Syntheses, structural variation and characterization of a series of crystalline coordination compounds with 4-benzene-1,2,4-triazole: Polymorph, Incomplete spin transition and Single Crystal to Single Crystal Transformation

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Based on the 4-substituted triazole ligand 4-benzene-1,2,4-triazole (L), a series of crystalline coordination complexes varying from mononuclear to trinuclear species, namely [Zn(L)<sub>2</sub>Br<sub>2</sub>] (1),  $[Zn(L)_2Br_2]$  (2),  $[Fe(L)_4(NCS)_2]_2$  (3),  $[Fe_2(\mu_2-L)_3(L)_2(NCS)_4]\cdot CH_3OH\cdot CH_3CH_2OH$  (4),  $[Fe_2(\mu_2-L)_3(L)_2(NCS)_4] \cdot 2CH_3CH_2OH$  (5),  $[Fe_2(\mu_2-L)_3(L)_2(NCS)_4] \cdot 2CH_3CH_2OH \cdot 1.5H_2O$  (6) and  $[Ni_3(\mu_2-L)_6(L)_4(H_2O)_2](NO_3)_6\cdot 15.5H_2O$  (7) have been isolated. 1 and 2 present temperature induced polymorphic phenomenon of two Zinc(II) coordination complexes with L. Solvent effect is quite important for the assembly and magnetic properties of these iron(II) compounds 3-6: 3 contains mononuclear Fe(L)<sub>4</sub>(NCS)<sub>2</sub> units without spin transition behavior while both 4 and 5 present dinuclear iron(II) complexes with three N<sub>1</sub>,N<sub>2</sub>-1,2,4-triazole bridges behaving incomplete spin transition phenomenon. Low temperature structural analysis (100 K) of 4 is consistent with one of the Fe(II) centers within the dinuclear complex at low spin (LS) state and the other at high-spin (HS) state resulting in a [LS:HS] species. Interestingly when the dinuclear iron(II) complex 5 was exposed in the water atmosphere, solvent induced single crystal to single crystal transformation can be observed, dinuclear iron(II) complex 6 behaving antiferromagnetic interactions can be isolated. Further a trinuclear crystalline compound is isolated when Nickel(II) salts were used to react with L. Variable temperature magnetic susceptibility measurement (2-300K) reveals antiferromagnetic interactions in 7. The polymorphic phenomenon (1 and 2), incomplete spin transition phenomenon (4 and 5) and single crystal to single crystal transformation (from 5 to 6) also reveal great potential in the construction of these novel functional materials with L.

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