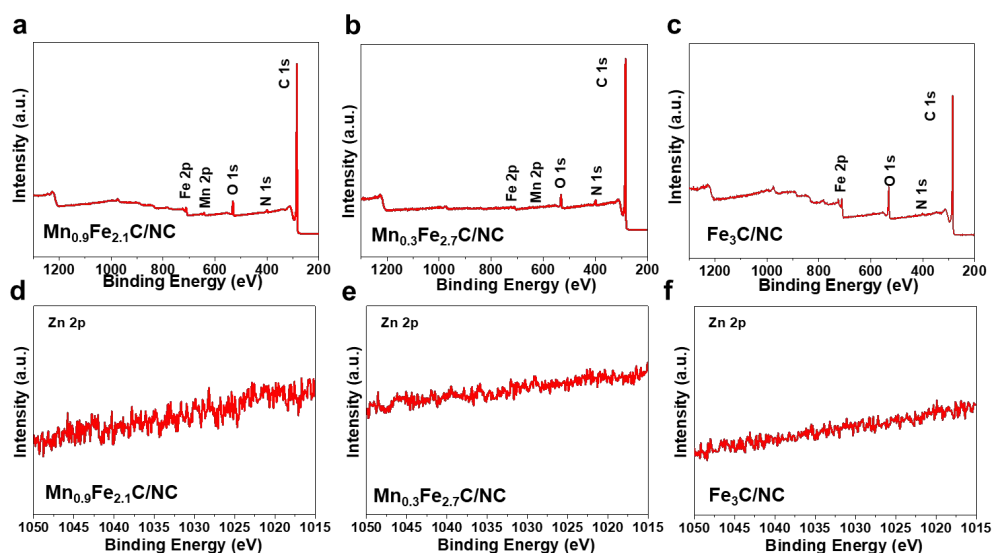


Figure S2. XPS survey and high-resolution spectra for (a, d) $\text{Mn}_{0.9}\text{Fe}_{2.1}\text{C}/\text{NC}$, (b, e) $\text{Mn}_{0.3}\text{Fe}_{2.7}\text{C}/\text{NC}$, and (c, f) $\text{Fe}_3\text{C}/\text{NC}$.

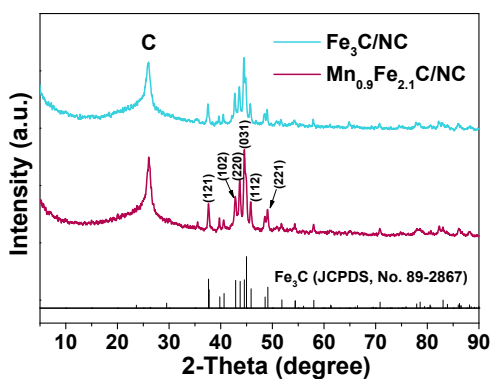


Figure S3. XRD patterns of $\text{Fe}_3\text{C}/\text{NC}$ and $\text{Mn}_{0.9}\text{Fe}_{2.1}\text{C}/\text{NC}$.

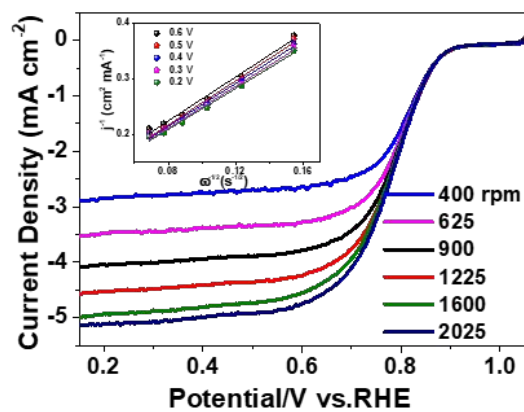


Figure S4. ORR polarization curves of Mn_xFe_{3-x}C/NC-2 at different rotation speeds. Inset shows the corresponding Koutecky–Levich plots.

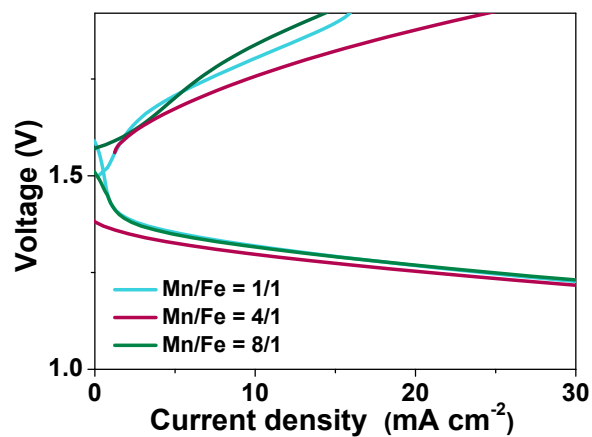


Figure S5. Charge-discharge polarization curves of Mn_xFe_{3-x}C/NC catalysts with Mn/Fe precursor ratios of 1/1, 4/1, and 8/1.

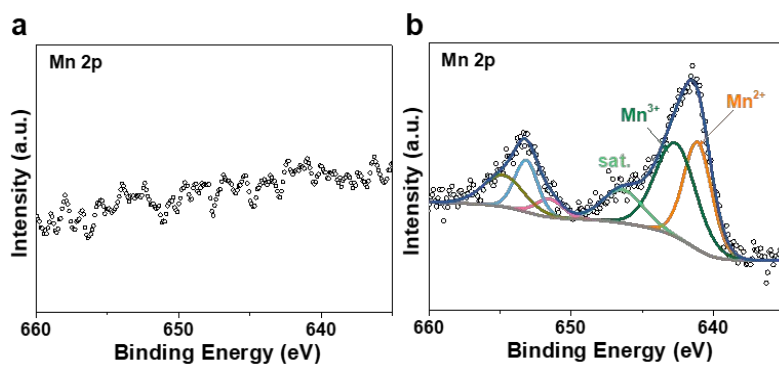


Figure S6. XPS spectra of Mn 2p for (a) $\text{Fe}_3\text{C}/\text{NC}$ and b) $\text{Mn}_{0.9}\text{Fe}_{2.1}\text{C}/\text{NC}$.

Table S1. Quantitative XPS analysis results for $\text{Mn}_x\text{Fe}_{3-x}\text{C}/\text{NC}$ and $\text{Fe}_3\text{C}/\text{NC}$.

	C (at.%)	Fe (at.%)	Mn (at.%)	O (at.%)	N (at.%)
$\text{Fe}_3\text{C}/\text{NC}$	86.12	2.67	-	9.72	1.49
$\text{Mn}_{0.3}\text{Fe}_{2.7}/\text{NC}$	91.65	0.44	0.05	4.57	3.29
$\text{Fe}_{0.9}\text{Mn}_{2.1}/\text{NC}$	91.52	0.87	0.38	5.17	2.06

Table S2. Atomic concentration of different N species in $\text{Mn}_x\text{Fe}_{3-x}\text{C}/\text{NC}$ and $\text{Fe}_3\text{C}/\text{NC}$.

	oxidized-N	pyrrolic-N	pyridinic-N	graphitic-N	M-N_x
Fe_3C	0.30	0.38	0.19	0.39	0.23
$\text{Mn}_{0.3}\text{Fe}_{2.7}/\text{NC}$	0.45	0.35	1.06	0.74	0.69
$\text{Mn}_{0.9}\text{Fe}_{2.1}/\text{NC}$	0.42	0.44	0.37	0.39	0.44