## Photocatalytic Dehydrogenative Lactonization of 2-Arylbenzoic Acids

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#### 1. General Remarks

TLC was performed on silica gel 60 F<sub>254</sub>, using aluminium plates and visualized by exposure to ultraviolet light. Flash chromatography was carried out on handpacked columns of silica gel 60 (230 – 400 mesh). Melting points are uncorrected. Infrared (IR) analysis was performed with a JASCO FT/IR 4100 spectrophotometer equipped with an ATR component; wavenumbers are given in cm<sup>-1</sup>. LRMS were performed in an AGILENT 6890N mass spectrometer coupled with a gas chromatographer (GC); the mobile phase was helium (2 mL/min); HP-1 column of 12 m was used; temperature program starts at 60 °C for 3 min, then up to 270 °C with a rate of 15 °C/min, and 10 min at 270 °C. HRMS analyses were carried out in an AGILENT 7200 using the Electron Impact (EI) mode at 70 eV by Q-TOF. <sup>1</sup>H NMR spectra were recorded at 300, 400 or 500 MHz for <sup>1</sup>H NMR and 75, 101 or 126 MHz for <sup>13</sup>C NMR, using CDCl<sub>3</sub> as the solvent and TMS as an internal Standard (0.00 ppm). The data is being reported as (s = singlet, d = doublet, t = triplet, m = multiplet or unresolved, br s = broad signal, coupling constant(s) in Hz, integration). <sup>13</sup>C NMR spectra were recorded with <sup>1</sup>Hdecoupling at 100 MHz and referenced to CDCl<sub>3</sub> at 77.16 ppm. DEPT-135 experiments were performed to assign CH, CH<sub>2</sub> and CH<sub>3</sub>.

#### 2. Set up of the Reaction



Photoreactor was made with flexible 2 m strip lights (30 microchips/m; 4.8 W/m) of multicolor LEDs (5122161 model), set up as an spiral of 12 cm of diameter as shown.

Reactions were performed in Pyrex tubes of 25 mL and 1.6 cm of diameter located 1-2 cm form LED chips.

Blue light was selected to perform the reactions.

A thermometer was placed in a parallel Pyrex tube containing the solvent of the reaction. The measured temperature in all reactions varied from 30 to 35 °C.

#### 3. Kinetic Isotopic Effect (KIE) Experiments

#### 3.1. Calibration Curve.

For this calibration, benzophenone was selected as internal standard (**IS**). In a vial, a known quantity of **IS** and coumarin **2a** were weight and dissolved in 2 mL of EtOAc. This procedure was repeated 4 times obtaining different molar **IS**:**2a** ratios. 1µL of each sample was injected into a GC and the areas of **IS** and **2a** were measured.

	IS	2a	IS	2a	Area	Area	mmol 2a	Area 2a
	(mg)	(mg)	(mmol)	(mmol)	IS	2a	/mmol <b>IS</b>	/Area <b>IS</b>
CAL 1	4,261	4,126	0,0234	0,0210	3983,6	3106,7	0,899	0,779
CAL 2	3,416	2,002	0,0187	0,0102	3402,5	1596,4	0,544	0,469
CAL 3	5,616	0,766	0,0308	0,0039	6297,2	705,2	0,126	0,112
CAL 4	4,066	1,020	0,0223	0,0052	4318	872,2	0,232	0,202
CAL 5	3,165	1,270	0,0173	0,0064	3651,2	1168,1	0,372	0,319

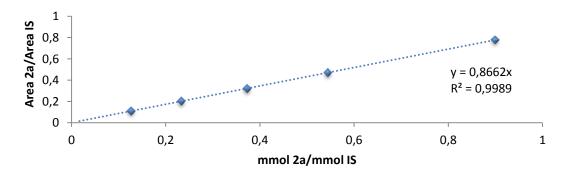


Figure S1: Calibration curve for coumarin 2a in GC

Therefore, the following relationship was obtained:

$$\frac{Area \ 2a}{Area \ IS} = 0.8662 \ x \ \frac{mmol \ 2a}{mmol \ IS}$$
 Eq S1

# 3.2. Measurement of the Kinetic Isotopic Effect (KIE) from Individual Experiments.

A reaction for the synthesis of benzocoumarin **2a** was performed. Into a 25 mL pyrex tube (1.5 cm of diameter) acid **1a** (39.6 mg, 0.2 mmol), photocatalyst [Acr-Mes]ClO<sub>4</sub> (2.1 mg, 2.5 mol%) and amonium persulfate (55 mg, 1.2 eq) were added and dissolved with a 4:1 mixture of MeCN:H<sub>2</sub>O (2 mL). The reaction mixture was stirred while irradiating with blue LEDs at 35 °C.

At the indicated time, 100  $\mu$ L of the reaction mixture was extracted and added to a vial with known quantity of **IS** and EtOAc (0.9 mL). This solution was dried over MgSO<sub>4</sub>, carefully filtered through cotton and introduced into a light-protected vial. 1  $\mu$ L of this solution was then injected into a GC and the areas of the **IS** and coumarin **2a** were measured (**Table S1**). By using **Eq S1**, the concentration of coumarin **2a** in the reaction vessel was calculated and represented  $\nu$ s the reaction time (**Figure S2**, blue line).

1a 2a

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t (min)	IS	Area <b>IS</b>	Area	area <b>2a</b>	2a produced	Yield 2a	[2a]
t (IIIII)	(mg)	Area 15	2a	/area <b>IS</b>	(mg)	(%)	(M)
0	-	-	-	-	0	0	0
10	2,03	4010,2	30,1	0,008	0,378	0,96	0,9664
17	1,99	3951,4	53,6	0,013	0,671	1,71	0,0017
33	2,02	3998,2	140,0	0,035	1,758	4,48	4,4864
54	2,05	3984,6	237,1	0,059	3,035	7,66	0,0077
68	2,10	4085,6	328,5	0,080	4,198	10,60	10,7097
82	2,13	4280,8	430,9	0,100	5,340	13,48	0,0136

Table S1: Use of GC and Eq 1 to calculate the concentration of 2a at initial stages of the reaction.

The same procedure described above was followed in order to synthesize benzocoumarin [ $\mathbf{D_4}$ ]- $\mathbf{2a}$  from acid [ $\mathbf{D_5}$ ]- $\mathbf{1a}$  (40.2 mg, 0.2 mmol). As before, by using the calibration curve, the concentration of benzocoumarin [ $\mathbf{D_4}$ ]- $\mathbf{2a}$  ( $\mathbf{Table}$  S2) in the reaction mixture was represented vs the reaction time (**Figure S2**, orange line).

t (min)	IS (mg)	Area IS	Area 2a-D <sub>4</sub>	area <b>2a-D</b> <sub>4</sub> /area <b>IS</b>	2a-D <sub>4</sub> produced (mg)	Yield <b>2a-D</b> <sub>4</sub> (%)	[2a-D <sub>4</sub> ] (M)
0	-	-	-	-	0	0	0
11	2,31	4511,1	31,2	0,0069	0,407	1,04	1,0133
19	2,36	4544,9	58,8	0,0129	0,778	1,98	0,0019
28	2,20	4270	85,7	0,0201	1,126	2,86	2,8006
45	2,21	4251,5	140,5	0,0330	1,862	4,73	0,0046
57	2,30	4500,2	190,1	0,0422	2,477	6,30	6,1625
74	2,36	5045,2	335,6	0,0665	4,003	10,18	0,0099

**Table S2**: Use of GC and Eq 1 to calculate the concentration of  $2a-D_4$  at initial stages of the reaction. GC response was proved to be the same for both, benzocoumarins 2a and  $2a-D_4$ .

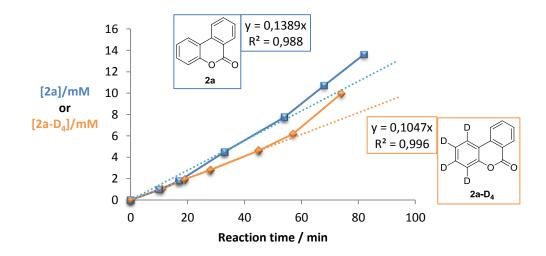
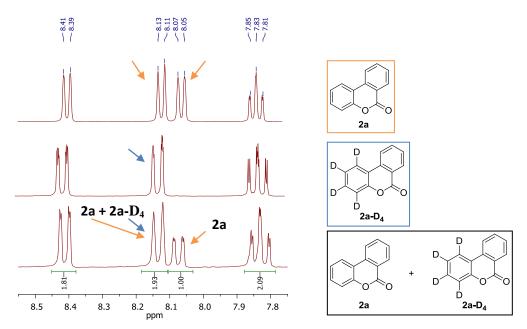


Figure S2. Initial rate of formation of 2a or 2a-D<sub>4</sub>.

For benzocoumarin **2a**: 
$$k_H = 0.1389 \text{ mM} \cdot \text{min}^{-1}$$
  
For benzocoumarin **2a**-**D**<sub>4</sub>:  $k_D = 0.1047 \text{ mM} \cdot \text{min}^{-1}$  }  $KIE = \frac{k_H}{k_D} = \frac{0.1389}{0.1047} = 1.3$  **Eq 2**

## 3.3. Measurement of the Kinetic Isotopic Effect (KIE) from Competition Experiments in an Equimolar Mixture of 1a and $1a-D_5$ .

Into a pyrex tube of 25 mL acid 1a (29.7 mg, 0.15 mmol), acid 1a- $D_5$  (30.5 mg, 0.15 mmol) and photocatalyst [Acr-Mes]ClO<sub>4</sub> (1.6 mg, 2.5 mol%) were dissolved in MeCN (1.2 mL). Then a solution of (NH<sub>4</sub>)<sub>2</sub>S<sub>2</sub>O<sub>8</sub> (41.1 mg, 1.2 eq,) in H<sub>2</sub>O (0.3 mL) was added and the reaction mixture was irradiated with blue LEDs lights and stirred at 35 °C. After 30 min, the reaction mixture was concentrated under reduced pressure to afford a yellow residue which was purified by column chromatography (95:5, Hexane/EtOAc). A mixture of benzocoumarins 2a and 2a- $D_4$  were isolated as a white solid (3 mg, ~ 5 % yield). <sup>1</sup>H-NMR, represented in **Figure S3**, showed quasi-equimolar mixture of benzocoumarins.



**Figure S3**. Part of the <sup>1</sup>H-NMR (300 MHz, CDCl<sub>3</sub>) obtained for pure benzocoumarins **2a**, **2a-D**<sub>4</sub>, and for the mixture obtained after purification from the reaction described above.

$$KIE = \frac{1.00}{1.93 - 1.00} = 1.1$$
 Eq 3

#### 4. Alternative pathway for 3'-methoxy-2-arylbenzoic acid 1i

For methoxy-substituted substrates an initial formation of the corresponding radical cation would be possible by SET [ $E^{red}$  (methoxybenzene) = 1.76 V vs SCE)<sup>1</sup>. In this case, the *5-exo-trig* cyclization would also be possible, followed by a SET oxidation and a [1,2]-shift to obtain the 3-methoxy isomer **2ic**. Remarkably, we did not observe the formation of this isomer, which was the main product observed in some cases by Martin's when iodine (III) catalysts were used.<sup>1</sup>

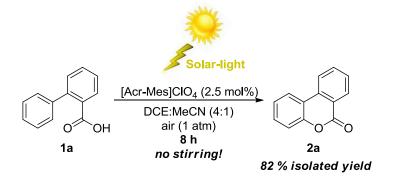
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<sup>&</sup>lt;sup>1</sup> Ohkubo, K.; Mizushima, K.; Iwata, R.; Fukuzumi, S. Chem. Sci. 2011, 2, 715.

#### 5. Reaction Performed Under Solar-Light Irradiation



Into a pyrex tube of 25 mL (1.5 cm of diameter) acid **1a** (39.6 mg, 0.2 mmol), and photocatalyst [Acr-Mes]ClO<sub>4</sub> (2.1 mg, 2.5 mol%) were dissolved in a 1:1 mixture of DCE:MeCN (2 mL). The tube was capped with a septa wearing a needle (air was allowed to enter the tube) and placed at the windowsill without stirring. After 8 h under solar-light irradiation,<sup>2</sup> the reaction mixture was concentrated under reduced pressure to afford a yellow residue which was purified by column chromatography (95:5 Hexane/EtOAc). The corresponding benzocoumarin **2a** was obtained as a white solid (32.2 mg, 82%). Characterization is described below.



**Scheme S3**: Formation of **2a** under solar-light irradiation and using air as terminal oxidant.

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<sup>&</sup>lt;sup>2</sup> The reaction was performed in Alicante on May 07, 2015. The reaction temperature was measured by introducing a thermometer into the reaction tube and varied from 25 to 35 °C.

#### 6. Syntheses of 2-Arylbenzoic Acids 1b – 1u

Procedure for the synthesis of PdCl<sub>2</sub>(PPh<sub>3</sub>)<sub>2</sub> catalyst<sup>3</sup>

Into an oven-dried *Schlenck* a suspension of PdCl<sub>2</sub> (53 mg, 0.3 mmol) in dry THF (3 mL) was prepared under Ar atmosphere. Then, LiCl (26 mg, 0.6 mmol) was added and the reaction mixture was stirred under Ar atmosphere for 5 minutes at room temperature. After this time, PPh<sub>3</sub> (157 mg, 0.60 mmol) was added to the resulted grey suspension. The reaction mixture was stirred under Ar atmosphere for at least 2 h at room temperature. The formed pale yellow suspension remained under Ar atmosphere and was directly used for the Suzuki Cross-coupling reactions.

#### 6.1. Synthesis of compounds 1b-1l.

• General procedure for the synthesis of biaryl 2-carboxylic acids 1b - 1l using Suzuki Cross-coupling reaction.<sup>4</sup>

To a solution of the desired arylboronic acid (2.1 mmol, 1.05 equiv.) in dry THF (4.5 mL) was added the prepared suspension of  $PdCl_2(PPh_3)_2$  in dry THF (0.5 mL, 0.1 M). The suspension was stirred and a  $Na_2CO_3$  aqueous solution (5 mL, 0.8 M) was added. The reaction mixture was stirred under Ar atmosphere for 2 minutes at room temperature. Finally, methyl 2-iodobenzoate (294  $\mu$ L, 2 mmol) was added and the reaction mixture was heated at 80 °C for 12 h under Ar atmosphere. The resultant mixture was extracted with EtOAc (3 × 10 mL). The combined layers were washed with brine (1 × 5 mL), dried over anhydrous MgSO<sub>4</sub>, filtered and concentrated under reduced pressure. The residue was purified by column chromatography (95:5 Hexane/EtOAc) to provide the corresponding methyl esters as pale yellow oils.

A NaOH aqueous solution (7 mL, 1 M) was added to a stirring mixture of the desired ester (1.8 mmol) in MeOH (7 mL) under Ar atmosphere. The reaction mixture was heated to 50 °C, observing complete hydrolysis by TLC after 12 h. The reaction mixture was concentrated under reduced pressure. The residue was quenched with 6 M HCl until

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<sup>&</sup>lt;sup>3</sup> Molander, G.A. Handbook of Reagents for Organic Synthesis. Catalyst Components for Coupling Reactions. Wiley **2008**, pp 332-343.

<sup>&</sup>lt;sup>4</sup> Wang, Y.; Gulevich, A.V.; Gevorgyan, V. Chem. Eur. J. 2013, 19, 15836.

pH < 3 and was extracted with EtOAc ( $3 \times 10$  mL). The combined layers were washed with brine ( $1 \times 5$  mL), dried over anhydrous MgSO<sub>4</sub>, filtered and concentrated under reduced pressure to afford the desired pure biaryl 2-carboxylic acids.

## 4'-fluoro-[1,1'-biphenyl]-2-carboxylic acid (1b)<sup>5</sup>:

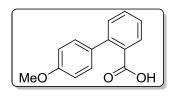
Compound **1b** was prepared from 4-fluorophenylboronic acid following the general procedure described above.

Compound **1b** was obtained as a white solid (361 mg, 1.67 mmol, 91% over two steps):  $R_{\rm f}$  0.37 (7:3 Hexane/(3:1:0.08) EtOAc/EtOH/AcOH<sup>6</sup>); <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  7.96 (dd, J = 7.8, 1.1 Hz, 1H), 7.57 (td, J = 7.6, 1.4 Hz, 1H), 7.43 (td, J =

7.6, 1.4 Hz, 1H), 7.33 (dd, J = 7.8, 1.1 Hz, 1H), 7.31 - 7.25 (m, 2H), 7.13 - 7.01 (m, 2H); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>)  $\delta$  173.4 (C), 162.5 (d, J = 244.5 Hz, C), 142.6 (C), 137.2 (d, J = 3.0 Hz, C), 132.4 (CH), 131.4 (CH), 131.0 (CH), 130.2 (d, J = 7.5 Hz, 2 × CH), 129.2 (C), 127.5 (CH), 115.2 (d, J = 21.8 Hz, 2 × CH).

## 4'-methoxy-[1,1'-biphenyl]-2-carboxylic acid (1c)<sup>7</sup>:

Compound **1c** was prepared from 4-methoxyphenylboronic acid following the general procedure described above, but in a larger scale, using 5 mmol of methyl 2-iodobenzoate.



Compound **1c** was obtained as a white solid (959 mg, 4.2 mmol, 85% over two steps):  $R_{\rm f}$  0.34 (7:3 Hexane/(3:1:0.08) EtOAc/EtOH/AcOH); <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  7.92 (dd, J = 7.8, 0.9 Hz, 1H), 7.54 (td, J = 7.5, 1.5 Hz, 1H), 7.40

(dd, J = 7.6, 1.4 Hz, 1H), 7.35 (dd, J = 7.7, 1.2 Hz, 1H), 7.29 - 7.24 (m, 2H), 6.96 - 6.89 (m, 2H), 3.84 (s, 3H); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>)  $\delta$  173.8 (C), 159.2 (C), 143.0 (C), 133.4 (C), 132.2 (CH), 131.3 (CH), 130.8 (CH), 129.8 (2 × CH), 129.4 (C), 127.0 (CH), 113.7 (2 × CH), 55.4 (CH<sub>3</sub>).

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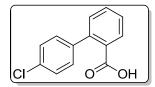
<sup>&</sup>lt;sup>5</sup> Li, Y.; Ding, Y.-J.; Wang, J.-Y.; Su, Y.-M.; Wang, X.-S. Org. Lett. 2013, 15, 2574.

<sup>&</sup>lt;sup>6</sup> Taygerly, J. P.; Miller, L. M.; Yee, A.; Peterson, E. A. *Green Chem.* **2012**, *14*, 3020.

<sup>&</sup>lt;sup>7</sup> Wang, Y.; Gulevich, A.V.; Gevorgyan, V. Chem. Eur. J. **2013**, 19, 15836.

## 4'-chloro-[1,1'-biphenyl]-2-carboxylic acid (1d)<sup>5</sup>:

Compound **1d** was prepared from 4-chlorophenylboronic acid following the general procedure described above, but in a larger scale, using 5 mmol of methyl 2-iodobenzoate.

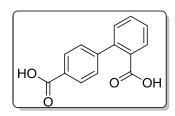


Compound **1d** was obtained as a white solid (977 mg, 4.2 mmol, 84% over two steps):  $R_{\rm f}$  0.53 (7:3 Hexane/(3:1:0.08) EtOAc/EtOH/AcOH); <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  7.97 (d, J = 7.7 Hz, 1H), 7.57 (t, J = 7.5 Hz, 1H), 7.44 (t, J = 7.6 Hz,

1H), 7.34 (t, J = 8.5 Hz, 3H), 7.29 – 7.23 (m, 2H); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>)  $\delta$  173.0 (C), 142.5 (C), 139.7 (C), 133.6 (C), 132.5 (CH), 131.3 (CH), 130.0 (2 × CH), 129.1 (C), 128.4 (2 × CH), 127.7 (CH), 77.6 (CH).

#### [1,1'-biphenyl]-2,4'-dicarboxylic acid (1e):

Compound **1e** was prepared from (4-(ethoxycarbonyl)phenyl)boronic acid following the general procedure described above.

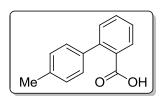


Compound **1e** was obtained as a white solid (193.8 mg, 0.8 mmol, 40% over two steps):  $R_{\rm f}$  0.16 (7:3 Hexane/(3:1:0.08) EtOAc/EtOH/AcOH); <sup>1</sup>H NMR (400 MHz, DMSO- $d_6$ )  $\delta$  12.92 (br s, 2H), 7.97 (d, J = 8.4 Hz, 2H), 7.80 (dd, J = 7.7, 1.2 Hz, 1H), 7.10 (td, J = 7.5, 1.4 Hz, 1H), 7.50 (td, J = 7.6,

1.2 Hz, 1H) 7.44 (d, J = 8.4 Hz, 2H), 7.41 (dd, J = 7.6, 0.9 Hz, 1H); <sup>13</sup>C NMR (75 MHz, DMSO- $d_6$ )  $\delta$  169.2 (C), 167.3 (C), 145.5 (C), 140.4 (C), 132.0 (C), 131.2 (CH), 130.5 (CH), 129.5 (C + CH). 129.1 (2 x CH), 128.6 (2 x CH), 127.8 (CH).

## 4'-methyl-[1,1'-biphenyl]-2-carboxylic acid (1f)<sup>7</sup>:

Compound **1f** was prepared from 4-methylphenylboronic acid following the general procedure described above.

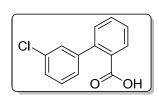


Compound **1f** was obtained as a white solid (350 mg, 1.65 mmol, 87% over two steps):  $R_{\rm f}$  0.56 (7:3 Hexane/(3:1:0.08) EtOAc/EtOH/AcOH); <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  7.92 (dd, J = 7.8, 1.1 Hz, 1H), 7.53 (td, J = 7.6, 1.4 Hz, 1H), 7.39

(td, J = 7.6, 1.4 Hz, 1H), 7.35 (dd, J = 7.7, 0.9 Hz, 1H), 7.26 - 7.14 (m, 4H), 2.38, (s, 3H); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>)  $\delta$  174.0 (C), 143.4 (C), 138.2 (C), 137.2 (C), 132.2 (CH), 131.4 (CH), 130.8 (CH), 129.5 (C), 129.0 (2 × CH), 128.5(2 × CH), 127.1 (CH), 21.4 (CH<sub>3</sub>).

## 3'-chloro-[1,1'-biphenyl]-2-carboxylic acid (1g)<sup>5</sup>:

Compound **1g** was prepared from 3-chlorophenylboronic acid following the general procedure described above, using a 1.82 mmol scale of methyl 2-iodobenzoate.

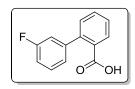


Compound **1g** was obtained as a white solid (123.1 mg, 0.53 mmol, 29% over two steps):  $R_{\rm f}$  0.38 (7:3 Hexane/(3:1:0.08) EtOAc/EtOH/AcOH); <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  7.99 (ddd, J = 7.9, 1.5, 0.5 Hz, 1H), 7.58 (td, J = 7.6, 1.5 Hz, 1H),

7.46 (td, J = 7.6, 1.4 Hz, 1H), 7.37 – 7.30 (m, 4H), 7.23 – 7.17 (m, 1H); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>)  $\delta$  171.9 (C), 142.9 (C), 142.1 (C), 133.8 (C), 132.3 (CH), 131.3 (CH), 130.9 (CH), 129.2 (CH), 128.9 (C), 128.5 (CH), 127.8 (CH), 127.4 (CH), 126.8 (CH).

## 3'-fluoro-[1,1'-biphenyl]-2-carboxylic acid (1h)<sup>5</sup>:

Compound **1h** was prepared from 3-fluorophenylboronic acid following the general procedure described above.



Compound **1h** was obtained as a white solid (318 mg, 1.47 mmol, 77% over two steps):  $R_{\rm f}$  0.53 (7:3 Hexane/(3:1:0.08) EtOAc/EtOH/AcOH); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  7.97 (dd, J = 7.8, 1.1 Hz, 1H), 7.57 (td, J = 7.6, 1.4 Hz, 1H), 7.44 (td, J = 7.6,

1.3 Hz, 1H), 7.34 (ddd, J = 7.7, 3.3, 1.4 Hz, 2H), 7.06 (dd, J = 20.4, 8.6 Hz, 3H); <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>)  $\delta$  173.2 (C), 162.6 (d, J = 246.4 Hz, C), 143.4 (d, J = 8.1 Hz, C), 142.4 (C), 142.3 (C), 132.4 (CH), 131.2 (CH), 131.0 (CH), 129.6 (d, J = 4.0 Hz, CH), 127.8 (CH), 124.4 (d, J = 3.0 Hz, CH), 115.7 (d, J = 22.2 Hz, CH), 114.4 (d, J = 21.2 Hz, CH).

## 3'-methoxy-[1,1'-biphenyl]-2-carboxylic acid (1i)<sup>5</sup>:

Compound **1i** was prepared from 3-methoxyphenylboronic acid following the general procedure described above.

Compound **1i** was obtained as a white solid (367 mg, 1.61 mmol, 85% over two steps):  $R_{\rm f}$  0.28 (7:3 Hexane/(3:1:0.08) EtOAc/EtOH/AcOH); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  7.92 (dd, J = 7.8, 1.1 Hz, 1H), 7.54 (td, J = 7.6, 1.4 Hz, 1H), 7.41

(td, J = 7.6, 1.3 Hz, 1H), 7.37 (dd, J = 7.7, 0.9 Hz, 1H), 7.31 – 7.23 (m, 1H), 6.94 - 6.86 (m, 3H), 3.80 (s, 3H); <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>)  $\delta$  173.8 (C), 159.4 (C), 143.2 (C),

142.5 (C), 132.2 (CH), 131.2 (CH), 130.7 (CH), 129.6 (C), 129.2 (CH), 127.4 (CH), 121.2 (CH), 114.3 (CH), 113.1 (CH), 55.4 (CH<sub>3</sub>).

## 2-(naphtalen-2-yl)benzoic acid (1j)<sup>5</sup>:

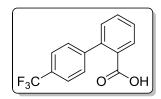
Compound **1j** was prepared from 2-naphtylboronic acid following the general procedure described above, using a 1.3 mmol scale of methyl 2-iodobenzoate.

Compound **1j** was obtained as a white solid (187 mg, 0.75 mmol, 58% over two steps):  $R_f$  0.43 (7:3 Hexane/(3:1:0.08) EtOAc/EtOH/AcOH); <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  7.98 (dd, J = 8.1, 1.4 Hz, 1H), 7.88 - 7.81 (m, 3H), 7.79 (s, 1H), 7.59 (td,

J = 7.7, 1.4 Hz, 1H), 7.52 - 7.42 (m, 5H); <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>)  $\delta$  172.4 (C), 143.5 (C), 138.9 (C), 133.4 (C), 132.7 (C), 132.3 (CH), 131.7 (CH), 131.0 (CH), 129.4 (C), 128.3 (CH), 127.9 (CH), 127.6 (CH), 127.3 (CH), 127.1 (CH), 126.3 (CH), 126.2 (CH), 126.0 (CH).

## 4'-(trifluoromethyl)-[1,1'-biphenyl]-2-carboxylic acid (1k)<sup>8</sup>:

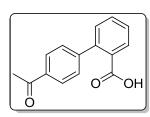
Compound **1k** was prepared from 4-(trifluoromethyl)phenylboronic acid following the general procedure described above, using a 1.3 mmol scale of methyl 2-iodobenzoate.



Compound **1k** was obtained as a white solid (72 mg, 0.27 mmol, 44% over two steps):  $R_{\rm f}$  0.57 (7:3 Hexane/(3:1:0.08) EtOAc/EtOH/AcOH); <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  8.03 (dd, J = 7.8, 1.1 Hz, 1H), 7.64 (d, J = 7.9 Hz, 2H), 7.59 (dd, J = 1.0 Hz, 2H), 7.50 (dd, J

7.6, 1.5 Hz, 1H), 7.48 (td, J = 7.6, 1.3 Hz, 1H), 7.43 (d, J = 8.0 Hz, 2H), 7.33 (dd, J = 7.6, 1.0 Hz, 1H); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>)  $\delta$  171.7 (C), 145.0 (C), 142.5 (C), 132.6, 131.2 (2 × CH), 129.7 (q, J = 32.7 Hz, C), 129.0 (2 × CH), 128.1 (2 × CH), 128.0 (C), 125.1 (q, J = 3.5 Hz, 2 × CH), 124.2 (q, J = 272.0 Hz, C).

## 4'-acetyl-[1,1'-biphenyl]-2-carboxylic acid (11)<sup>8</sup>:



Compound **11** was prepared from 4-(acetyl)phenylboronic acid following the general procedure described above, using a 1.3 mmol scale of methyl 2-iodobenzoate. Compound **11** was obtained as a brown pale solid (210 mg, 0.92 mmol, 71% over two steps):  $R_{\rm f}$  0.39 (7:3 Hexane/(3:1:0.08)

<sup>&</sup>lt;sup>8</sup> Day, J.-J.; Xu, W.- T.; Wu, Y.- D.; Zhang, W. – M.; Gong, Y.; He, X.-P.; Zhang, X. – Q.; Xu, H. J. *J. Org. Chem.* **2015**, 2, 911.

EtOAc/EtOH/AcOH); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  7.99 (dd, J = 11.3, 8.2 Hz, 3H), 7.60 (td, J = 7.60, 1.4 Hz, 1H), 7.47 (td, J = 7.7, 1.3 Hz, 1H), 7.44 (dd, J = 13.8, 4.9 Hz, 2H), 7.35 (dd, J = 7.6, 1.0 Hz, 1H), 2.64 (s, 3H); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>)  $\delta$  198.2 (C), 172.6 (C), 146.4 (C), 142.6 (C), 136.0 (C), 132.5 (CH), 131.1 (2 x CH), 129.2 (CH), 128.9 (C), 128.3 (2 x CH), 128.1 (2 x CH), 26.8 (CH<sub>3</sub>).

#### 6.2. Procedure for the synthesis of acid 1m.

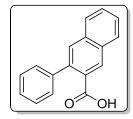
## Methyl 3-(((trifluoromethyl)sulfonyl)oxy)-2-naphthoate (3a)<sup>5</sup>:

To a solution of 3-hydroxi-2-naphtoic acid (564 mg, 3 mmol) in dry DMF (30 mL) was added  $Cs_2CO_3$  (489 mg, 1.5 mmol) under Ar atmosphere, followed by MeI (196  $\mu$ L, 3.15 mmol). The mixture reaction was stirred for 90 minutes under Ar atmosphere

at room temperature. At this time, the reaction was distributed between phases using a  $EtOAc/H_2O$  mixture (30 mL:30 mL). The aqueous layer was extracted with EtOAc (4 × 20 mL). The combined layers were washed with  $H_2O$  (2 × 10 mL),  $NaHCO_3$  (aq. sat.) (2 × 10 mL) and brine (1 × 5 mL), dried over anhydrous  $MgSO_4$ , filtered and concentrated under reduced pressure to afford the desired hydroxiester intermediate.

The hydroxiester afforded in the previously step was dissolved in dry CH<sub>2</sub>Cl<sub>2</sub> (30 mL). This reaction mixture was cooled to -78 °C under Ar atmosphere and Et<sub>3</sub>N (6.6 mmol, 921  $\mu$ L) followed by Tf<sub>2</sub>O (4.5 mmol, 760  $\mu$ L) were both added dropwise. After complete addition, the reaction mixture was stirred for 30 minutes at -78 °C, and allowed to reach room temperature for 12 hours under Ar atmosphere. The reaction mixture was extracted with diethyl ether (4 × 25 mL). The combined layers was washed with HCl 1M (4 × 25 mL) and brine (1 × 5 mL). The organic layer was dried over anhydrous MgSO<sub>4</sub>, filtered and concentrated under reduced pressure affording the reaction crude. This residue was purified by column chromatography (95:5 hexane/EtOAc) to provide the desired ester **3a** as a pale yellow solid (480 mg, 1.45 mmol, 48 %): CG:  $R_T$  15.06 min; LRMS (EI): m/z (%) = 334 (M<sup>+</sup>, 100), 303 (20), 201 (67), 173 (32), 143 (84), 114 (38), 69 (14).

## 3-phenyl-2-naphthoic acid (1m)<sup>5</sup>:



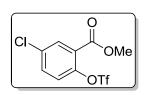
The desired biaryl 2-carboxilic acid 1m was obtained following the general procedure described for the synthesis of acids 1b - 1l, from 1.27 mmol of ester 3a instead of using methyl 2-iodobenzoate. Compound 1m was obtained as a white solid (132 mg, 0.53 mmol, 52% over two steps):  $R_f$  0.28 (7:3)

Hexane/(3:1:0.08) EtOAc/EtOH/AcOH);  $^{1}$ H NMR (300 MHz, CDCl<sub>3</sub>) δ 8.54 (s, 1H), 7.95 (d, J = 8.3 Hz, 1H), 7.87 (d, J = 8.3 Hz, 1H), 7.81 (s, 1H), 7.67 – 7.49 (m, 2H), 7.49 – 7.32 (m, 5H);  $^{13}$ C NMR (75 MHz, CDCl<sub>3</sub>) δ 172.6 (C), 141.4 (C), 139.3 (C) 134.9 (CH), 132.4 (C), 131.7 (CH), 130.3 (2 × C), 130.0 (CH), 128.9 (CH), 128.8 (2 × CH), 128.2 (2 × CH), 127.9 (CH), 127.3 (CH), 126.9 (CH).

#### 6.3. Procedure for the synthesis of acid 1n and 1o.

$$\begin{array}{c} \text{PhB}(\text{OH})_2 \\ \text{(1.2 equiv.)} \\ \text{"Suzuki"} \\ \text{HO} \\ \text{O} \\ \text{OH} \\ \end{array}$$

## Methyl 5-chloro-2-(((trifluoromethyl)sulfonyl)oxy)benzoate (3b):<sup>5</sup>



Methylester **3b** was prepared from 5-chloro-2-hydroxibenzoic acid (704.33 mg, 4 mmol) following the same procedure described for the synthesis of compound **3a**, obtaining product **3b** after column chromatography (95:5 Hexane/EtOAc) as a

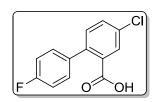
colorless liquid (815 mg, 2.56 mmol, 64 %): GC:  $R_T$  15.67 min; LRMS (EI): m/z (%) = 318/320 (54/21, M<sup>+</sup>), 287/289 (35/12), 223/225 (100/35), 185/187 (52/18), 169/171 (18/6), 154/156 (42/30), 129/131 (36/12), 63 (40).

## 4-chloro-[1, 1'-biphenyl]-2-carboxylic acid (1n):<sup>5</sup>

Compound **1n** was prepared from 1.25 mmol of methylester **3b** instead of using methyl 2-iodobenzoate, following the general procedure for the synthesis of biaryl 2-carboxylic acids **1b** - **1l**. Compound **1n** was obtained as a white solid (242 mg, 1.04 mmol,

84% over two steps):  $R_{\rm f}$  0.53 (7:3 Hexane/(3:1:0.08) EtOAc/EtOH/AcOH); <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  7.92 (d, J = 2.3 Hz, 1H), 7.53 (dd, J = 8.3, 2.3 Hz, 1H), 7.41 – 7.35 (m, 3H), 7.34 – 7.27 (m, 3H); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>)  $\delta$  172.4 (C), 141.9 (C), 139.9 (C), 133.5 (C), 132.7 (CH), 132.2 (CH), 130.8 (C), 130.7 (CH) 128.5 (2 × CH), 128.3 (2 × CH), 127.8 (CH).

#### 4-chloro-4'-fluoro-[1,1'-biphenyl]-2-carboxylic acid (10):



Compound **10** was prepared from 1.25 mmol of methylester **3b** and 1.5 mmol of 4-fluorophenylboronic acid, following the typical procedure for the synthesis of biaryl 2-carboxylic acids **1b** - **1l**. Compound **1o** was obtained as a white solid (122 mg,

0.49 mmol, 45% over two steps):  $R_{\rm f}$  0.31 (7:3 Hexane/(3:1:0.08) EtOAc/EtOH/AcOH); <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  7.94 (d, J = 2.3 Hz, 1H), 7.53 (dd, J = 8.3, 2.3 Hz, 1H), 7.31 - 7.18 (m, 3H), 7.14 - 7.00 (m, 2H); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>)  $\delta$  171.6 (C), 162.7 (d, J = 246 Hz, C), 141.1 (C), 136.0 (d, J = 3.8 Hz, C), 133.7 (C), 132.7 (CH), 132.4 (CH) 130.9 (CH), 130.5 (C), 130.2 (d, J = 8.3 Hz, 2 ×CH), 115.3 (d, J = 21 Hz, 2 × CH).

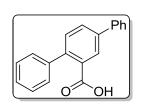
#### 6.4. Procedure for the synthesis of acid 1p.

## Methyl 5-bromo-2-(((trifluoromethyl)sulfonyl)oxy) benzoate (3c):<sup>5</sup>

Methylester **3c** was prepared from 5-chloro-2-hydroxibenzoic acid (836 mg, 2.3 mmol) following the same procedure described for the synthesis of methylester **3a**, obtaining product **3c** as pale yellow oil (856 mg, 2.36 mmol, 79 %) after column

chromatography (97:3, Hexane/EtOAc): GC:  $R_T$  12.51 min; LRMS (EI): m/z (%) = 364/363 (84/80,  $M^+$ ), 333/332 (32/31), 269/268 (100/96).

## [1, 1':4', 1"-terphenyl]-2'-carboxylic acid (1p):<sup>8</sup>



Compound **1p** was prepared from 2.3 mmol of methylester **3c** and 3.8 mmol (2 eq) of phenylboronic acid, following the general procedure for the synthesis of biaryl 2-carboxylic acids **1b** - **1l**. Compound **1p** was obtained as a white solid (466.3 mg, 1.7 mmol,

74%):  $R_f$ : 0.31 (7:3 Hexane/(3:1:0.08) EtOAc/EtOH/AcOH); <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  8.19 (d, J = 1.9 Hz, 1H), 7.79 (dd, J = 8.0, 2.0 Hz, 1H), 7.65 (dd, J = 8.3, 1.3 Hz, 2H), 7.54 – 7.34 (m, 9H); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>)  $\delta$  173.3 (C), 142.3 (C), 140.8 (C), 140.4 (C), 139.7 (C), 131.9 (CH), 130.7 (CH), 129.8 (C), 129.5 (CH), 129.1 (2 × CH), 128.7 (2 × CH), 128.3 (2 × CH), 128.0 (CH), 127.6 (CH), 127.2 (2 × CH).

## 6.5. Procedure for synthesis of acids 1q-1s<sup>5,9</sup>:

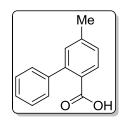
• General procedure for the synthesis of biaryl 2-carboxylic acids 1q - 1s

Into an oven dried pyrex tube, 2-bromo-4-methylbenzoic acid (329 mg, 1.5 mmol), the desired phenylboronic acid (1.8 mmol) and LiOH·H<sub>2</sub>O (141 mg, 3.3 mmol) were added under Ar atmosphere. Then, a NMP:H<sub>2</sub>O (4.5 mL:4.5 mL) desgassed mixture was added, followed by  $Pd_2(dba)_3$ -CHCl<sub>3</sub> (21 mg, 1.5 mol%) The reaction mixture was stirred for 24 hours under Ar atmosphere at 70 °C until conversion into a major product was observed (monitored by GC). At this time, the reaction mixture was quenched with HCl 2 M until pH < 3. The aqueous layer was extracted with EtOAc (3 × 10 mL). The organic extracts were combined, washed with brine (1 × 5 mL), dried over anhydrous MgSO<sub>4</sub>, filtered and concentrated under reduced pressure allowing a yellow viscous solid as the crude reaction. This residue was purified by column chromatography (8:2, Hexane/[EtOAc/EtOH/AcOH (3:1:0.08)]).

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<sup>&</sup>lt;sup>9</sup> Houpis, I. N.; Huang, C.; Nettekove, U.; Chen, J. G.; Liu, R.; Canters, M. Org. Lett. **2008**, 10, 5601.

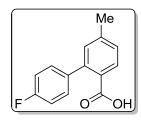
## 5-methyl-[1,1'-biphenyl]-2-carboxylic acid (1q)<sup>5</sup>:



Compound **1q** was obtained following the general procedure for the synthesis of biaryl 2-carboxylic acids **1q** - **1s**, from 1.8 mmol (252 mg) of phenylboronic acid. Compound **1q** was obtained as a yellow solid (213 mg, 1 mmol, 67%  $^{10}$ ):  $R_{\rm f}$  0.34 (7:3 Hexane/(3:1:0.08) EtOAc/EtOH/AcOH);  $^{1}$ H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  7.88 (dd, J =

8.0, 4.3 Hz, 1H), 7.43 - 7.07 (m, 7H), 2.42 (s, 3H);  $^{13}$ C NMR (75 MHz, CDCl<sub>3</sub>)  $\delta$  173.0 (C), 143.8 (C), 142.9 (C), 141.4 (C), 132.2 (CH), 131.2 (CH), 128.6 (2 × CH), 128.1(2 × CH), 128.0 (CH), 127.4 (CH), 126.4 (C), 21.6 (CH<sub>3</sub>).

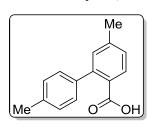
#### 4'-fluoro-5-methyl-[1,1'-biphenyl]-2-carboxylic acid (1r):



Compound **1r** was obtained following the typical procedure for the synthesis of biaryl 2-carboxylic acids **1q** - **1s**, from 0.90 mmol (126 mg) of 4-fluorophenylboronic acid and 0.75 mmol (165 mg) of 2-bromo-4-methylbenzoic acid. Compound **1r** was obtained as a yellow solid (103.4 mg, 0.45 mmol 60  $\%^{10}$ );  $R_f$  0.33

(7:3 Hexane/(3:1:0.08) EtOAc/EtOH/AcOH); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  7.90 (d, J = 8.3 Hz, 1H), 7.31 - 6.98 (m, 6 H), 2.42 (s, 3H); <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>)  $\delta$  173.2 (C), 162.4 (d, J = 246.4 Hz, C), 144.9 (C), 143.2 (CH), 137.4 (d, J = 3.0 Hz, C), 135.6 (CH), 132.8 (CH), 131.4 (C), 130.2 (d, J = 7.1 Hz, 2 × CH), 122.8 (C), 115.0 (d, J = 21.2 Hz, 2 × CH), 21.6 (CH<sub>3</sub>).

#### 4'-5-dimethyl-[1, 1'-biphenyl]-2-carboxylic acid (1s):



Compound **1s** was obtained following the typical procedure for the synthesis of biaryl 2-carboxylic acids **1q** - **1s**, from 0.54 mmol (122 mg) of 3-methylboronic acid acid and 0.75 mmol (165 mg) of 2-bromo-4-methylbenzoic acid. Compound **1s** was obtained as a yellow solid (116 mg, 0.51 mmol 68 %  $^{10}$ ):  $R_{\rm f}$  0.37

(7:3 Hexane/(3:1:0.08) EtOAc/EtOH/AcOH);  $^{1}$ H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  7.90 (d, J = 8.3 Hz, 1H), 7.31 - 6.98 (m, 6H), 2.41 (s, 3H), 2.39 (s, 3H);  $^{13}$ C NMR (101 MHz, CDCl<sub>3</sub>)  $\delta$  173.4 (C), 143.8 (C), 142.9 (C), 138.5 (C), 137.1 (C), 132.3 (CH), 131.2 (CH), 128.9 (2 x CH), 128.5 (2 x CH), 127.9 (CH), 126.4 (C), 21.7 (CH<sub>3</sub>), 21.4 (CH<sub>3</sub>).

According to <sup>1</sup>H NMR spectrum, biaryl 2-carboxylic acids **1q**, **1r** and **1s** were contaminated with 2-bromo-4-methylbenzoic acid. These results were corrected according to the integration of the methyl signals at <sup>1</sup>H NMR spectrum.

#### 6.6. Procedure for synthesis of acid 1t:

Compound 1t was prepared from 3-pyridinylboronic acid following the general procedure described for the synthesis of acids 1b-1l. After hydrolysis, the aqueous phase was basified and washed once with EtOAc (10 mL). The aqueous phase was acidified to pH 6 and the product was extracted with CH<sub>2</sub>Cl<sub>2</sub> (4 x 10 mL). Organic phases were collected, dried over MgSO<sub>4</sub>, filtered and concentrated.

#### 2-(3`-Pyridinyl)benzoic acid (1t):

Compound **1t** was obtained as a white solid (95.5 mg, 0.48 mmol, 24 %): 
$$R_f$$
: 0.07 (7:3 Hexane/(3:1:0.08) EtOAc/EtOH/AcOH); <sup>1</sup>H NMR (400 MHz, DMSO- $d_6$  + D<sub>2</sub>O)  $\delta$  8.50 – 8.40 (m, 1H), 8.40 (s, 1H), 7.80 (d,  $J$  = 7.7 Hz, 1H), 7.74 (d,  $J$  = 7.9 Hz, 1H), 7.60 (t,  $J$  = 7.5 Hz,

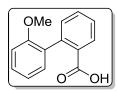
1H), 7.50 (t, J = 7.6 Hz, 1H), 7.45 (dd, J = 7.9, 4.9 Hz, 1H), 7.36 (d, J = 7.6 Hz, 1H); <sup>13</sup>C NMR (101 MHz, DMSO- $d_6 + D_2O$ )  $\delta$  169.9 (C), 148.2 (CH), 148.0 (CH), 138.0 (C), 137.5 (C), 136.9 (CH), 132.1 (CH), 132.0 (C), 131.2 (CH), 130.2 (CH), 128.8 (CH), 124.1 (CH).

## 6.7. Procedure for synthesis of acid 1u <sup>5,7</sup>:

To a KOH suspension (280 mg, 5 mmol) in dry MeCN (10 mL) was added 2a (196 mg, 1 mmol) under Ar atmosphere. The suspension was stirred for 2 hours under Ar atmosphere at room temperature. After this time, MeI (6.2 mL, 10 mmol) was added dropwise and the reaction mixture was stirred overnight under Ar atmosphere at reflux. The solvent was evaporated, affording a white wax. An EtOAc/H<sub>2</sub>O mixture was added (10 mL/10 mL) and the aqueous layer was extracted with EtOAc (4 × 10 mL). The organic extracts were combined, washed with brine (1 × 5 mL), dried over anhydrous MgSO<sub>4</sub>, filtered and concentrated under reduced pressure giving a yellow oil (241 mg, 99 %).

The desired biaryl 2-carboxilic acid **1u** was obtained following the general procedure described for the synthesis of acids **1b-1l**.

#### 2'-methoxy-[1, 1'-biphenyl]-2carboxylic acid (1u):



Compound **1u** was obtained as a white solid (84 mg, 0.84 mmol, 84 %):  $R_{\rm f}$ : 0.47 (7:3 Hexane/(3:1:0.08) EtOAc/EtOH/AcOH); <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  7.92 (dd, J = 7.7, 1.4 Hz, 1H), 7.57 (td, J = 7.5, 1.4 Hz, 1H), 7.39 (td, J = 7.6, 1.3 Hz, 1H), 7.33 (d, J = 7.3 Hz,

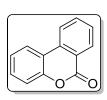
2H), 7.28 (dd, J = 8.8, 1.7 Hz, 1H), 7.03 (td, J = 7.4, 0.9 Hz, 1H), 6.86 (d, J = 8.8 Hz, 1H), 3.69 (s, 3H); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>)  $\delta$  173.5 (C), 156.3 (C), 139.2 (C), 132.4 (CH), 131.7 (C), 130.8 (C), 130.3 (CH), 130.0 (2 x CH), 129.2 (CH), 127.3 (CH), 121.0 (CH), 110.7 (CH), 55.2 (CH<sub>3</sub>).

#### 7. Synthesis of benzocoumarins 2a-2s.

• General procedure for the synthesis of benzocoumarins 2a - 2s by visible-light photoredox catalysis

Into a 25 mL pyrex tube, a solution of the desired acid **1** (0.20 mmol) in MeCN (1.6 mL) was added, followed by [Acr-Mes]ClO<sub>4</sub> (2.1 mg, 2.5 mol%). Then, a solution of  $(NH_4)_2S_2O_8$  (69 mg, 0.30 mmol) in  $H_2O$  (0.4 mL) was added. The reaction mixture was irradiated with blue LED's and stirred at room temperature<sup>11</sup> until complete conversion of starting material (monitored by TLC). The reaction mixture was concentrated under reduced pressure to afford a yellow residue which was purified by column chromatography (a gradient from 95:5 to 80:20, Hexane/EtOAc was used).

## 6H-benzo[c]chromen-6-one (2a)8:



Compound **2a** was prepared following the general procedure and was obtained after 8 h as a white solid (34 mg, 0.17 mmol, 87%):  $R_f$  0.14 (95:5 Hexane/EtOAc); GC:  $R_T$  15.7 min; IR: v 3075, 3026, 1725,

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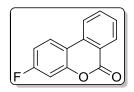
<sup>&</sup>lt;sup>11</sup> Upon irradiation the internal temperature increased to 30-35 °C.

1606, 1077, 744, 718, 681 cm<sup>-1</sup>; LRMS (EI): m/z (%) = 196 (M<sup>+</sup>, 100), 168 (40), 139 (40); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  8.38 (d, J = 7.9 Hz, 1H), 8.10 (d, J = 8.1 Hz, 1H), 8.04 (d, J = 7.8 Hz, 1H), 7.81 (td, J = 7.8, 10 Hz, 1H), 7.57 (t, J = 7.6 Hz, 1H), 7.49 - 7.44 (m, 1H), 7.39 - 7.29 (m, 2H); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>)  $\delta$  116.3 (C), 151.4 (C), 134.9 (CH), 134.8 (C), 130.6 (CH), 130.5 (CH), 128.9 (CH), 124.6 (CH), 122.6 (CH), 121.8 (CH), 121.3 (C), 118.1 (C), 117.9 (CH).

#### Scaled-up to 2 mmol:

Into a 25 mL pyrex tube, a solution of the desired acid 1 (2 mmol) in MeCN:DCE (10 mL: 10 mL) was added, followed by [Acr-Mes]ClO<sub>4</sub> (2.1 mg, 2.5 mol%). The reaction mixture was irradiated with blue LED's and stirred at room temperature for 22 h (complete conversion of starting material by TLC). The reaction mixture was concentrated under reduced pressure to afford a yellow residue which was purified by column chromatography (95:5, Hexane/EtOAc), giving the corresponding product 2a as a white solid (230 mg, 60 %).

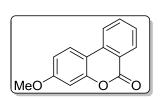
#### 3-Fluoro-6*H*-benzo[c]chromen-6-one (2b)<sup>5</sup>:



Compound **2b** was prepared following the general procedure and was obtained after 9 h as a white solid (30 mg 0.14 mmol, 70%):  $R_f$  0.20 (95:5 Hexane/EtOAc); GC:  $R_T$  15.5 min; IR: v 1746, 1607, 1456, 1273, 1154, 1052, 875, 764 cm<sup>-1</sup>; LRMS (EI): m/z (%) = 214

(M<sup>+</sup>, 100), 186 (44), 157 (35); <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  8.38 (ddd, J = 8.0, 1.4, 0.6 Hz, 1H), 8.07 - 8.00 (m, 2H), 7.83 (ddd, J = 8.1, 7.3, 1.4 Hz, 1H), 7.58 (ddd, J = 8.3, 7.3, 1.1 Hz, 1H), 7.11 - 7.05 (m, 2H); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>)  $\delta$  163.6 (d, J = 249.8 Hz, C), 160.9 (C), 152.3 (d, J = 12 Hz, C), 135.2 (CH), 134.4 (C), 130.8 (CH), 128.9 (CH), 124.5 (d, J = 9.8 Hz, CH), 121.6 (CH), 120.6 (C), 114.7 (d, J = 3 Hz, C), 112.6 (d, J = 21.8 Hz, CH), 105.2 (d, J = 24.8 Hz, CH).

## 3-methoxy-6*H*-benzo[c]chromen-6-one (2c)<sup>7</sup>:

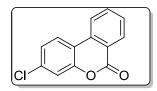


Compound **2c** was prepared following the general procedure, but employing 3 equiv. of  $(NH_4)_2S_2O_8$ . Compound **2c** was obtained after 65 h as a white solid (33 mg 0.15 mmol, 73%):  $R_f$  0.10 (95:5 Hexane/EtOAc); GC:  $R_T$  17.9 min; IR: v 1732,

1619, 1609, 1277, 1265, 1034, 760 cm<sup>-1</sup>; LRMS (EI): m/z (%) = 226 (M<sup>+</sup> 100), 183 (37), 127 (24); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  8.36 (dd, J = 8.0, 0.9 Hz, 1H), 8.00 (d, J = 8.1

Hz, 1H), 7.95 (d, J = 8.8 Hz, 1H), 7.82 - 7.75 (m, 1H), 7.54 - 7.48 (m, 1H), 6.92 (dd, J = 8.8, 2.6 Hz, 1H), 6.87 (d, J = 2.5 Hz, 1H), 3.86 (s, 3H); <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>) δ 161.6 (2 x C), 152.8 (C), 135.3 (C), 135.0 (CH), 130.7 (CH), 127.9 (CH), 123.9 (CH), 121.2 (CH), 120.1 (C), 112.6 (CH), 111.3 (C), 101.8 (CH), 55.8 (CH<sub>3</sub>).

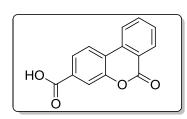
## 3-chloro-6*H*-benzo[c]chromen-6-one (2d)<sup>5</sup>:



Compound **2d** was prepared following the general procedure and was obtained after 7 h as a white solid (32 mg 0.14 mmol, 70%):  $R_f$  0.17 (95:5 Hexane/EtOAc); GC:  $R_T$  17.2 min; IR: v 1754, 1605, 1262, 1067, 761, 709 cm<sup>-1</sup>; LRMS (EI): m/z (%) =

232/230 (M<sup>+</sup>, 32/100), 204/202 (11/34), 139 (45); <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  8.41 (ddd, J = 7.9, 1.4, 0.5 Hz, 1H), 8.08 (d, J = 8.5 Hz, 1H), 8.00 (d, J = 8.6 Hz, 1H), 7.91 – 7.79 (m, 1H), 7.62 (ddd, J = 8.4, 7.3, 1.1 Hz, 1H), 7.40 (d, J = 2.0, 1H), 7.33 (dd, J = 8.5, 2.1 Hz, 1H); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>)  $\delta$  160.7 (C), 151.7 (C), 136.1 (C), 135.2 (CH), 134.1 (C), 130.9 (CH), 129.3 (CH), 125.2 (C), 123.9 (CH), 121.8 (C), 121.1 (CH), 118.1 (CH), 116.9 (CH).

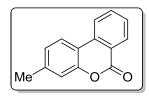
#### 6-oxo-6*H*-benzo[c]chromene-3-carboxylic acid (2e):



For the synthesis of compound **2e** a different procedure was followed: Into a pyrex tube, a solution of **1e** (47.7 mg, 0.25 mmol) in a 1:1:1 mixture of DCE/MeOH/MeCH (3 mL) was added, followed by [Acr-Mes]ClO<sub>4</sub> (2.625 mg 2.5 mol%). The reaction mixture was irradiated with blue

LED's lights and stirred at room temperature for 24 hours after which time, a precipitate appeared into the reaction vessel. At this time, the precipitate was filtered under vacuum, affording the desired pure product **2e** after 24 h as a white solid (24 mg 0.1 mmol, 53 %):  $R_f$  0.23 [7:3 Hexane: (3:1:0.08, EtOAc/EtOH/AcOH)]; IR: v 1756, 1673, 1424, 1072, 752 cm<sup>-1</sup>; LRMS (EI) m/z (%) = 240 (M<sup>+</sup>, 100), 223 (28), 195 (13), 167 (12), 139 (23), 44 (21); HRMS: *Calcd.* for  $C_{14}H_8O_4$  240.0423, *found* 240.0427; <sup>1</sup>H NMR (300 MHz, DMSO- $d_6$ )  $\delta$  13.42 (s, 1H), 8.50 (d, J = 5.0 Hz, 2H), 8.28 (d, J = 7.3 Hz, 1H), 8.19 - 7.52 (m, 4H); <sup>13</sup>C NMR (75 MHz, DMSO- $d_6$ )  $\delta$  166.2 (C), 159.9 (C), 150.5 (C), 135.4 (CH), 133.3 (C), 132.5 (C), 130.3 (CH), 129.8 (CH), 125.0 (CH), 124.1 (CH), 123.3 (CH), 121.5 (C), 121.2 (C), 117.8 (CH).

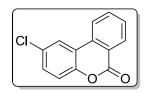
## 3-methyl-6*H*-benzo[c]chromen-6-one $(2f)^7$ :



Compound **2f** was prepared following the general procedure, but employing a DCE/H<sub>2</sub>O mixture (1.6 mL, 0.4 mL) as solvent. Compound **2f** was obtained after 14 h as a white solid (32 mg 0.16 mmol, 76%):  $R_f$  0.17 (95:5 Hexane/EtOAc); CG:

 $R_T$  16.8 min; IR: v 1727, 1606, 1458, 1254, 1081, 1046, 760 cm<sup>-1</sup>; LRMS (IE): m/z (%) = 210 (M<sup>+</sup>, 100), 212 (16), 208 (12), 182 (11), 181 (42), 154 (13), 152 (25), 76 (10); <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  8.39 (dd, J = 7.9 Hz, 1H), 8.09 (d, J = 8.1 Hz, 1H), 7.94 (d, J = 8.0 Hz, 1H), 7.81 (td, J = 8.3, 7.9, 1.4 Hz, 1H), 7.55 (td, J = 8.1, 0.9, 1H), 7.16 (d, J = 9.1 Hz, 2H), 2.46 (s, 3H); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>)  $\delta$  161.6 (C), 151.5 (C), 141.5 (C), 135.2 (C), 134.9 (CH), 130.7 (CH), 128.5 (CH), 125.8 (CH), 122.7 (CH), 121.6 (CH), 121.1 (C), 118.1 (CH), 115.6 (C), 21.6 (CH<sub>3</sub>).

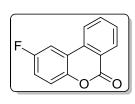
## **2-chloro-6H-benzo[c]chromen-6-one** (2g)<sup>5</sup>:



Compound **2g** was prepared following the general procedure and was obtained after 48 h as a mixture of regioisomers as a white solid (36 mg, 0.16 mmol, 80%, 90:10 rr<sup>12</sup> according to <sup>1</sup>HNMR): For major regioisomer:  $R_f$  0.36 (9:1 Hexane/EtOAc); GC:  $R_T$ 

17.1 min; IR: v 2973, 2926, 2884, 1604, 1047, 1038 cm<sup>-1</sup>; LRMS (IE): m/z (%) =232/230 (M<sup>+</sup>, 32/100), 204/202 (10/34), 139 (48); HRMS: Calcd. for C<sub>13</sub>H<sub>7</sub>ClO<sub>2</sub> 230.0135, found 230.0133; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  8.42 (dd, J = 8.0, 0.9 Hz, 1H), 8.08 (d, J = 8.1 Hz, 1H), 8.03 (d, J = 2.4 Hz, 1H), 7.93 - 7.81 (m, 1H), 7.73 - 7.60 (m, 1H), 7.44 (dd, J = 8.8, 2.4 Hz, 1H), 7.33 (d, J = 8.8 Hz, 1H); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>)  $\delta$  160.7 (C), 149.8 (C), 135.2 (CH), 133.7 (C), 130.9 (CH), 130.5 (CH), 130.0 (C), 129.7 (CH), 122.7 (CH), 121.9 (CH), 121.4 (C), 119.5 (C), 119.3 (CH).

## 2-fluoro-6*H*-benzo[c]-chromen-6-one (2h)<sup>5</sup>:



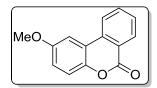
Compound **2h** was prepared following the general procedure and was obtained as a mixture of regioisomers after 48 h as a white solid (35 mg 82%, 96:4 rr<sup>12</sup> according to  $^{1}$ H NMR): For major regioisomer:  $R_f$  0.14 (95:5 Hexane/EtOAc); GC:  $R_T$  15.6 min; IR:

v 1717, 1496, 1185, 1076, 871, 763, 684 cm<sup>-1</sup>; LRMS (IE): m/z (%) = 214 (M<sup>+</sup>, 100), 186 (50), 157 (43); <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>) δ 8.43 (ddd, J = 7.9, 1.4, 0.6 Hz, 1H),

<sup>&</sup>lt;sup>12</sup> Regioselectivity was calculated according to <sup>1</sup>H NMR spectra of a regioisomers mixture, which was purified by column chromatography.

8.05 (d, J = 8.1 Hz, 1H), 7.90 - 7.83 (m, 1H), 7.73 (dd, J = 9.1, 2.9 Hz, 1H), 7.68 - 7.60 (m, 1H), 7.37 (dd, J = 9.0, 4.7 Hz, 1H), 7.20 (ddd, J = 9.1, 7.7, 2.9 Hz, 1H); <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>)  $\delta$  160.9 (C), 159.4 (d, J = 244.4 Hz, C), 147.5 (C), 135.1 (CH), 134.1 (d, J = 3.0 Hz, C), 130.9 (CH), 129.7 (CH), 122.0 (CH), 121.4 (C), 119.5 (d, J = 8.0 Hz, C), 119.4 (d, J = 8.0 Hz, CH), 117.8 (d, J = 24.2 Hz, CH), 108.9 (d, J = 25.3 Hz, CH).

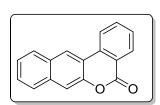
## 2-methoxy-6*H*-benzo[c]chromen-6-one (2i)<sup>5</sup>:



Compound **2i** was prepared following the general procedure and was obtained as a mixture of regioisomers after 48 h as a white solid (32.5 mg 72%, 92:8 rr<sup>12</sup> according to  $^{1}$ H NMR): For the major regioisomer:  $R_{\rm f}$  0.08 (95:5 Hexane/EtOAc); GC:

 $R_T$  17.8 min; IR: v 1710, 1091, 1048, 879, 681 cm<sup>-1</sup>; LRMS (IE): m/z (%) = 226 (M<sup>+</sup>, 100), 211 (46), 183 (32), 155 (11), 127 (26), 101 (11); <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  8.41 (d, J = 0.7 Hz, 1H), 8.06 (d, J = 8.1 Hz, 1H), 7.82 (td, J = 8.1, 1.3 Hz, 1H), 7.62 - 7.56 (m, 1H), 7.49 (d, J = 2.9 Hz, 1H), 7.30 (d, J = 9.0 Hz, 1H), 7.05 (dd, J = 9.0, 2.9 Hz, 1H), 3.91 (s, 3H); <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>)  $\delta$  161.4 (C), 156.5 (C), 145.7 (C), 134.9 (CH), 134.8 (C), 130.8 (CH), 129.1 (CH), 121.8 (CH), 121.5 (C), 118.8 (CH), 118.7 (C), 117.2 (CH), 106.5 (CH), 56.0 (CH<sub>3</sub>).

## 6H-dibenzo[c,h]chromen-6-one (2j)<sup>13</sup>:



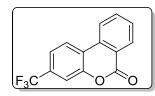
Compound 2j was prepared following the general procedure, but adding a DCE/H<sub>2</sub>O mixture (1.6 mL:0.4 mL) as solvent and AcOH (100  $\mu$ L) in order to improve the solubility. Product 2j was obtained after 24 h as a white solid (39.9 mg,

0.15 mmol, 75%):  $R_f$  0.26 (95:5 Hexane/EtOAc); GC:  $R_T$  21.15 min; IR: v 1717, 1610, 1270, 1064, 784 cm<sup>-1</sup>; LRMS (IE): m/z (%) = 246 (M<sup>+</sup>, 100), 218 (20), 189 (35), 95 (11); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  8.53 (d, J = 7.6 Hz, 1H), 8.42 (d, J = 7.9 Hz, 1H), 8.12 (d, J = 8.1 Hz, 1H), 7.98 (d, J = 8.8 Hz, 1H), 7.82 (t, J = 7.3 Hz, 2H), 7.70 (d, J = 8.8 Hz, 1H), 7.62 – 7.54 (m, 3H); <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>)  $\delta$  161.3 (C), 147.3 (C), 135.4 (C), 135.0 (CH), 134.3 (C), 130.7 (CH), 128.7 (CH), 128.0 (CH), 127.7 (CH), 127.2 (CH), 124.6 (CH), 123.9 (C), 122.4 (CH), 122.1 (CH), 121.2 (C), 119.2 (CH), 113.1 (C).

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<sup>&</sup>lt;sup>13</sup> Gallardo - Donaire, J.; Martin, R. J. Am. Chem. Soc. 2013, 135, 9350.

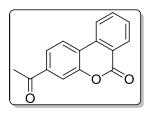
## 3-(trifluoromethyl)-6H-benzo[c]chromen-6-one (2k)<sup>8</sup>:



Compound **2k** was prepared following the general procedure, but employing 3 equiv. of  $(NH_4)_2S_2O_8$  and was obtained after 57 h as a white solid (18 mg, 0.07 mmol, 70%):  $R_f$  0.30 (95:5 Hexane/EtOAc); GC:  $R_T$  15.33 min; IR: v 1728, 1407, 1336,

1045, 722 cm<sup>-1</sup>; LRMS (IE): m/z (%) = 264 (M<sup>+</sup>, 100), 236 (33), 217 (11), 186 (11), 139 (14); <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  8.46 (ddd, J = 7.9, 1.4, 0.5 Hz, 1H), 8.23-8.16 (m, 2H), 7.94-7.87 (m, 1H), 7.73-7.66 (m, 1H), 7.66-7.58 (m, 2H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub>)  $\delta$  160.4 (C), 151.2 (C), 135.5 (CH), 133.6 (C), 132.5 (q, J = 33.5 Hz, C), 131.1 (CH), 130.3 (CH), 123.8 (CH), 123.5 (q, J = 273.0, C), 122.4 (CH), 121.9 (C), 121.3 (C), 121.2 (q, J = 3.4 Hz, CH), 115.4 (q, J = 4.0 Hz, CH).

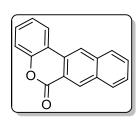
## 3-acetyl-6*H*-benzo[c]chromen-6-one (21)<sup>8</sup>:



Compound **2l** was prepared following the general procedure and was obtained after 39 h as a white solid (24 mg, 0.1 mmol, 50%):  $R_f$  0.06 (95:5 Hexane/EtOAc); GC:  $R_T$  19.16 min; IR: v 1736, 1686, 1476, 1262, 1217, 1032, 767, 716 cm<sup>-1</sup>; LRMS (IE): m/z (%) = 238 (M<sup>+</sup>, 47), 223 (100), 167 (25), 139 (33); <sup>1</sup>H

NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  8.44 (dd, J = 7.9, 1.0 Hz, 1H), 8.17 (t, J = 7.4 Hz, 2H), 7.97 - 7.85 (m, 3H), 7.72 - 7.63 (m, 1H), 2.67 (s, 3H); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>)  $\delta$  196.5 (C), 160.6 (C), 151.1 (C), 138.4 (C), 135.1 (CH), 133.6 (C), 130.8 (CH), 130.1 (CH), 123.7 (CH), 123.2 (CH), 122.4 (CH), 122.0 (C), 121.8 (C), 117.9 (CH), 26.7 (CH<sub>3</sub>).

## 6H-naphtho[2,3-c]chromen-6-one (2m)<sup>5</sup>:

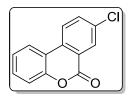


Compound **2m** was prepared from **1m** (0.17 mmol, 41.5 mg) following the general procedure, but using a DCE/H<sub>2</sub>O mixture (1.6 mL:0.4 mL) as solvent. Compound **2m** was obtained after 14 h as a white solid (30 mg 0.12 mmol, 72%):  $R_f$  0.14 (95:5 Hexane/EtOAc); GC:  $R_T$  21.6 min; IR: v 1728, 1173, 752 cm<sup>-1</sup>.

LRMS (IE): m/z (%) = 246 (M<sup>+</sup>, 100), 218 (34), 189 (40), 95 (14); <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  9.04 (s, 1H), 8.55 (s, 1H), 8.35 – 8.19 (m, 1H), 8.12 - 7.99 (m, 2H), 7.70 (ddd, J = 8.3, 6.8, 1.4 Hz, 1H), 7.60 (ddd, J = 8.1, 6.8, 1.3 Hz, 1H), 7.53 - 7.45 (m, 1H), 7.43 - 7.33 (m, 2H); <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>)  $\delta$  161.6 (C), 150.9 (C), 136.3 (C), 132.9

(CH), 132.5 (C), 130.3 (CH), 129.7 (C), 129.6 (2 x CH), 128.2 (CH), 127.3 (CH), 124.7 (CH), 123.0 (CH), 120.8 (CH), 119.3 (C), 118.4 (C), 118.0 (CH).

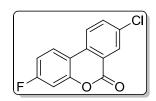
## 8-chloro-6*H*-benzo[c]chromen-6-one (2n)<sup>5</sup>:



Compound **2n** was prepared following the general procedure, and was obtained after 15 h as a white solid (37 mg 0.16 mmol, 80%):  $R_f$  0.30 (95:5 Hexane/EtOAc); GC:  $R_T$  17.1 min; IR: v 1718, 1604, 1307, 1089, 827 cm<sup>-1</sup>; LRMS (IE): m/z (%) =232/230 (M<sup>+</sup>,

32/100), 204/202 (11/32), 139 (46);  ${}^{1}$ H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  8.38 (d, J = 2.2 Hz, 1H), 8.08 (d, J = 8.6 Hz, 1H), 8.03 (dd, J = 7.9, 1.5 Hz, 1H), 7.78 (dd, J = 8.6, 2.3 Hz, 1H), 7.58 - 7.47 (m, 1H), 7.44 - 7.33 (m, 2H);  ${}^{13}$ C NMR (101 MHz, CDCl<sub>3</sub>)  $\delta$  160.2 (C), 151.3 (C), 135.3 (CH), 135.1 (C), 133.4 (C), 131.0 (CH), 130.2 (CH), 125.0 (CH), 123.6 (CH), 122.9 (CH), 122.7 (C), 118.0 (CH), 117.4 (C).

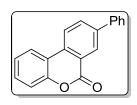
#### 8-chloro-3-fluoro-6*H*-benzo[c]chromen-6-one (2o):



Compound **20** was prepared following the general procedure using a DCE/ $H_2O$  mixture (1.6 mL:0.4 mL) as solvent. Compound **20** was obtained after 22 h as a white solid (40 mg 0.16 mmol, 79%):  $R_f$  0.27 (95:5 Hexane/EtOAc); CG:  $R_T$  16.8

min; IR: v 1742, 1608, 1475, 1275, 1151, 1116, 805, 786, 772 cm<sup>-1</sup>; LRMS (IE): m/z (%) = 250/248 (M<sup>+</sup>, 33/100), 222/220 (13/36), 157 (43); HRMS: Calcd. for  $C_{13}H_6CIFO_2$  248.0040, found 248.0042; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  8.32 (d, J = 2.3 Hz, 1H), 7.98 (t, J = 8.1 Hz, 2H), 7.77 (dd, J = 8.6, 2.3 Hz, 1H), 7.09 (t, J = 8.2 Hz, 2H); <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>)  $\delta$  163.6 (d, J = 253.5 Hz, C), 159.6 (C), 152.0 (d, J = 12.1 Hz, C), 135.4 (CH), 134.9 (C), 132.7 (C), 130.3 (CH), 124.5 (d, J = 10.1 Hz, CH), 123.4 (CH), 121.8 (C), 114.0 (d, J = 3.0 Hz, C), 112.9 (d, J = 23.0 Hz, CH), 105.4 (d, J = 25.3 Hz, CH).

## 8-phenyl-6*H*-benzo[c]chromen-6-one (2p)<sup>8</sup>:

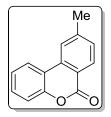


Compound **2p** was prepared following the general procedure and was obtained as a white solid after 11 h (35 mg 0.13 mmol, 65%):  $R_f$  0.15 (95:5 Hexane/EtOAc);  $R_T$  24.7 min; IR: v 1727, 1442, 1235, 840, 743 cm<sup>-1</sup>; LRMS (IE): m/z (%) = 272 (M<sup>+</sup>, 100), 244

(12), 215 (20), 189 (9);  ${}^{1}$ H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  8.65 (s, 1H), 8.21 (d, J = 8.4 Hz, 1H), 8.14 - 8.06 (m, 2H), 7.72 (dd, J = 8.3, 1.3 Hz, 2H), 7.56 - 7.32 (m, 6H);  ${}^{13}$ C NMR

(75 MHz, CDCl<sub>3</sub>) δ 161.3 (C), 151.2 (C), 141.8 (C), 139.0 (C), 133.5 (CH), 133.5 (C), 130.4 (CH), 129.1 (2 x CH), 128.6 (CH), 128.3 (CH), 127.1 (2 x CH), 124.6 (CH), 122.8 (CH), 122.4 (CH), 121.7 (C), 117.9 (C), 117.8 (CH).

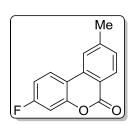
## 9-methyl-6*H*-benzo[c]chromen-6-one (2q)<sup>5</sup>:



Compound **2q** was prepared from **1q** (0.17 mmol, 39.1 mg), following the general procedure, but adding 1.2 equiv. of  $(NH_4)_2S_2O_8$ , and using a DCE/H<sub>2</sub>O mixture (1.6 mL:0.4 mL) as a solvent. Compound **2q** was obtained after 20 h as a white solid (25 mg 0.12 mmol, 71%):  $R_f$  0.14 (95:5 Hexane/EtOAc); GC:  $R_T$  16.8 min; IR: v

1722, 1613, 1268, 743 cm<sup>-1</sup>; LRMS (IE): m/z (%) = 272 (M<sup>+</sup>, 100), 244 (12), 215 (20), 189 (9); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  8.27 (d, J = 8.1 Hz, 1H), 8.04 (d, J = 8.0 Hz, 1H), 7.89 (s, 1H), 7.51 - 7.43 (m, 1H), 7.41 - 7.28 (m, 3H), 2.55 (s, 3H); <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>)  $\delta$  161.4 (C), 151.6 (C), 146.0 (C), 134.8 (C), 130.7 (CH), 130.4 (CH), 130.3 (CH), 124.5 (CH), 122.8 (CH), 122.0 (CH), 118.9 (C), 118.2 (C), 117.9 (CH), 22.4 (CH<sub>3</sub>).

## 3-fluoro-9-methyl-6*H*-benzo[c]chromen-6-one (2r)<sup>14</sup>:



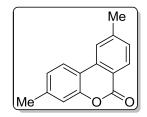
Benzocoumarin **2r** was prepared from **1r** (0.17 mmol, 39.1 mg), following the general procedure, but using a DCE/H<sub>2</sub>O mixture (1.6 mL:0.4 mL) as a solvent. Compound **2r** was obtained after 24 h as a white solid; (22 mg, 0.1 mmol, 59%):  $R_{\rm f}$  0.20 (95:5 Hexane/EtOAc); GC:  $R_{\rm T}$  16.4 min; IR: v 1717, 1599, 1266, 1106,

769 cm<sup>-1</sup>; LRMS (IE): m/z (%) = 228 (M<sup>+</sup>, 100), 213 (11), 199 (45), 170 (16); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  8.26 (d, J = 8.1 Hz, 1H), 8.05 - 7.99 (m, 1H), 7.82 (s, 1H), 7.38 (d, J = 8.1 Hz, 1H), 7.06 (m, 2H), 2.56 (s, 3H); <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>)  $\delta$  163.5 (d, J = 251.5 Hz, C), 161.0 (C), 152.5 (d, J = 12.1 Hz, C), 146.4 (C), 134.4 (C), 130.8 (CH), 130.2 (CH), 124.4 (d, J = 10.1 Hz, CH), 121.8 (CH), 118.2 (C), 114.8 (d, J = 2.0 Hz, C) 112.4 (d, J = 22.2 Hz, CH), 106.2 (d, J = 25.3 Hz, CH), 22.4 (CH<sub>3</sub>).

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<sup>&</sup>lt;sup>14</sup> Lee, T. – H.; Jayakumar, J.; Cheng, C. – H.; Chuang, S. – S. *Chem. Comm.* **2013**, *40*, 11797.

#### 3, 9-dimethyl-6*H*-benzo[c]chromen-6-one (2s):



Benzocoumarin **2s** was prepared from **1s** (0.17 mmol, 38.4 mg), following the general procedure, but using a DCE/H<sub>2</sub>O mixture (1.6 mL:0.4 mL) as a solvent. Compound **2s** was obtained after 18 h as a white solid (29 mg, 0.13 mmol, 77%):  $R_f$  0.29 (95:5 Hexane/EtOAc); GC:  $R_T$  17.7 min; IR: v 1717, 1087, 1039, 890

cm<sup>-1</sup>; LRMS (IE): m/z (%) = 224 (M<sup>+</sup>, 100), 195 (24), 181 (10), 152 (15); HRMS *Calcd*. for C<sub>15</sub>H<sub>11</sub>O<sub>2</sub> 223.0759, found 223.0767; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  8.26 (d, J = 8.1 Hz, 1H), 7.92 (d, J = 8.0 Hz, 1H), 7.86 (s, 1H), 7.35 (d, J = 8.1 Hz, 1H), 7.18 - 7.11 (m, 2H), 2.55 (s, 3H), 2.45 (s, 3H); <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>)  $\delta$  161.7 (C), 151.6 (C), 146.0 (C), 141.2 (C), 135.1 (C), 130.7 (CH), 129.8 (CH), 125.7 (CH), 122.6 (CH), 121.7 (CH), 118.6 (C), 118.0 (CH), 115.6 (C), 22.4 (CH<sub>3</sub>), 21.6 (CH<sub>3</sub>).

# 8. Synthesis of Deuterated Phenylboronic $Acid^{15}$ 1a- $D_5$ and Benzo-3,4-coumarin 2a- $D_4$ . $^{16}$

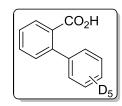
Into an oven-dried one-necked 100 mL flask a solution of polideuterated bromobencene (10 mmol, 2.03 mL) in dry THF under Ar atmosphere was prepared. The solution was cooled at -78 °C and BuLi (3.8 mL of a solution 3.2 M, 12 mmol) was carefully charged dropwise over the stirred solution ca. 15 minutes. The reaction mixture was stirred under Ar atmosphere at -78 °C for 1 h. Then, B(O<sup>i</sup>Pr)<sub>3</sub> (26 mmol, 6.12 mL) was added dropwise and the reaction mixture was allowed to reach room temperature and stirred overnight. After this time, complete conversion was observed (monitored by TLC) and the reaction mixture was quenched with HCl (10 %, 30 mL) until pH<1 and stirred for 3 h. The reaction mixture was concentrated and H<sub>2</sub>O (10 mL) was added. The aqueous layer was extracted with EtOAc (4 × 10 mL). The combined organic layers were washed with brine (1 × 5 mL), dried over anhydrous MgSO<sub>4</sub> and filtered under reduced pressure affording a viscous yellow oil. The oil was crystallized with hexane (18 mL) forming a white solid which was used without further purification at the Suzuki Cross-Coupling reaction. The desired deuterated phenyl 2-carboxylic acid **1a-D<sub>5</sub>** was obtained following the typical procedure described for the synthesis of acids 1b-11 by using 0.86 mmol of methyl 2-iodobenzoate.

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<sup>&</sup>lt;sup>15</sup> (a) Gandeepan P.; Cheng, C. – H. *Org. Lett*, **2013**, *15*, 2084; (b) Zhou, B.; Chen, H.; Wang, C. *J. Am. Chem. Soc.* **2013**, *135*, 1264.

<sup>&</sup>lt;sup>16</sup> Wang, C.; Rakshit, S.; Glorus, F. J. Am. Chem. Soc. 2010, 132, 14006

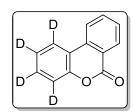
#### (1,1'-Biphenyl-2',3',4',5',6'- $d_5$ )-2-carboxylic acid $(1a-D_5)$ :



Compound **1a-D**<sub>5</sub> was obtained as a white solid (116 mg, 0.54 mmol, 63%):  $R_f$  0.4 (95:5 Hexane/EtOAc); GC:  $R_T$  14.26 min. LRMS (IE): m/z (%) = 203 (M<sup>+</sup>, 100), 185 (37), 157 (40), 78 (11); <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  7.95 (dd, J = 7.8, 1.1 Hz, 1H), 7.56 (td, J = 7.5, 1.4 Hz, 1H), 7.47-7.35 (m, 2H); <sup>13</sup>C NMR (75 MHz,

CDCl<sub>3</sub>) δ 173.4 (C), 143.5 (C),141.0 (C), 132.2 (CH), 131.3 (CH), 130.8 (CH), 129.5 (C), 127.3 (CH).

#### 6*H*-Benzo[c]chromen-6-one-1,2,3,4-d<sub>4</sub> (2a-D<sub>4</sub>):



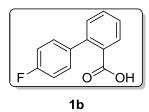
Compound **2a-D**<sub>4</sub> was prepared following the general procedure, and was obtained as a white solid (20 mg, 0.1 mmol, 50 %):  $R_f$  0.23 (95:5 Hexane/EtOAc); GC:  $R_T$  15.85 min; IR: v 1721, 1604, 1582, 1257, 1114, 679 cm<sup>-1</sup>; LRMS (IE): m/z (%) = 200 (M<sup>+</sup>,

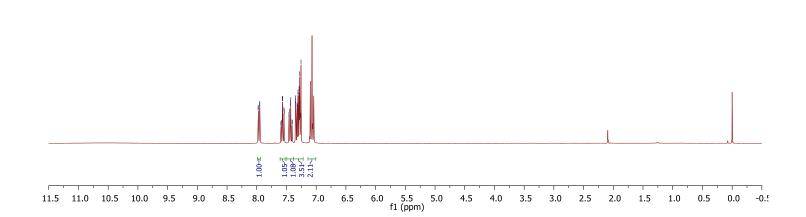
100), 172 (41), 143 (25); HRMS *Calcd.* for  $C_{13}H_4D_4O_2$  200.0775, found 200.0777 <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  8.41 (ddd, J = 7.9, 1.4, 0.6 Hz, 1 H), 8.13 (ddd, J = 8.1, 1.1, 0.6 Hz, 1H), 7.83 (ddd, J = 8.1, 7.3, 1.4 Hz, 1H), 7.59 (ddd, J = 8.0, 7.3, 1.1 Hz, 1H); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>)  $\delta$  161.3 (C), 151.4 (C), 135.0 (CH), 134.9 (C), 130.7 (CH), 129.0 (CH), 121.8 (CH), 121.4 (C), 118.1 (C).

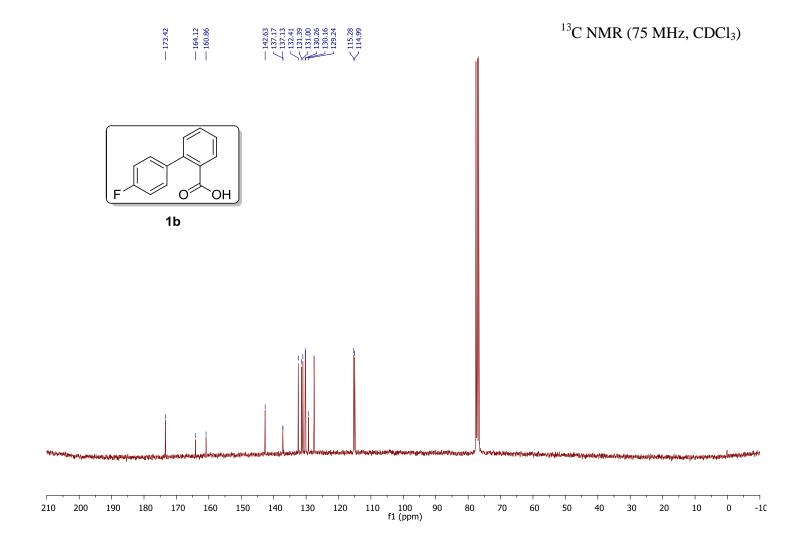
## 9. <sup>1</sup>H NMR and <sup>13</sup>C NMR Spectra of 2-Biarylcarboxylic Acids 1b - 1t

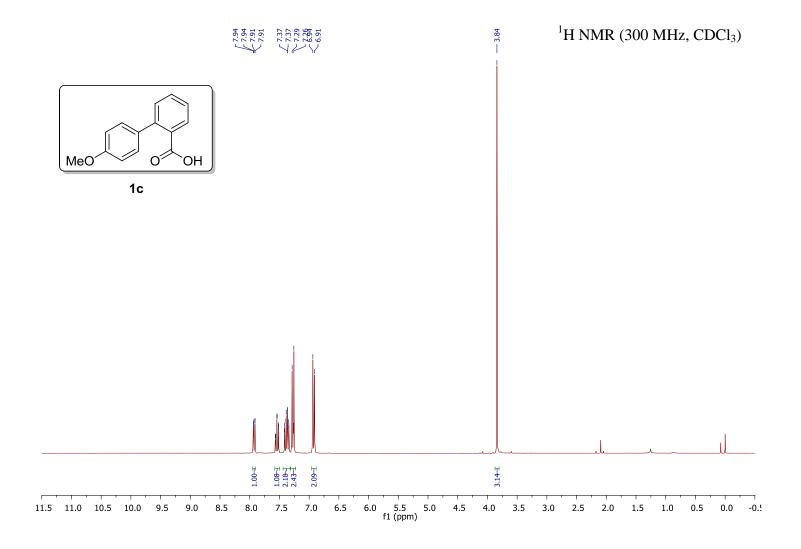


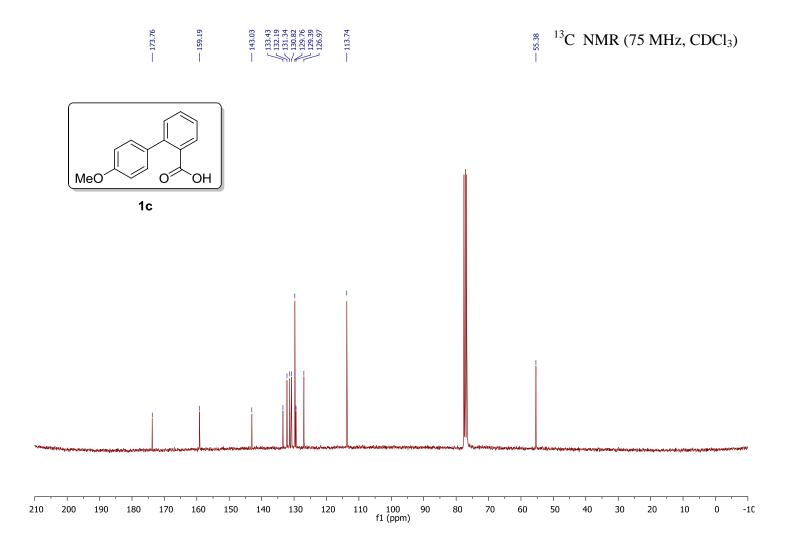
<sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)





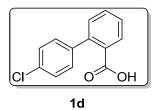


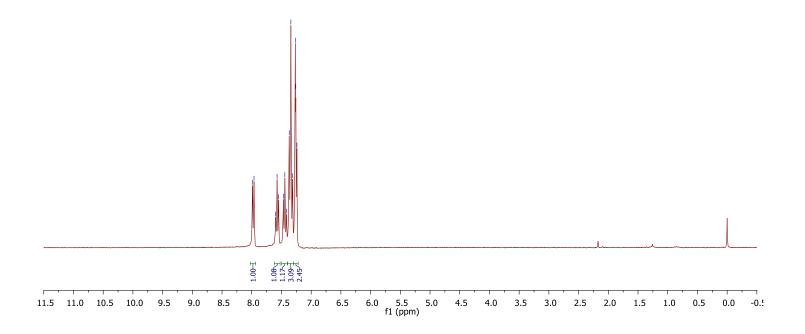


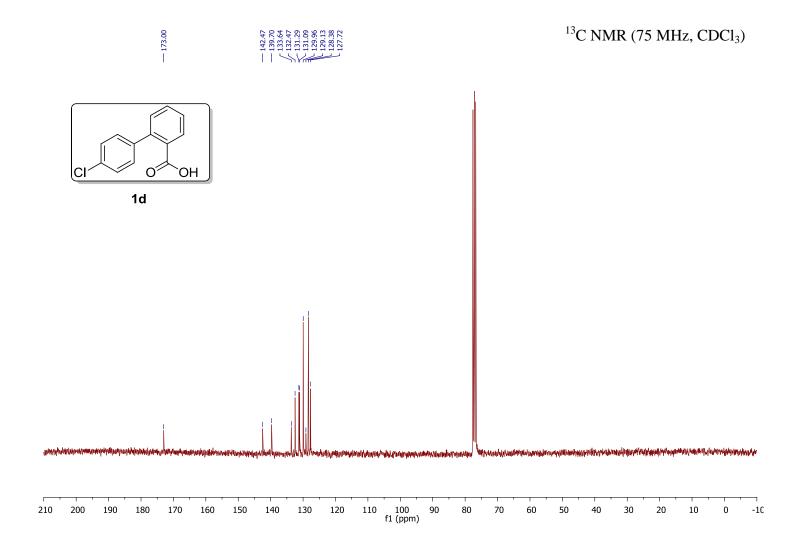




<sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)

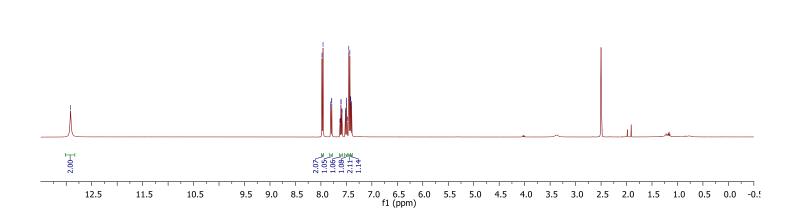


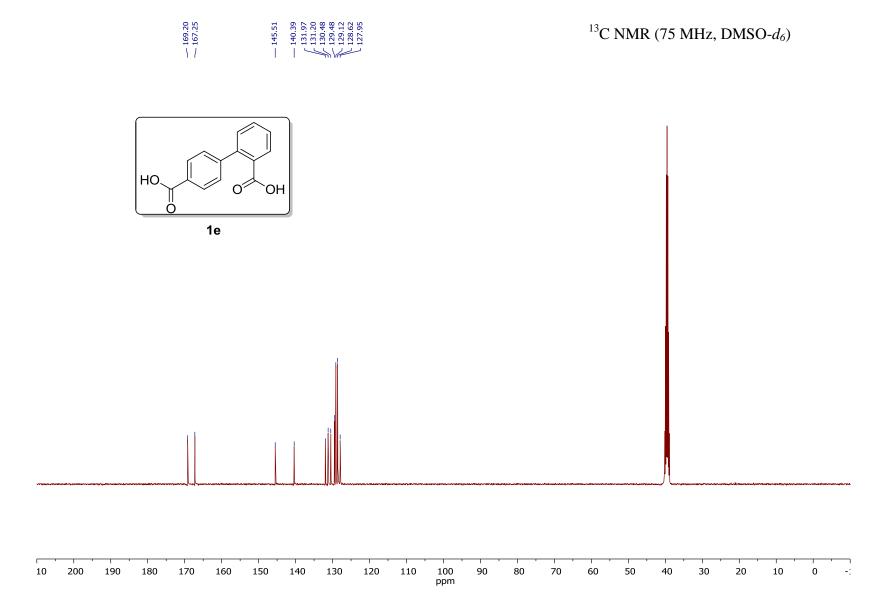




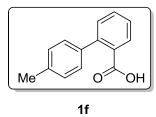


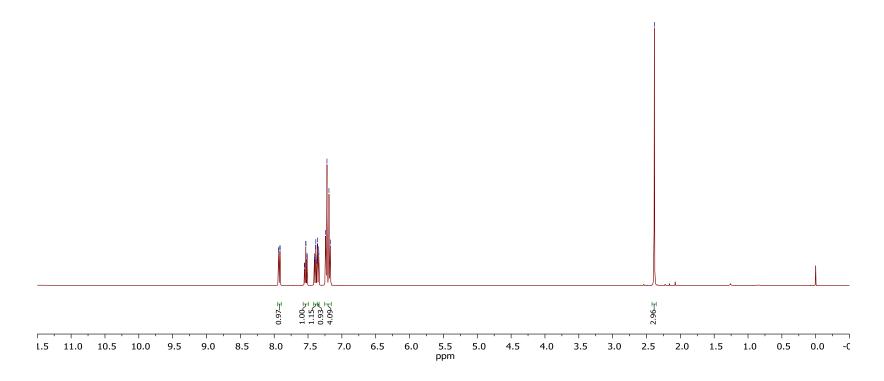
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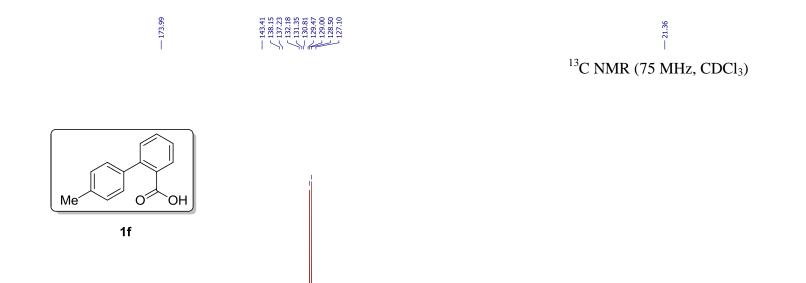




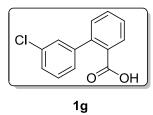


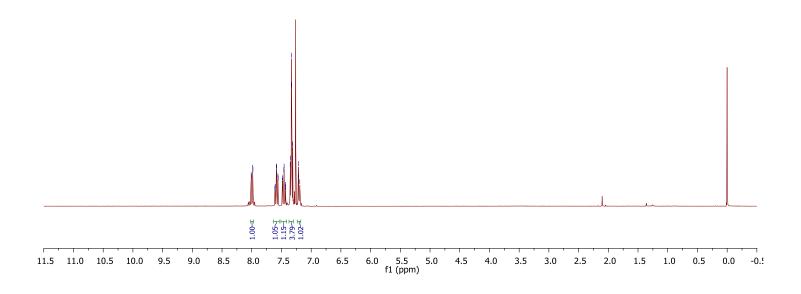


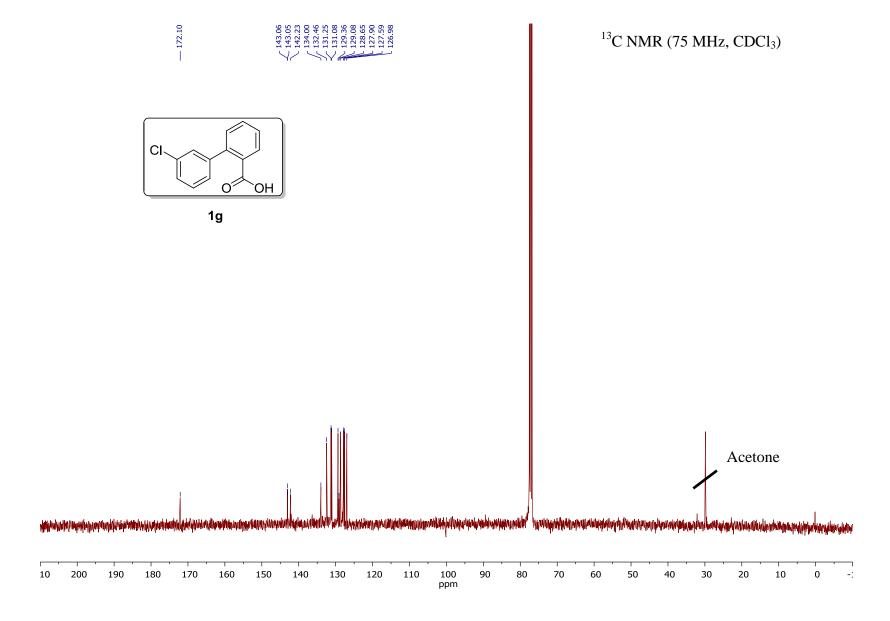




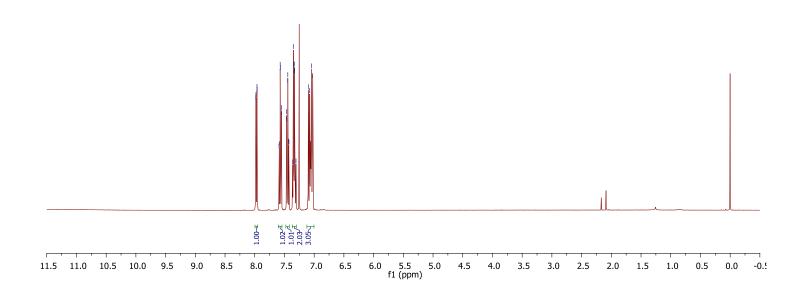


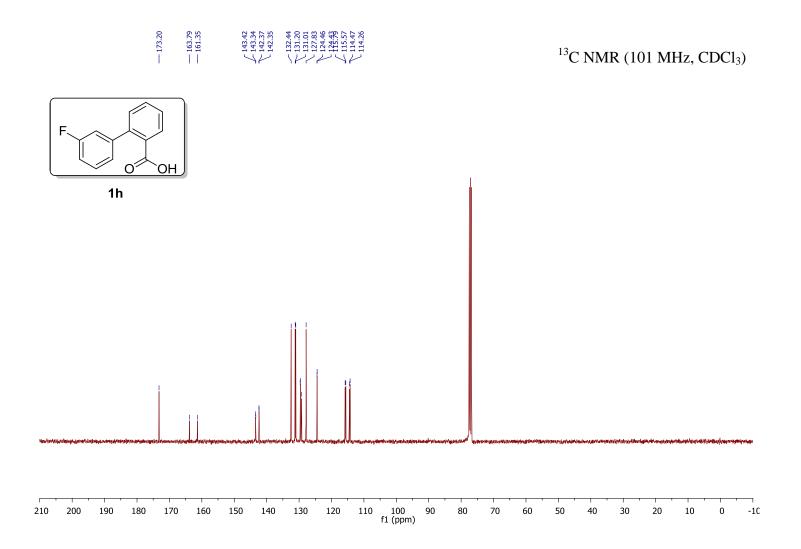




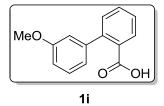


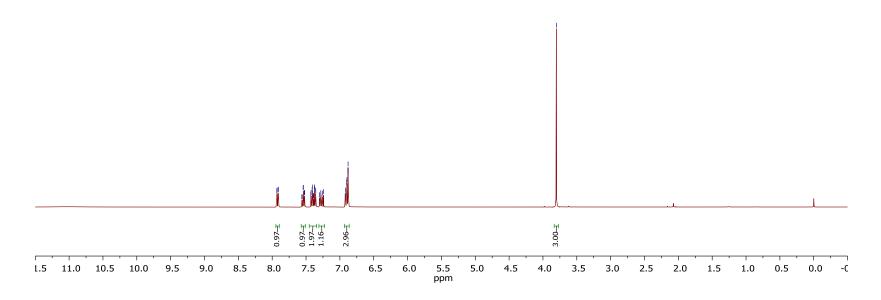


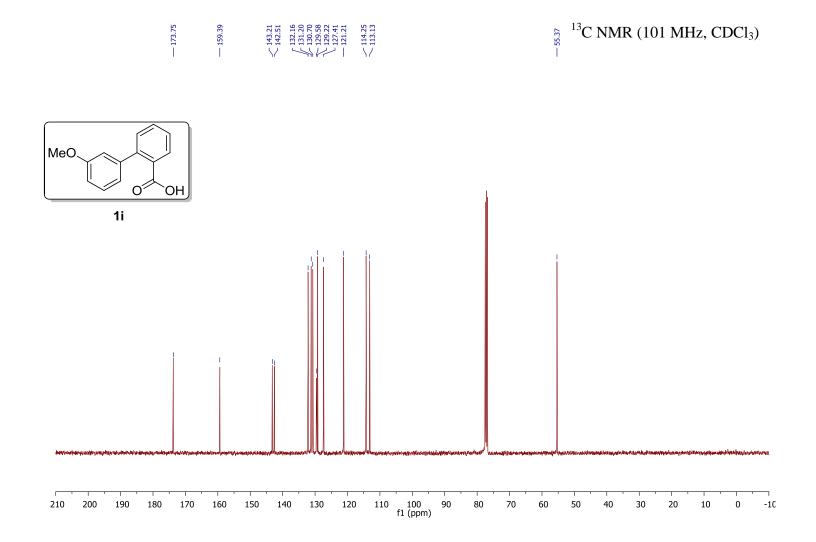


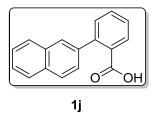


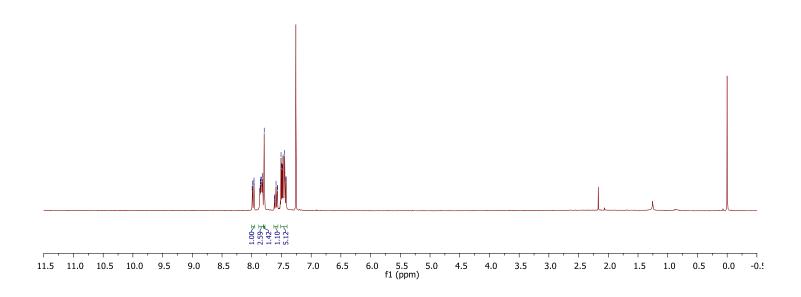


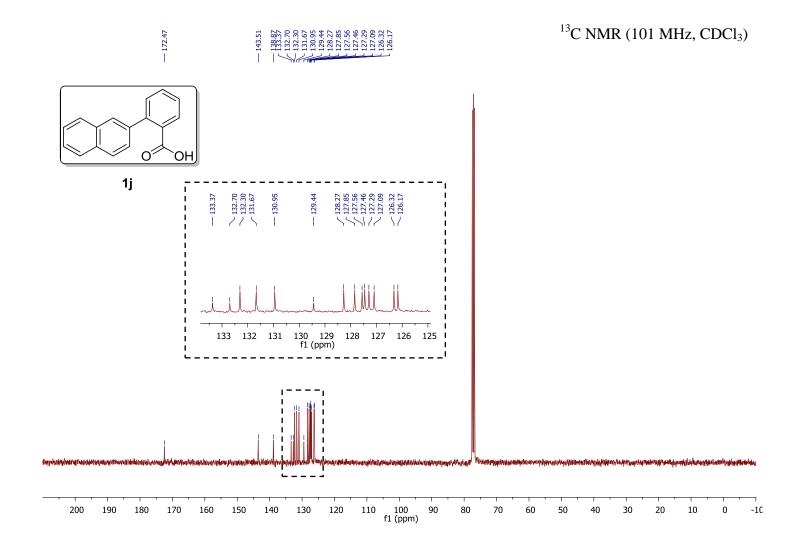


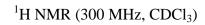




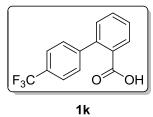


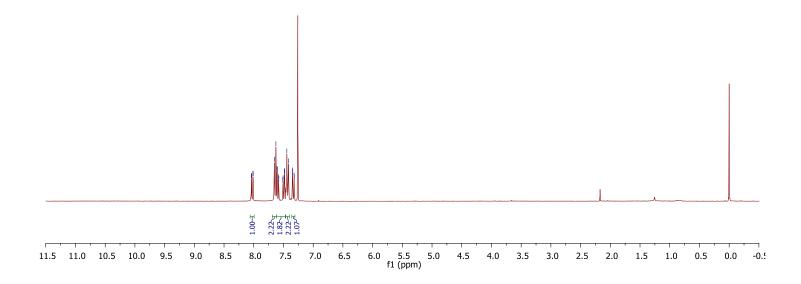


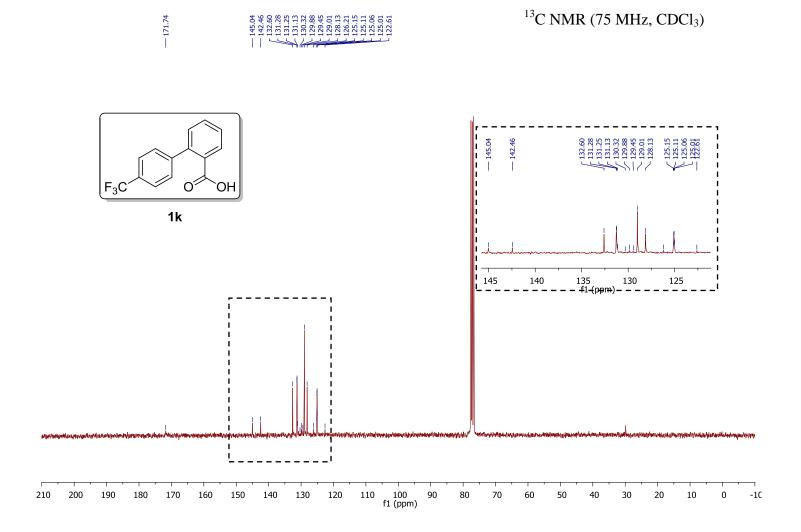


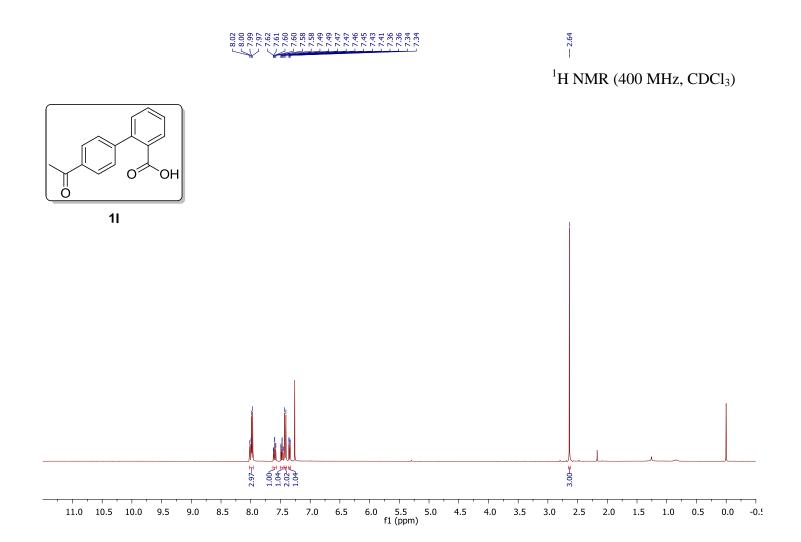


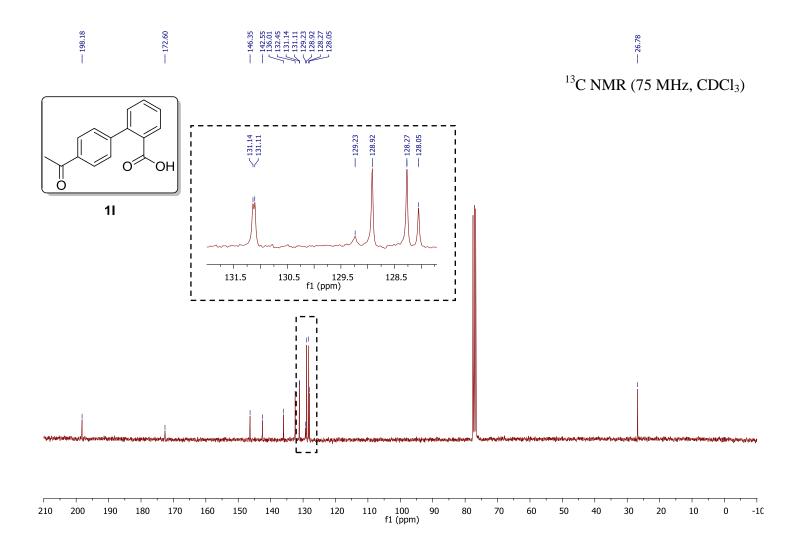




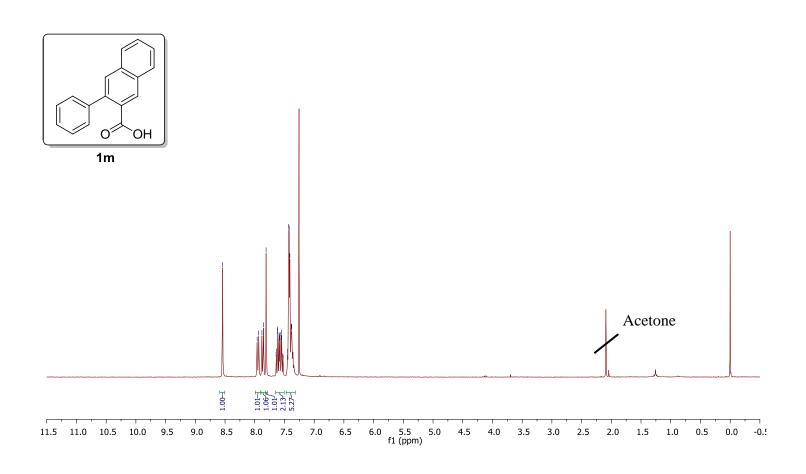


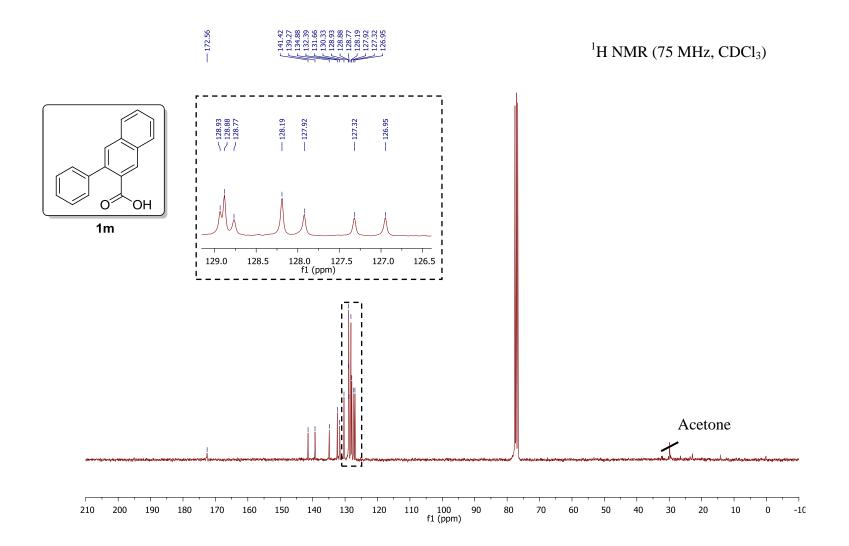


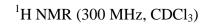






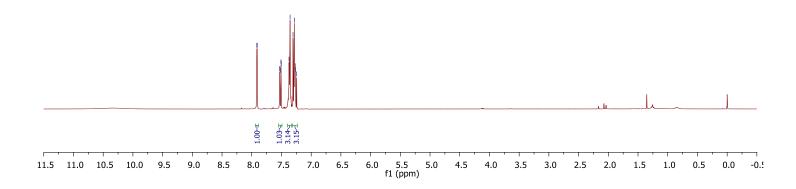


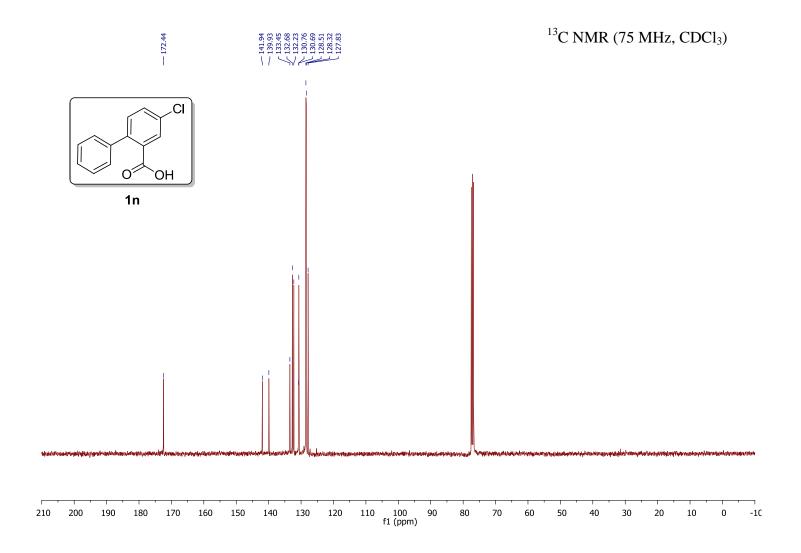




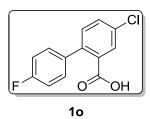


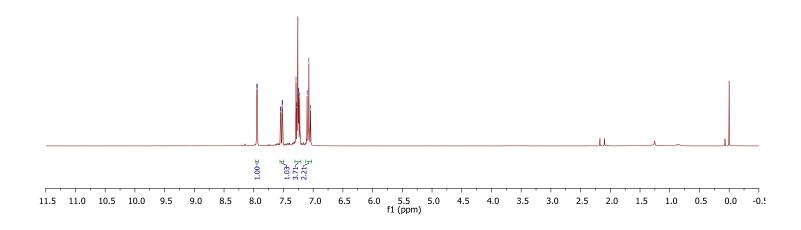
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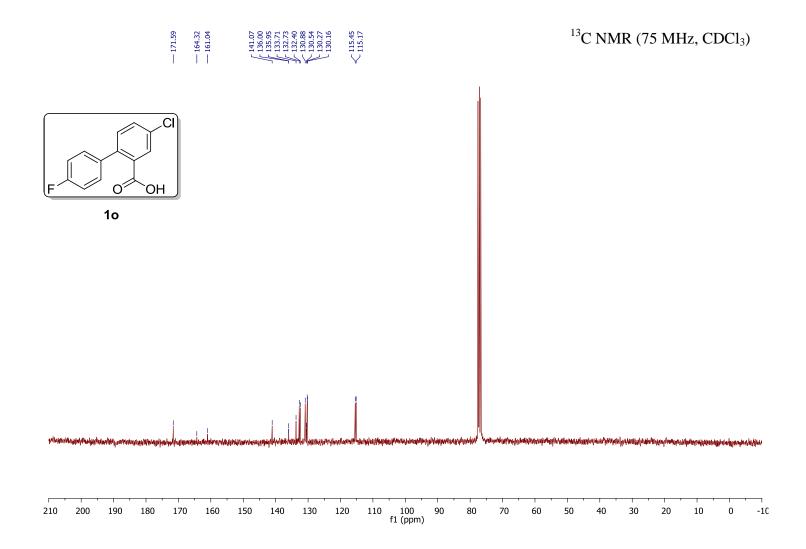


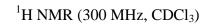


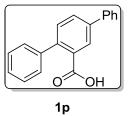


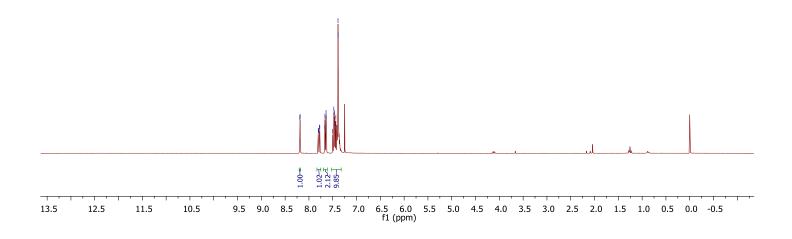


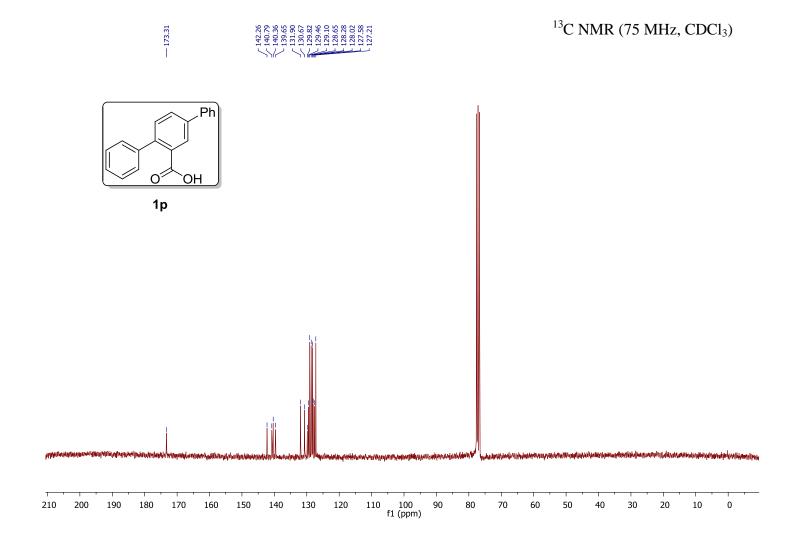






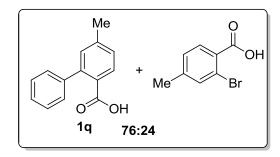


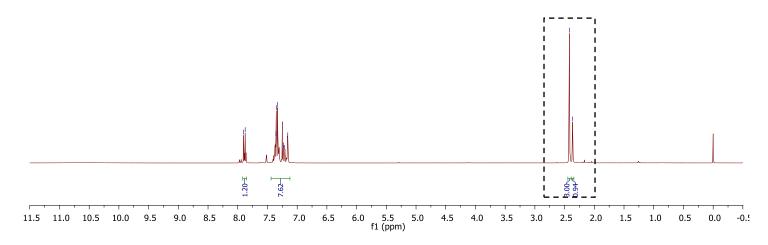


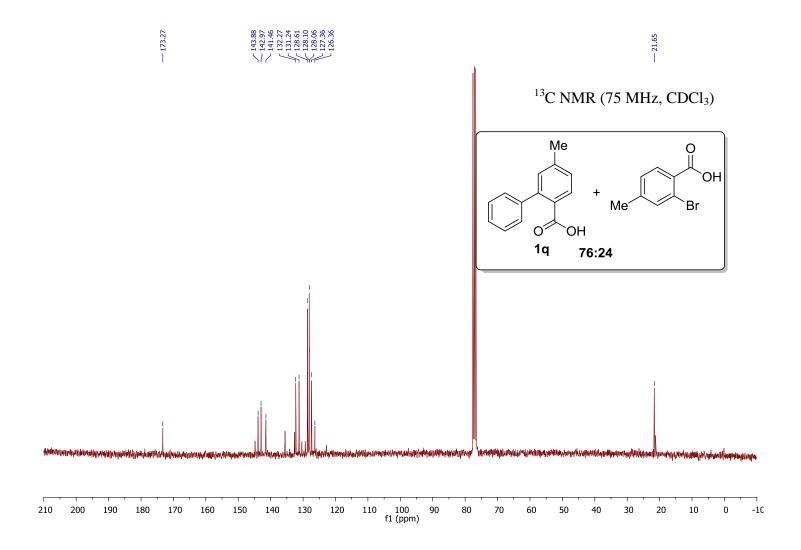




# 2.42

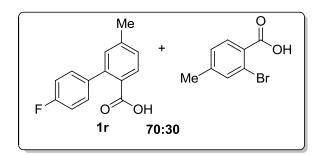


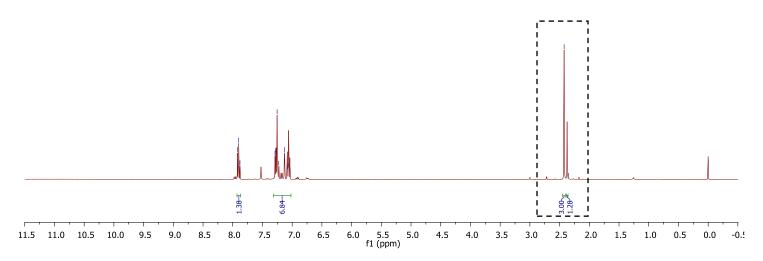


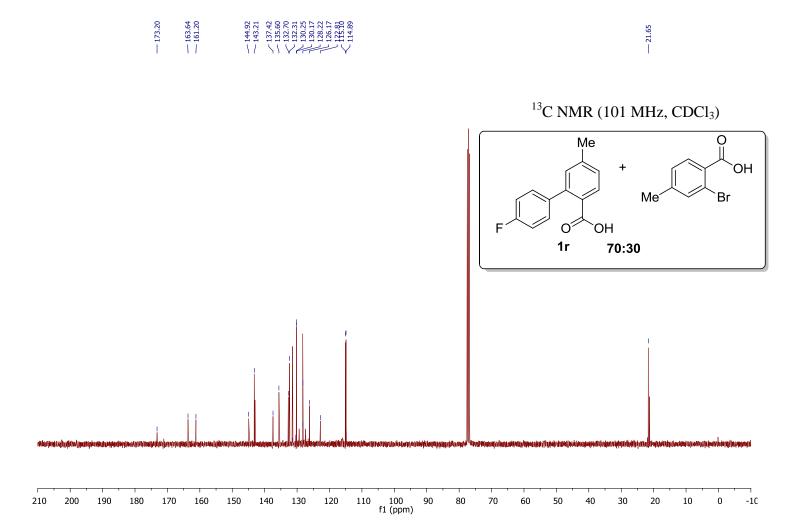




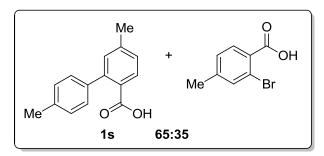
#### \_\_\_2.42 \_\_\_2.35

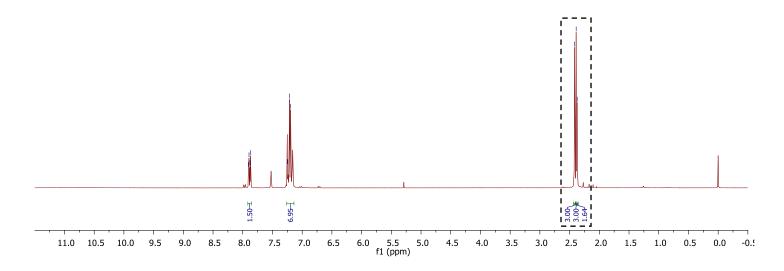


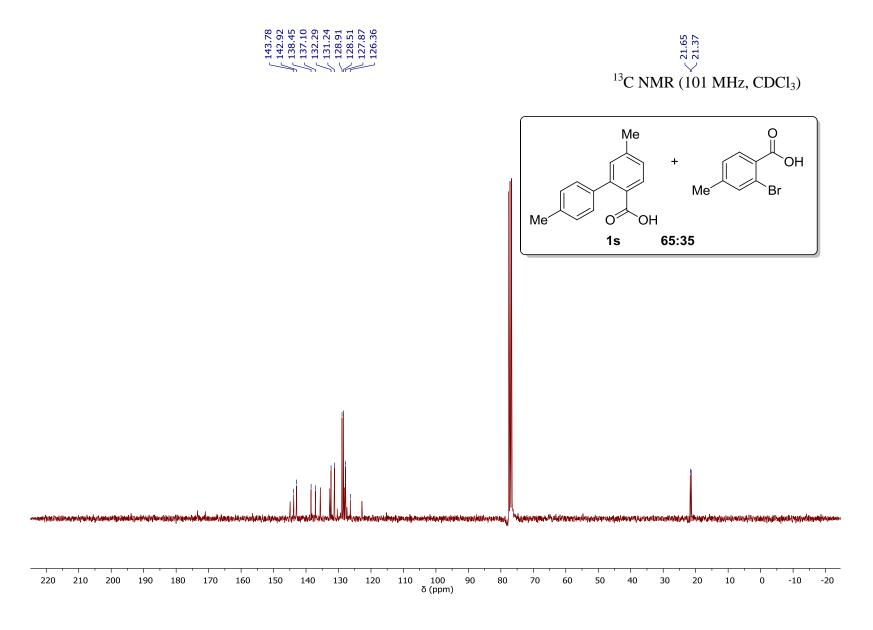


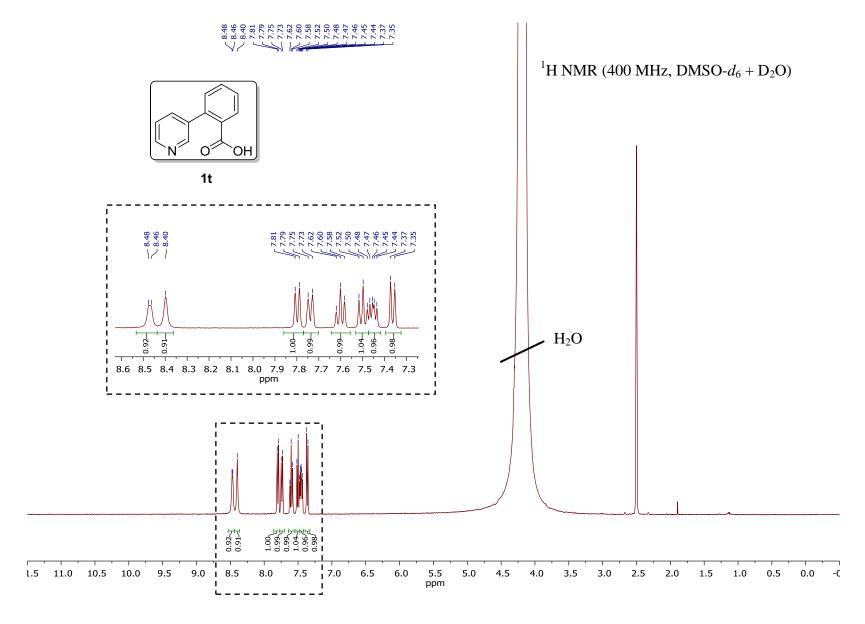


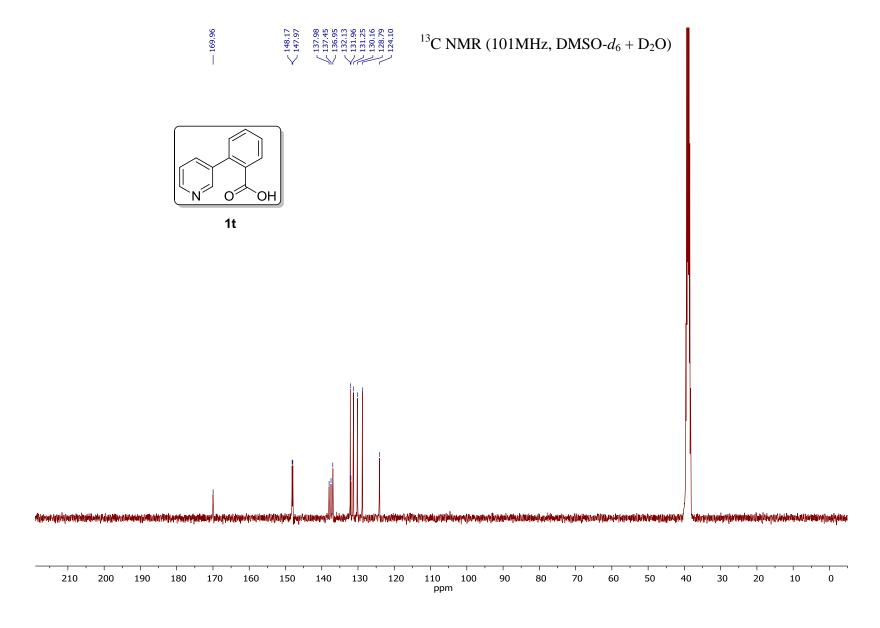


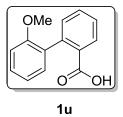


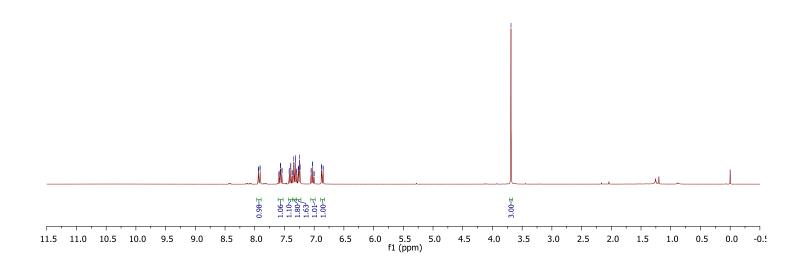


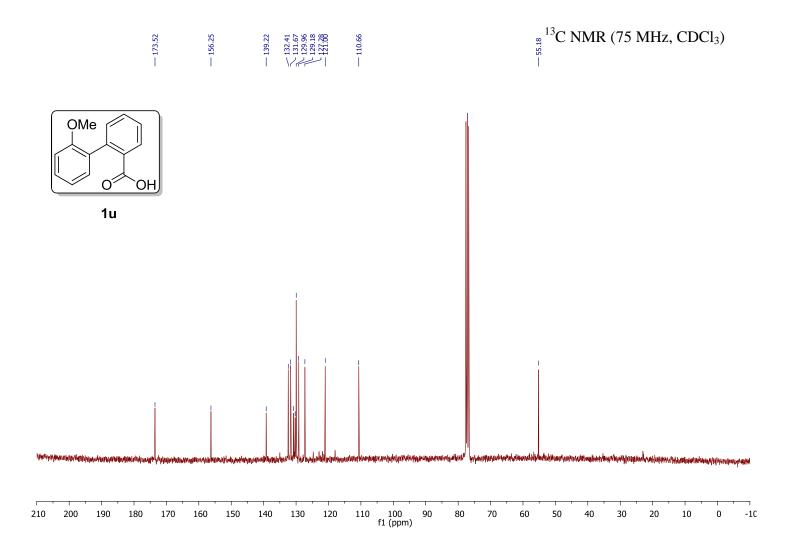






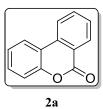


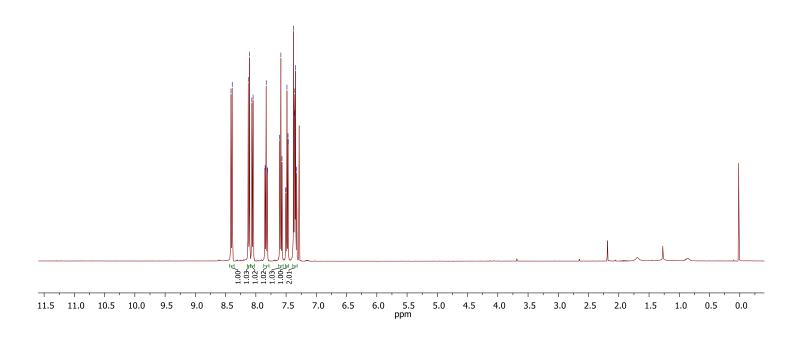




#### 10. <sup>1</sup>H NMR and <sup>13</sup>C NMR Spectra of Benzo-3,4-coumarins 2a - 2s

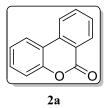


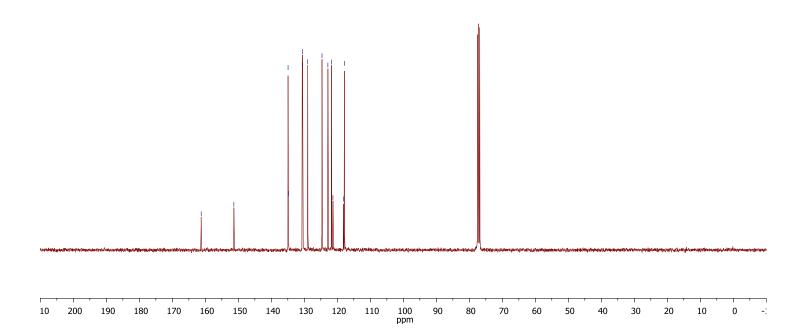


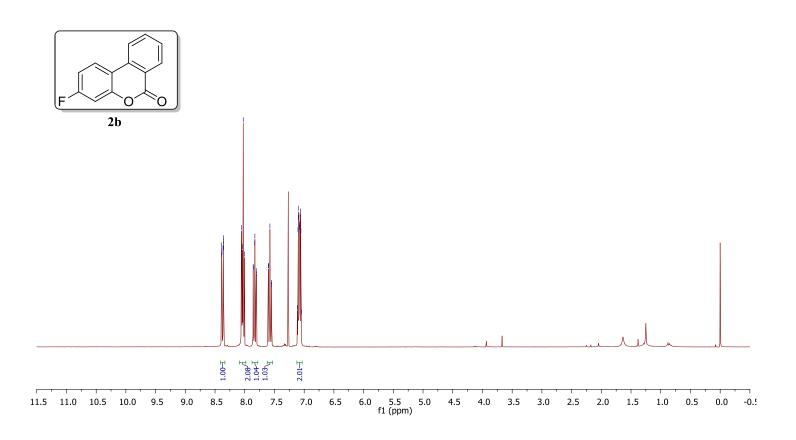


161.26	151.35	134.93 134.83 130.64 130.53 128.97 124.64 122.85 121.77 121.32 111.31
	1	

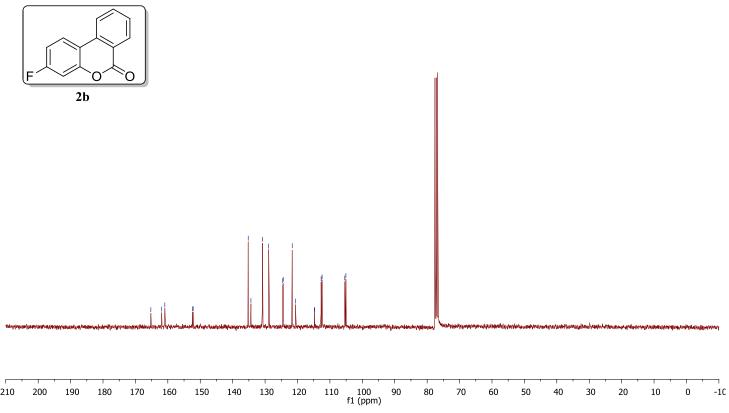
<sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>)

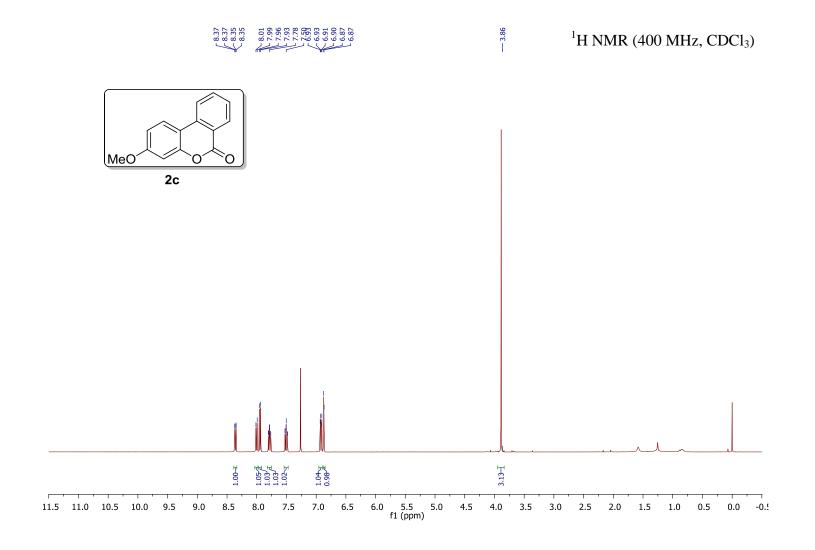


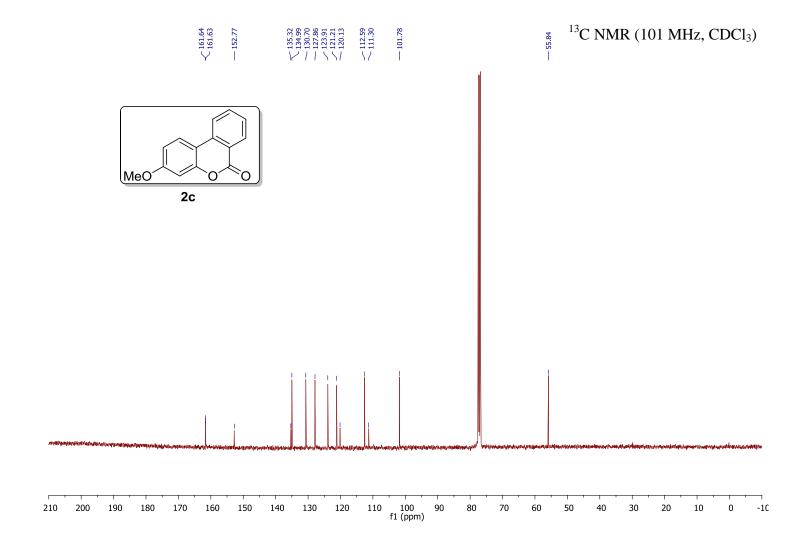


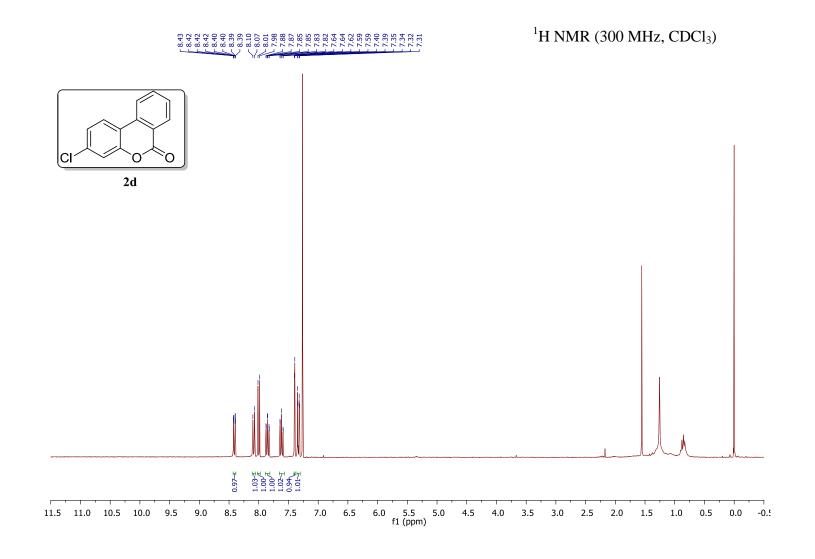






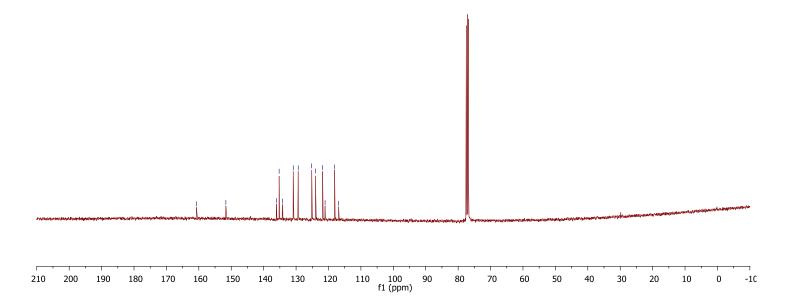


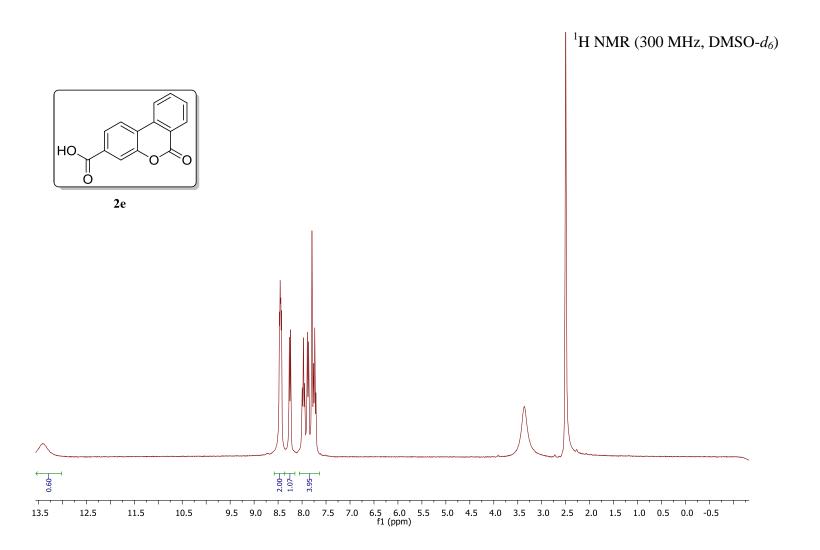


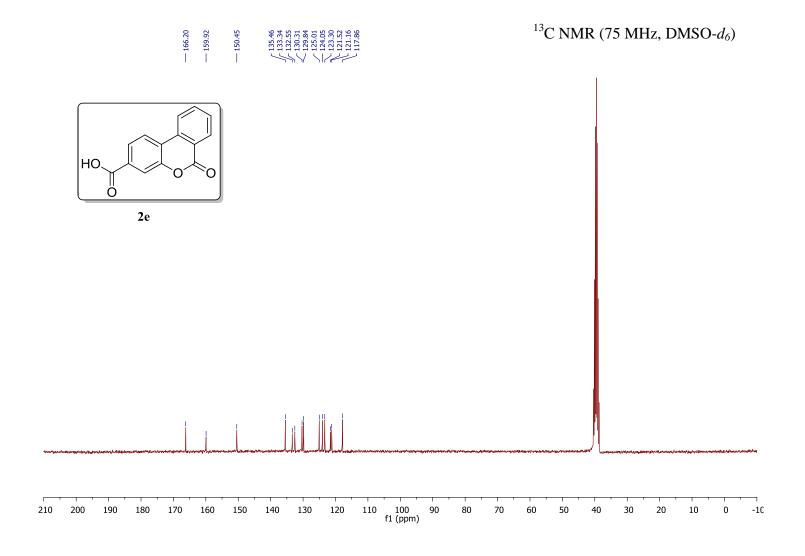


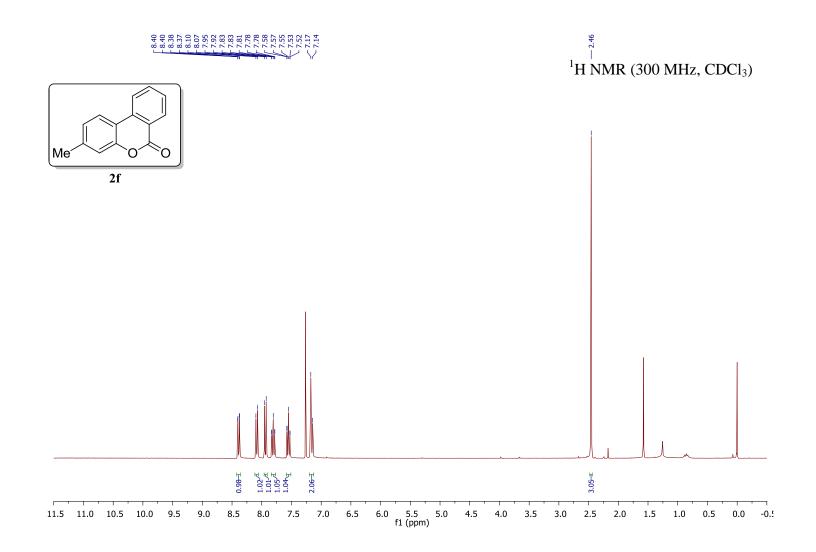


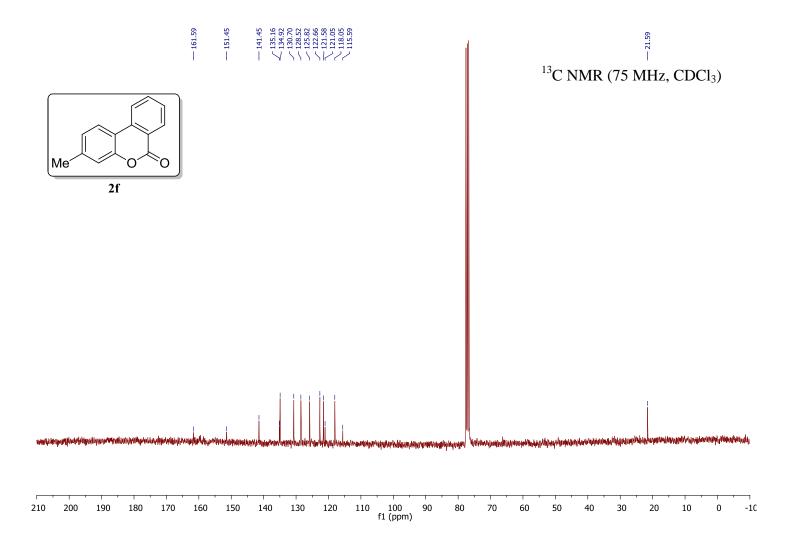


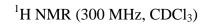




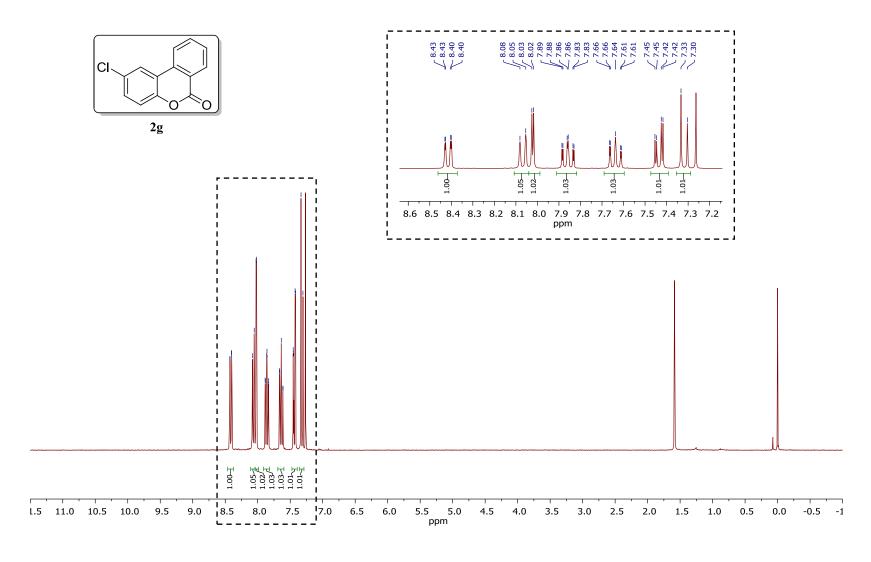


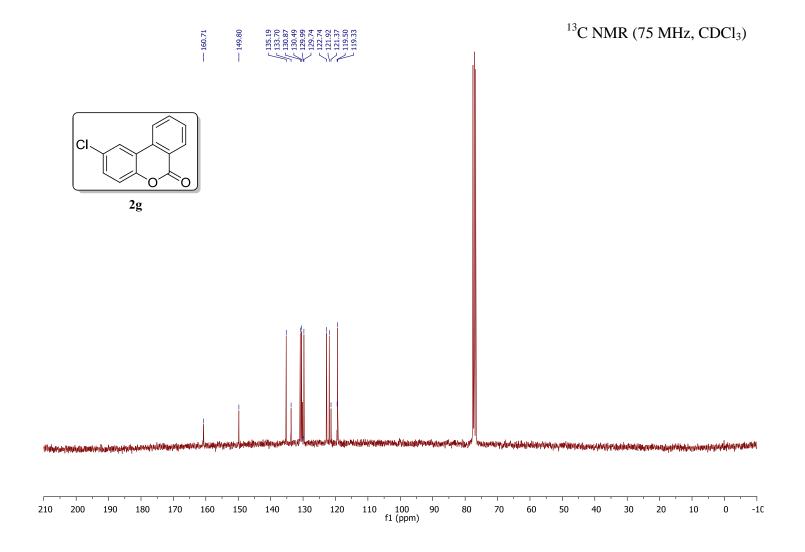






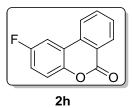


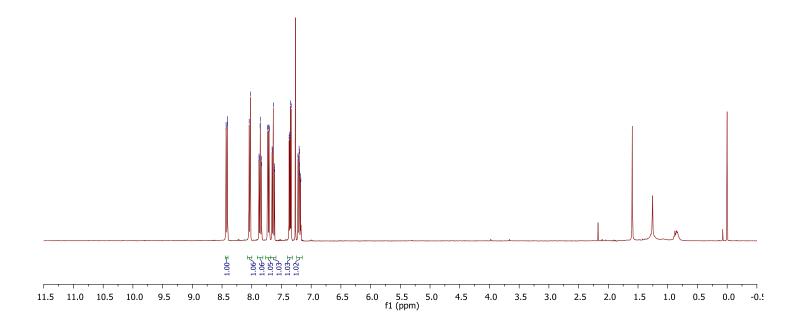






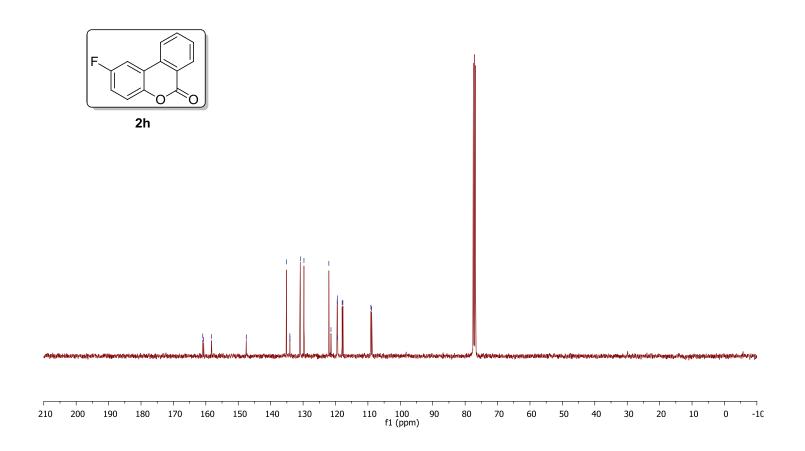
<sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)

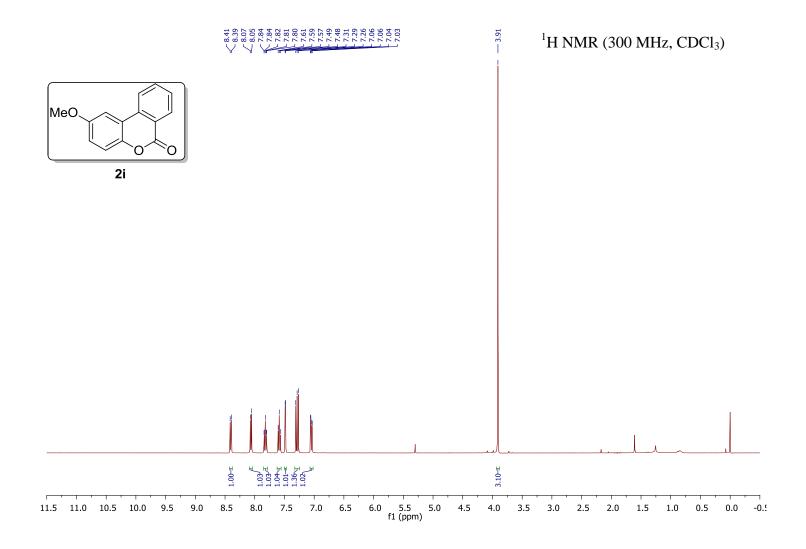


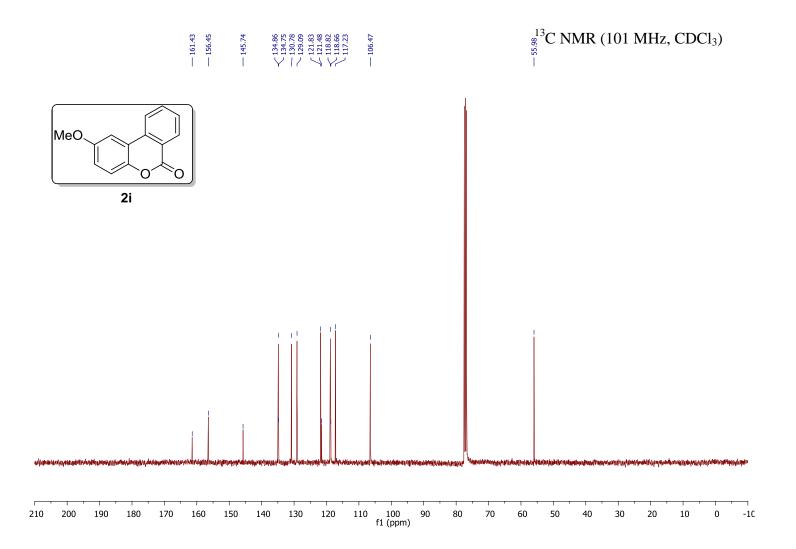


<sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>)

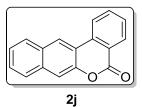


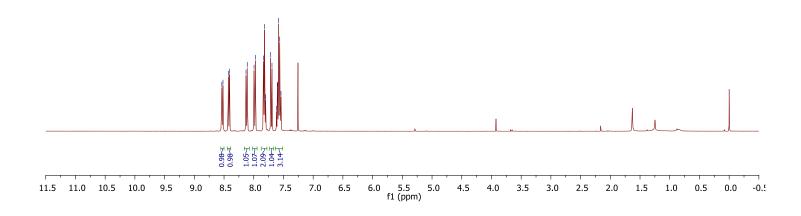


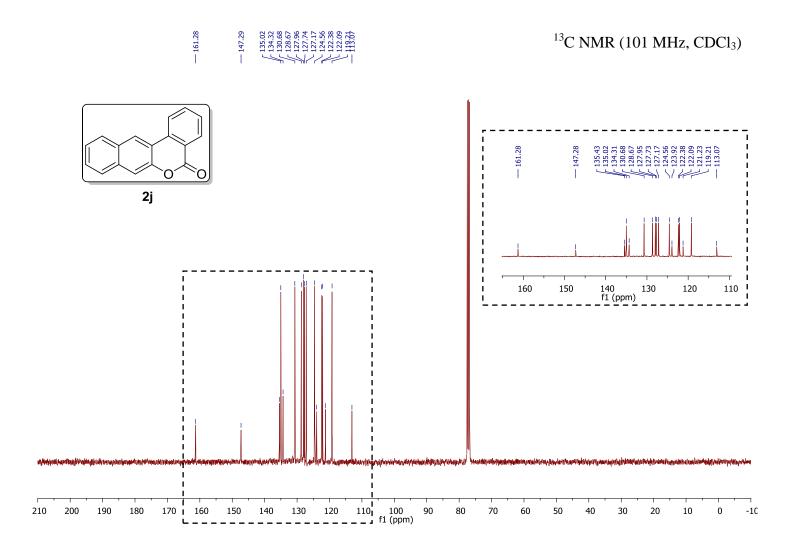






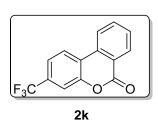


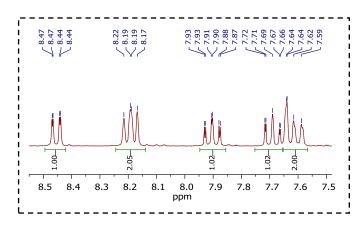


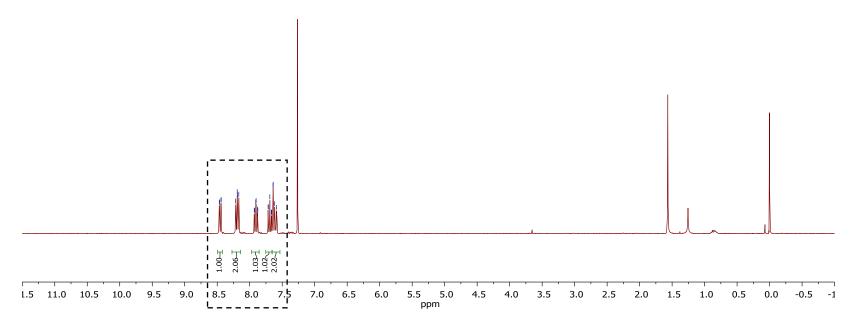


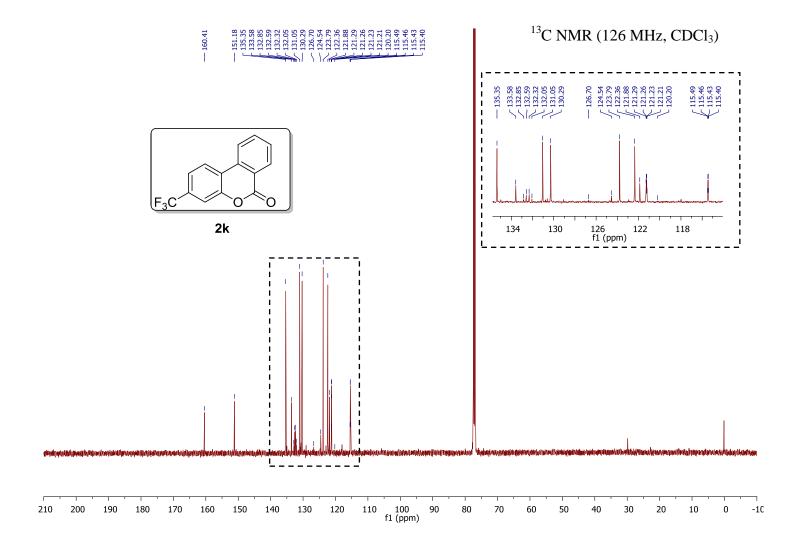


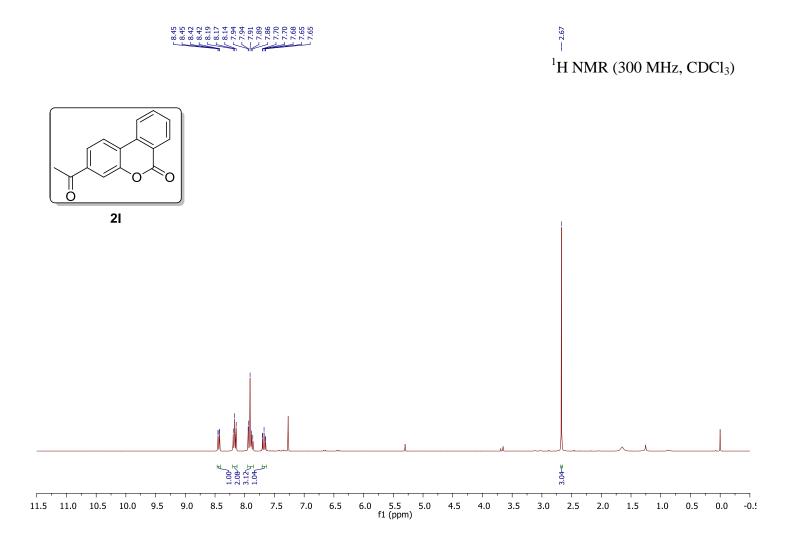
## <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)

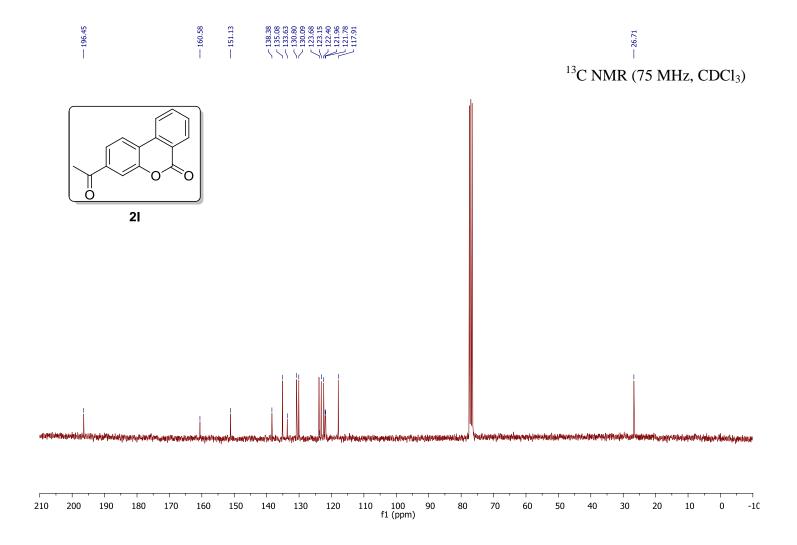


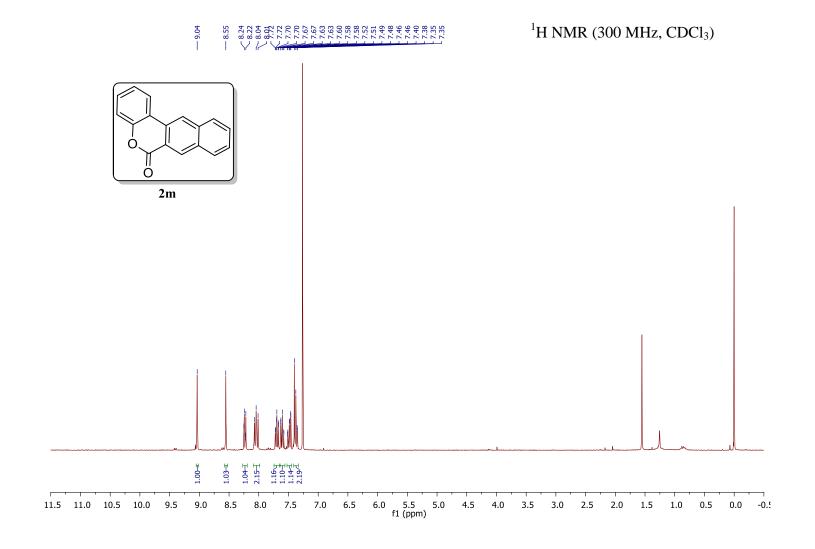


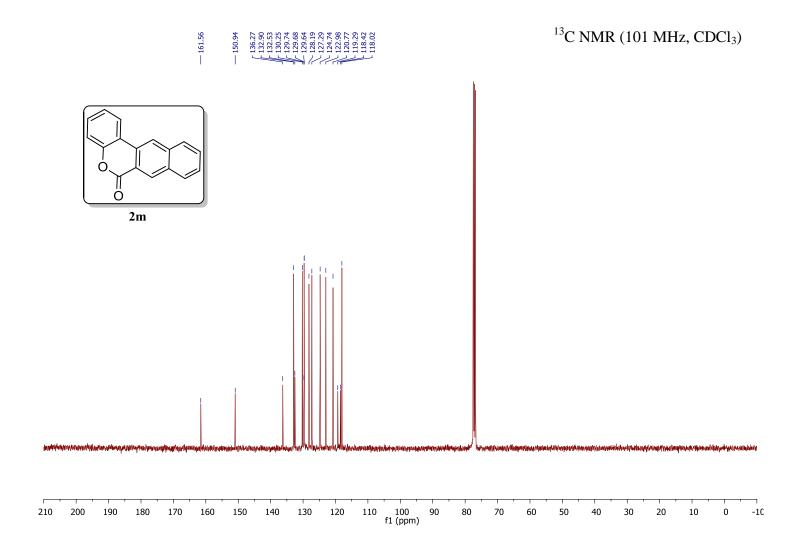




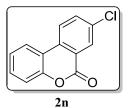


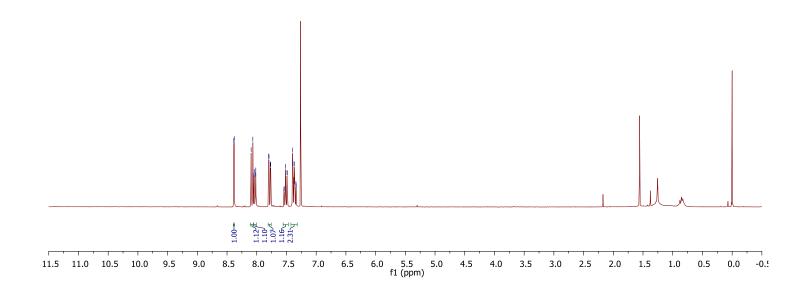


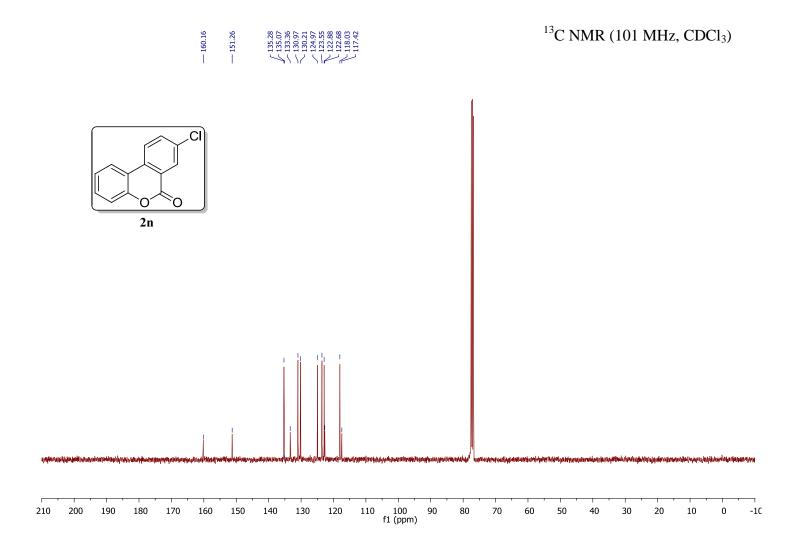




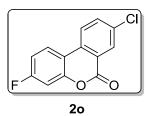


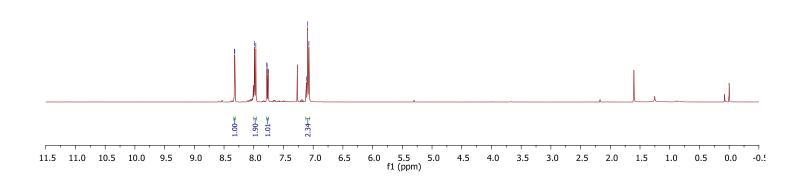


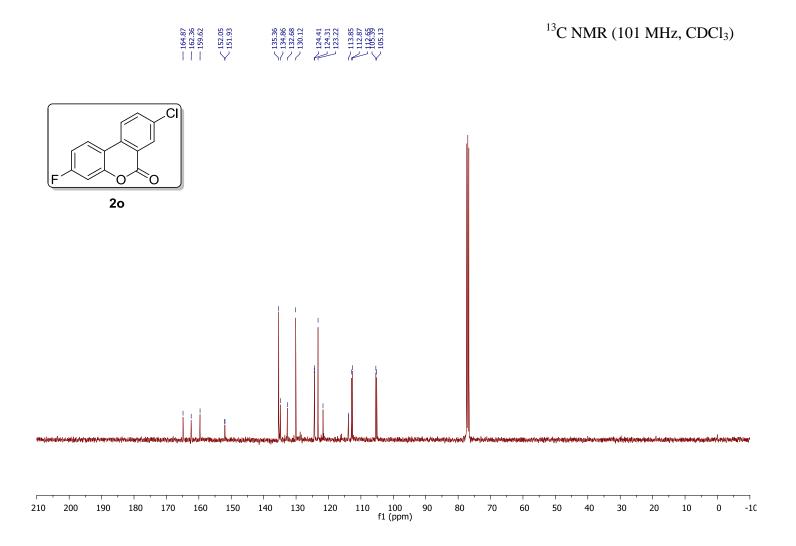


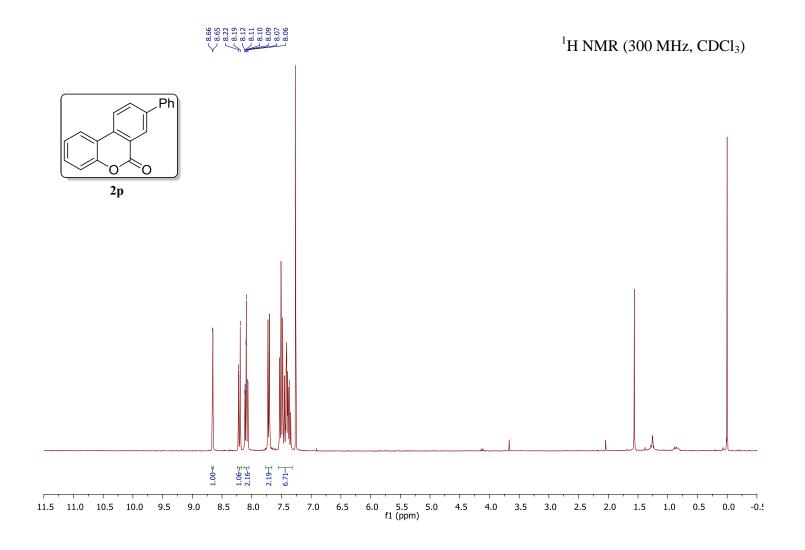


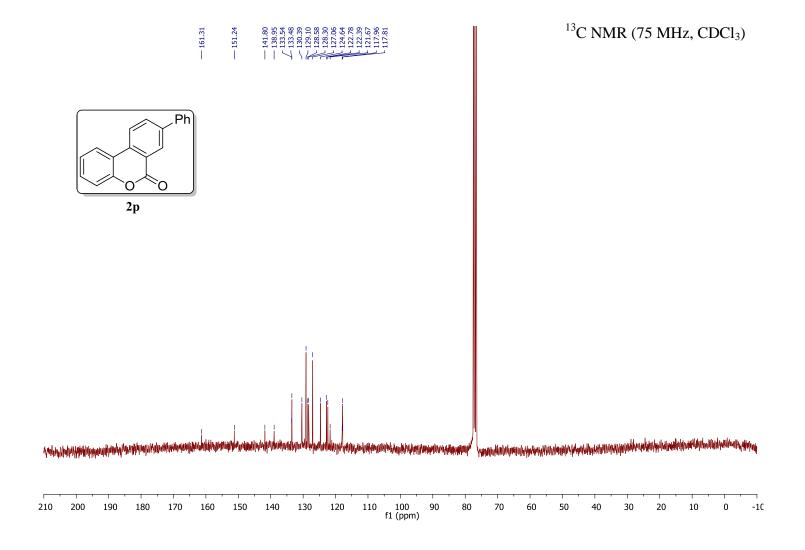


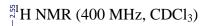




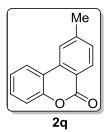


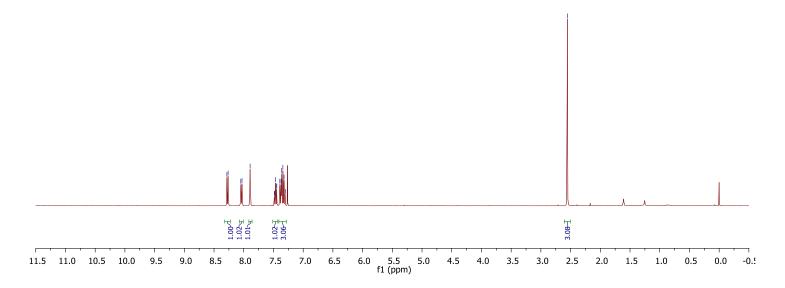


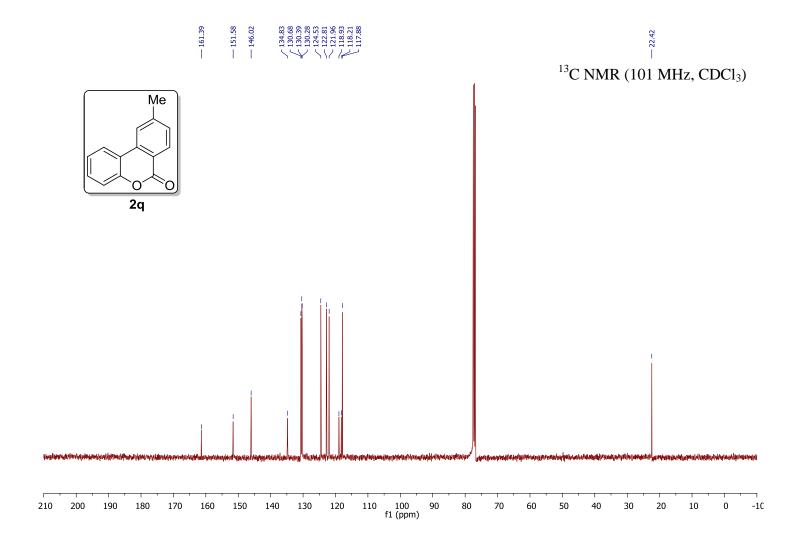


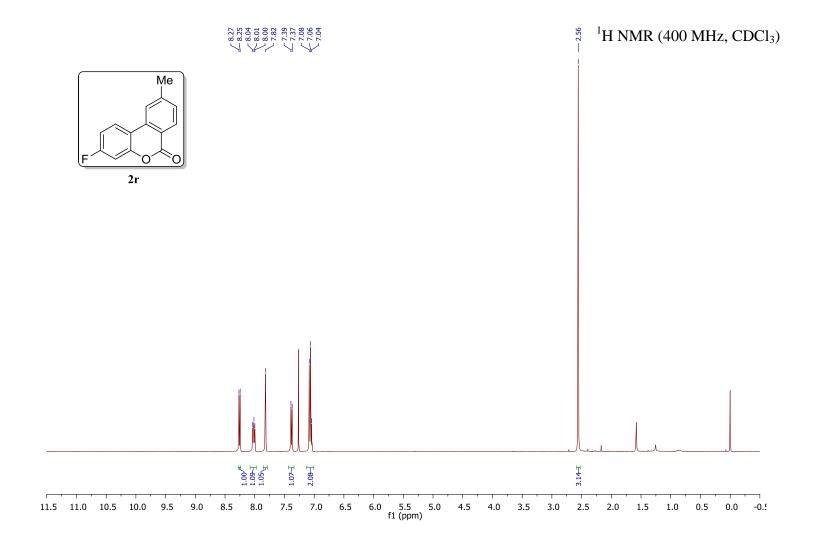






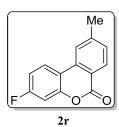


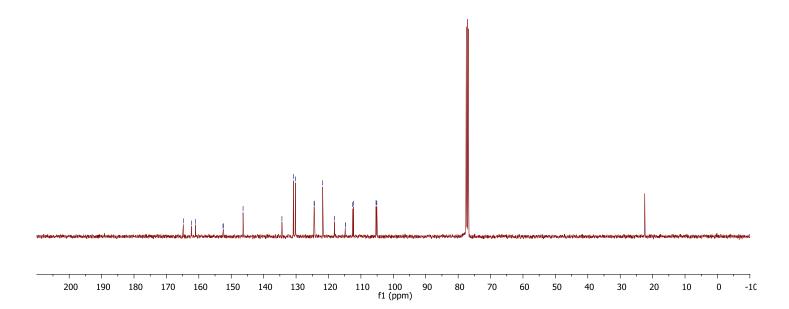


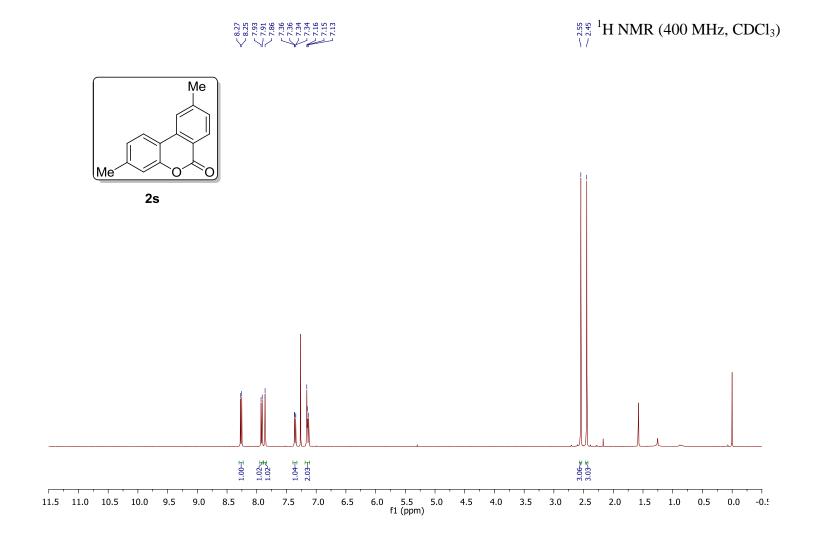


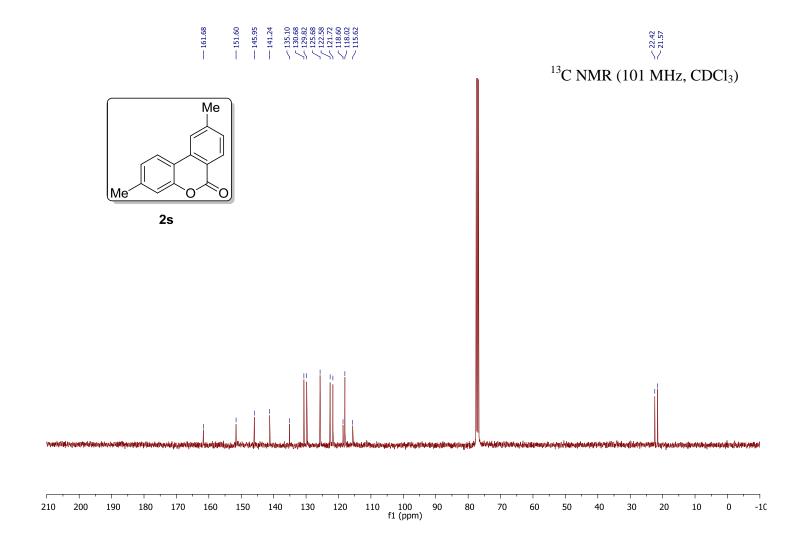


<sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>)





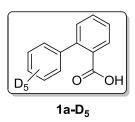


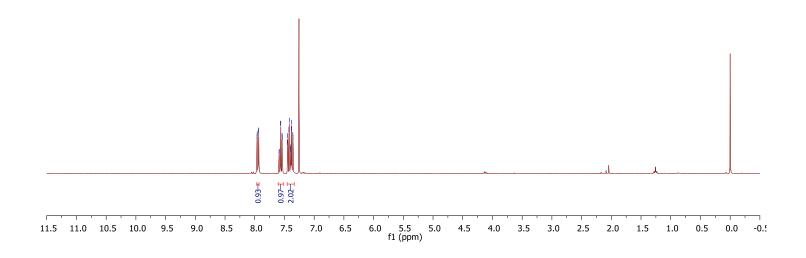


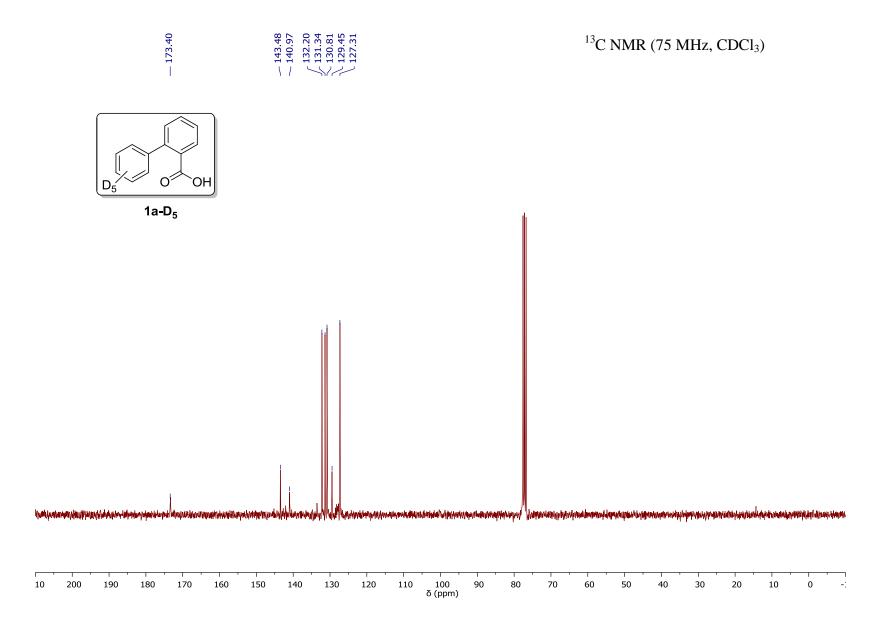
## 11. <sup>1</sup>H NMR and <sup>13</sup>C NMR Spectra of Deuterated Acid 1a-D<sub>5</sub> and Deuterated Benzo-3,4-coumarin 2a-D<sub>4</sub>



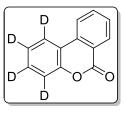
<sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)











2a-D<sub>4</sub>

