

Supporting Information

Interfacial Mass Transfer in Water - Toluene Systems

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Table S1. Starting compositions for mass transfer experiments where the transferring component was added to either the aqueous or both phases at $T = 298.15$ K and $p = 0.1$ MPa^a

Transferring component i	Injection phase	water-rich phase			toluene-rich phase		
		W_{water}	$W_{toluene}$	W_i	W_{water}	$W_{toluene}$	W_i
acetone	aqueous	0.886	0.0008	0.112	0.000	0.9996	0.000
	s	8		4	4		0
acetone	both	0.942	0.0005	0.057	0.000	0.9316	0.068
		4		1	3		1
ethanol	aqueous	0.884	0.0006	0.115	0.000	0.9996	0.000
	s	0		4	4		0
ethanol	both	0.940	0.0007	0.059	0.000	0.9323	0.067
		3		0	4		3
tetrahydrofuran	aqueous	0.885	0.0006	0.114	0.000	0.9994	0.000
	s	4		0	6		0
tetrahydrofuran	both	0.940	0.0005	0.059	0.000	0.9317	0.067
		0		5	6		7
acetonitrile	aqueous	0.887	0.0004	0.112	0.000	0.9995	0.000
	s	3		3	5		0
acetonitrile	both	0.940	0.0005	0.059	0.000	0.9330	0.066
		3		2	3		7

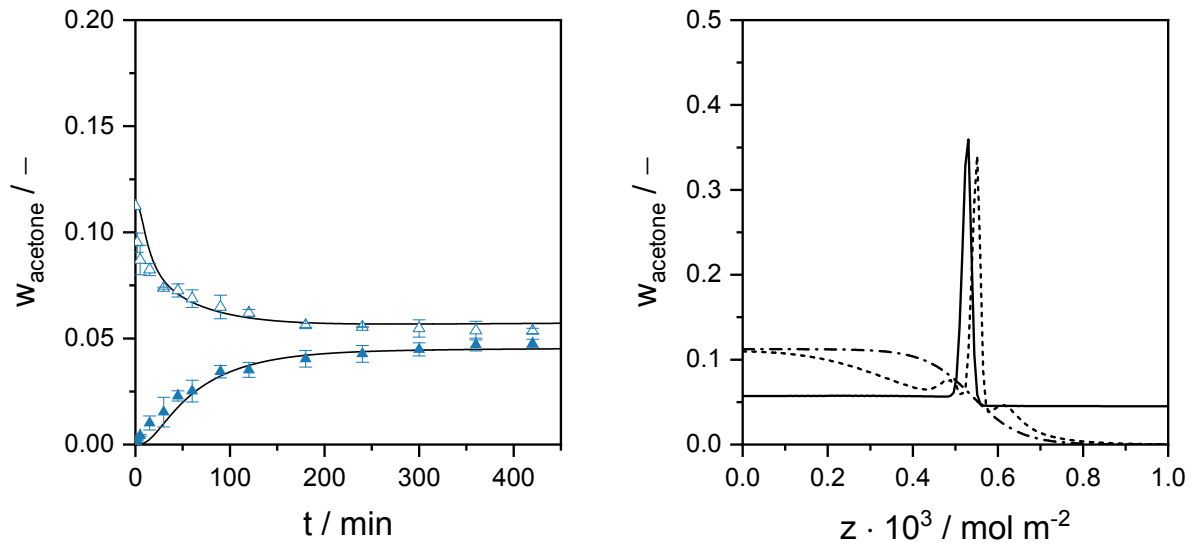


Figure S1. Interfacial mass transfer of acetone in the system water + toluene + acetone at $T = 298.15$ K. Acetone was injected in the aqueous phase at $t = 0$ min. Left: Development of the bulk phase composition over time: \blacktriangle , weight fraction of acetone in toluene-rich phase; \triangle , weight fraction of acetone in water-rich phase; lines are calculated by stationary DGT. Right: Concentration profile across the interface calculated by stationary DGT: dash-dotted line, $t = 0$ min; dotted line, $t = 5$ min; straight line, $t = 450$ min.

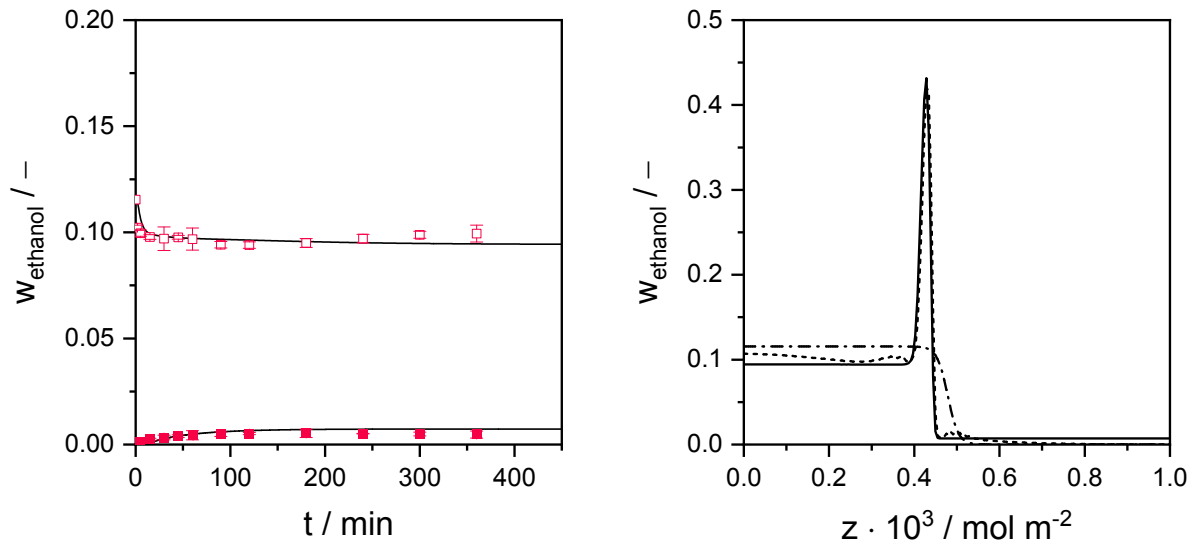


Figure S2. Interfacial mass transfer of ethanol in the system water + toluene + ethanol

$T = 298.15$ K. Left: Ethanol was injected in the aqueous phase at $t = 0$ min.

Development of the bulk phase composition over time: \blacksquare , weight fraction of ethanol in toluene-rich phase; \square , weight fraction of ethanol in water-rich phase; lines are calculated by stationary DGT. Right: Concentration profile across the interface calculated by stationary DGT: dash-dotted line, $t = 0$ min; dotted line, $t = 5$ min; straight line, $t = 450$ min.

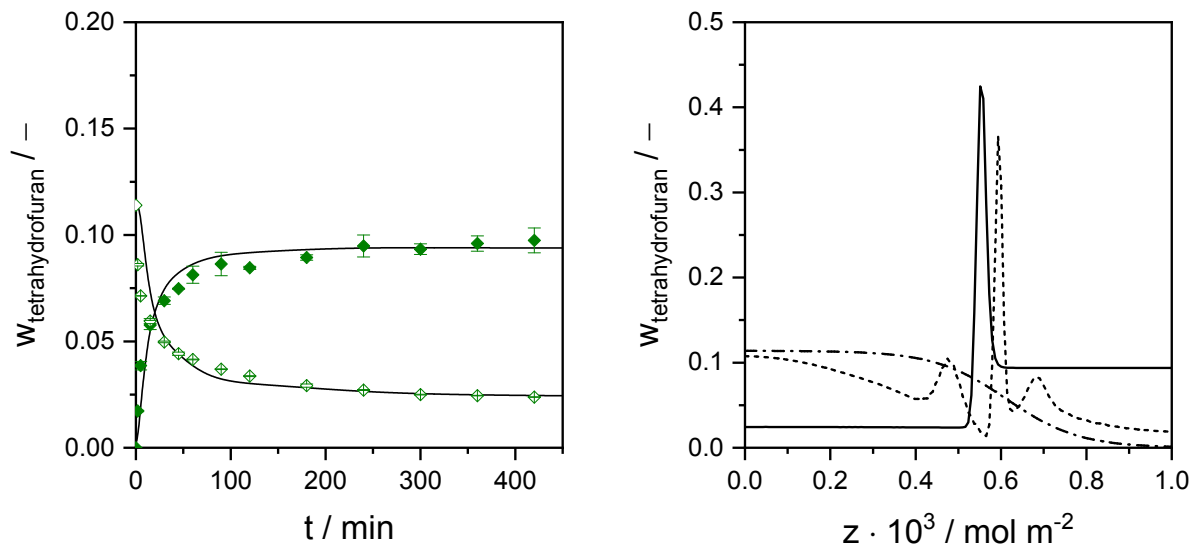


Figure S3. Interfacial mass transfer of tetrahydrofuran in the system water + toluene + tetrahydrofuran $T = 298.15$ K. Tetrahydrofuran was injected in the aqueous phase at $t = 0$ min. Left: Development of the bulk phase composition over time: \blacklozenge , weight fraction of tetrahydrofuran in toluene-rich phase; \diamond , weight fraction of tetrahydrofuran in water-rich phase; lines are calculated by stationary DGT. Right: Concentration profile across the interface calculated by stationary DGT: dash-dotted line, $t = 0$ min; dotted line, $t = 5$ min; straight line, $t = 450$ min.

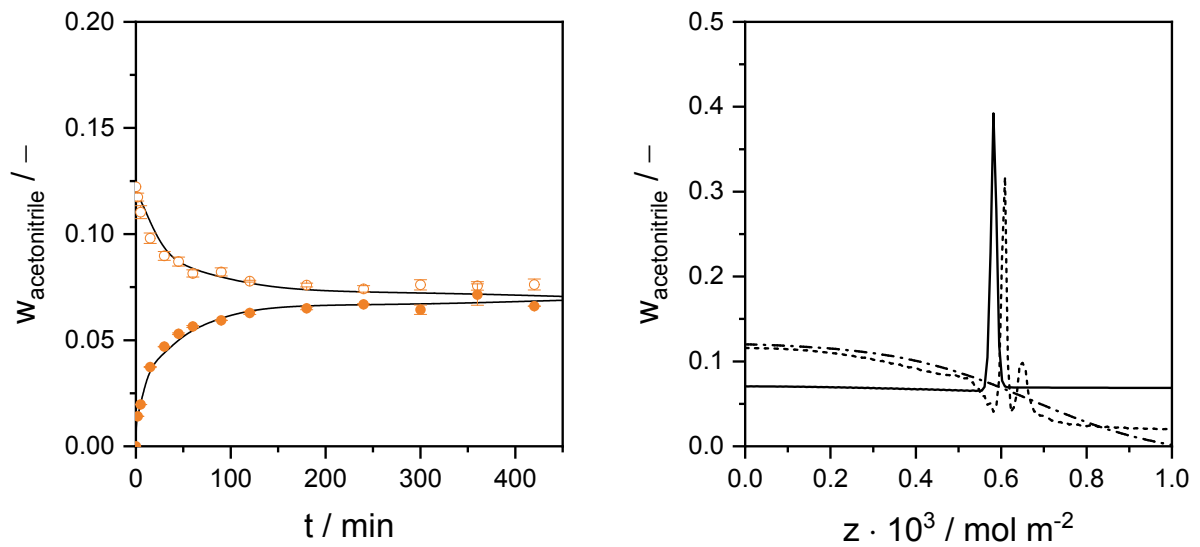


Figure S4. Interfacial mass transfer of acetonitrile in the system water + toluene + acetonitrile $T = 298.15$ K. Acetonitrile was injected in the aqueous phase at $t = 0$ min.

Left: Development of the bulk phase composition over time: ●, weight fraction of acetonitrile in toluene-rich phase; ○, weight fraction of acetonitrile in water-rich phase; lines are calculated by stationary DGT. Right: Concentration profile across the interface calculated by stationary DGT: dash-dotted line, $t = 0$ min; dotted line, $t = 5$ min; straight line, $t = 450$ min.

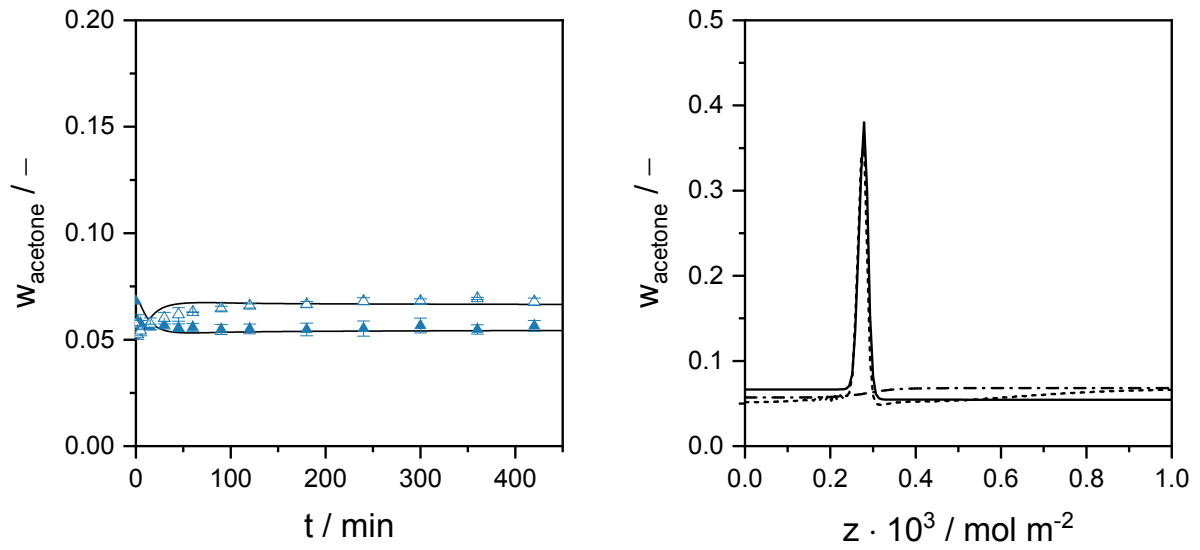


Figure S5. Interfacial mass transfer of acetone in the system water + toluene + acetone at $T = 298.15$ K. Acetone was injected into both phases at $t = 0$ min. Left: Development of the bulk phase composition over time: \blacktriangle , weight fraction of acetone in toluene-rich phase; \triangle , weight fraction of acetone in water-rich phase; lines are calculated by instationary DGT. Right: Concentration profile across the interface calculated by instationary DGT: dash-dotted line, $t = 0$ min; dotted line, $t = 5$ min; straight line, $t = 450$ min.

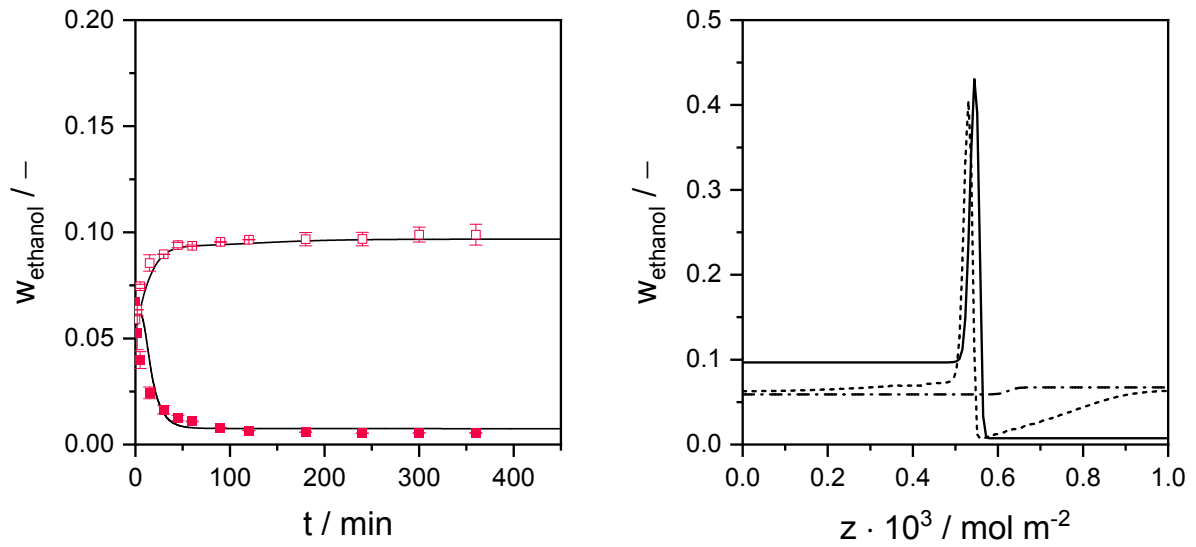


Figure S6. Interfacial mass transfer of ethanol in the system water + toluene + ethanol $T = 298.15$ K. Ethanol was injected into both phases at $t = 0$ min. Left: Development of the bulk phase composition over time: ■, weight fraction of ethanol in toluene-rich phase; □, weight fraction of ethanol in water-rich phase; lines are calculated by instationary DGT. Right: Concentration profile across the interface calculated by instationary DGT: dash-dotted line, $t = 0$ min; dotted line, $t = 5$ min; straight line, $t = 450$ min.

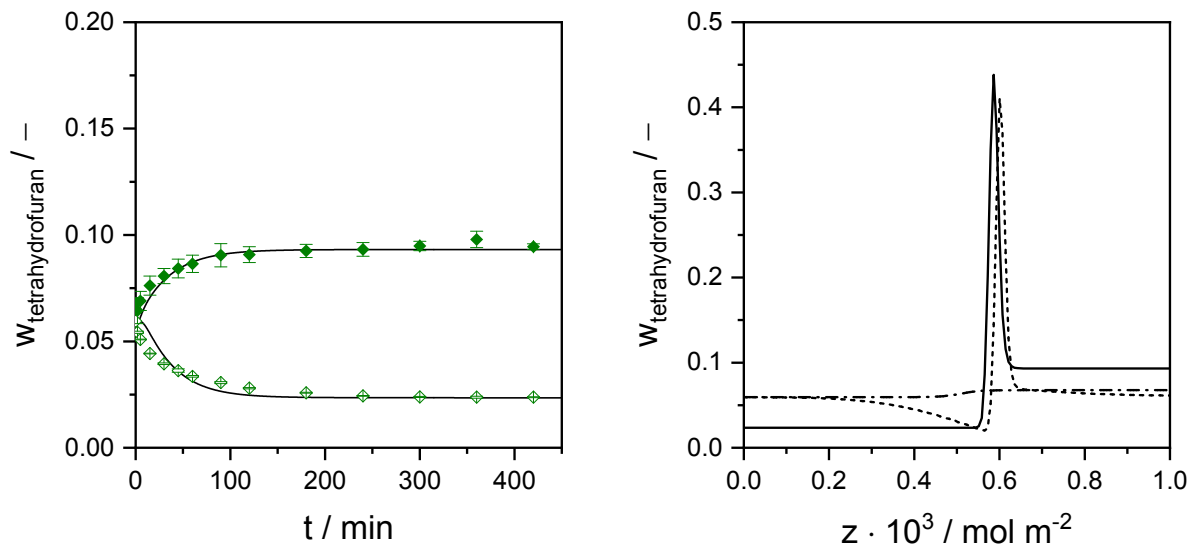


Figure S7. Interfacial mass transfer of tetrahydrofuran in the system water + toluene + tetrahydrofuran $T = 298.15$ K. Tetrahydrofuran was injected into both phases at $t = 0$ min. Left: Development of the bulk phase composition over time: \blacklozenge , weight fraction of tetrahydrofuran in toluene-rich phase; \blacklozenge , weight fraction of tetrahydrofuran in water-rich phase; lines are calculated by instationary DGT. Right: Concentration profile across the interface calculated by instationary DGT: dash-dotted line, $t = 0$ min; dotted line, $t = 5$ min; straight line, $t = 450$ min.

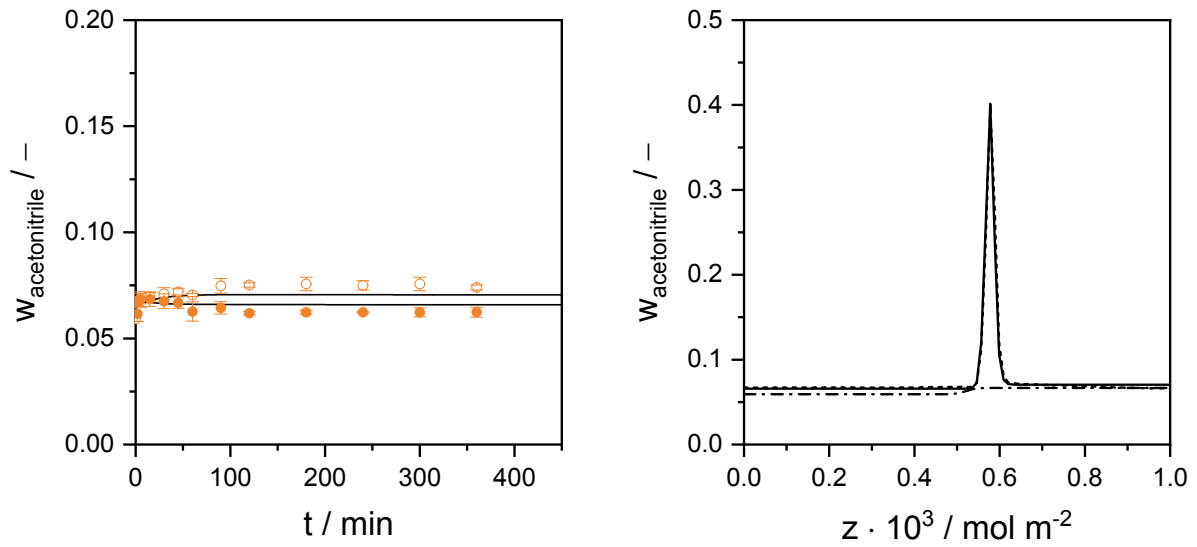


Figure S8. Interfacial mass transfer of acetonitrile in the system water + toluene + acetonitrile $T = 298.15$ K. Acetonitrile was injected into both phases at $t = 0$ min. Left: Development of the bulk phase composition over time: ●, weight fraction of acetonitrile in toluene-rich phase; ○, weight fraction of acetonitrile in water-rich phase; lines are calculated by instationary DGT. Right: Concentration profile across the interface calculated by instationary DGT: dash-dotted line, $t = 0$ min; dotted line, $t = 5$ min; straight line, $t = 450$ min.