

# A Matrix Isolation ESR and Theoretical Study of ZnN

Thomas S. Hearne, Sally A. Yates, Duncan A. Wild, Allan J. McKinley

Chemistry, School of Molecular Sciences, The University of Western Australia, 35 Stirling Highway, Crawley, Western Australia 6009, Australia

**Table S1.** Observed and Simulated Line

Positions in Gauss for the  $xy_1$ ,  $oa$  and  $xy_3$

Lines of  $Zn^{15}N$  and the  $xy_1$  lines of  $^{67}ZnN$ .

$Zn^{15}N$			
frequency (MHz)	$M_I^a$	Obs. <sup>b</sup>	Calc. <sup>c</sup>
9714.181	1/2	1887.6(3)	1887.53
	-1/2	1896.6(3)	1896.51
9714.483	1/2	5035.5(3)	5035.45
	-1/2	5040.3(3)	5040.30
9714.379	1/2	5888.8(3)	5888.80
	-1/2	5897.6(3)	5897.76
$^{67}ZnN^d$			
$M_I^e$	$M_I^a$	Obs. <sup>b</sup>	Calc. <sup>c</sup>
5/2	1		1737.83
	0		1744.22
	-1	1750.7(3)	1750.59
3/2	1	1791.7(3)	1791.45
	0	1797.9(3)	1797.83
	-1	1804.1(3)	1804.23
1/2	1	1846.1(3)	1846.25
	0	1852.8(3)	1852.64

	-1	1859.1(3)	1859.02
-1/2	1	1902.1(3)	1902.23
	0	1908.8(3)	1908.61
	-1	1915.2(3)	1914.99
-3/2	1	1958.9(3)	1959.40
	0	1965.3(3)	1965.78
	-1	1971.9(3)	1972.16
-5/2	1	2017.9(3)	2017.71
	0	2024.2(3)	2024.09
	-1	2030.4(3)	2030.47

<sup>a</sup>The magnetic quantum number for the N (I=1/2) nucleus or the <sup>15</sup>N (I=1) nucleus.

<sup>b</sup>Peak positions as measured at the apex. The radical was trapped in a neon matrix at 4.3 K. Uncertainty in position is  $\pm 0.3$  G.

<sup>c</sup>Peak positions of a simulated spectra generated by direct diagonalisation of the spin Hamiltonian using the magnetic parameters found in Table 2.

<sup>d</sup>Microwave frequency of 9683.086 MHz

<sup>e</sup>The magnetic quantum number for the Zn-67 (I=5/2) nucleus.

**Table S2.** Comparison of Potential Energy Surface dissociation energy levels to experimentally determined energy level combinations for atomic Zn and N in Hartree.

Zn	N	S <sub>z</sub>	Experiment <sup>a</sup>	PES <sup>b</sup>
<b>1S</b>	<b>4S<sup>0</sup></b>	1.5	0 <sup>c</sup>	0 <sup>c</sup>
<b>1S</b>	<b>2D<sup>0</sup></b>	0.5	0.0876	0.100
<b>1S</b>	<b>2P<sup>0</sup></b>	0.5	0.1314	0.110
<b>3P<sup>0</sup></b>	<b>4S<sup>0</sup></b>	2.5	0.1484	0.144
<b>1P<sup>0</sup></b>	<b>4S<sup>0</sup></b>	1.5	0.2130	0.1687
<b>3P<sup>0</sup></b>	<b>2D<sup>0</sup></b>	1.5	0.23481	0.244

<sup>a</sup>Experimental values from Ref. 55 and 56

<sup>b</sup>Potential energy values are taken from the PES shown in Figure 5 at a bond length of 6.0 Å

<sup>c</sup>Set to 0