A New Method for the Rapid Extraction of Natural Products: Efficient Isolation of Shikimic Acid from Star Anise

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Supporting Information

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I. General

Plant material was ground using a Sunbeam spice/coffee bean grinder. Pressurized hot water extraction was undertaken employing a Breville Espresso Machine Model 800ES. Unless otherwise specified, reactions were conducted with magnetic stirring in oven-dried glassware in air. Flash chromatography was carried out using a Reveleris® X2 Flash Chromatography System using silica gel cartridges. All NMR spectra were recorded on a Bruker NMR spectrometer operating at 400 MHz (¹H) and 100 MHz (¹³C).

II. Isolation of Shikimic Acid and Synthesis of Derivatives

Shikimic acid (1)

Dried Chinese star anise (*Illicium verum*) pods (20 g) were finely ground using a coffee bean grinder. This material was mixed with sand (2 g), placed into the portafilter (sample compartment) of an espresso machine and extracted using 30% EtOH/H₂O (200 mL of a hot solution) (Figure S1). This was repeated with another sample of star anise (20 g). The

⁽¹⁾ The ethanol content was not increased beyond 30% in order to minimize the flammability of the solvent mixture.

extracts were combined and silica gel (20 g) was added. The ensuing suspension was evaporated to dryness on a rotary evaporator (50 °C water bath temperature). The ensuing brown solid was then loaded onto a sintered glass funnel and washed with CH_2Cl_2 (150 mL) then EtOAc (150 mL) in order to remove anethole and p-anisaldehyde (see Appendix I for the 1H NMR spectrum of the residue obtained from the evaporation of the combined CH_2Cl_2 and EtOAc washings). These washings were discarded. The resulting solid was then extracted with 10% AcOH/EtOAc (250 mL) and the filtrate was evaporated to dryness (see Appendix II for the 1H NMR spectrum of this residue). The ensuing solid was then washed with CH_2Cl_2 and then dried under vacuum to provide the title compound as an off-white solid (2.21 g, 5.50 % w/w) (Figure S2).

¹H NMR (400 MHz, D₂O) δ 6.81 (m, 1H), 4.47 (m, 1H), 4.05 (m, 1H), 3.80 (dd, J = 8.2, 4.2 Hz, 1H), 2.77 (ddt, J = 18.1, 5.2, 1.4 Hz, 1H), 2.24 (ddt, J = 18.2, 6.5, 1.8 Hz, 1H). ¹³C NMR (100 MHz, D₂O) δ 171.1, 136.8, 130.8, 71.5, 66.9, 66.1, 30.9.



Figure S1: Extraction of star anise using an espresso machine.



Figure S2: Isolated shikimic acid.

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⁽²⁾ It is well-established that anethole and *p*-anisaldehyde are present in star anise. See: Sy, L.-K.; Brown, G. D. *J. Nat. Prod.* **1998**, *61*, 987 and references cited therein.

⁽³⁾ Payne, R.; Edmonds, M. J. Chem. Educ. 2005, 82, 599.

Methyl shikimate (4)

Dried Chinese star anise (*Illicium verum*) pods (20 g) were finely ground using a coffee bean grinder. This material was mixed with sand (2 g), placed into the portafilter of an espresso machine and extracted using 30% EtOH/ H_2O (200 mL of a hot solution). This was repeated with another sample of star anise (20 g). The extracts were combined and were reduced to 200 mL on a rotary evaporator (to remove EtOH). The ensuing aqueous phase was washed with CH_2Cl_2 (3 x 80 mL) and the organic phases were discarded. The aqueous layer was then evaporated to dryness on a rotary evaporator (50 °C water bath temperature) to yield a red/brown sticky solid. MeOH (100 mL) was added to the ensuing residue and then $SOCl_2$ (3 mL) was added dropwise. The mixture was heated at reflux for 1.5 h, cooled and then the solvent was a removed on rotary evaporator. The resulting dark red/black oil was dissolved in 30% MeOH/EtOAc, filtered through a silica plug eluting with 30% MeOH/EtOAc (300 mL) and the solvent removed on rotary evaporator. Automated gradient flash chromatography (silica; MeOH/EtOAc elution) provided the title compound as a light-pink solid (2.03 g, equivalent to 4.7 % yield of shikimic acid w/w).⁴

¹H NMR (400 MHz, D_2O) δ 6.72 (m, 1H), 4.35 (m, 1H), 3.93 (m, 1H), 3.68 (m, 1H), 3.66 (s, 3H), 2.65 (dd, J = 18.2, 5.0 Hz, 1H), 2.14 (dd, J = 18.3, 6.3 Hz, 1H).

¹³C NMR (100 MHz, DMSO-d₆) δ 166.7, 139.8, 127.3, 70.0, 66.8, 65.4, 51.5, 29.6.

Ethyl shikimate (5)

Dried Chinese star anise (*Illicium verum*) pods (20 g) were finely ground using a coffee bean grinder. This material was mixed with sand (2 g), placed into the portafilter of an espresso machine and extracted using 30% EtOH/H₂O (200 mL of a hot solution). This was repeated with another sample of star anise (20 g). The extracts were combined and were reduced to 200 mL on a rotary evaporator (to remove EtOH). The ensuing aqueous phase was washed with CH₂Cl₂ (3 x 80 mL) and the organic phases were discarded. The aqueous layer was then evaporated to dryness on a rotary evaporator (50 °C water bath temperature) to yield a red/brown sticky solid. EtOH (100 mL) was added to the ensuing residue and then SOCl₂ (3 mL) was added dropwise. The mixture was heated at reflux for 1.5 h, cooled and then the solvent was a removed on rotary evaporator. The resulting dark red/black oil was dissolved in 30% MeOH/EtOAc, filtered through a silica plug eluting with 30% MeOH/EtOAc (300 mL) and the solvent removed on rotary evaporator. Automated

(4) Wu, W.; Zou, Y.; Chen, Y.; Li, J.; Lv, Z.; Wei, W.; Huang, T.; Liu, X. Green Chem. 2012, 14, 363.

gradient flash chromatography (silica; MeOH/EtOAc elution) provided the title compound as a light-brown solid (2.49 g, equivalent to 5.4% yield of shikimic acid w/w).⁵

¹H NMR (400 MHz, DMSO-d₆) δ 6.62 (m, 1H), 4.82 (m, 2H, OH), 4.61 (d, J = 4.3 Hz, 1H, OH), 4.23 (m, 1H), 4.12 (q, J = 7.1 Hz, 2H), 3.86 (m, 1H), 3.57 (m, 1H), 2.43 (m, 1H), 2.05 (m, 1H), 1.22 (t, J = 7.1 Hz, 3H).

¹³C NMR (100 MHz, DMSO-d₆) δ 166.7, 140.0, 128.1, 70.5, 67.3, 65.9, 60.5, 30.1, 14.6.

Methyl 3,4-O-isopropylideneshikimate (6)

Dried Chinese star anise (Illicium verum) pods (15 g) were finely ground using a coffee bean grinder. This material was mixed with sand (2 g), placed into the portafilter of an espresso machine and extracted using 30% EtOH/H₂O (200 mL of a hot solution). This was repeated with a further 5 samples of star anise (15 g). The extracts were combined and were reduced to 800 mL on a rotary evaporator (to remove EtOH). The ensuing aqueous phase was washed with CH₂Cl₂ (4 x 80 mL) and the organic phases were discarded. The aqueous layer was then evaporated to dryness on a rotary evaporator (50 °C water bath temperature) to yield a red/brown sticky solid. MeOH (300 mL) was added to the ensuing residue and then SOCl₂ (5 mL) was added dropwise. The mixture was heated at reflux for 1.5 h, cooled and then the solvent was a removed on rotary evaporator. Acetone (100 mL), 2,2dimethoxypropane (80 mL), and Amberlyst® 15 acidic ion exchange resin (0.8 g) were added to ensuing dark red/black oil. After 3 h at r.t., the mixture was filtered and evaporated to dryness to provide a dark red/black residue which was then dissolved in CH₂Cl₂ (100 mL) and loaded onto a silica plug (4.5 inch diameter, 1.5 inch depth). 7 fractions were collected: $(20\% \text{ EtOAc/hexanes} (5 \times 100 \text{ mL}), \text{ EtOAc} (1 \times 250 \text{ mL}, 1 \times 100 \text{ mL}). \text{ The}$ title compound was present in the final 2 fractions. After the solvent was removed on a rotary evaporator, automated gradient flash chromatography (silica; EtOAc/hexane elution) provided the title compound as a red oil (4.76 g, equivalent to 4.0 % yield of shikimic acid w/w).6

 1 H NMR (400 MHz, CDCl₃) δ 6.91 (m, 1H), 4.74 (m, 1H), 4.08 (m, 1H), 3.89 (m, 1H), 3.76 (s, 3H), 2.78 (dd, J = 17.46, 4.7 Hz, 1H), 2.22 (ddt, J = 17.4, 8.4, 1.8 Hz, 1H), 2.20 (br s, 1H, OH), 1.44 (s, 3H), 1.39 (s, 3H).

¹³C NMR (100 MHz, CDCl₃) δ 166.8, 134.2, 130.9, 110.0, 78.2, 72.5, 69.1, 52.4, 29.6, 28.2, 26.0.

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⁽⁵⁾ Kalashnikov, A. I.; Sysolyatin, S. V.; Sakovich, G. V.; Sonina, E. G.; Shchurova, I. A. Russ. Chem. Bull. 2014, 62, 163.

⁽⁶⁾ Tavassoli, A.; Duffy, J. E. S.; Young, D. W. Org. Biomol. Chem. 2006, 4. 569.

Ethyl 3,4-O-(3-pentylidene)shikimate (7)

Dried Chinese star anise (*Illicium verum*) pods (20 g) were finely ground using a coffee bean grinder. This material was mixed with sand (2 g), placed into the portafilter of an espresso machine and extracted using 30% EtOH/H₂O (200 mL of a hot solution). This was repeated with a further 9 samples of star anise (20 g). The extracts were combined and were reduced to 800 mL on a rotary evaporator (to remove EtOH). The ensuing aqueous phase was washed with CH₂Cl₂ (4 x 100 mL) and the organic phases were discarded. The aqueous layer was then evaporated to dryness on a rotary evaporator (50 °C water bath temperature) to yield a red/brown sticky solid. EtOH (300 mL) was added to the ensuing residue and then SOCl₂ (10 mL) was added dropwise. The mixture was heated at reflux for 1.5 h, cooled and then the solvent was a removed on rotary evaporator. 3-Pentanone (200 mL), and Amberlyst® 15 acidic ion exchange resin (1.6 g) were added to ensuing dark red/black oil. After 2.5 h at r.t., the mixture was evaporated to dryness to provide a dark red/black residue which was then dissolved in CH₂Cl₂ (100 mL) and loaded onto a silica plug (4.5 inch diameter, 2 inch depth). 7 fractions were collected: (30% EtOAc/hexanes (2 x 150 mL), 50% EtOAc/hexanes (2 x 200 mL), EtOAc (1 x 100 mL). Evaporation of fractions 2–4 provided the title compound as a yellow oil (13.15 g, equivalent to 4.2 % yield of shikimic acid w/w). ¹H NMR (400 MHz, CDCl₃) δ 6.93 (m, 1H), 4.75 (m, 1H), 4.21 (q, J = 7.1 Hz, 2H), 4.11 (m, 1H), 3.91 (m, 1H), 2.77 (dd, J = 17.3, 4.6 Hz, 1H), 2.25 (ddt, J = 17.3, 8.2, 1.9 Hz, 1H), 1.67 (q, J = 7.5Hz 4H), 1.65 (q, I = 7.5 Hz, 2H), 1.29 (t, I = 7.1 Hz, 3H), 0.92 (t, I = 7.5 Hz, 3H), 0.88 (t, I = 7.5Hz, 3H).

 $^{13}\text{C NMR}$ (100 MHz, CDCl₃) δ 166.4, 134.3, 130.6, 113.9, 78.1, 72.5, 69.2, 61.3, 29.9, 29.5, 29.3, 14.4, 8.7, 8.1.

Methyl 3,4-O-benzylideneshikimate (8)

Dried Chinese star anise (*Illicium verum*) pods (20 g) were finely ground using a coffee bean grinder. This material was mixed with sand (2 g), placed into the portafilter of an espresso machine and extracted using 30% EtOH/H₂O (200 mL of a hot solution). This was repeated with another sample of star anise (20 g). The extracts were combined and were reduced to 200 mL on a rotary evaporator (to remove EtOH). The ensuing aqueous phase was washed with CH₂Cl₂ (3 x 80 mL) and the organic phases were discarded. The aqueous layer was then evaporated to dryness on a rotary evaporator (50 °C water bath temperature) to yield a

⁽⁷⁾ Carr, R.; Ciccone, F.; Gabel, R.; Guinn, M.; Johnston, D.; Mastriona, J.; Vandermeer, T.; Groaning, M. *Green Chem.* **2008**, *10*, 743.

red/brown sticky solid. MeOH (100 mL) was added to the ensuing residue and then $SOCl_2$ (3 mL) was added dropwise. The mixture was heated at reflux for 1.5 h, cooled and then the solvent was a removed on rotary evaporator. THF (100 mL), benzaldehyde (3 mL), and Amberlyst® 15 acidic ion exchange resin (1.6 g) were added to ensuing black oil. After 2 h at r.t., the mixture was evaporated to dryness to provide a black residue which was then and loaded onto a silica plug (4.5 inch diameter, 1.5 inch depth). 4 fractions were collected: 50% EtOAc/hexanes (3 x 150 mL), 70% EtOAc/hexanes (1 x 150 mL). The title compound was present in fractions 2–4. After the solvent was removed on a rotary evaporator, automated gradient flash chromatography (silica; EtOAc/hexane elution) provided the title compound as a mixture of diastereoisomers as a yellow oil (2.34 g, equivalent to 3.5 % yield of shikimic acid w/w). These diastereoisomers were partially separable by chromatography.

(R)-Methyl 3,4-O-benzylideneshikimate

¹H NMR (400 MHz, CDCl₃) δ 7.43 (m, 2H), 7.37 (m, 3H), 7.00 (m, 1H), 5.94 (s, 1H), 4.80 (m, 1H), 4.18 (m, 1H), 3.93 (m, 1H), 3.78 (s, 3H), 2.83 (dd, J = 17.4, 4.8 Hz, 1H), 2.27 (ddt, J = 17.5, 9.0, 1.9 Hz, 1H).

¹³C NMR (100 MHz, CDCl₃) δ 166.6, 136.6, 133.0, 131.7, 129.9, 128.7, 127.0, 104.6, 78.8, 74.2, 69.3, 52.4, 29.7.

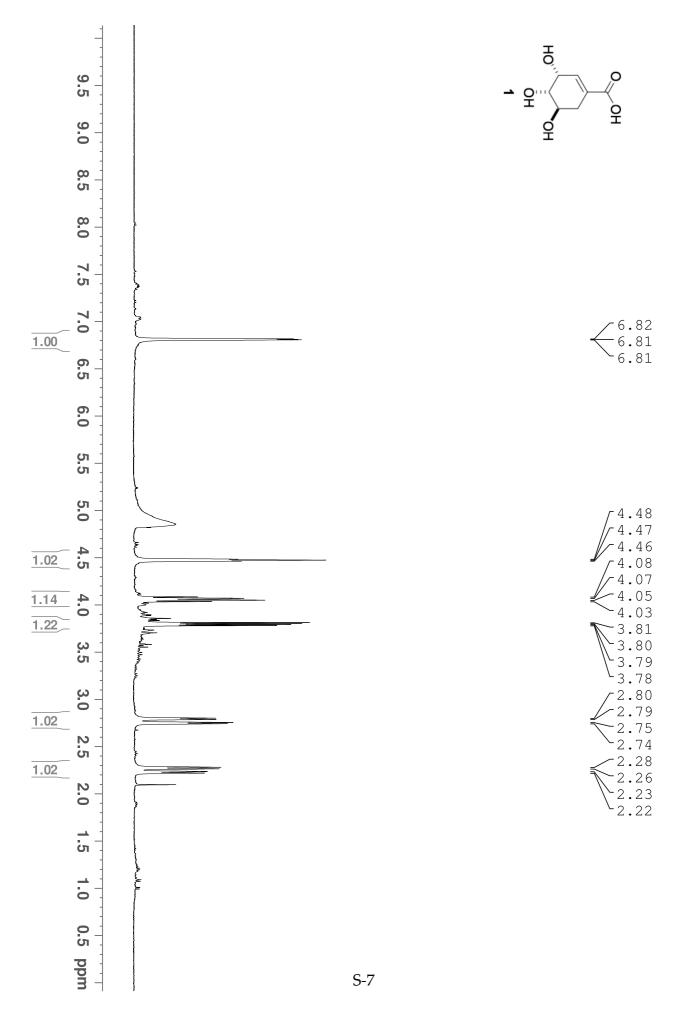
(S)-Methyl 3,4-O-benzylideneshikimate

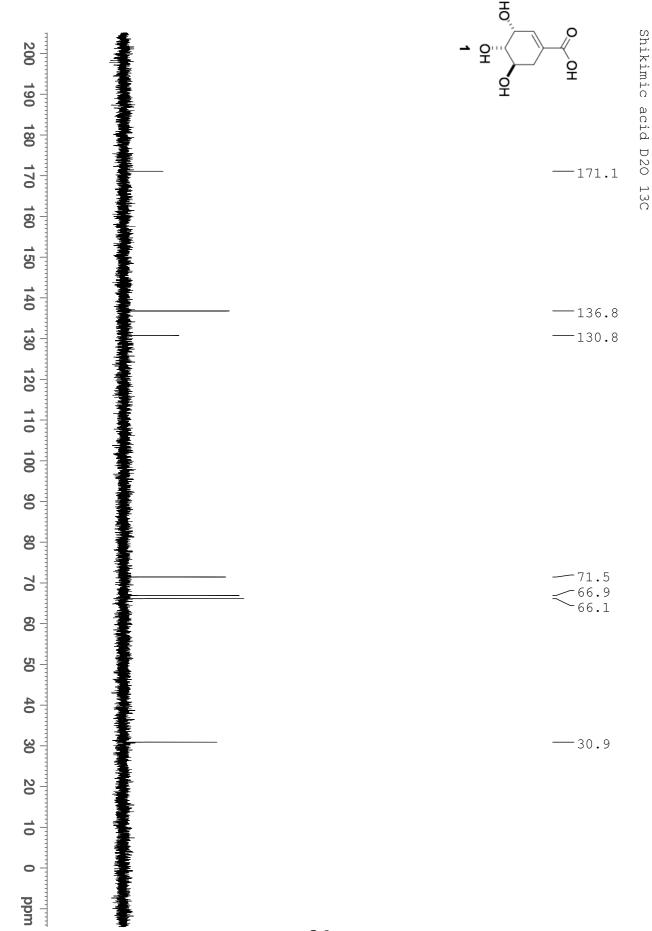
¹H NMR (400 MHz, CDCl₃) δ 7.45 (m, 2H), 7.38 (m, 3H), 6.93 (m, 1H), 5.97 (s, 1H), 4.87 (m, 1H), 4.34 (m, 1H), 4.14 (m, 1H), 3.79 (s, 3H), 2.84 (dd, J = 17.3, 4.5 Hz, 1H), 2.34 (ddt, J = 17.5, 7.2, 1.8 Hz, 1H).

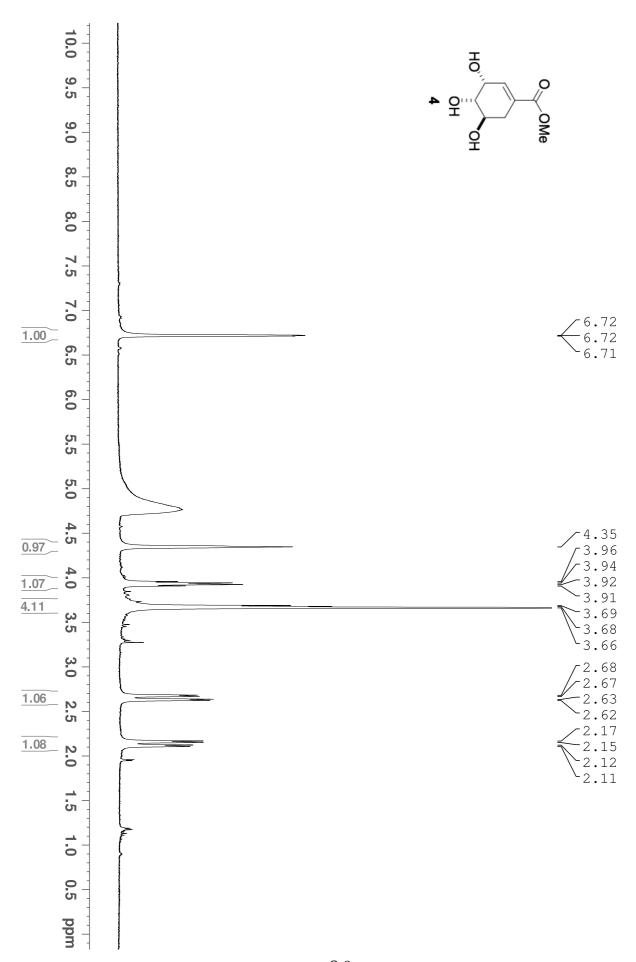
¹³C NMR (100 MHz, CDCl₃) δ 166.7, 137.9, 133.7, 131.3, 129.5, 128.6, 126.5, 102.9, 78.1, 72.7, 67.3, 52.4, 29.1.

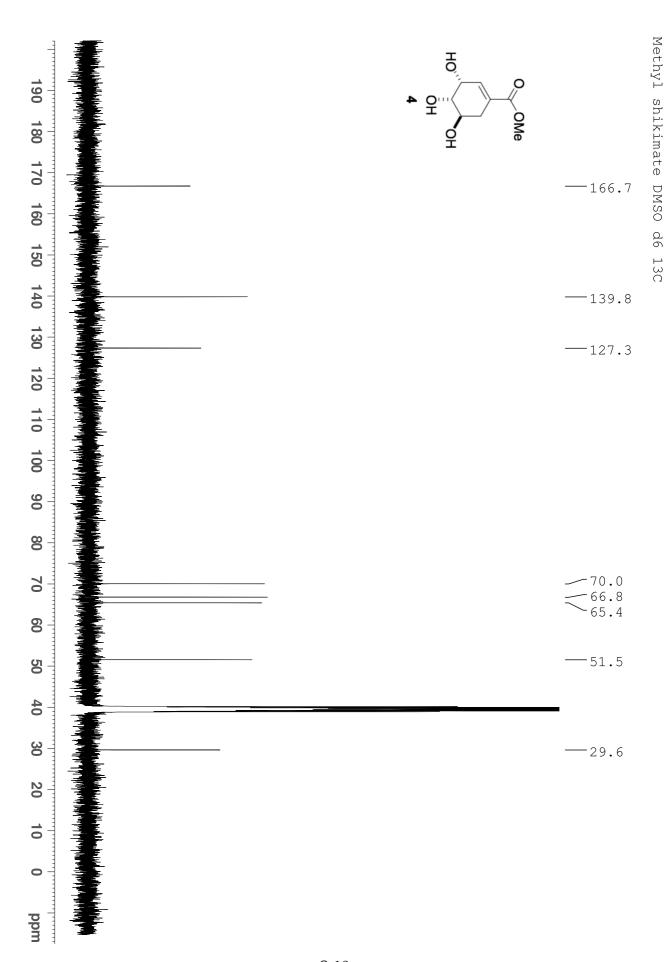
S-6

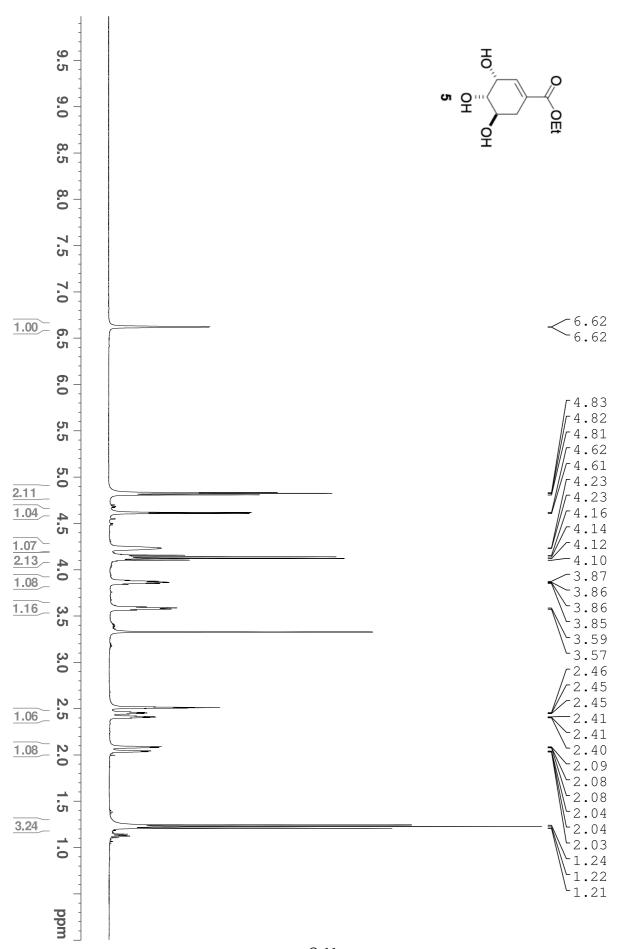
⁽⁸⁾ Adams, H.; Bailey, N. A.; Brettle, R.; Cross, R.; Frederickson, M.; Haslam, E.; MacBeath, F. S.; Davies, G. M. *Tetrahedron* **1996**, *52*, 8565.

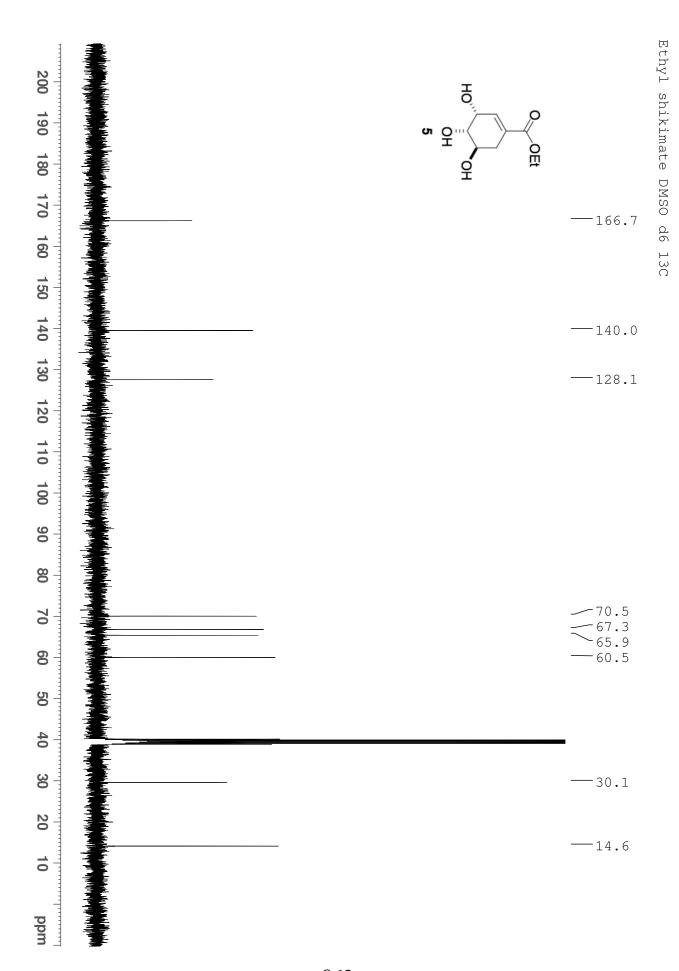












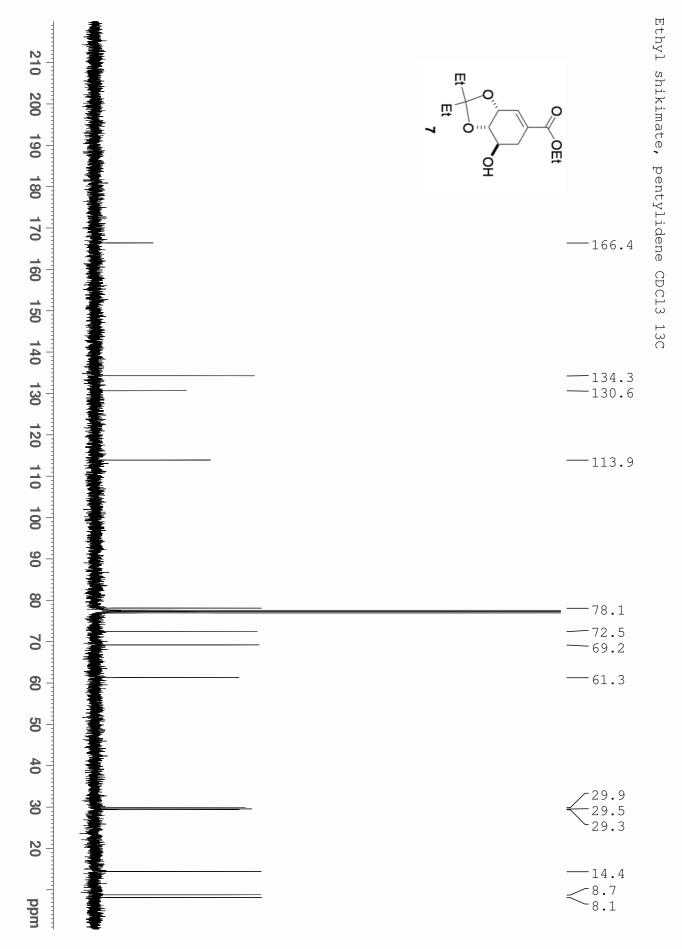
Methyl shikimate, propylidene CDC13

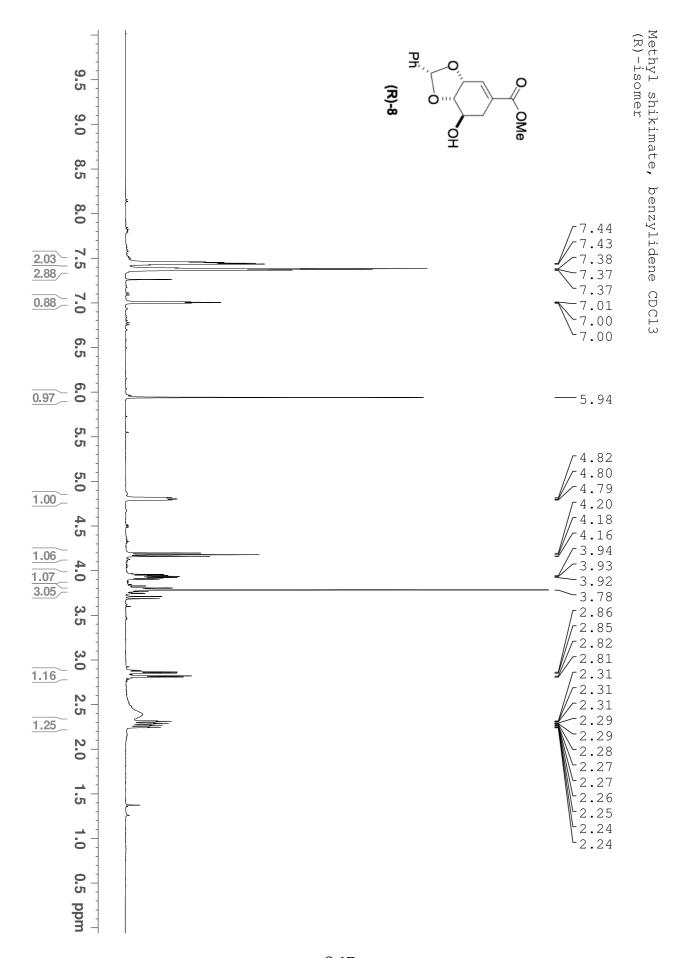
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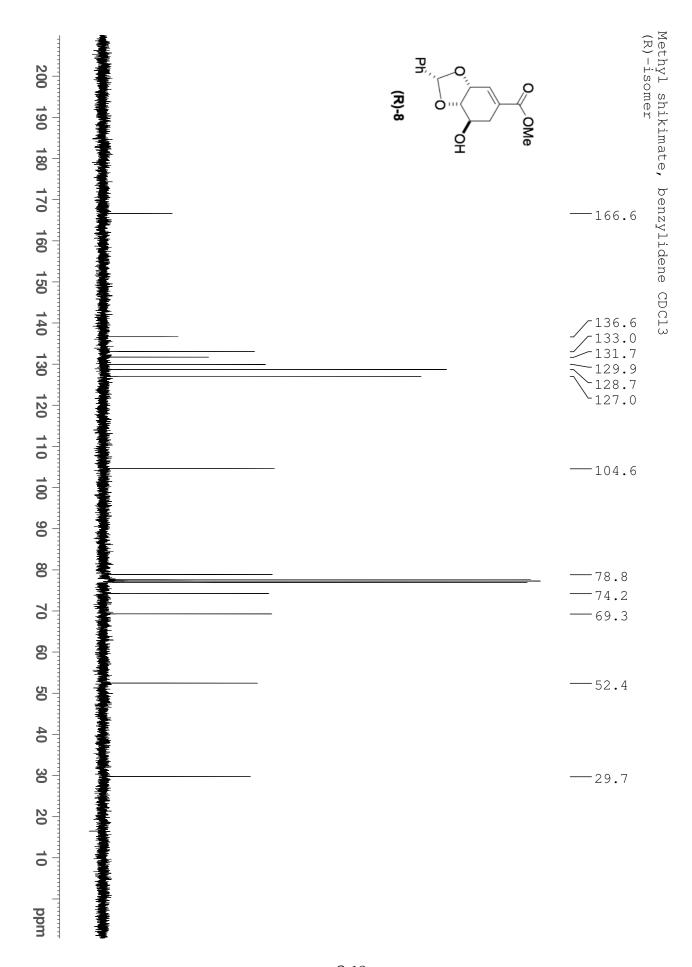
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S-14

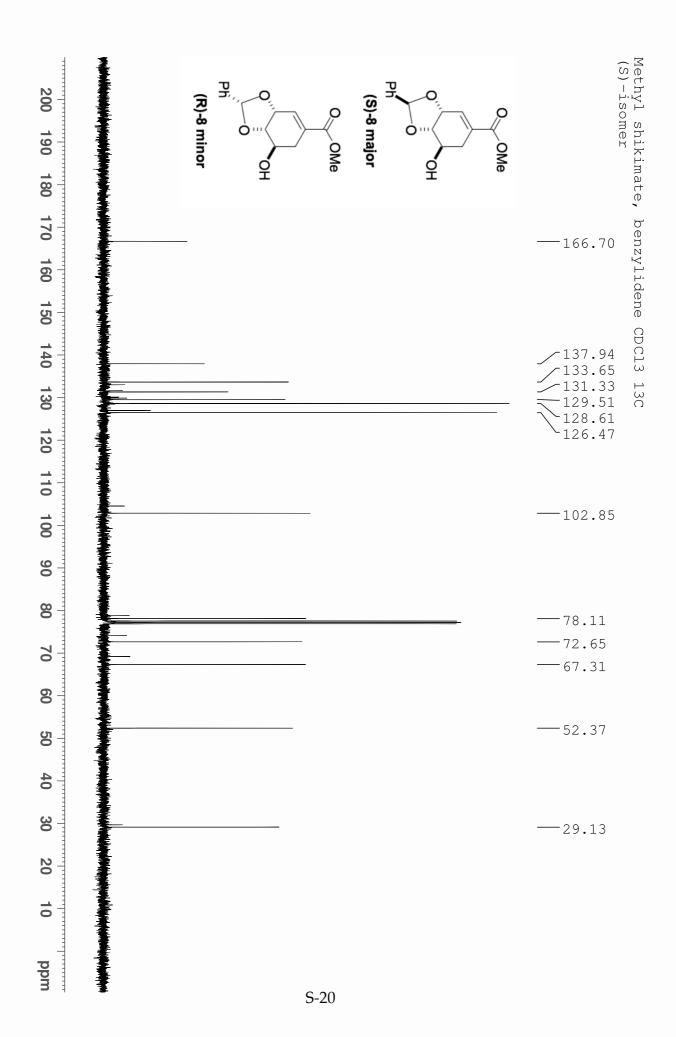
S-15







Methyl shikimate, benzylidene CDC13 (S)-isomer (R)-8 minor 9.5 (S)-8 major OMe 9.0 <u>8</u> 7.46 7.45 7.45 7.39 7.38 7.37 7.37 6.94 6.93 6.93 7.5 2.17 7.0 0.85 <u>6.</u>5 5.97 4.88 4.87 4.87 4.86 4.86 4.85 4.34 4.32 4.16 4.13 4.12 4.11 3.79 2.87 2.81 2.37 2.35 2.33 2.31 0.84 <u>5</u>.5 5.0 0.92 4.5 0.95 1.00 4.0 2.98 <u>3.5</u> 3.0 1.09 2.5 1.06 2.0 1.5 1.0 0.5 ppm



Star anise extract combined CH2Cl2, EtOAC washings CDCl3

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