

Supporting Information

Aerobic Oxidation of Formaldehyde Catalyzed by Polyvanadotungstates

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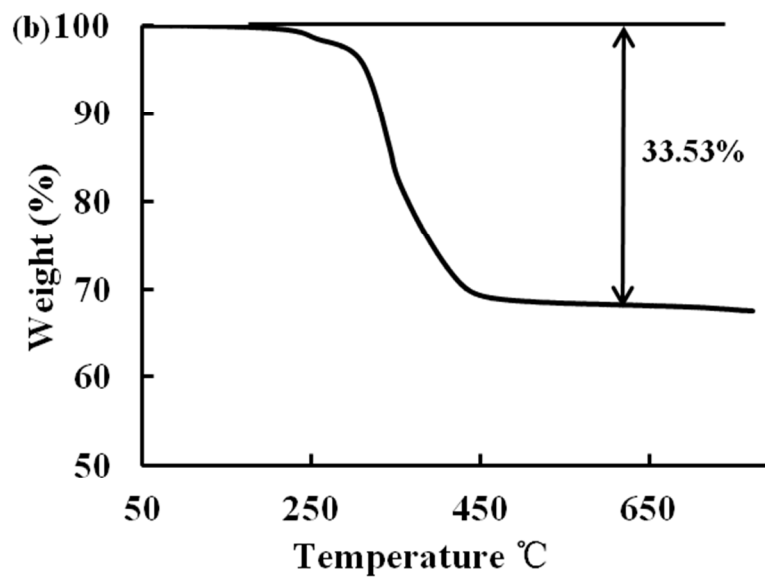
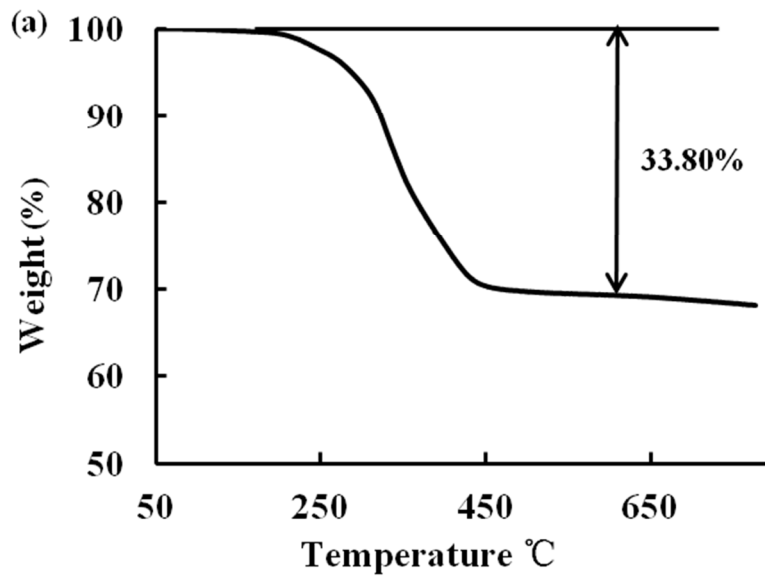
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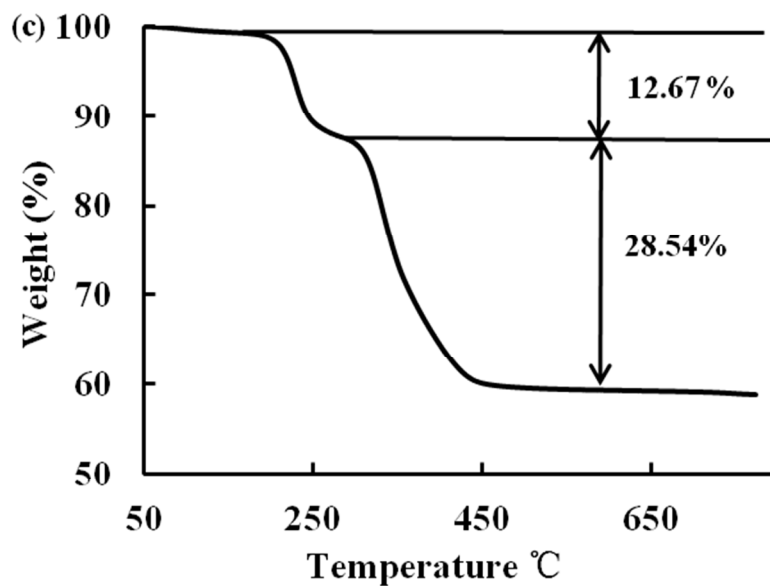
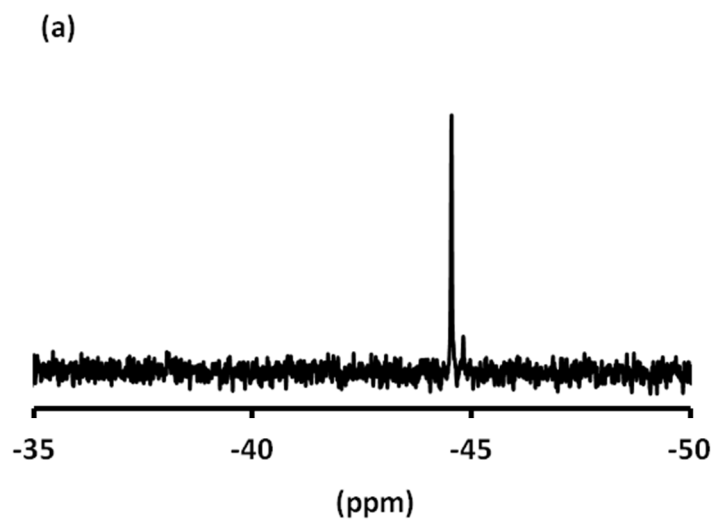


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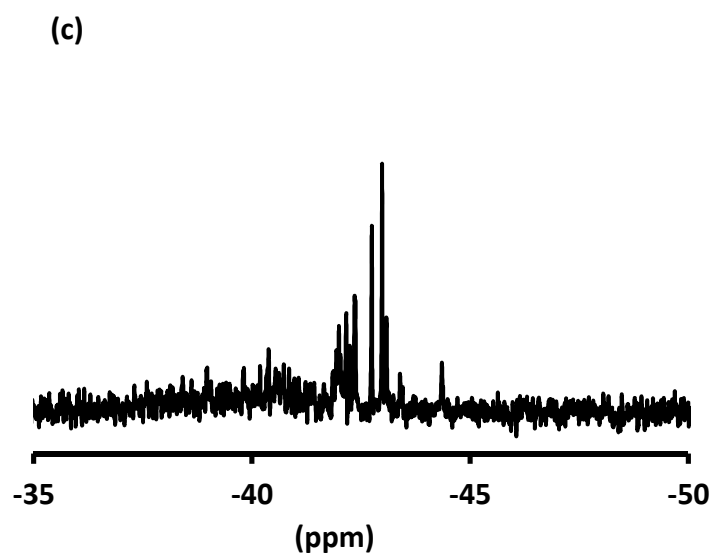
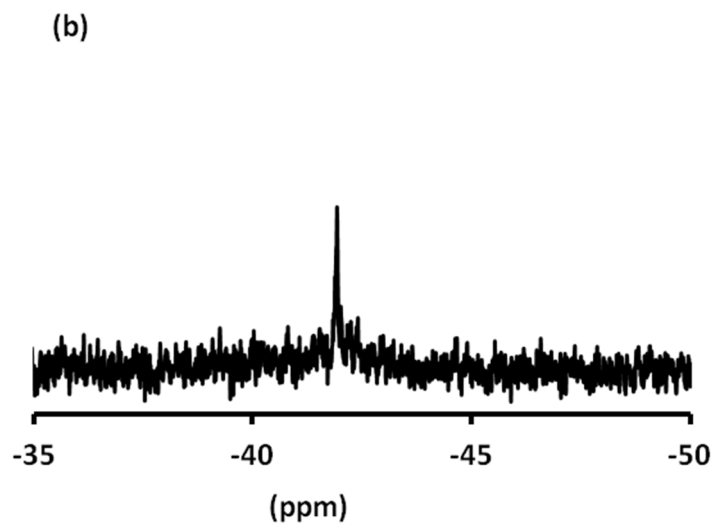


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Entry	$[\text{PW}_8\text{V}_4]$	Initial $[\text{CH}_2\text{O}]$	CH_2O conversion (%) ^b	TON ^c
1	3.8 mM	0.065 M	12	2
2	3.8 mM	0.130 M	21	7
3	3.8 mM	0.260 M	31	21
4	3.8 mM	0.520 M	35	47
5	7.6 mM	0.520 M	45	31
6	11.4 mM	0.520 M	48	22
7	15.3 mM	0.520 M	47	16
8	19.1 mM	0.520 M	45	12

^a Reaction Conditions: concentration of formaldehyde $[\text{CH}_2\text{O}] = 0.065 \text{ M} - 0.52 \text{ M}$, concentration of catalyst $[\text{PW}_8\text{V}_4] = 3.8 \text{ mM} - 19.1 \text{ mM}$, 1 atm of air, 2 mL of solvent, ambient temperature, 144 h. ^b CH_2O conversion (%) = moles of CH_2O consumed/moles of initial CH_2O . ^c TON = moles of formaldehyde consumed/moles of catalyst.

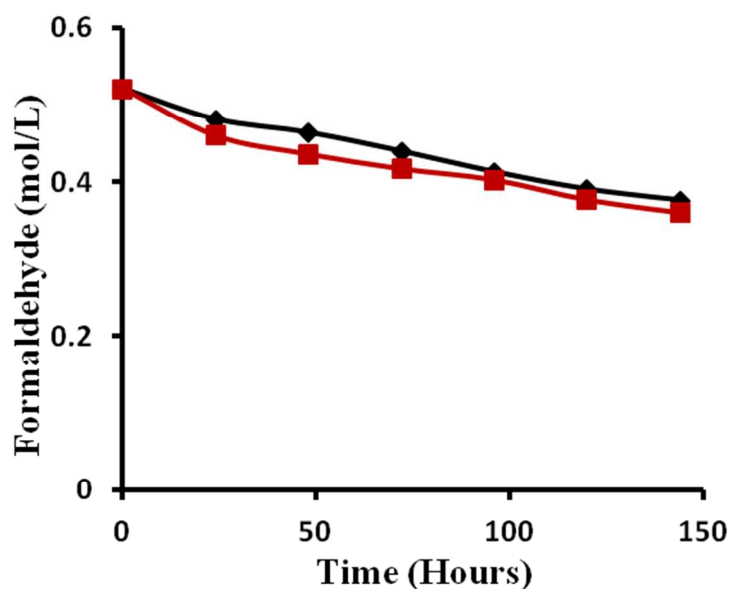


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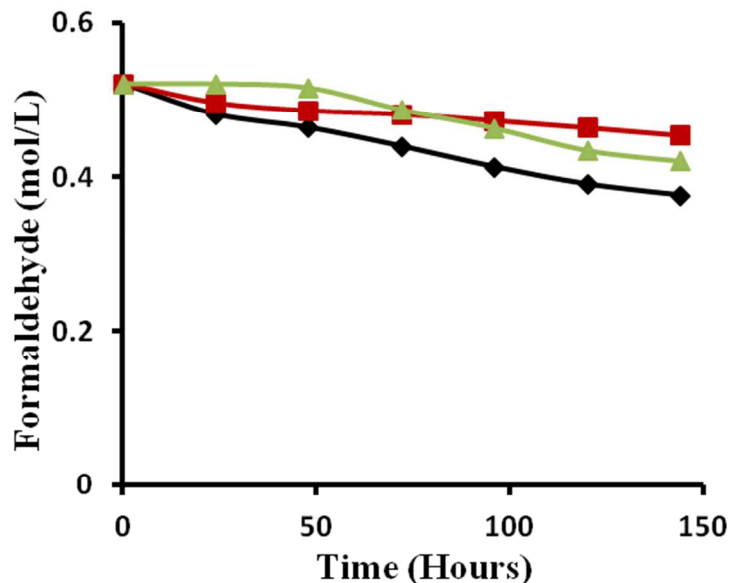


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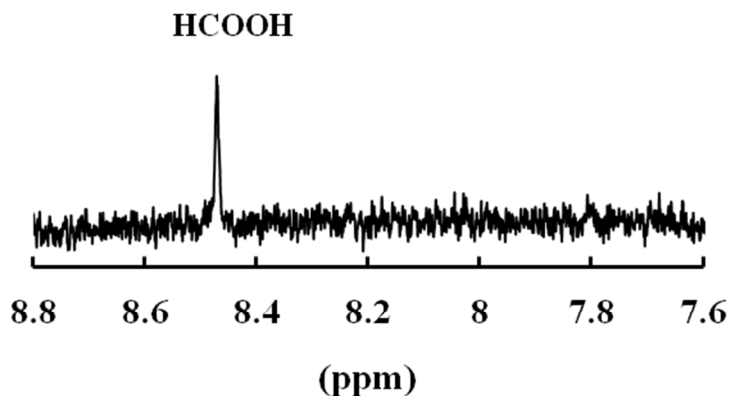


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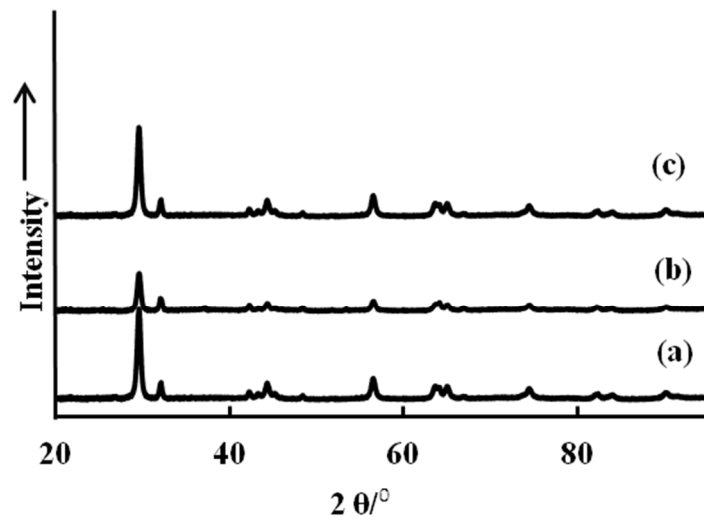
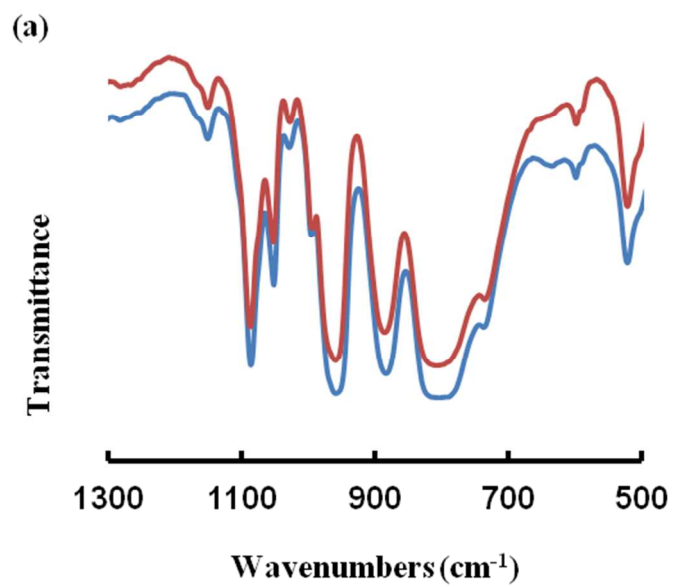


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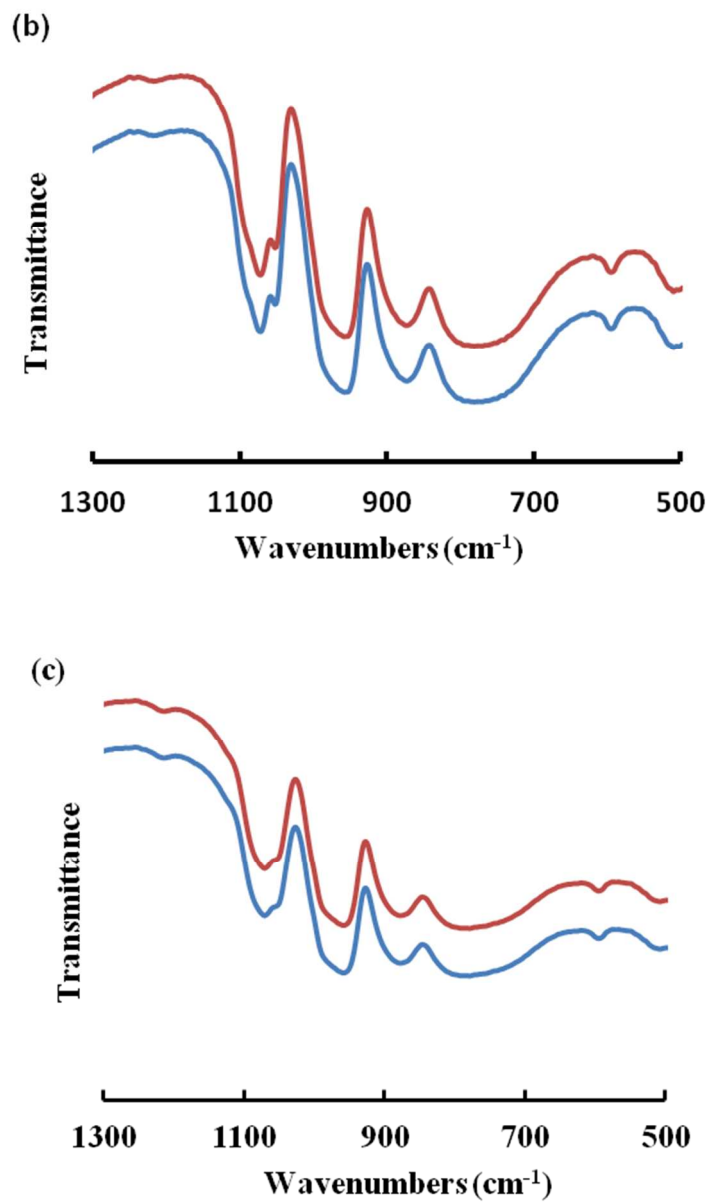


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