

A multi-scale simulation of vinyl acetate systems applied in the industrial gas separation column

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Table S1 Vapor-liquid equilibrium data for the C₂H₂-VAc binary mixture at 101.325 kPa

<i>T</i> /K	<i>x</i> _{C₂H₂}	<i>y</i> _{C₂H₂}
188.30	1	1
208.15	0.426(0.004)	0.999(0.021)
218.15	0.315(0.006)	0.999(0.020)
223.15	0.256(0.003)	0.998(0.023)
243.15	0.175(0.005)	0.994(0.019)
263.15	0.087(0.007)	0.984(0.011)
273.15	0.061(0.007)	0.964(0.012)
283.15	0.054(0.002)	0.943(0.024)
293.15	0.042(0.006)	0.928(0.013)
303.15	0.033(0.006)	0.912(0.043)
345.85	0	0

Table S2 Vapor-liquid equilibrium data for the CO₂-VAc binary mixture at 101.325 kPa

$T(K)$	x_{CO_2}	y_{CO_2}
216.65	1	1
223.15	0.342 (0.008)	0.994 (0.033)
233.15	0.185 (0.006)	0.990 (0.061)
243.15	0.075 (0.006)	0.985 (0.053)
263.15	0.053 (0.004)	0.980 (0.056)
278.15	0.047 (0.006)	0.967 (0.071)
293.15	0.041 (0.003)	0.905 (0.023)
345.85	0	0

Table S3 Vapor-liquid equilibrium data for the CH₃CHO-VAc mixture at 101.325 kPa

T/K	x_{CH_3CHO}	y_{CH_3CHO}
293.00	1	1
295.00	0.950 (0.004)	0.992 (0.023)
300.00	0.775 (0.002)	0.959 (0.019)
305.00	0.625 (0.003)	0.918 (0.037)
310.00	0.475 (0.004)	0.854 (0.025)
315.00	0.375 (0.004)	0.788 (0.038)
320.00	0.300 (0.003)	0.720 (0.015)
325.00	0.225 (0.005)	0.625 (0.045)
330.00	0.150 (0.006)	0.490 (0.042)
335.00	0.102 (0.003)	0.368 (0.035)
340.00	0.050 (0.001)	0.209 (0.026)
345.85	0	0

Table S4 Binary interaction parameters of the UNIQUAC-RK model for the light components

Component i	C ₂ H ₂	CO ₂	C ₂ H ₂
Component j	CO ₂	CH ₃ CHO	CH ₃ CHO
b_{ij}	-58.95	78.54	230.49
b_{ji}	36.58	49.96	-108.32

Note: b_{ij} , b_{ji} are the binary interaction parameters of the UNIQUAC-RK model.

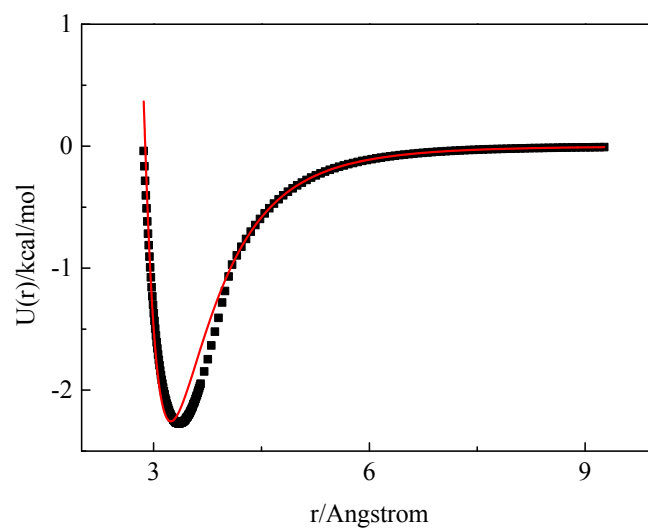


Fig. S1 L-J 12-6 potential energy fitting curve of the interaction energy between C_2H_2 and Ne^{20}
(black points: quantum chemistry calculation results, red line: fitting curve).