

Submitted to *Inorganic Chemistry*

Supporting Information

Non-Innocence of 1,4-Dicyanamidobenzene Bridging Ligands in Dinuclear Ruthenium Complexes.

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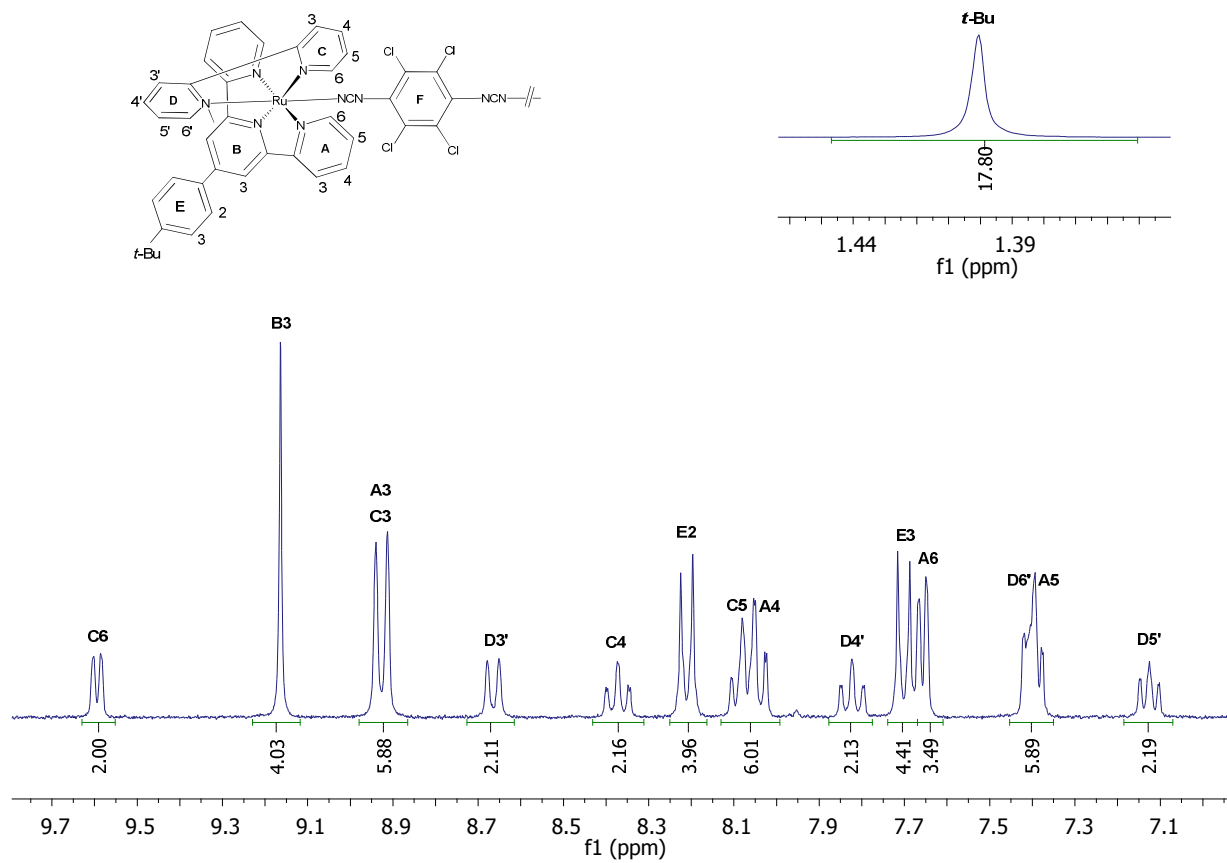


Figure S1 ^1H NMR of complex 4 in $\text{DMSO-}d_6$

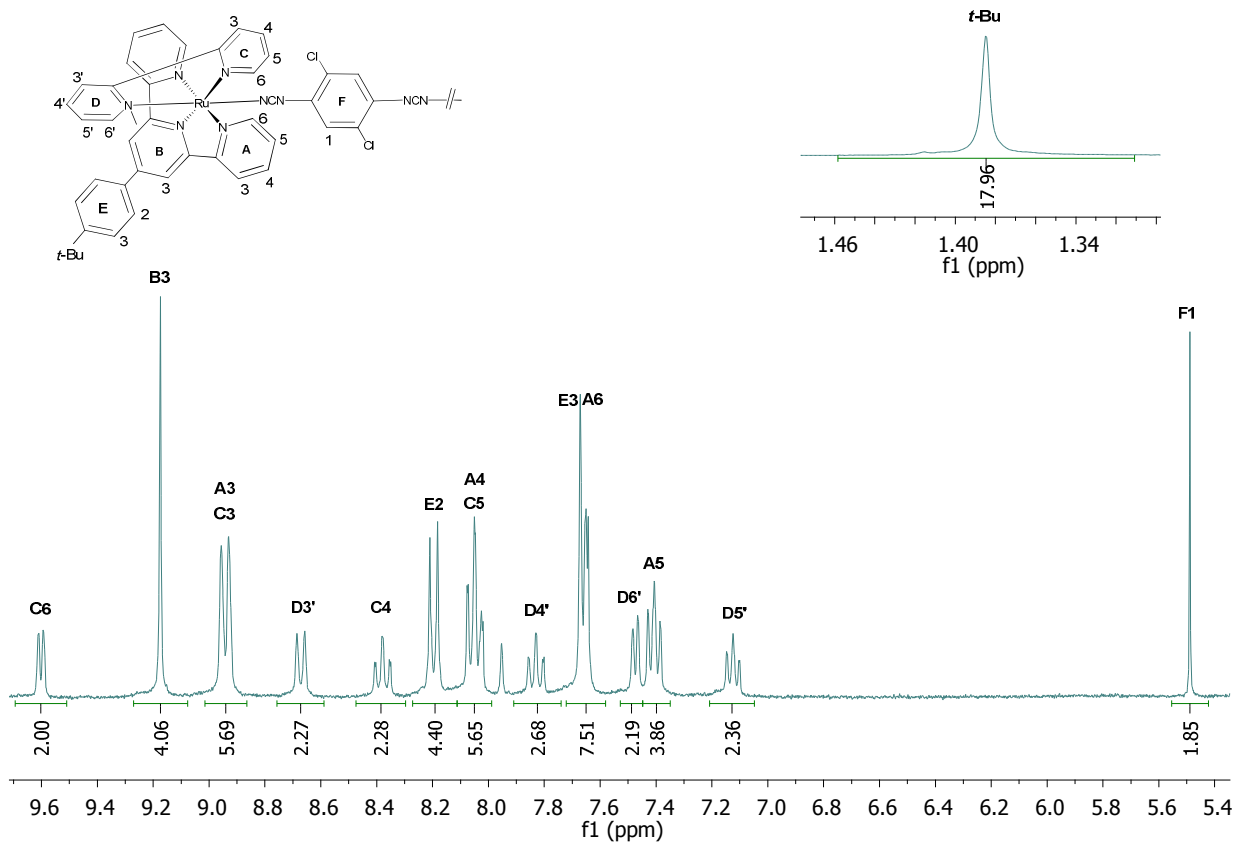


Figure S2 ^1H NMR of complex **3** in $\text{DMSO-}d_6$

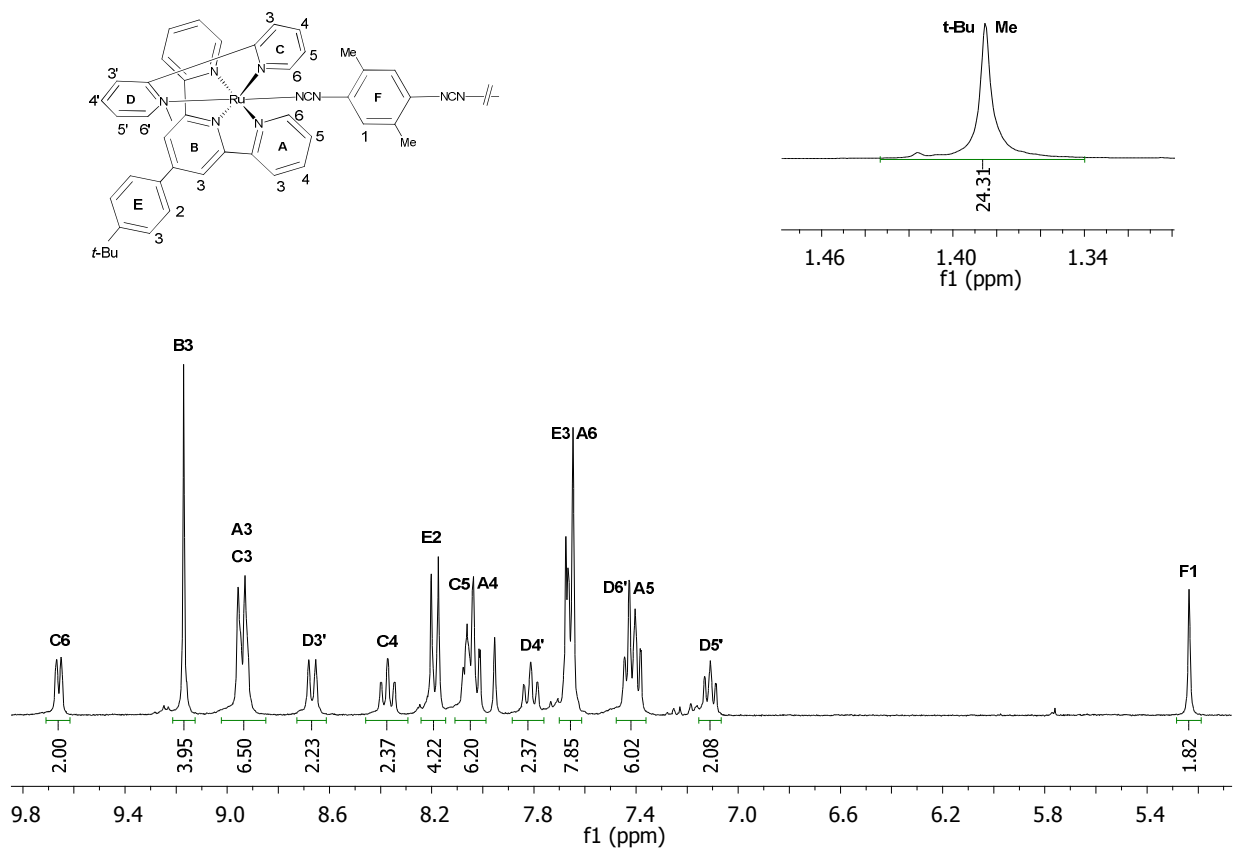


Figure S3 ^1H NMR of complex **1** in $\text{DMSO-}d_6$

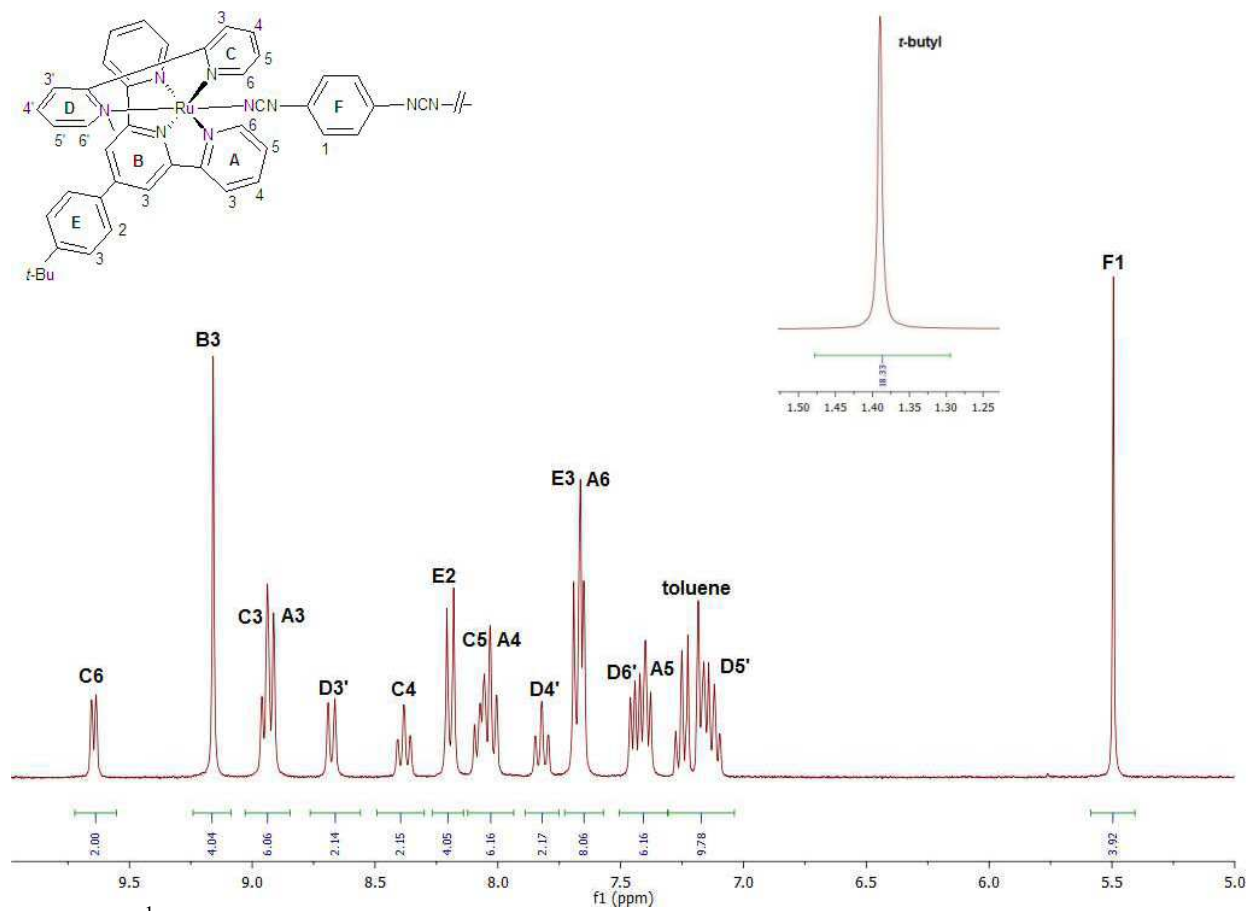


Figure S4 ^1H NMR of complex **2** in $\text{DMSO-}d_6$

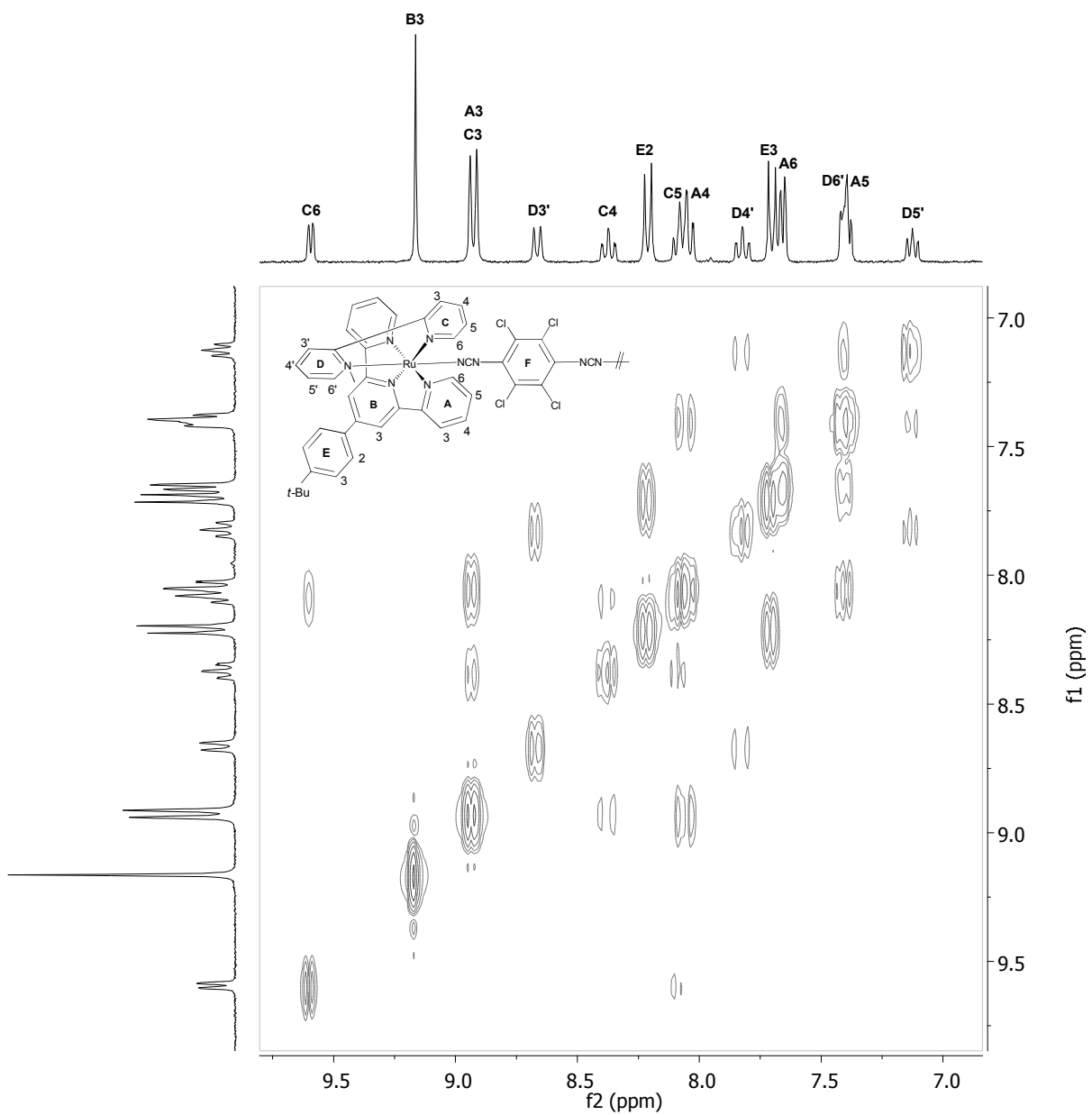


Figure S5 COESY experiment for complex **4** in DMSO-*d*₆

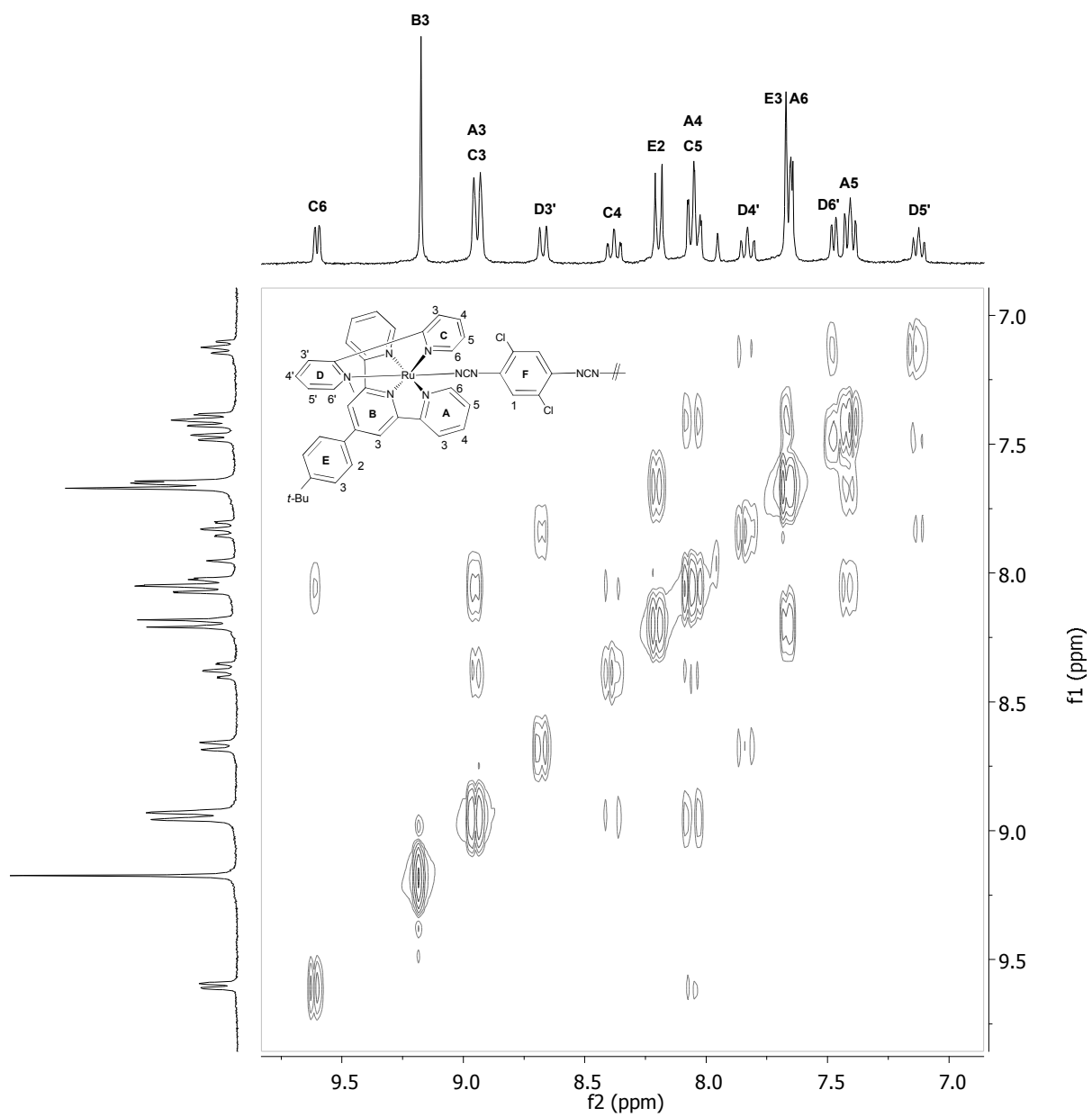


Figure S6 COESY experiment for complex **3** in DMSO-*d*₆

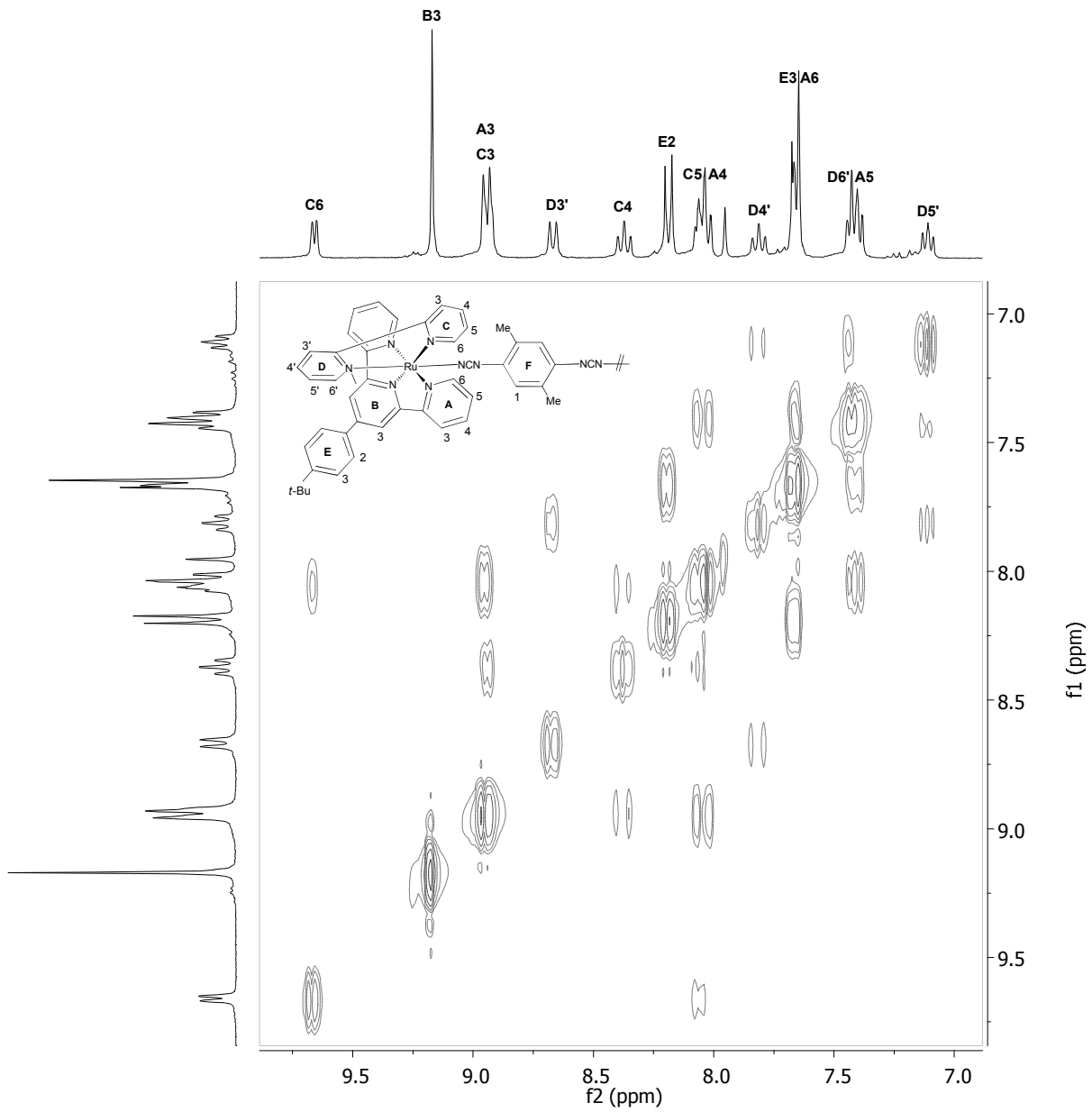


Figure S7 COESY experiment for complex **1** in DMSO- d_6

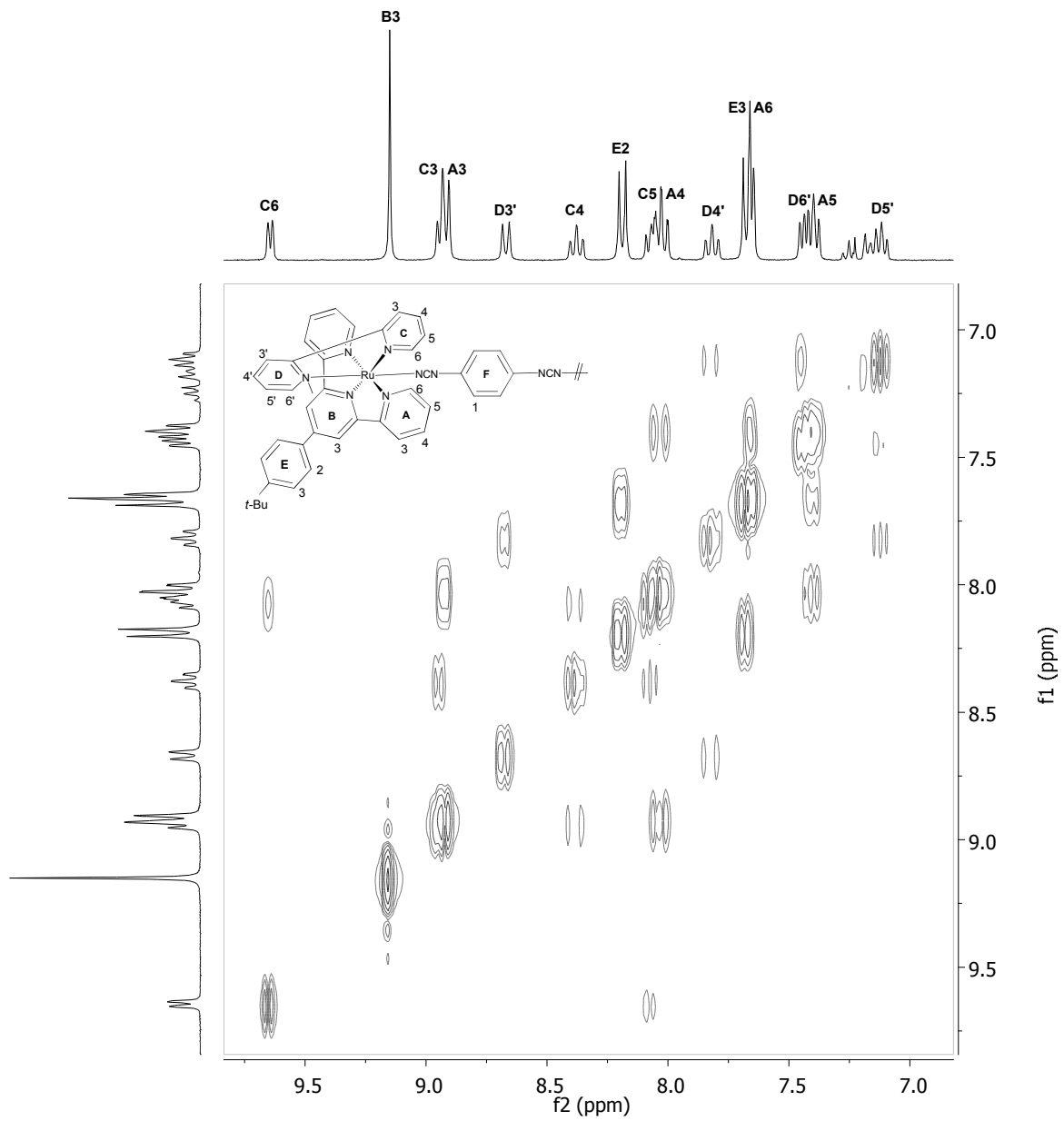


Figure S8 COESY experiment for complex **2** in DMSO- d_6

Table S1. ¹H NMR data of complexes 1-4 in DMSO-*d*₆

Complexes	Chemical Shift (ppm)/type	<i>J</i> values (Hz)
[$\{\text{Ru}(\text{ttpy})(\text{bpy})\}_2(\mu\text{-Me}_2\text{dicyd})\][\text{PF}_6]_2$	9.66 (d, 2H) (C6)	5.2
	9.17 (s, 4H) (B3)	
	8.94 (d, 6H) (C3, A3)	8.0
	8.67 (d, 2H) (D3')	8.3
	8.37 (t, 2H) (C4)	7.9
	8.19 (d, 4H) (E2)	8.5
	8.08-8.01 (m, 6H) (C5, A4)	
	7.81 (t, 2H) (D4')	8.3
	7.68-7.64 (m, 8H) (E3, A6)	
	7.44-7.38 (m, 6H) (D6', A5)	
	7.11 (t, 2H) (D5')	6.7
	5.24 (s, 2H) (F1)	
	1.39 (s, 24H) (<i>t</i> -Bu, Me)	
[$\{\text{Ru}(\text{ttpy})(\text{bpy})\}_2(\mu\text{-dicyd})\][\text{PF}_6]_2$	9.64 (d, 2H) (C6)	5.2
	9.15 (s, 4H) (B3)	
	8.95-8.91 (m, 6H) (C3, A3)	
	8.67 (d, 2H) (D3')	8.2
	8.38 (t, 2H) (C4)	7.9
	8.19 (d, 4H) (E2)	8.5
	8.11 – 7.97 (m, 6H) (C5, A4)	
	7.82 (t, 2H) (D4')	7.9
	7.69-7.65 (m, 8H) (E3, A6)	
	7.48 – 7.34 (m, 6H) (D6', A5)	
	7.12 (t, 2H) (D5')	7.2
	5.49 (s, 4H) (F1)	
	1.39 (s, 18H) (<i>t</i> -Bu)	
[$\{\text{Ru}(\text{ttpy})(\text{bpy})\}_2(\mu\text{-Cl}_2\text{dicyd})\][\text{PF}_6]_2$	9.60 (d, 2H) (C6)	4.9
	9.17 (s, 4H) (B3)	
	8.94 (d, 6H) (C3, A3)	7.9
	8.67 (d, 2H) (D3')	8.2
	8.38 (t, 2H) (C4)	7.2
	8.20 (d, 4H) (E2)	8.5
	8.08-8.02 (m, 6H) (C5, A4)	
	7.83 (t, 2H) (D4')	8.5
	7.67-7.64 (m, 8H) (E3, A6)	
	7.47 (d, 2H) (D6')	5.1
	7.41 (t, 4H) (A5)	6.9
	7.12 (t, 2H) (D5')	6.6
	5.49 (s, 2H) (F1)	
1.38 (s, 18H) (<i>t</i> -Bu)		

$[\{\text{Ru}(\text{tpy})(\text{bpy})\}_2(\mu\text{-Cl}_4\text{dicyd})][\text{PF}_6]_2$	<p>9.59 (d, 2H) (C6) 9.16 (s, 4H) (B3) 8.93 (d, 6H) (C3, A3) 8.66 (d, 2H) (D3') 8.37 (t, 2H) (C4) 8.21 (d, 4H) (E2) 8.10-8.02 (m, 6H) (C5, A4) 7.82 (t, 2H) (D4') 7.70 (d, 4H) (E3) 7.66 (d, 4H) (A6) 7.45 – 7.35 (m, 6H) (D6', A5) 7.13 (t, 2H) (D5') 1.40 (s, 18H) (t-Bu)</p>	<p>5.1 8.2 8.3 8.0 8.5 8.5 8.5 4.8 7.1</p>
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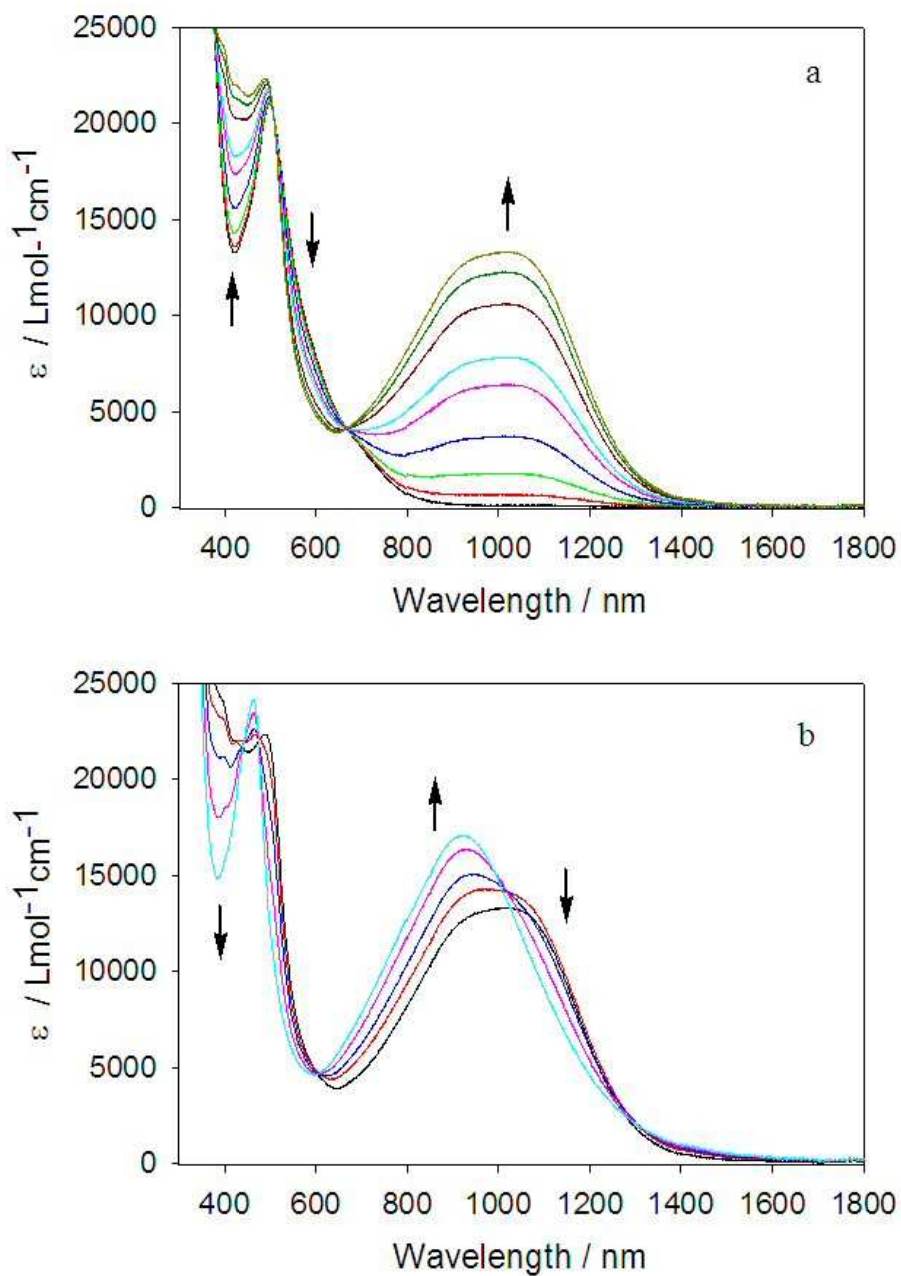


Figure S9. Spectroelectrochemistry of $[\{Ru(tpy)(bpy)\}_2(\mu-Me_2dicyd)][PF_6]_2$ in DMF, 0.1 M TBAH a) oxidation to 3+ cation b) oxidation to 4+ cation.

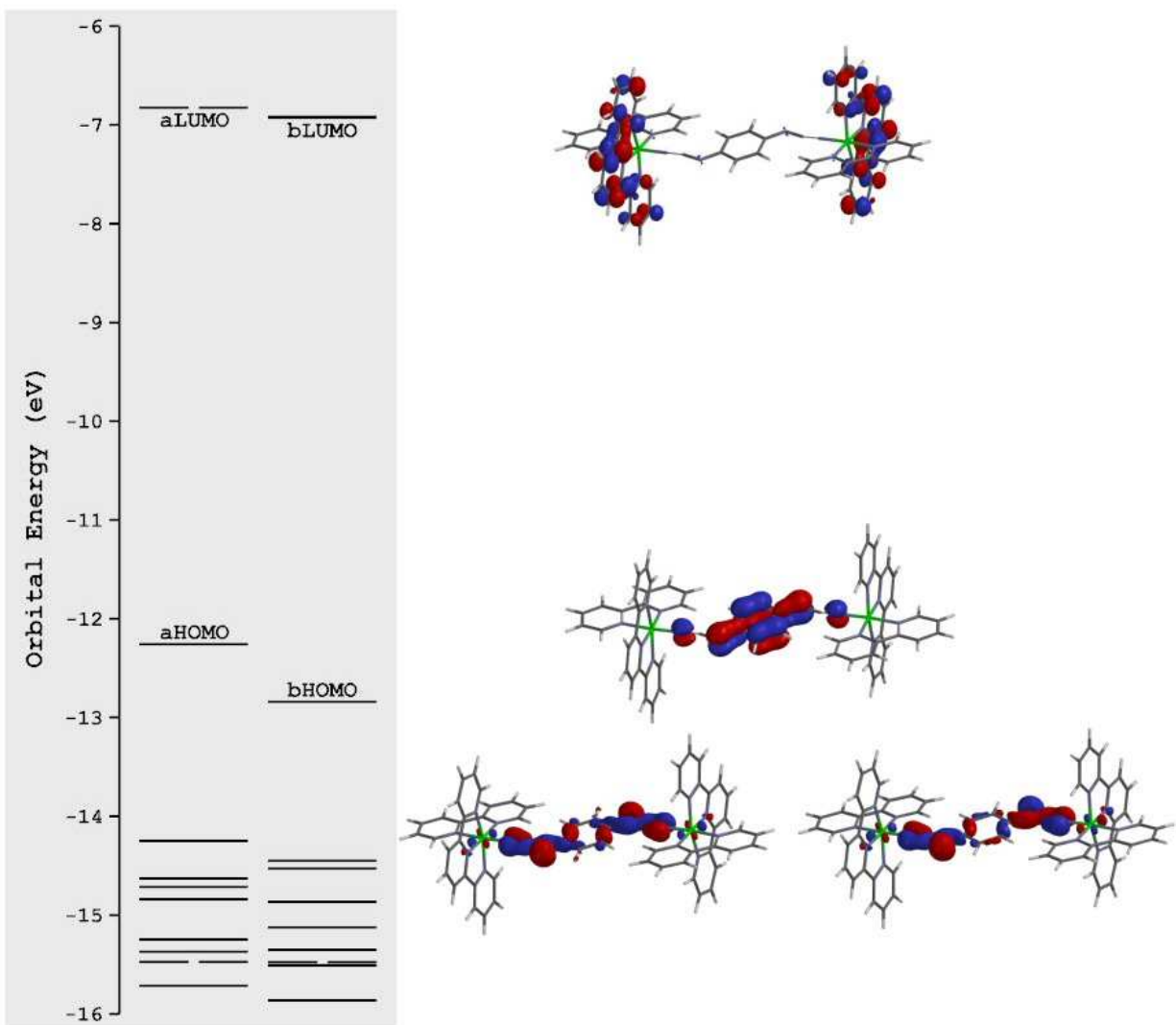


Figure S10. Left side: the gas phase semi-empirical calculation (PM3) of the orbital energies of $[\{Ru(terpyridine)(bpy)\}_2(\mu\text{-dicyd})]^{3+}$, $\mathbf{2}^+$. Right side, beta wavefunctions from the top LUMO, HOMO and the two nearly degenerate HOMO(-1) and HOMO(-2).

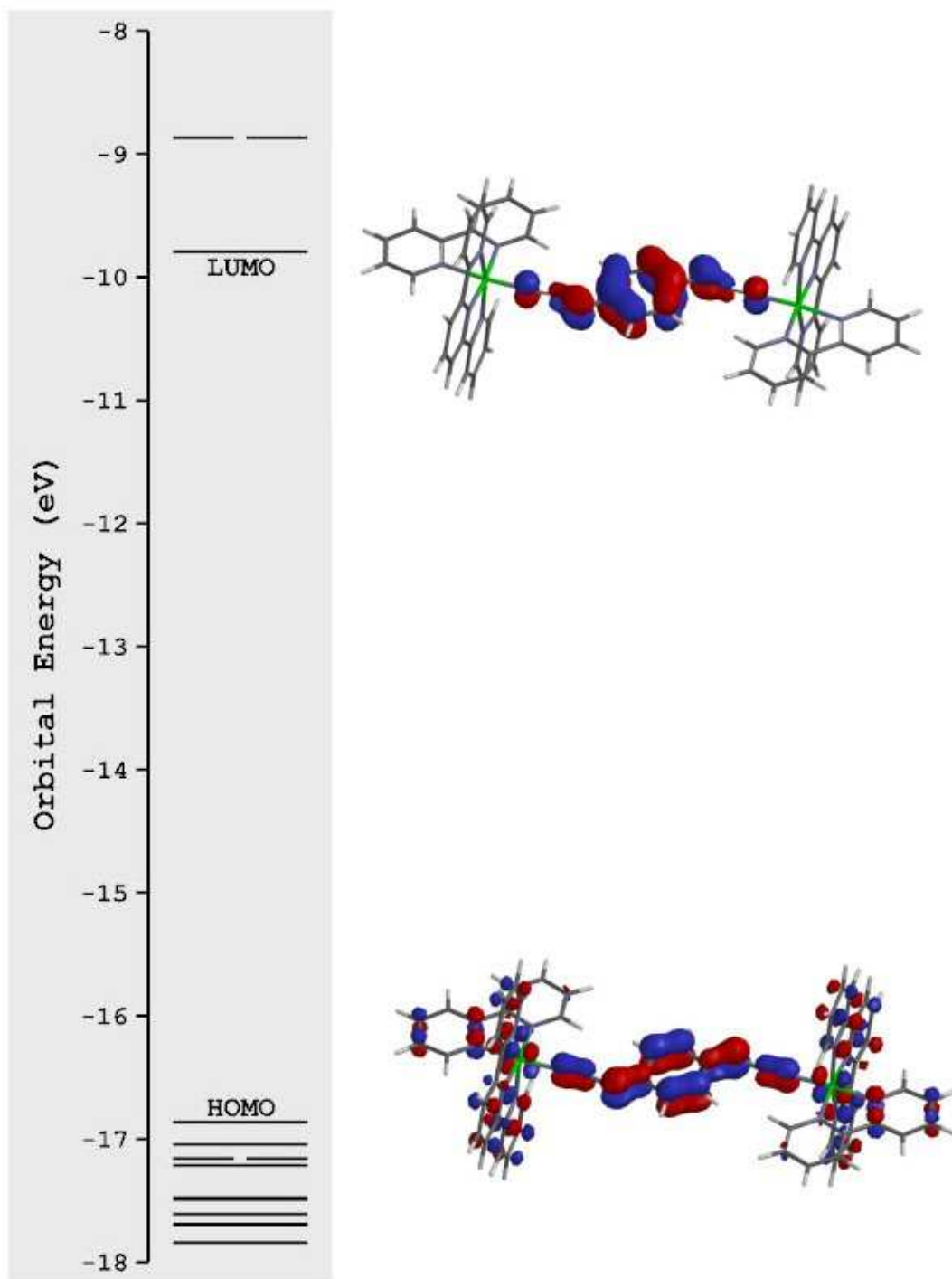


Figure S11. Left side: the gas phase semi-empirical calculation of the orbital energies of $[\{\text{Ru}(\text{terpyridine})(\text{bpy})\}_2(\mu\text{-dicyd})]^{4+}$, $\mathbf{2}^{2+}$. Right side, LUMO and HOMO wavefunctions.

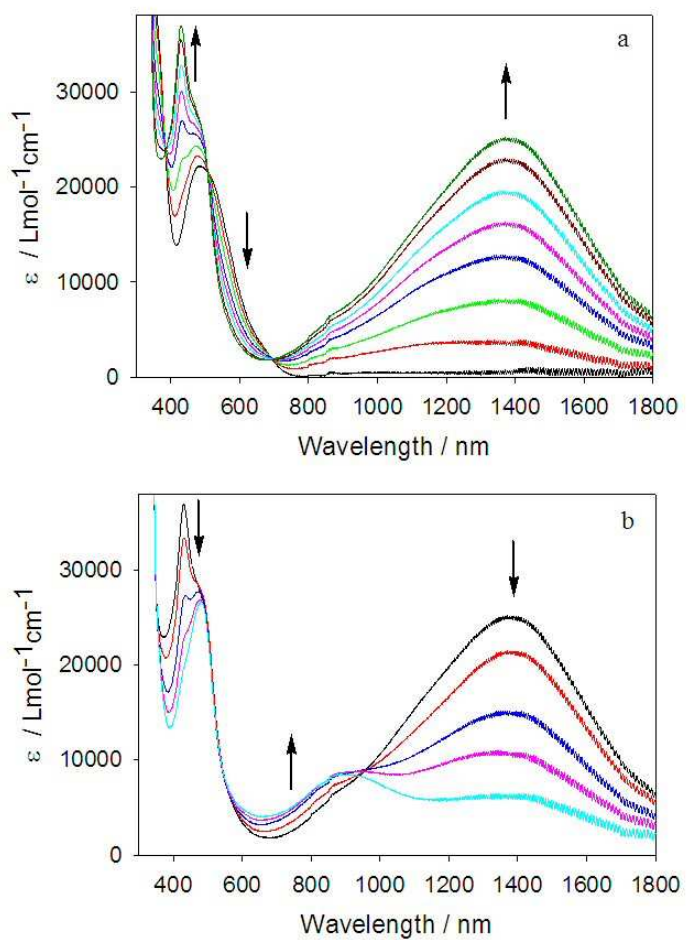


Figure S12. Spectroelectrochemistry of $[\{\text{Ru}(\text{ttpy})(\text{bpy})\}_2(\mu\text{-Cl}_4\text{dicyd})][\text{PF}_6]_2$ in DMF, 0.1 M TBAH
 a) oxidation to 3+ cation b) partial oxidation to 4+ cation, loss of isosbestic point indicating irreversibility.

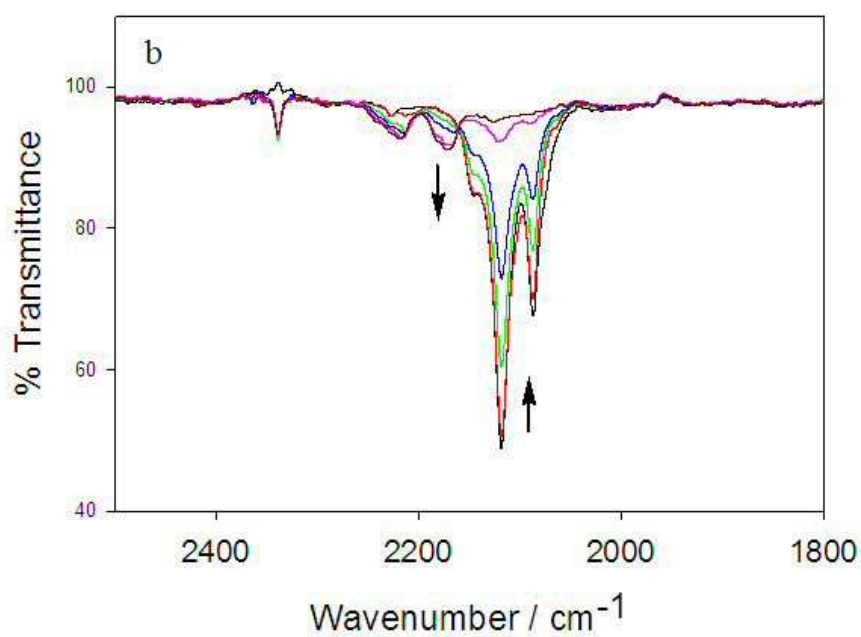
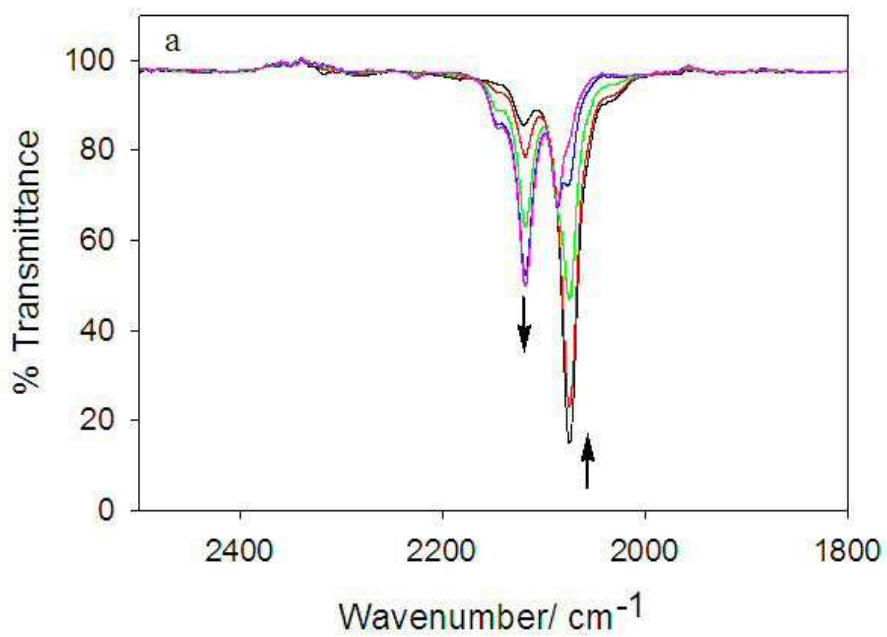


Figure S13. IR spectroelectrochemistry of the tetraphenylarsonium salt of $\text{Me}_2\text{dicyd}^{2-}$ in DMF, 0.1 M TBAH, a) oxidation to $\text{Me}_2\text{dicyd}^-$ and b) oxidation to $\text{Me}_2\text{dicyd}^0$

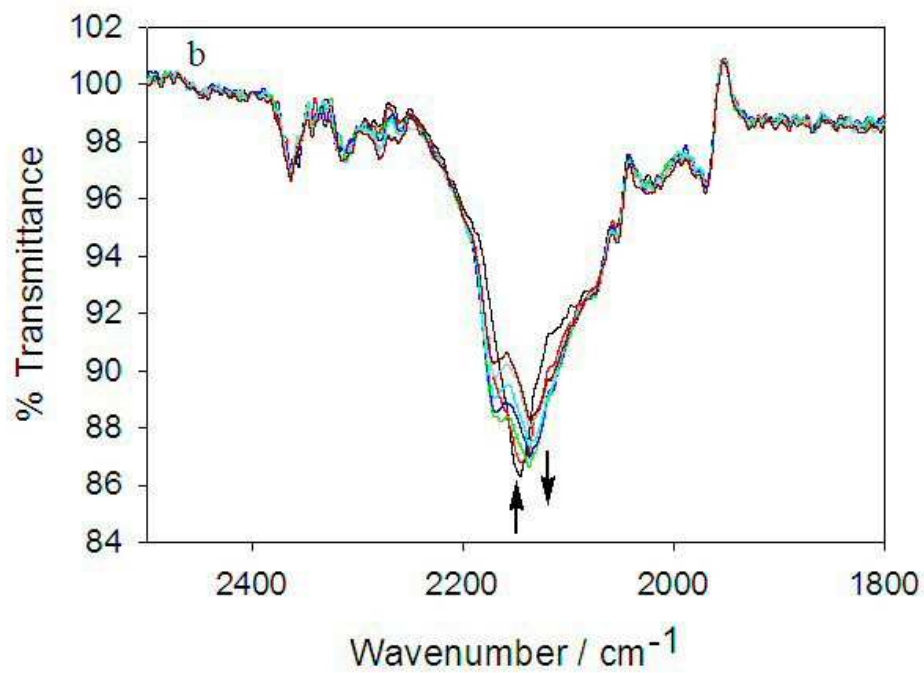
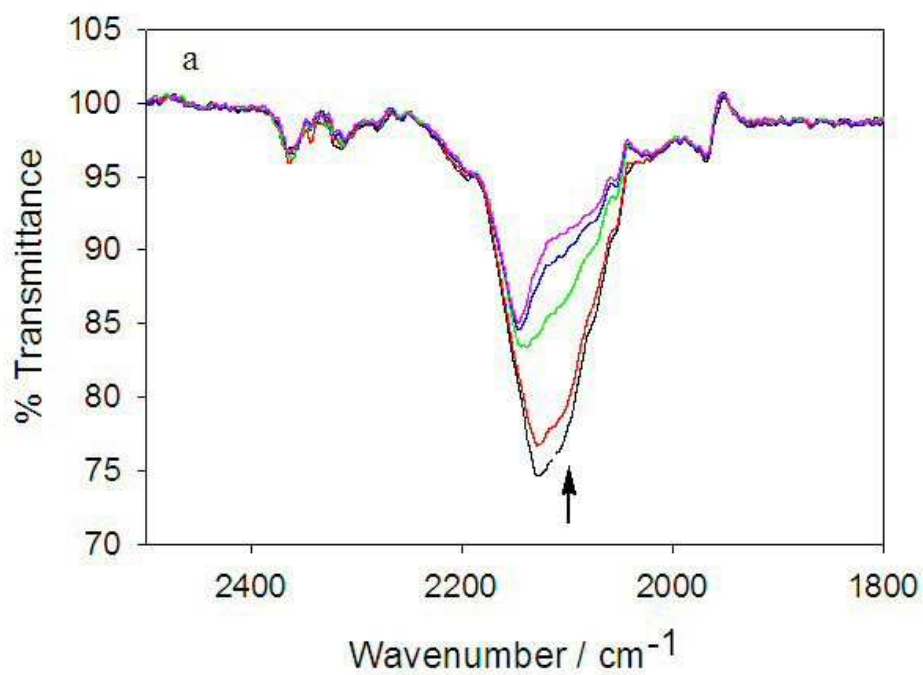


Figure S14. IR spectroelectrochemistry of $[\{Ru(tpy)(bpy)\}_2(\mu-Me_2dicyd)][PF_6]_2$ in DMF, 0.1 M TBAH a) oxidation to 3+ cation b) oxidation to 4+ cation.

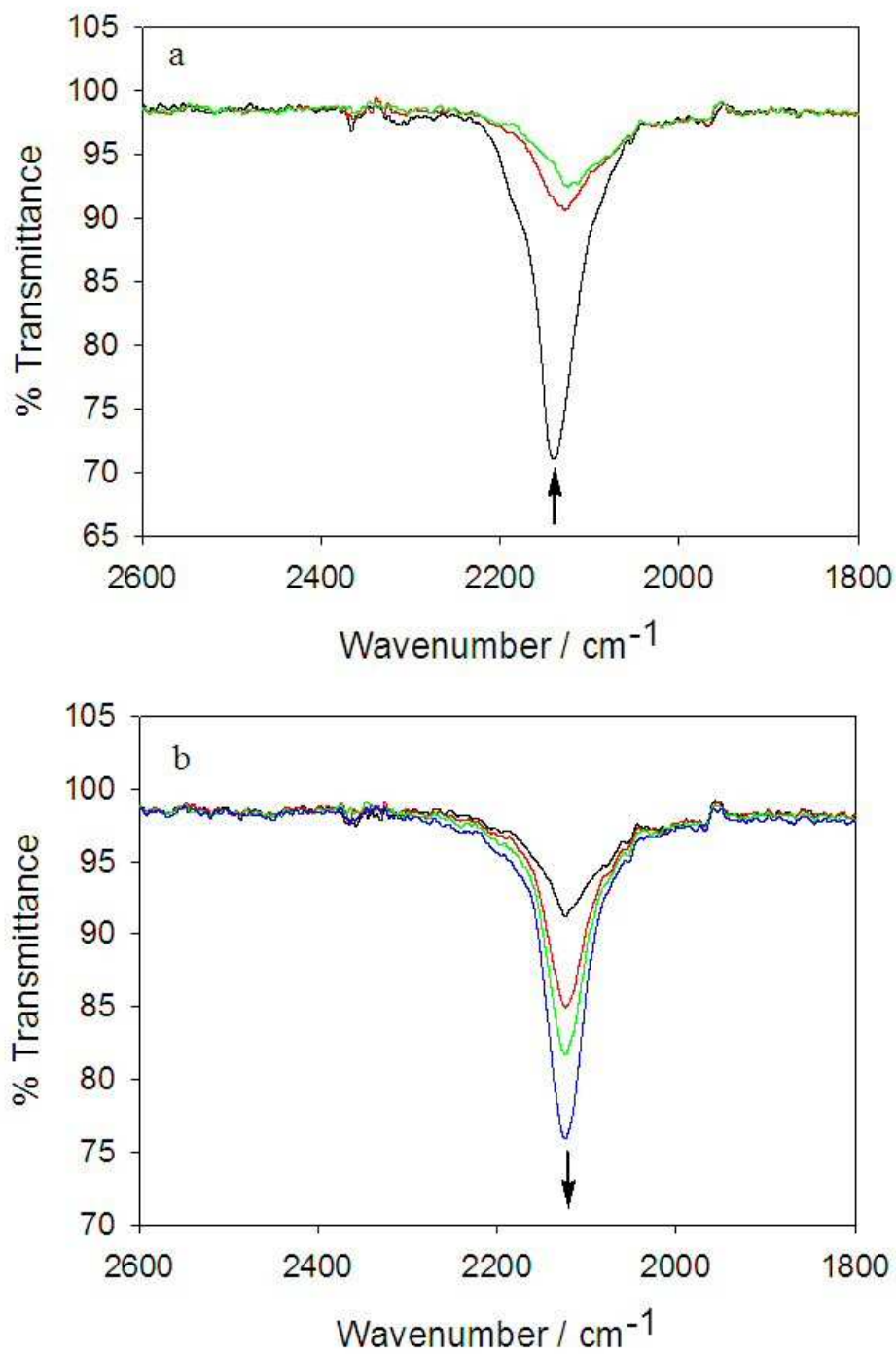


Figure S15. IR spectroelectrochemistry of [$\{\text{Ru}(\text{ttpy})(\text{bpy})\}_2(\mu\text{-Cl}_2\text{dicyd})\}\text{[PF}_6\text{]}_2$] in DMF, 0.1 M TBAH a) oxidation to 3+ cation b) oxidation to 4+ cation.