

Supporting Information

Fate and Transport of Phthalates in Indoor Environments and the Influence of Temperature: A Case Study in a Test House

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Table S1. Typical indoor temperature range

Country	Reference	Number of buildings	Building type	Temperature range (°C)
Korea	[1]	713	Multi-use facilities	21.8-24.6 (season)
Japan	[1]	213	Multi-use facilities	20.1-26 (season)
U.S.	[1]	807	Multi-use facilities	20.9-23.2 (season)
Turkey	[2]	119	Education	22-30 (year)
Nigeria	[3]	528	Residence	29-33 (day)
U.S.	[4]	356	Residence	18.2-27.2 (year)
U.K.	[5]	909	Office building	16-33 (season)
India	[6]	200	Residence	28-32 (season)

Table S2. Measured indoor air concentration and surface/gas partition coefficients^a.

		BBzP		DEHP	
		21 °C	30 °C	21 °C	30 °C
Indoor air					
Steady-state airborne concentration (ng/m ³)		149±63.7	419±64.8	110±77.1	300±116
Calculated gas-phase concentration ^b (ng/m ³)		123±52.6	391±60.5	46.0±32.1	209±80.9
Surfaces					
Mirror	K _{surf} (m)	33.5±22.0	5.78±1.04	288±227	135±55.9
	Log K _{film} ^c	10.2	9.46	11.2	10.8
Plate	K _{surf} (m)	86.9±42.9	21.8±4.24	582±433	253±148
	Log K _{film} ^c	10.6	10.0	11.5	11.1
Window	K _{surf} (m)	123±73.0	83.6±39.2	758±556	413±198
	Log K _{film} ^c	10.8	10.6	11.6	11.3
Dust	K _{dust} ^d (m ³ /g)	962±567	202±61.8	1970±1720	258±132

- a. The calculation of the standard deviations of the partition coefficients were based on the uncertainties of measurements in airborne concentrations and surface concentrations. For $K = \frac{A}{B}$, $\sigma_K = \frac{\bar{A}}{\bar{B}} \sqrt{\left(\frac{\sigma_A}{\bar{A}}\right)^2 + \left(\frac{\sigma_B}{\bar{B}}\right)^2}$
- b. The measured concentrations in indoor air are actually the total airborne concentrations; that is, the sum of the gas-phase (C_g) and particle-phase concentration (F), $C_g + F$. Based on the definition of particle/gas partition coefficient ($K_p = (F/TSP)/C_g$), the gas-phase concentration can be calculated using equation: $C_g = (C_g + F)/(1 + K_p TSP)$. Since $C_g + F$ and TSP were measured and the values of K_p were estimated in Table 1, we calculated gas-phase phthalate concentration (C_g) and used it to further obtain surface/gas partition coefficients (K_{surf}).
- c. Assuming the thickness of organic film is 10 nm and the film contains 20% organics, we calculated the dimensionless film/gas partition coefficients using equation 3.12 in ref 7.
- d. Non-floor settled dust. Caution should be take when comparing results, because the dust/gas partitioning may have not reached steady state at 30 °C due to the short waiting time (~several weeks).

Table S3. The values of Log K_{oa} in literature.

	Log K_{oa}					
	BBzP			DEHP		
	21°C	25°C	30°C	21°C	25°C	30°C
Cousins and Mackay (2000) ⁸	9.2 ^a	8.8	8.2 ^a	11.0 ^a	10.5	9.9 ^a
Weschler and Nazaroff (2008) ⁷	11.6 ^a	11.2	10.7 ^a	13.1 ^a	12.7	12.2 ^a
Xiao and Wania (2003) ⁹	10.6 ^b	10.4 ^b	10.1 ^b	11.5 ^b	11.3 ^b	11.0 ^b

- Based on the K_{oa} value of BBzP and DEHP reported in 25°C, the values at 21°C and 30°C were extrapolated with their corresponding vapor pressures at different temperatures.
- Calculated using Equation 6 in ref 9 with vapor pressure estimated using Equation 9 in ref 10.

Table S4. The values of dimensionless partition coefficients for fabric cloth (K_{cloth})

	Cotton		Polyester	
	21°C	30°C	21°C	30°C
BBzP	2.93×10^6	3.14×10^6	3.35×10^6	3.44×10^6
DEHP	2.00×10^6	1.83×10^6	2.17×10^6	1.99×10^6

Note: The cloth/gas partitioning possibly had not reached equilibrium for DEHP



Figure S1. Exterior, interior, and floor plan of the UTest House

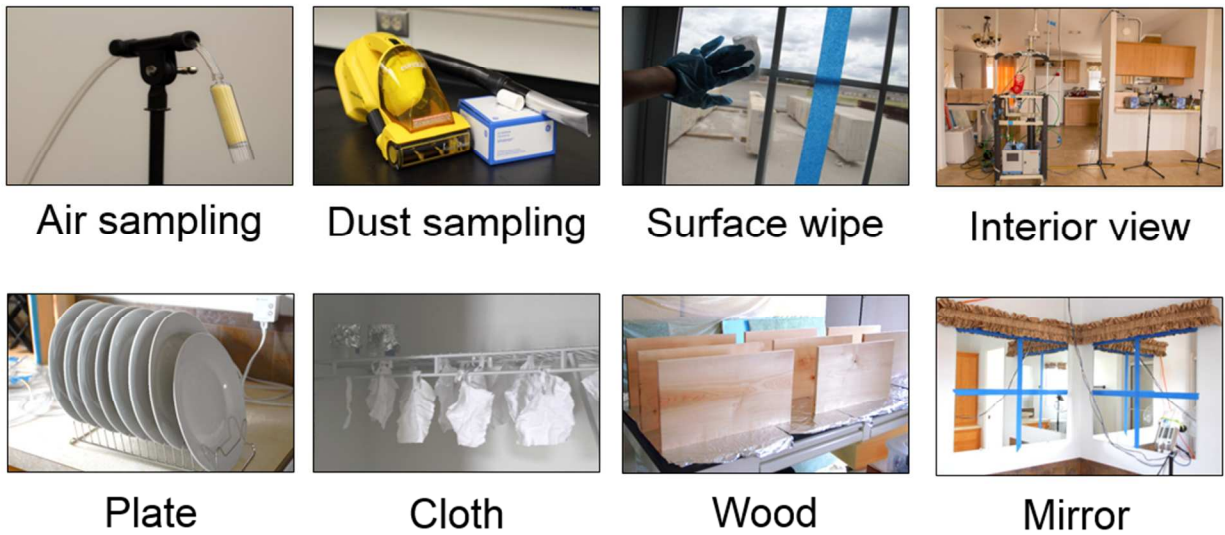


Figure S2. Photos of sampling method and materials

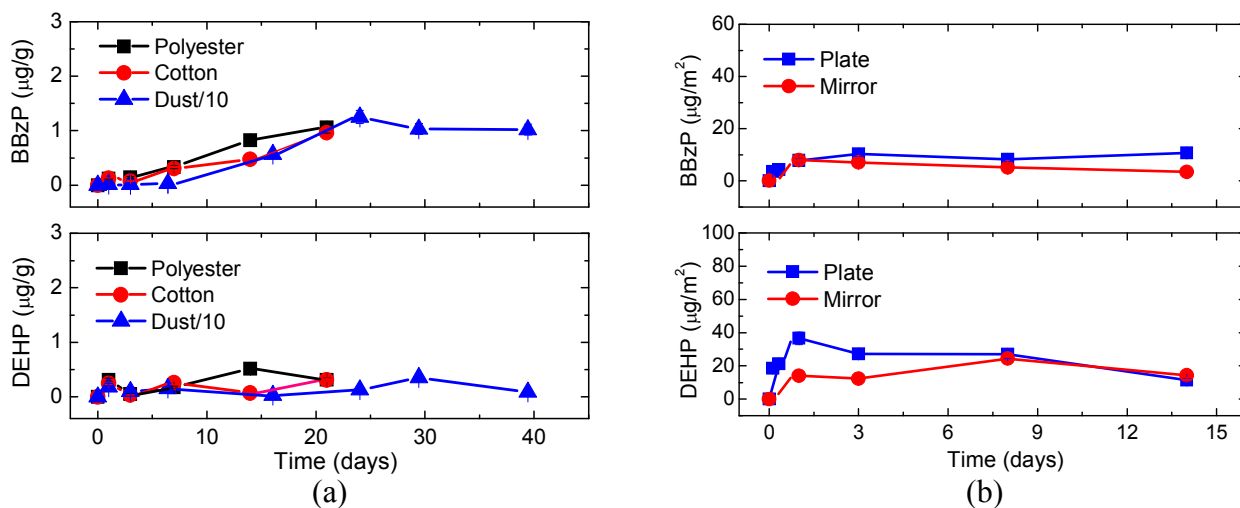


Figure S3. (a) Sorption kinetics for cloth and dust. (b) Sorption kinetics for plates and mirrors.

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