

Supporting Information

Mechanistic investigation of Palladium Catalyzed Allylic C-H Activation

Casper Junker Engelin, Thomas Jensen, Sergio Rodriguez-Rodriguez and Peter Fristrup*

Department of Chemistry, Technical University of Denmark, Kemitorvet, DK-2800, Lyngby, Denmark

Email: pf@kemi.dtu.dk

Table of contents

General information	S2
GC retention times	S2
Competition experiments for the Hammett Study	S4
KIE determination and deuterium experiments	S11
NMR data for deuterium marked compounds	S21
XYZ coordinates and energies from computational study	S25
Calculation of theoretical KIE	S52
Calculated charges	S53

General Information

GCMS analysis was performed on a Shimadzu QP5000 GCMS instrument using electron impact as the ionization method and a 30 m x 0.25 mm x 0.25 μ m Supelco Equity-1 capillary column.

Flash column chromatography was performed with silica gel 60 (35-70 μ m) with solvents of HPLC quality. ^1H and ^{13}C NMR spectra were recorded at 25 $^{\circ}\text{C}$ on a Varian Mercury 300 at 300 MHz and 75 MHz, respectively. Chemical shifts (δ) are represented in parts per million (ppm). ^1H NMR spectra are referenced to residual CHCl_3 (7.26 ppm). ^{13}C NMR spectra are referenced to CDCl_3 (77.0 ppm).

Dodecane, $\text{Pd}[1,2\text{-bis(phenylsulfonyl)ethane}](\text{OAc})_2$, 2,6-dimethyl benzoquinone, 4-allyltoluene and 1-allyl-4-methoxybenzene were purchased from Sigma-Aldrich and used without further purification. Allylbenzene and methyl nitroacetate were distilled before use.

GC retention times

Injector temperature 200 $^{\circ}\text{C}$, detector temperature 200 $^{\circ}\text{C}$.

Table S1 Programmed oven temperature: start 60 $^{\circ}\text{C}$ for 5 min, ramp 20 $^{\circ}\text{C}/\text{min}$ to 250 $^{\circ}\text{C}$, 6 min @ 250 $^{\circ}\text{C}$.

<i>Compound</i>	<i>Retention Time</i>
Methyl nitroacetate	6.5 min
Allylbenzene	6.9 min
1-allyl-4-(trifluoromethyl)benzene	7.6 min
1-allyl-4-Methylbenzene	8.5 min
2,6-dimethylbenzoquinone	9.3 min
1-allyl-4-chlorobenzene	9.5 min
Dodecane (Int. Std.)	10.0 min
1-allyl-4-Methoxybenzene	10.1 min
4-allylbenzonitrile	10.8 min
Methyl 4-allylbenzoate	11.7 min

Table S2 Programmed oven temperature: start 60 °C for 5 min, ramp 10 °C/min to 300 °C, 5 min @ 300 °C.

<i>Compound</i>	<i>Retention Time</i>
Methyl nitroacetate	6.7 min
<i>d</i> ₂ -allylbenzene	7.3 min
Allylbenzene	7.3 min
2,6-dimethylbenzoquinone	11.1 min
Dodecane (Int. Std.)	12.5 min
Methyl N-tosylcarbamate	19.1 min

Table 3 Programmed oven temperature: start 60 °C for 5 min, ramp 12 °C/min to 280 °C, 2 min @ 280 °C.

<i>Compound</i>	<i>Retention Time</i>
Methyl nitroacetate	6.6 min
<i>d</i> ₂ -allylbenzene	7.2 min
1-allyl-4-Methylbenzene	9.4 min
2,6-dimethylbenzoquinone	10.6 min
Dodecane (Int. Std.)	11.8 min
Methyl N-tosylcarbamate	17.3 min

Competition experiments for the Hammett Study

General

To a screw-cap vial was added Pd[1,2-bis(phenylsulfonyl)ethane](OAc)₂ (0.05 mmol, 25 mg), 2,6-dimethylbenzoquinone (0.75 mmol, 102 mg) and a stir bar. Allylbenzene (0.25 mmol, 30 mg), *para*-substituted allylbenzene (0.25 mmol) under investigation, methyl nitroacetate (1.5 mmol, 178 mg), AcOH (0.25 mmol, 15 mg) and dodecane as internal standard (0.125 mmol, 21 mg) were weighed out in a 1.0 mL vial and transferred to the catalyst mixture using dioxane:DMSO (4:1) (1.5 mL). The reaction was sealed with a screw cap fitted with a PTFE septum and placed in an aluminum block preheated to 45 °C. Samples of 0.05 mL each were withdrawn from the reaction mixture after 0, 0.5, 1, 2, 4, 6, 8, and 24 hrs reaction time, diluted with diethylether (1 mL) and analyzed by GCMS.

Table S4 Data for the competition experiment involving allylbenzene and 1-allyl-4-methoxybenzene.

Reaction time (min)	Conversion allylbenzene	Conversion 1-allyl-4-methoxybenzene	Kinetics allylbenzene	Kinetics 1-allyl-4-methoxybenzene
0	0.00	0.00	0	0
40	0.27	0.27	0.310	0.319
60	0.39	0.37	0.496	0.467
120	0.60	0.57	0.921	0.839
250	0.81	0.78	1.651	1.500
360	0.89	0.85	2.172	1.911
480	0.93	0.91	2.608	2.385
1410	0.96	0.94	3.220	2.869

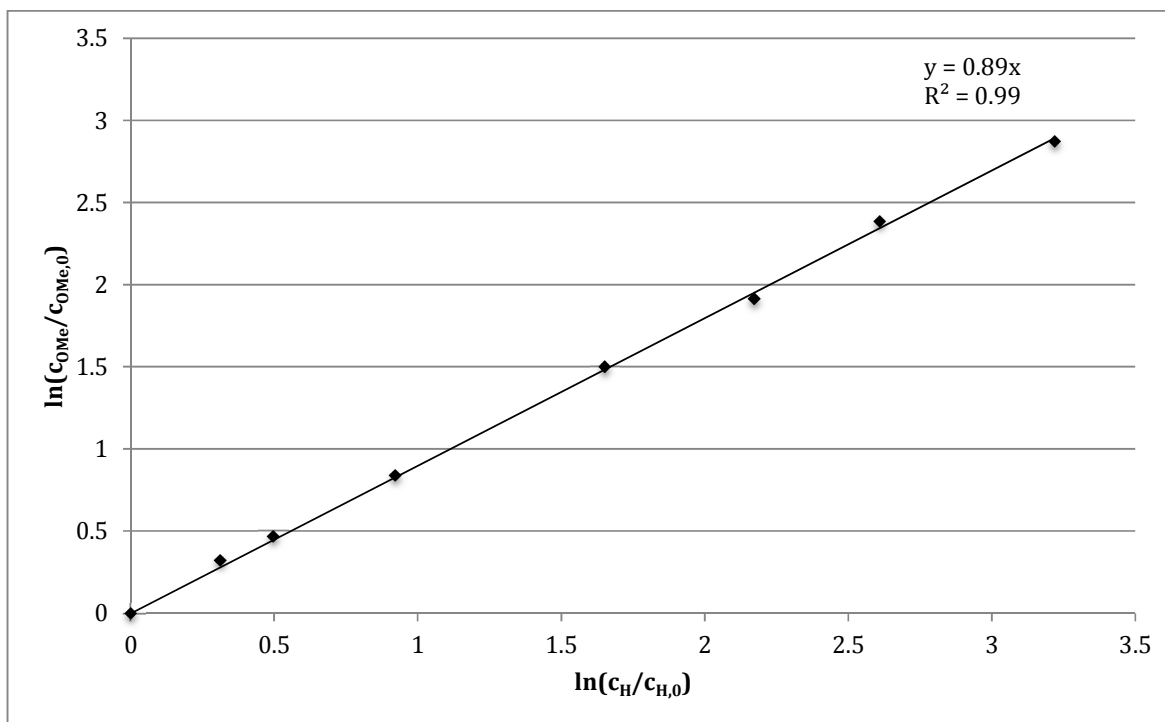


Figure S1 Kinetic plot for the competition experiment with allylbenzene and 1-allyl-4-methoxybenzene. The slope represents the reactivity of 1-allyl-4-methoxybenzene relative to allylbenzene.

Table S5 Data for the competition experiment involving allylbenzene and 1-allyl-4-(trifluoromethyl)benzene.

Reaction time (min)	Conversion allylbenzene	Conversion 1-allyl-4-(trifluoromethyl)benzene	Kinetics allylbenzene	Kinetics 1-allyl-4-(trifluoromethyl)benzene
0	0.00	0.00	0	0
30	0.21	0.29	0.234	0.339
70	0.31	0.43	0.370	0.568
120	0.41	0.56	0.528	0.820
240	0.51	0.66	0.705	1.093
360	0.54	0.69	0.767	1.167
480	0.54	0.70	0.782	1.188
1335	0.56	0.71	0.829	1.225

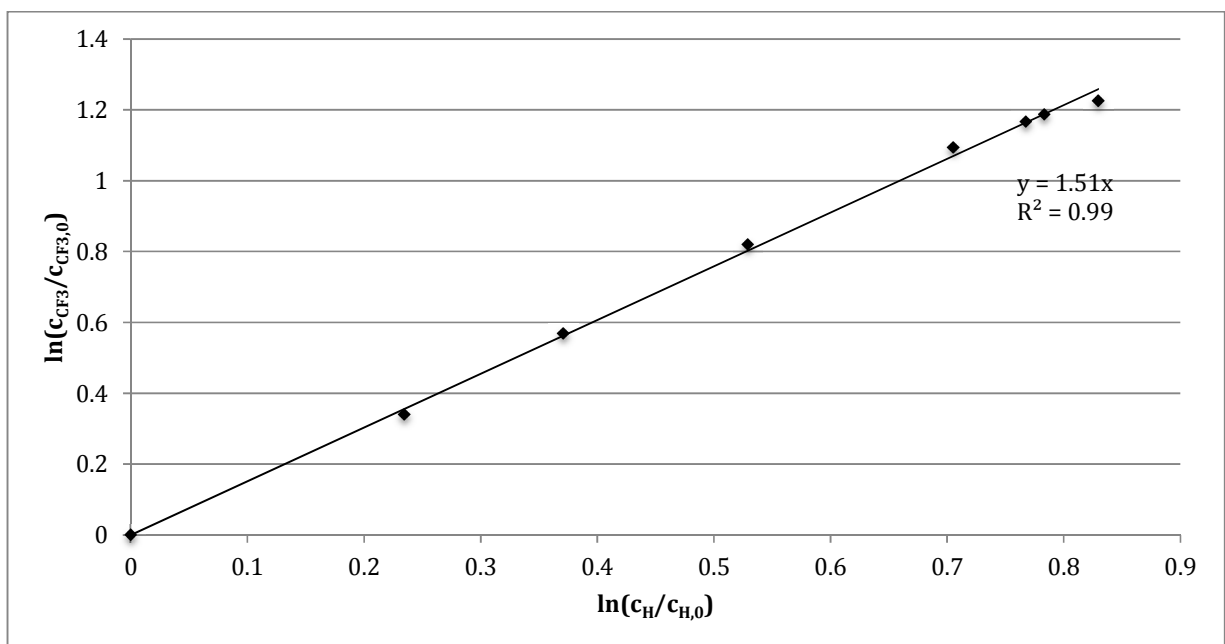


Figure S2 Kinetic plot for the competition experiment with allylbenzene and 1-allyl-4-(trifluoromethyl)benzene. The slope represents the reactivity of 1-allyl-4-(trifluoromethyl)benzene relative to allylbenzene.

Table S6 Data for the competition experiment involving allylbenzene and 1-allyl-4-chlorobenzene.

Reaction time (min)	Conversion allylbenzene	Conversion 1-allyl-4-chlorobenzene	Kinetics allylbenzene	Kinetics 1-allyl-4-chlorobenzene
0	0.00	0.00	0	0
30	0.20	0.20	0.217	0.228
60	0.32	0.33	0.382	0.402
120	0.50	0.52	0.683	0.738
240	0.68	0.70	1.129	1.214
367	0.76	0.78	1.410	1.512
480	0.78	0.81	1.530	1.662
1329	0.81	0.83	1.672	1.768

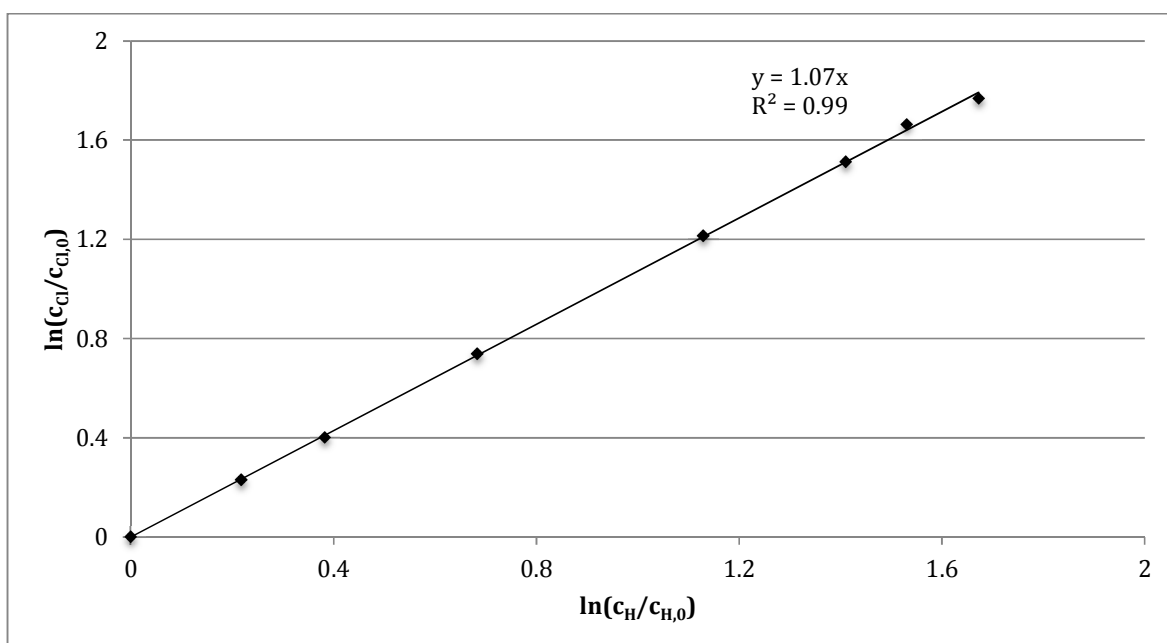


Figure S3 Kinetic plot for the competition experiment with allylbenzene and 1-allyl-4-chlorobenzene. The slope represents the reactivity of 1-allyl-4-chlorobenzene relative to allylbenzene.

Table S7 Data for the competition experiment involving allylbenzene and 1-allyl-4-methylbenzene.

Reaction time (min)	Conversion allylbenzene	Conversion 1-allyl-4-methylbenzene	Kinetics allylbenzene	Kinetics 1-allyl-4-methylbenzene
0	0.00	0.00	0	0
38	0.28	0.28	0.333	0.335
60	0.42	0.41	0.544	0.533
120	0.61	0.60	0.934	0.914
240	0.80	0.80	1.606	1.595
360	0.87	0.87	2.046	2.041
420	0.89	0.89	2.232	2.197
1560	0.93	0.92	2.604	2.581

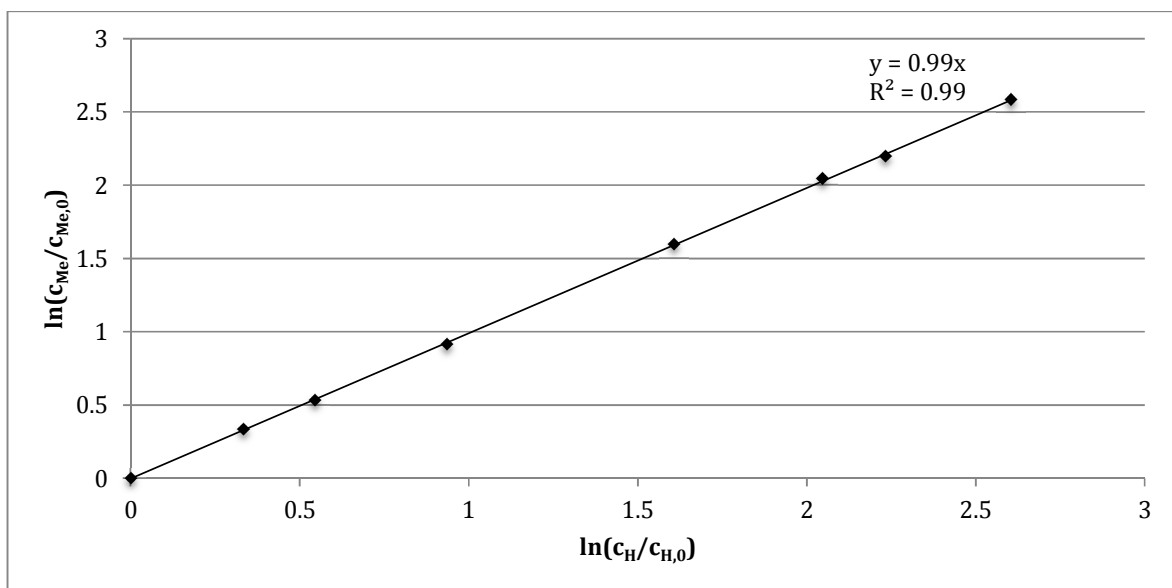


Figure S4 Kinetic plot for the competition experiment with allylbenzene and 1-allyl-4-methylbenzene. The slope represents the reactivity of 1-allyl-4-methylbenzene relative to allylbenzene.

Table S8 Data for the competition experiment involving allylbenzene and 4-allylbenzonitrile.

Reaction time (min)	Conversion allylbenzene	Conversion 4-allylbenzonitrile	Kinetics allylbenzene	Kinetics 4-allylbenzonitrile
0	0.00	0.00	0	0
5	0.03	0.03	0.026	0.029
10	0.08	0.12	0.085	0.127
15	0.11	0.19	0.112	0.210
30	0.19	0.36	0.215	0.452
60	0.35	0.65	0.427	1.048
90	0.46	0.80	0.622	1.595
120	0.54	0.87	0.774	2.003

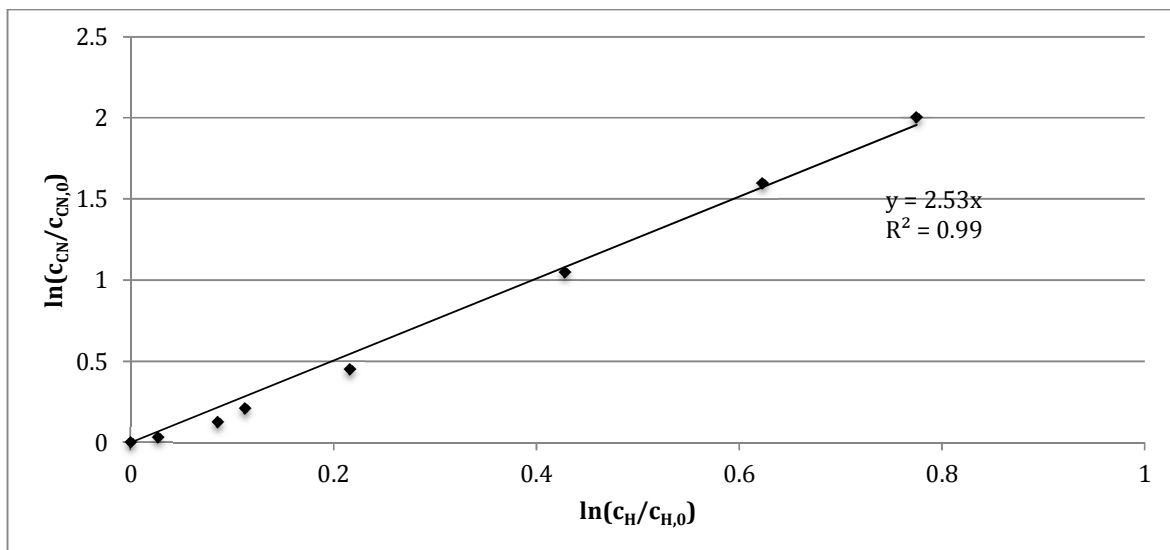


Figure S5 Kinetic plot for the competition experiment with allylbenzene and 4-allylbenzonitrile. The slope represents the reactivity of 4-allylbenzonitrile relative to allylbenzene.

Table S9 Data for the competition experiment involving 1-allyl-4-methylbenzene and methyl 4-allylbenzoate. Due to overlap between peaks during the reaction, a competitive experiment between allylbenzene and methyl 4-allylbenzoate could not be preformed. Therefore a competitive experiment between 1-allyl-4-methylbenzene and methyl 4-allylbenzoate was performed and the obtained values for the kinetics of methyl 4-allylbenzoate were corrected afterwards.

Reaction time (min)	Conversion 1-allyl-4-methylbenzene	Conversion methyl 4-allylbenzoate	Kinetics 1-allyl-4-methylbenzene	Kinetics methyl 4-allylbenzoate	Corrected kinetic methyl 4-allylbenzoate values
0	0.00	0.00	0	0	0
5	0.05	0.02	0.048	0.022	0.022
10	0.16	0.27	0.169	0.315	0.311
15	0.21	0.37	0.237	0.457	0.453
30	0.28	0.49	0.330	0.667	0.660
60	0.47	0.74	0.637	1.334	1.320
90	0.57	0.82	0.839	1.732	1.715
120	0.66	0.90	1.079	2.322	2.298

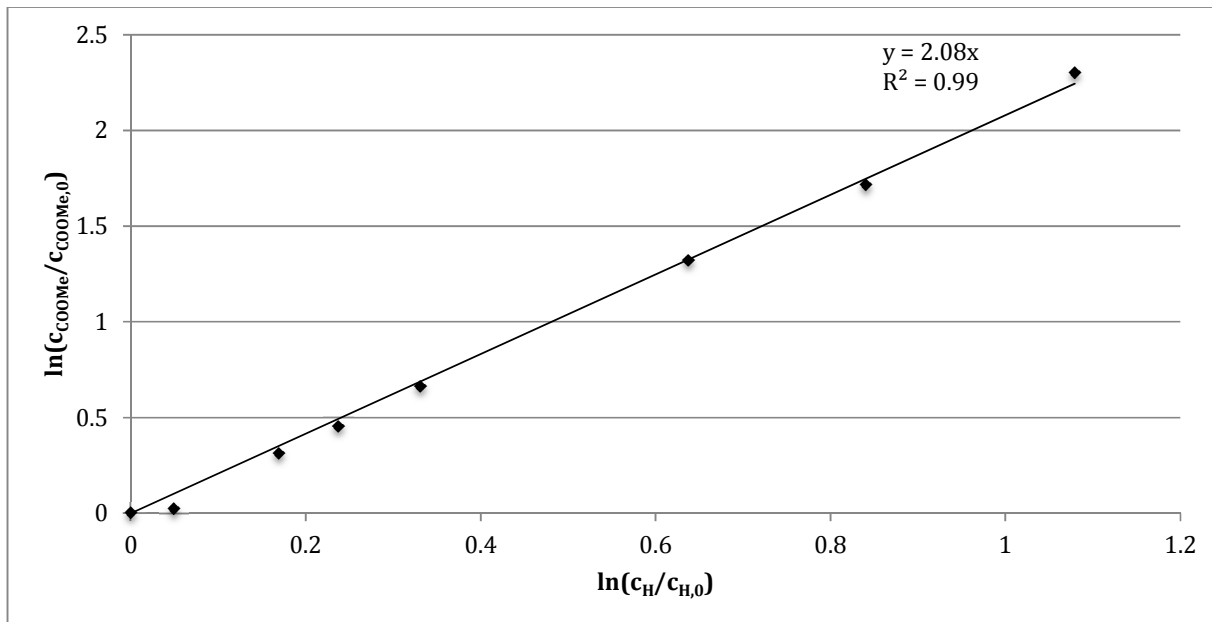


Figure S6 Kinetic plot for the competition experiment with 1-allyl-4-methylbenzene and methyl 4-allylbenzoate. The slope represents the reactivity of methyl 4-allylbenzoate relative to allylbenzene.

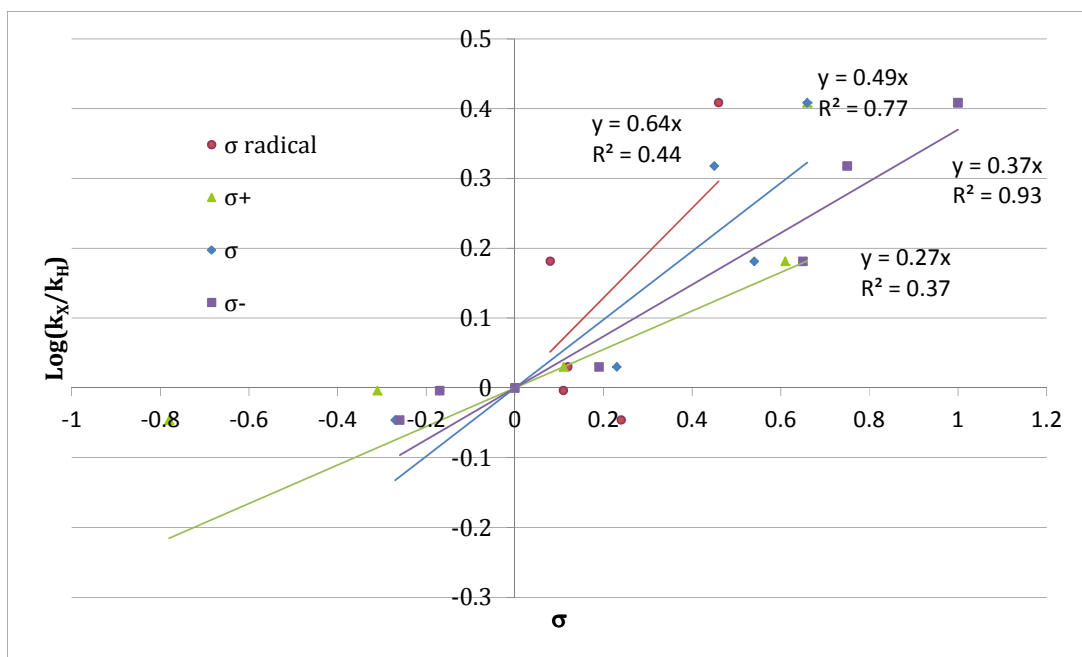


Figure S7 Hammett plots for the alkylation of *para*-substituted allylbenzenes using σ values from the literature. $\sigma^\bullet$ gives a slope of 0.64, σ gives 0.77, σ^- gives 0.37 and σ^+ gives a slope of 0.27.

KIE determination and deuterium experiments

Synthesis of d_2 -allylbenzene (**2**):

Methyl benzoate was reduced with LiAlD_4 ¹ to the corresponding benzylic alcohol and then halogenated with Br_2 and PPh_3 ² to benzyl bromide with deuterium in both benzylic positions. d_2 -allylbenzene was then obtained by a Kumada-Corriu reaction with vinylmagnesium bromide³.

Synthesis of deuterium marked (E)-methyl 2-nitro-5-phenylpent-4-enoate (**SI-1**):

To a 7 mL screw-cap vial was added $\text{Pd}[1,2\text{-bis(phenylsulfonyl)ethane}](\text{OAc})_2$ (0.05 mmol, 25 mg) and 2,6-dimethylbenzoquinone (0.75 mmol, 102 mg). d_2 -allylbenzene (0.5 mmol, 60 mg), methyl nitroacetate (1.5 mmol, 178 mg), and AcOH (0.25 mmol, 15 mg) were weighed out in a 1.0 mL vial and transferred to the catalyst mixture using dioxane (3 x 0.4 mL). DMSO (0.3 mL) was added followed by a stir bar. The reaction was sealed with a screw cap fitted with a PTFE septum and placed in an aluminum block preheated to 45 °C. The reaction mixture was stirred for 96 h, before the vial was allowed to cool to room temperature. The reaction mixture was diluted with saturated aqueous NH_4Cl (20 mL) and extracted with EtOAc (3 x 20 mL). The combined organics were dried over MgSO_4 , filtered, and concentrated under reduced pressure. The crude product was purified using silica-gel chromatography (15% EtOAc/hexane) giving the pure linear product (48 mg, 41 % yield) as a pale yellow oil. GC-MS: m/z 236. ^1H NMR (300 MHz, CDCl_3) δ 7.32 (m, 4H), 7.25 (m, 1H), 6.07 (m, 1H), 5.23 (dd, $J = 9.0, 5.6$ Hz, 1H), 3.85 (s, 3H), 3.10 (m, 2H); ^{13}C NMR (75 MHz, CDCl_3) δ 164.7, 136.3, 135.1 (t, $J = 92$ Hz), 128.8, 128.2, 126.6, 121.1, 87.7, 53.9, 33.9.

Synthesis of bis[chloro(d_2 -1,2,3-trihapto-allylbenzene)palladium (II)] (**SI-2**):

Procedure reported by White and coworkers⁴ was utilized with d_2 -allylbenzene, and a longer reaction time of 5 h, affording the product as a yellow solid (32 mg, 62 % yield).

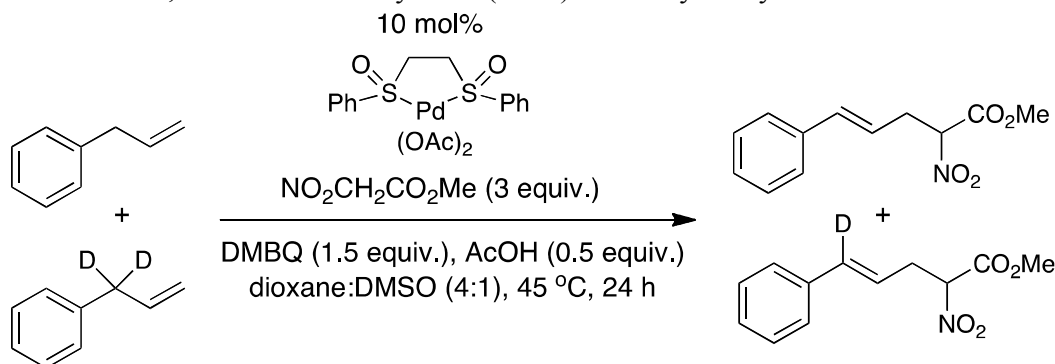
^1H NMR (300 MHz, CDCl_3) δ 7.47-7.40 (m, 2H), 7.30-7.17 (m, 3H), 5.73 (dd, $J = 11.9, 6.9$ Hz, 1H), 3.90 (d, $J = 6.9$ Hz, 1H), 2.97 (d, $J = 11.9$ Hz, 1H); ^{13}C NMR (75 MHz, CDCl_3) δ 136.8, 128.9, 128.4, 127.8, 105.7, 81.5 (t, $J = 86.6$ Hz), 59.3.

Synthesis of (E)-methyl 5-(4-chlorophenyl)-2-nitropent-4-enoate (**SI-3**):

To a 7 mL screw-cap vial was added Pd[1,2-bis(phenylsulfonyl)ethane](OAc)₂ (0.05 mmol, 25 mg) and 2,6-dimethylbenzoquinone (0.75 mmol, 102 mg). 1-allyl-4-chlorobenzene (0.5 mmol, 66 mg), methyl nitroacetate (1.5 mmol, 178 mg), and AcOH (0.25 mmol, 15 mg) were weighed out in a 1.0 mL vial and transferred to the catalyst mixture using dioxane (3 x 0.4 mL). DMSO (0.3 mL) was added followed by a stir bar. The reaction was sealed with a screw cap fitted with a PTFE septum and placed in an aluminum block preheated to 45 °C. The reaction mixture was stirred for 24 h, before the vial was allowed to cool to room temperature. The reaction mixture was diluted with saturated aqueous NH₄Cl (20 mL) and extracted with EtOAc (3 x 20 mL). The combined organics were dried over MgSO₄, filtered, and concentrated under reduced pressure. The crude product was purified using silica-gel chromatography (15% EtOAc/hexane) giving the pure linear product (73 mg, 54 % yield) as a pale yellow oil. GC-MS: *m/z* 269 (³⁵Cl), 271 (³⁷Cl). ¹H NMR (300 MHz, CDCl₃) δ 7.3-7.2 (m, 4H), 6.51 (dd, *J* = 15.8, 1.2 Hz, 1H), 6.07 (dt, *J* = 15.8, 7.2 Hz, 1H), 5.24 (dd, *J* = 9.0, 5.6 Hz, 1H), 3.85 (s, 3H), 3.25 – 2.94 (m, 2H); ¹³C NMR (75 MHz, CDCl₃) δ 164.5, 134.7, 134.3, 133.8, 128.9, 127.7, 121.8, 87.4, 53.8, 33.8.

Procedure for determining the deuterium-isotope-effect of allylbenzene in Palladium Catalyzed Allylic CH Alkylation

To a screw-cap vial was added Pd[1,2-bis(phenylsulfonyl)ethane](OAc)₂ (0.05 mmol, 25 mg), 2,6-dimethylbenzoquinone (0.75 mmol, 102 mg) and a stir bar. Allylbenzene (0.25 mmol, 30 mg), d₂-allylbenzene (0.25 mmol, 30 mg), methyl nitroacetate (1.5 mmol, 178 mg), AcOH (0.25 mmol, 15 mg) and dodecane as internal standard (0.125 mmol, 21 mg) were weighed out in a 1.0 mL vial and transferred to the catalyst mixture using dioxane:DMSO (4:1) (1.5 mL). The reaction was sealed with a screw cap fitted with a PTFE septum and placed in an aluminum block preheated to 45 °C. Samples of 0.05 mL each were withdrawn from the reaction mixture after 0, 0.5, 1, 2, 4, 6, 8, and 24 hrs reaction time, diluted with diethylether (1 mL) and analyzed by GCMS.



Scheme S1 The experiment for determination of the kinetic isotope effect (*k_H*/*k_D*) for allylbenzene in the allylic CH alkylation reaction.

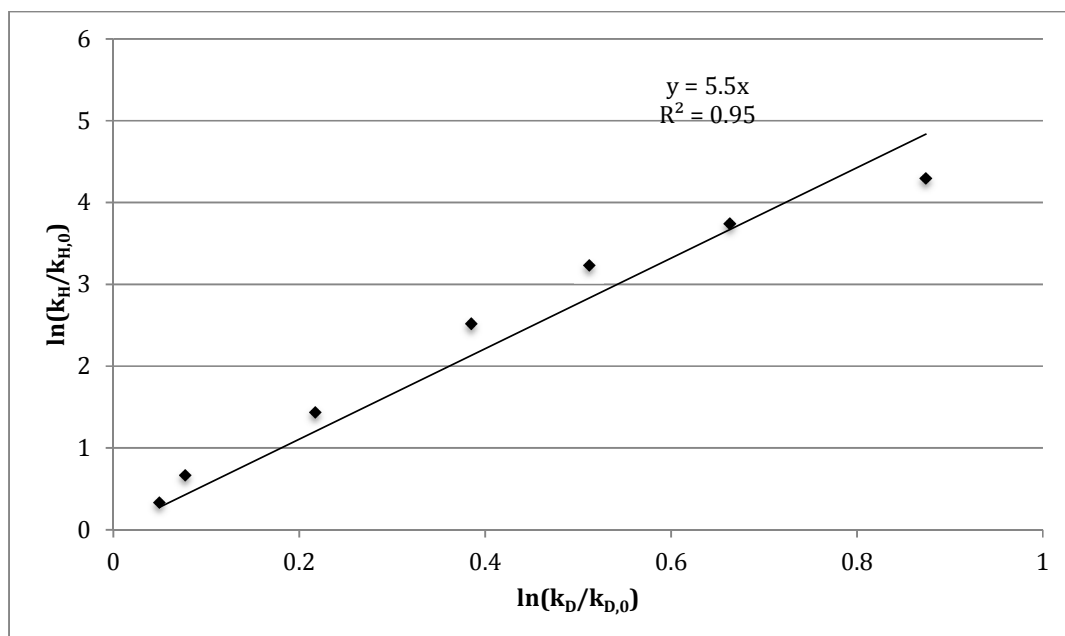


Figure S8 Deuterium-isotope-effect of allylbenzene in the allylic CH alkylation reaction.

Due to overlap in retention time between allylbenzene and d_2 -allylbenzene the deuterium-isotope-effect was determined from a distinct molecular ion of each of the substrates, $m/z=115$ and $m/z=120$ respectively, by SIM (Single Ion Monitoring).

The deuterium-isotope-effect was also determined by two separate experiments run in parallel, one with allylbenzene and the other with d_2 -allylbenzene as the substrate. Samples were taken every 10 minutes for the first two hours reaction time to determine the initial reaction rates. The resulting KIE was determined from the average of two runs.

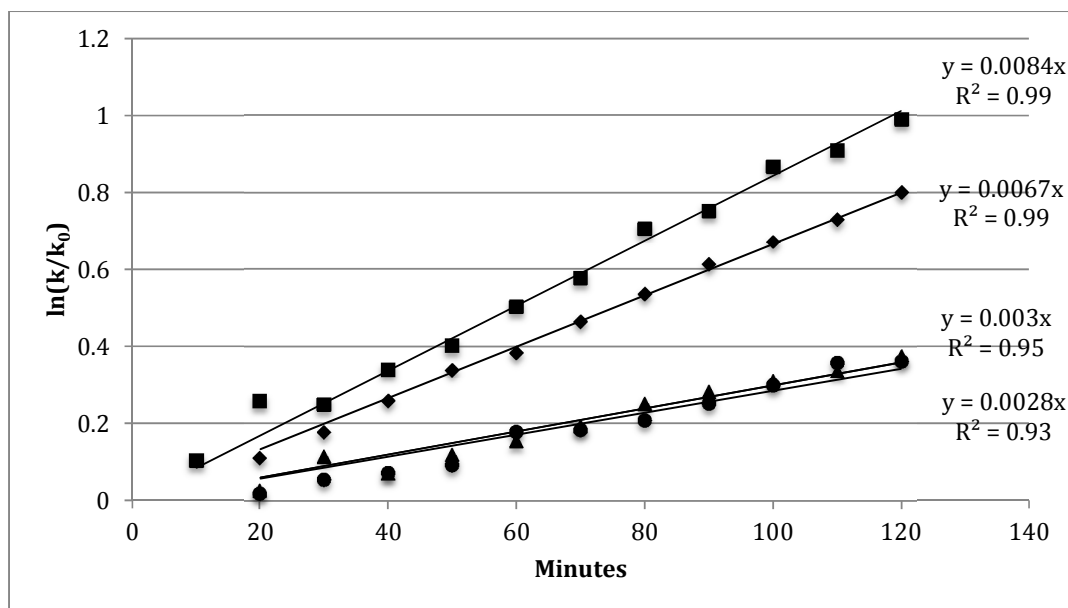
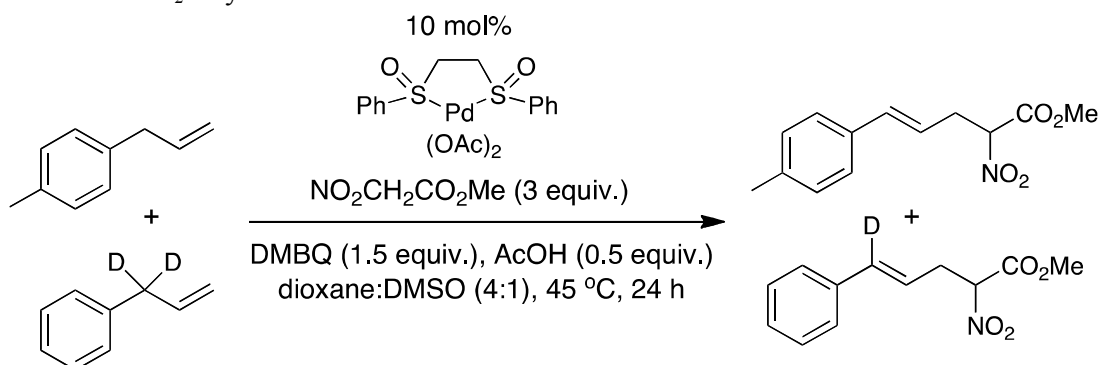


Figure S9 Deuterium-isotope-effect determined from two separate experiments run in parallel. The two top lines are experiments with allylbenzene and the two others are experiments with d_2 -allylbenzene.

The competitive experiment between d_2 -allylbenzene and 4-allyltoluene (0.25 mmol, 33 mg) in the allylic CH alkylation reaction was performed with the same conditions as the reaction between allylbenzene and d_2 -allylbenzene.



Scheme S2 Kinetic isotope effect (k_H/k_D) for allylbenzene determined by comparison to 4-allyltoluene in the allylic CH alkylation reaction.

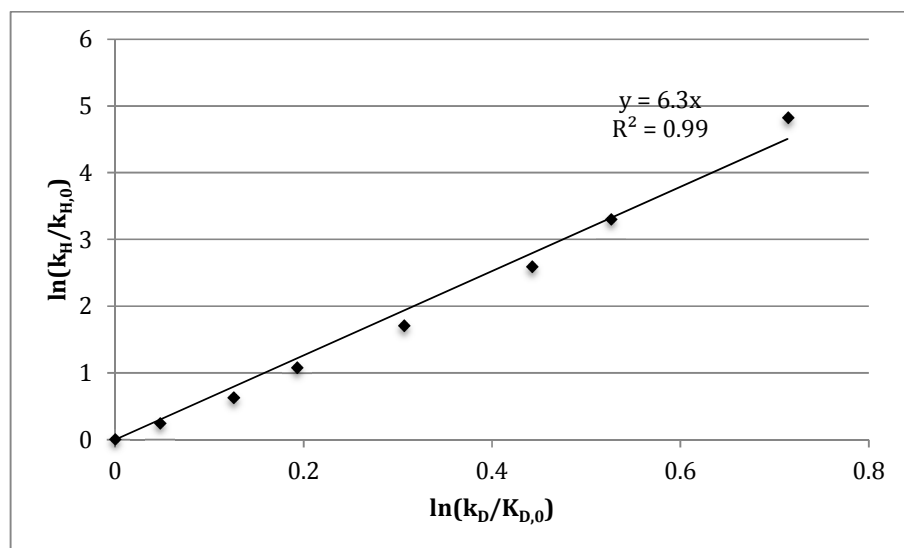


Figure S10 Deuterium-isotope-effect of allylbenzene based on comparison with 4-allyltoluene in the allylic CH alkylation reaction.

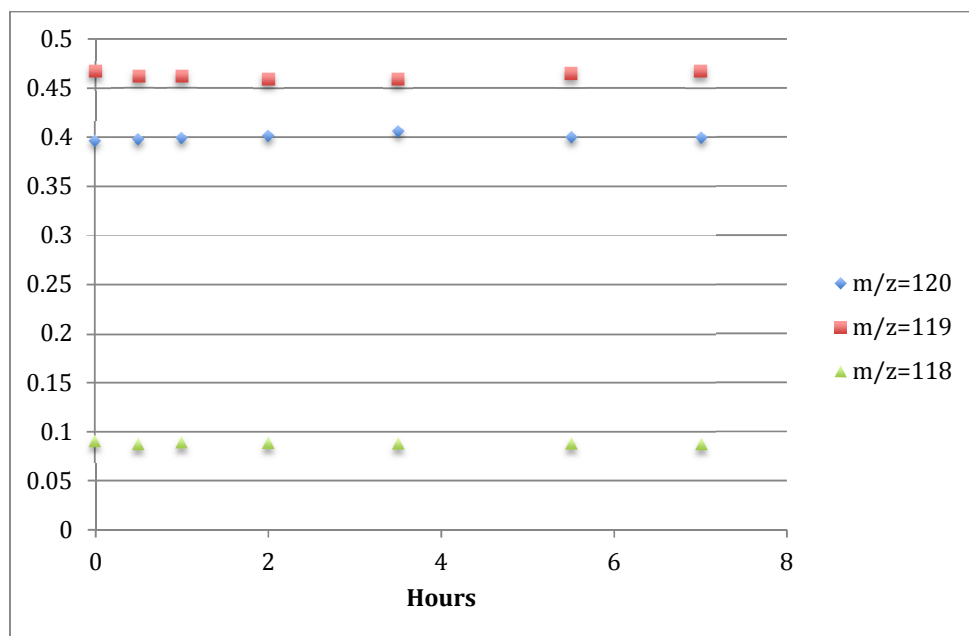


Figure S11 GC-MS data showing that the d_2 -allylbenzene ($m/z=120$) loses no deuterium content during the reaction.

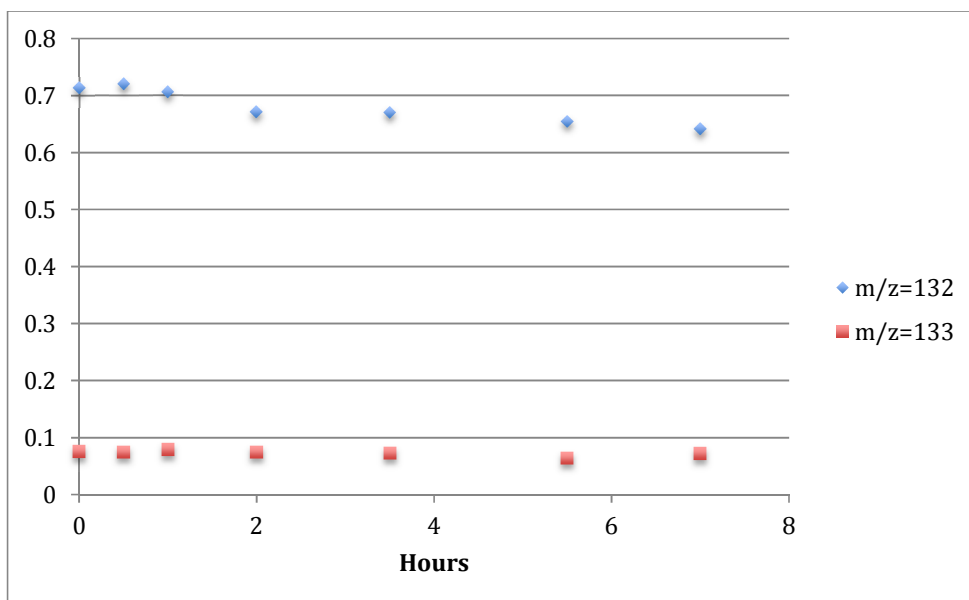
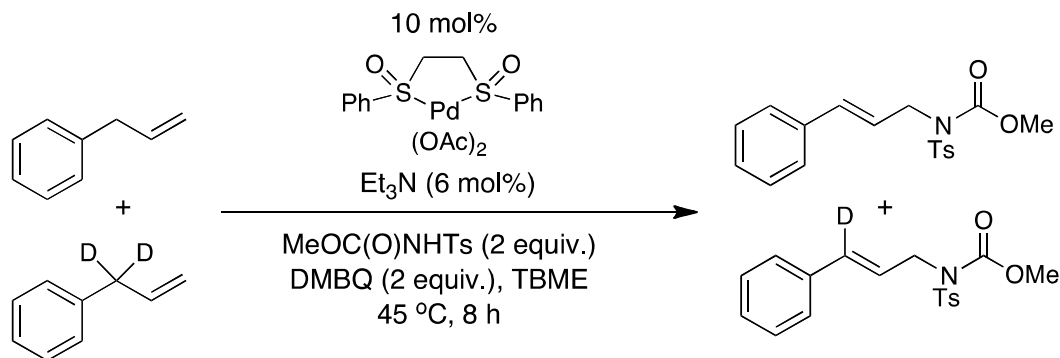


Figure S12 GC-MS data showing that no deuterium is incorporated in 4-allyltoluene ($m/z=132$) during the reaction.

Procedure for determining the deuterium-isotope-effect of allylbenzene in Palladium Catalyzed Allylic CH Amination

To a screw-cap vial was added Pd[1,2-bis(phenylsulfonyl)ethane](OAc)₂ (0.05 mmol, 25 mg), methyl tosylcarbamate (1.0 mmol, 229 mg), 2,6-dimethylbenzoquinone (1 mmol, 136 mg) and a stir bar. Allylbenzene (0.25 mmol, 30 mg), d_2 -allylbenzene (0.25 mmol, 30 mg), Et₃N (0.03 mmol, 4 μ L) and dodecane as internal standard (0.125 mmol, 21 mg) were weighed out in a 1.0 ml vial and

transferred to the catalyst mixture using *tert*-butyl methyl ether (1.0 mL). The reaction was sealed with a screw cap fitted with a PTFE septum and placed in an aluminum block preheated to 45 °C. Samples of 0.05 mL each were withdrawn from the reaction mixture after 0, 0.5, 1, 2, 3, 4, 6 and 8 hrs reaction time, diluted with diethylether (1 mL) and analyzed by GCMS.



Scheme S3 The experiment for determination of the kinetic isotope effect ($k_{\text{H}}/k_{\text{D}}$) for allylbenzene in the allylic CH acetoxylation reaction.

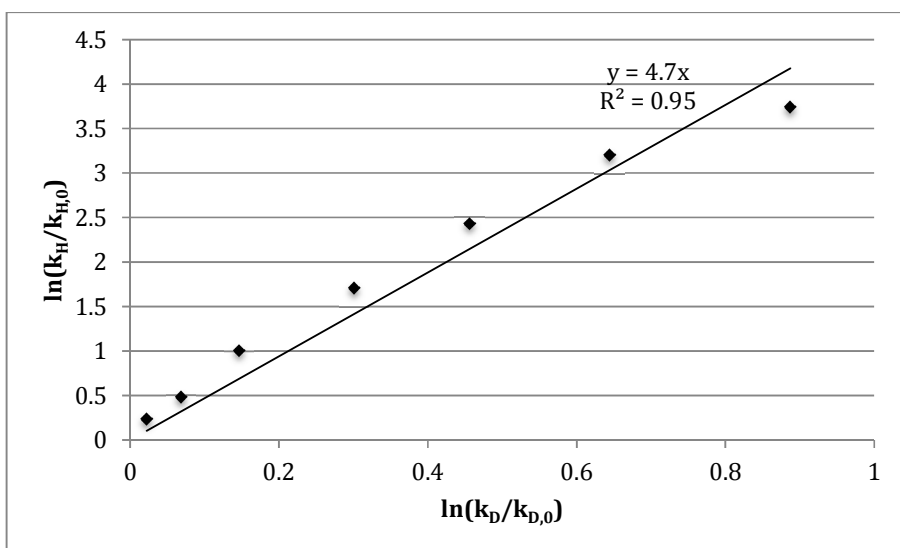
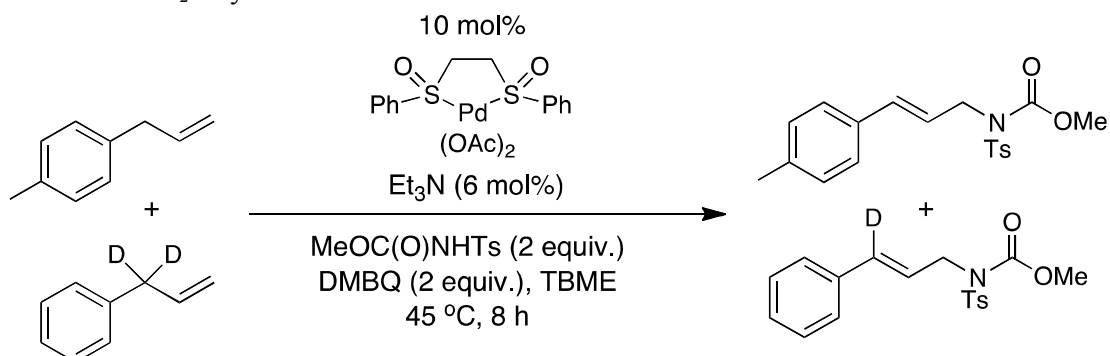


Figure S13 Deuterium-isotope-effect of allylbenzene in the allylic CH amination reaction.

Competitive experiment between d_2 -allylbenzene and 4-allyltoluene (0.25 mmol, 33 mg) in the allylic CH amination reaction was performed with the same conditions as the reaction between allylbenzene and d_2 -allylbenzene.



Scheme S4 Competitive experiment between d_2 -allylbenzene and 4-allyltoluene in the allylic CH amination reaction.

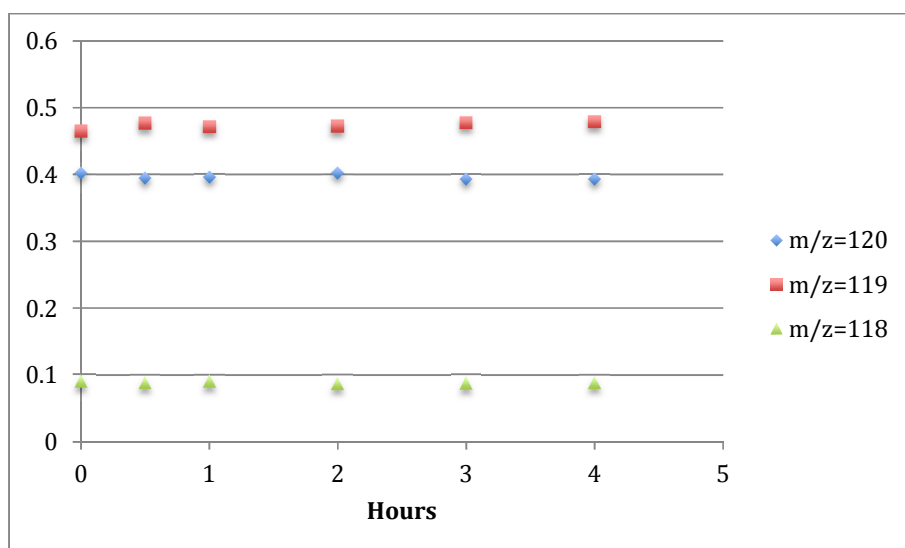


Figure S14 GC-MS data showing that the d_2 -allylbenzene ($m/z=120$) loses no deuterium content during the reaction.

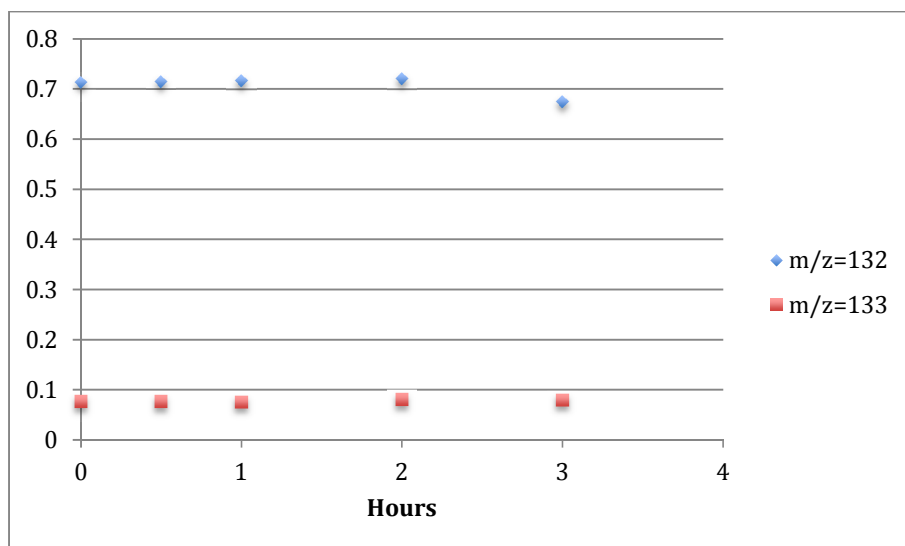
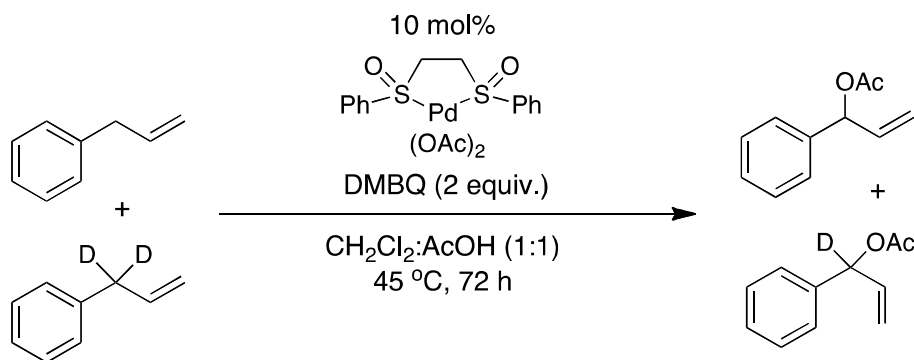


Figure S15 GC-MS data showing that no deuterium is incorporated in 4-allyltoluene ($m/z=132$) during the reaction.

Procedure for determining the deuterium-isotope-effect of allylbenzene in Palladium Catalyzed Allylic CH Acetoxylation

To a screw-cap vial was added $\text{Pd}[1,2\text{-bis(phenylsulfonyl)ethane}](\text{OAc})_2$ (0.05 mmol, 25 mg), 2,6-dimethylbenzoquinone (1 mmol, 136 mg) and a stir bar. Allylbenzene (0.25 mmol, 30 mg), d_2 -allylbenzene (0.25 mmol, 30 mg) and dodecane as internal standard (0.125 mmol, 21 mg) were weighed out in a 1.0 ml vial and transferred to the catalyst mixture using $\text{CH}_2\text{Cl}_2:\text{AcOH}$ (3 mL). The reaction was sealed with a screw cap fitted with a PTFE septum and placed in an aluminum block preheated to 45 °C. Samples of 0.05 mL each were withdrawn from the reaction mixture after 0, 1, 3, 6, 24, 28, 32 and 48 hrs reaction time, diluted with diethylether (1 mL) and analyzed by GCMS.



Scheme S5 The experiment for determination of the kinetic isotope effect (k_H/k_D) for allylbenzene in the allylic CH amination reaction.

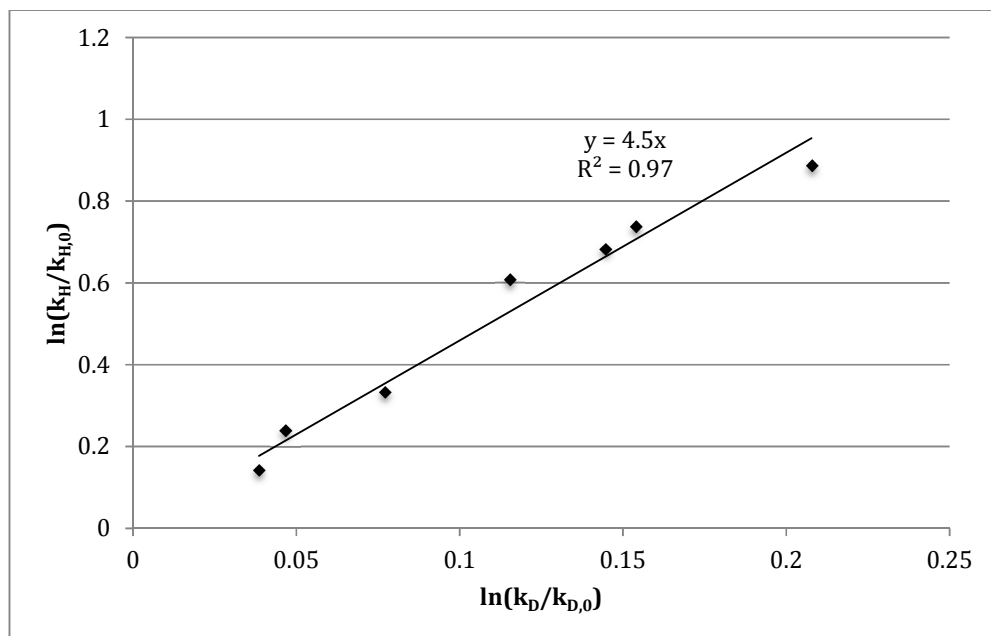


Figure S16 Deuterium-isotope-effect of allylbenzene in the allylic CH acetoxylation reaction. Competitive experiment between d_2 -allylbenzene and 4-allyltoluene (0.25 mmol, 33 mg) in the allylic CH acetoxylation reaction was performed with the same conditions as the reaction between allylbenzene and d_2 -allylbenzene.

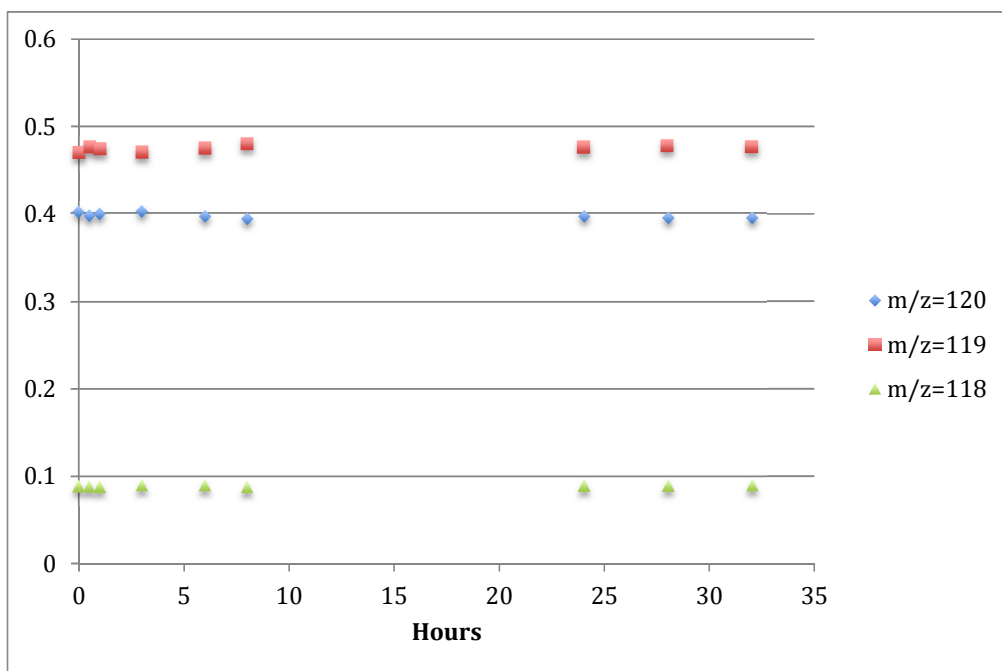


Figure S17 GC-MS data showing that the d_2 -allylbenzene ($m/z=120$) loses no deuterium content during the reaction.

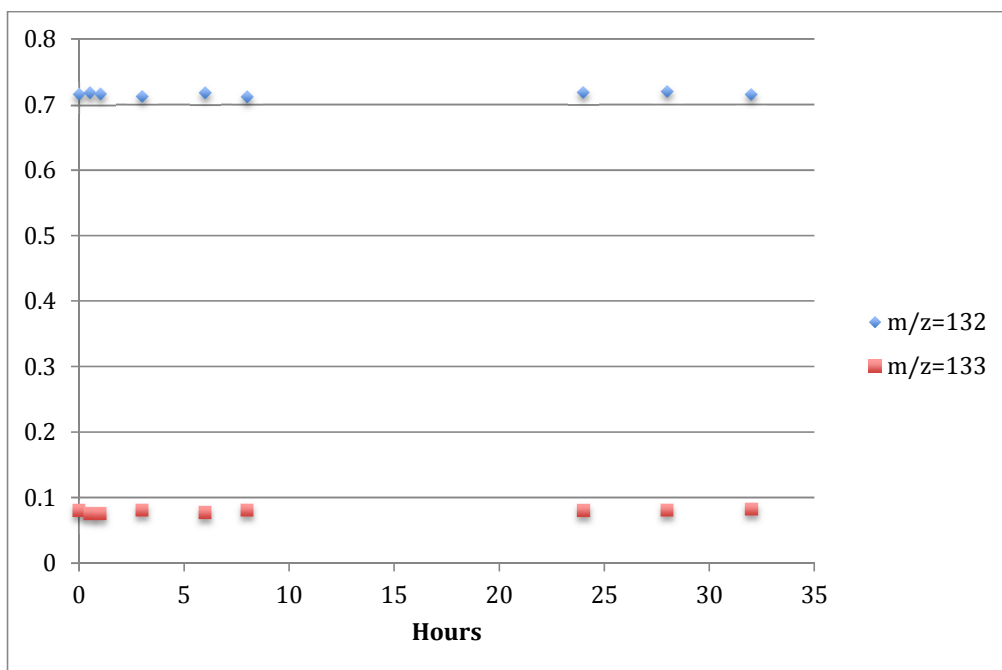
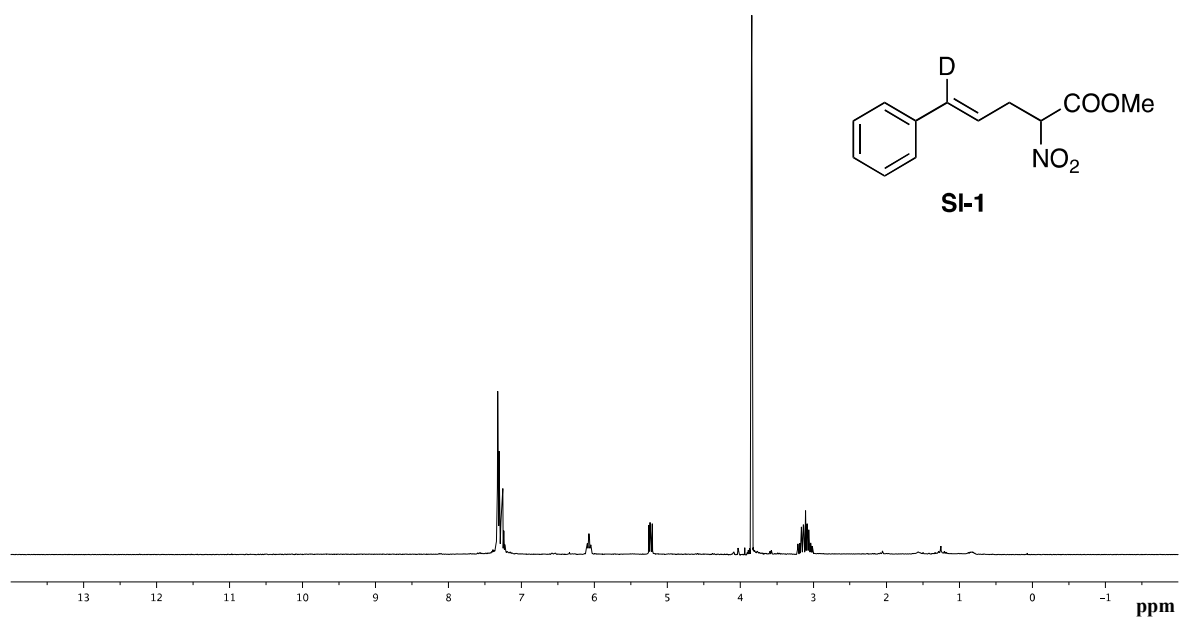
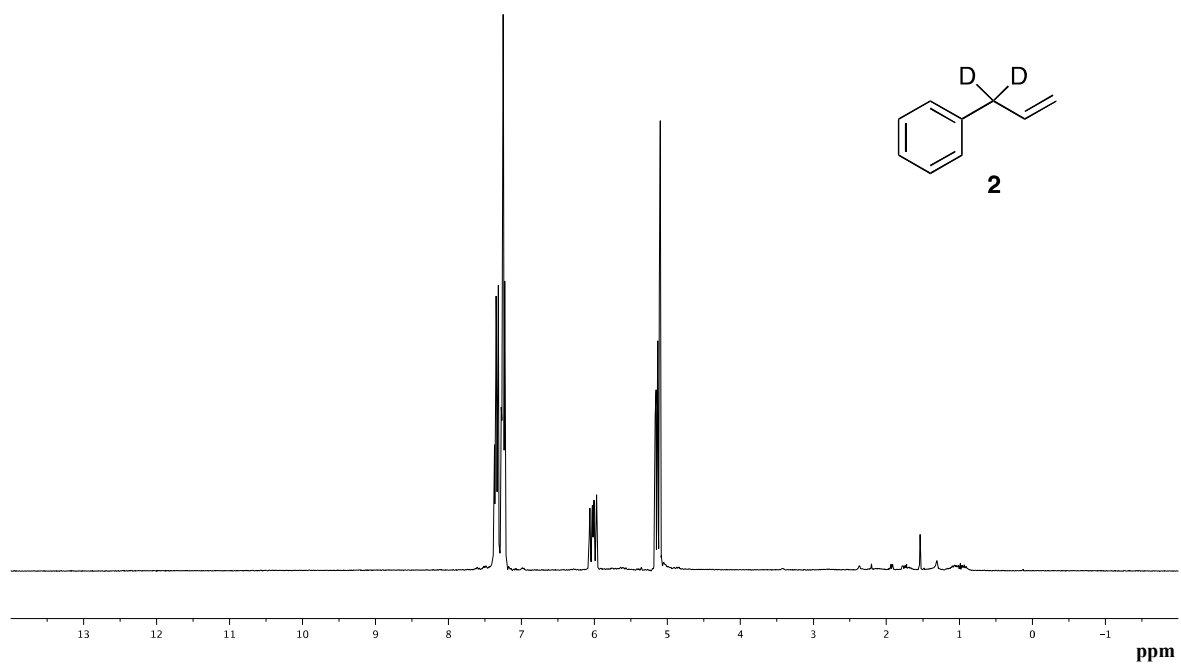
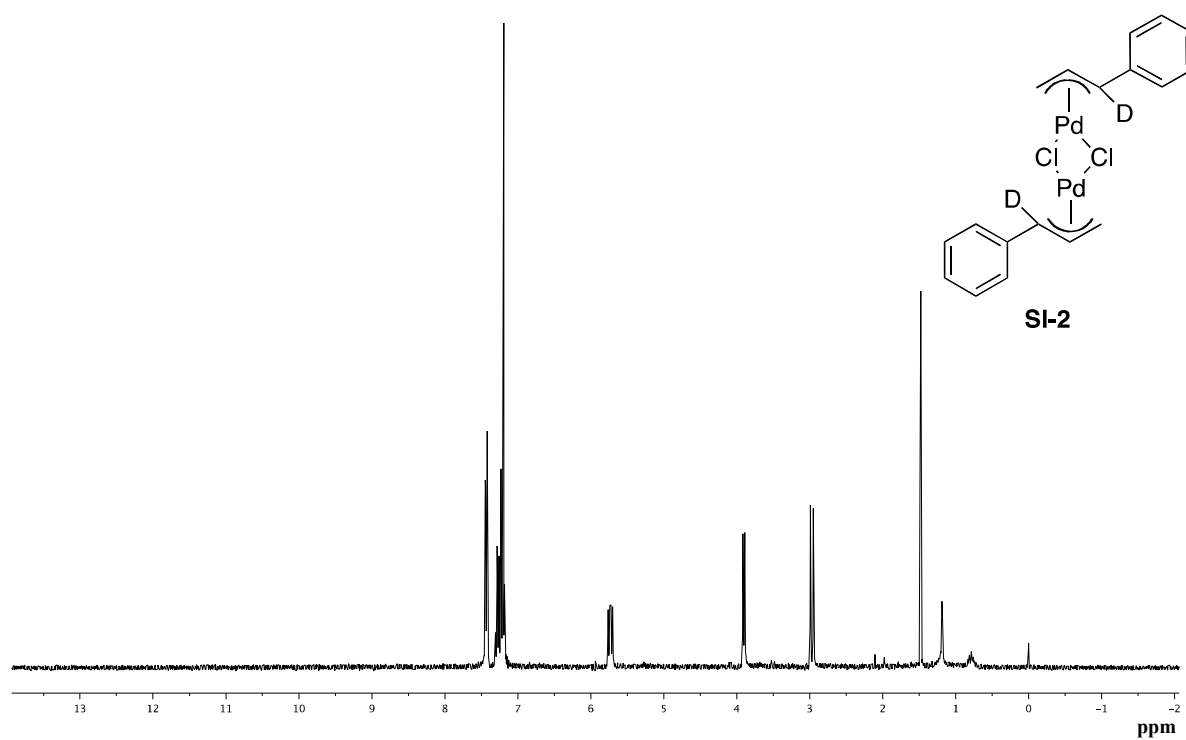
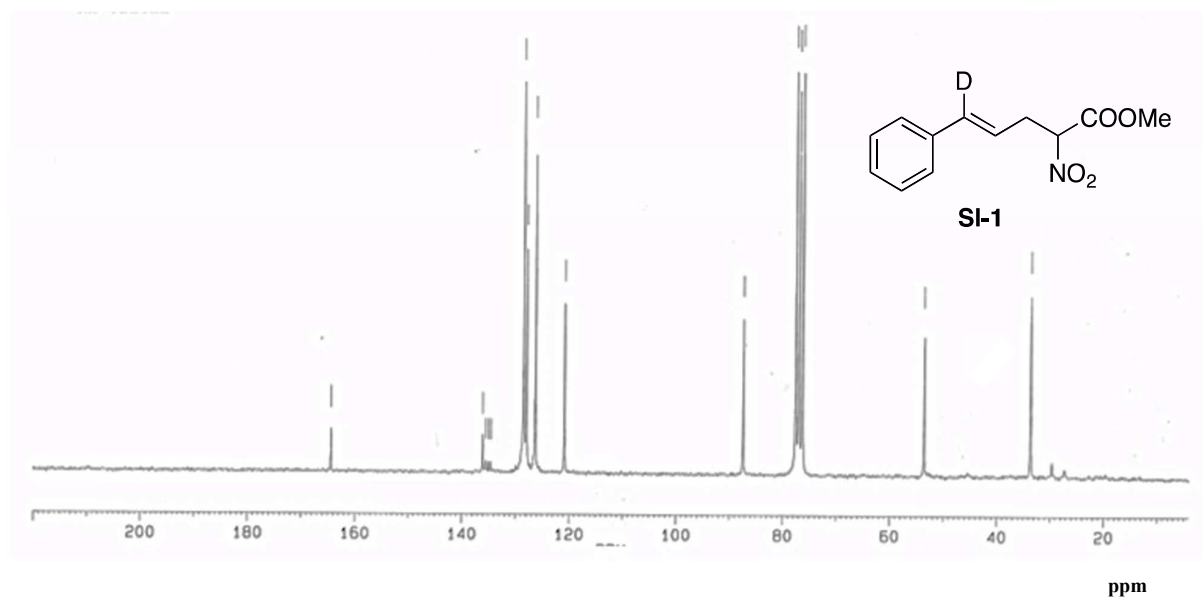
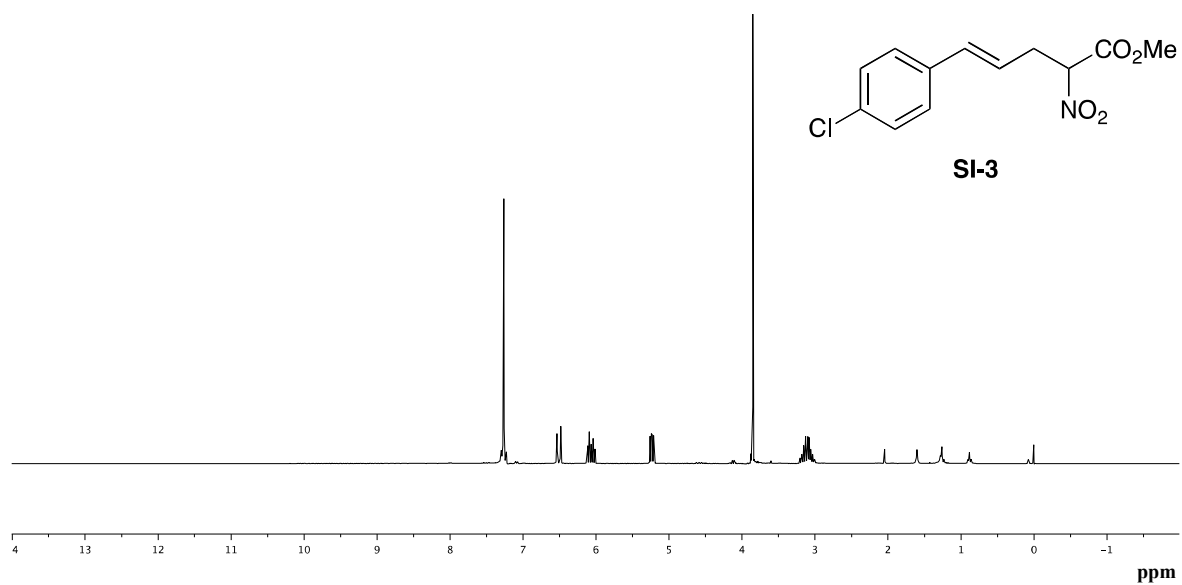
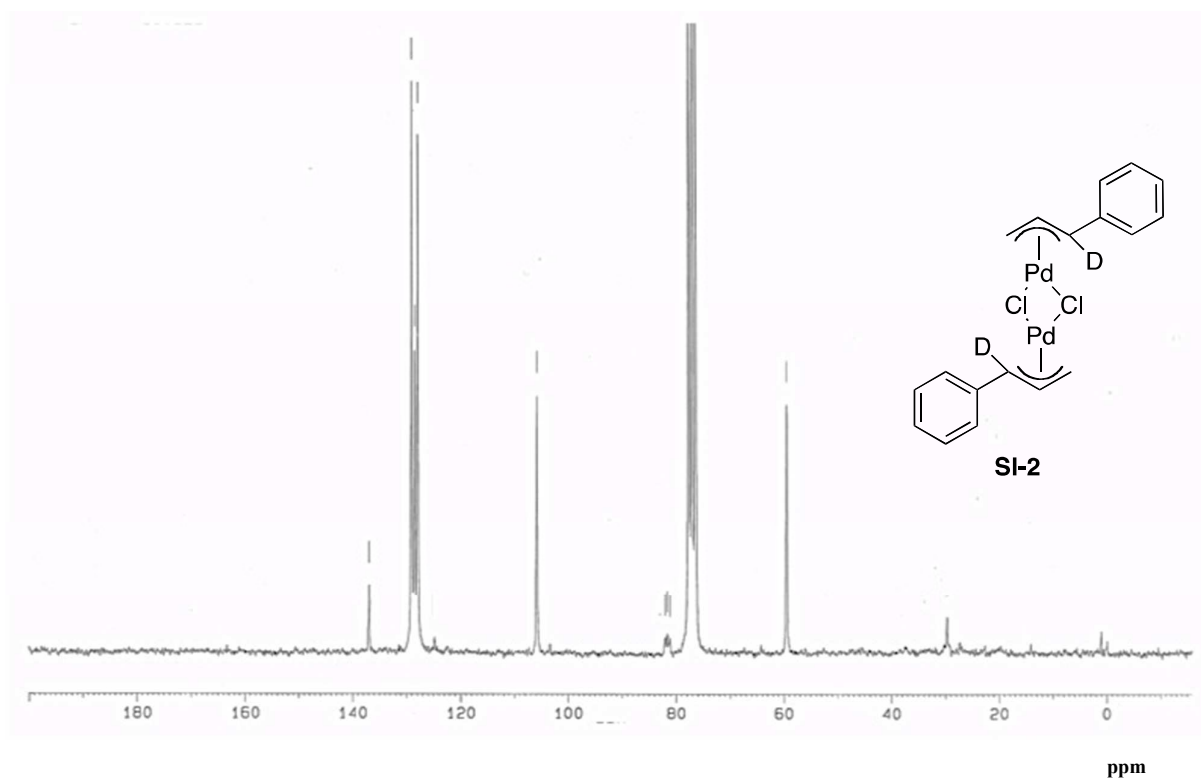


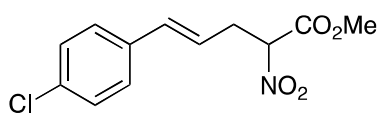
Figure S18 GC-MS data showing that no deuterium is incorporated in 4-allyltoluene ($m/z=132$) during the reaction.

NMR data

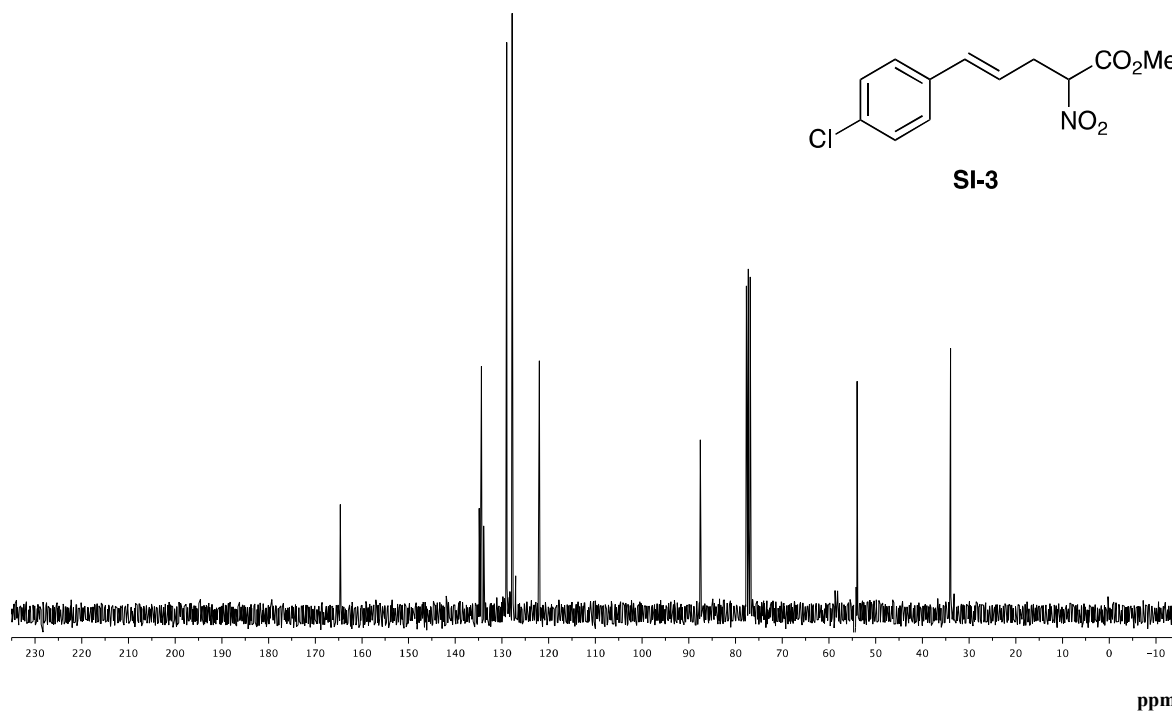








SI-3



XYZ coordinates and energies from computational study

The description of the computational method is included in the main text of the manuscript.

Pd_Bisulfoxide_2Acetate

E_scf (B3LYP/LACVP*) = -2072.35905820344 a.u.

E_scf (M06/LACVP*) = -2071.57851982228 a.u.

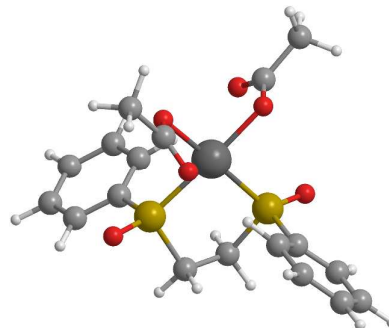
E_Gibbs = -2072.067993 a.u.

E_Gibbs(318K) = -2072.074334 a.u.

E_solv = -2072.40209458304 a.u.

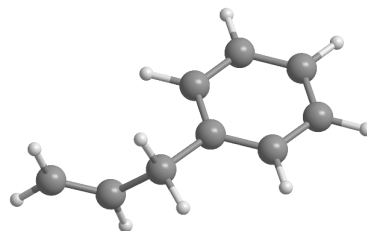
ZPE = 222.482 kcal/mol

Pd	0.17330	-0.35000	-0.04100
S	0.21020	1.35800	1.57480
S	-2.13920	-0.02230	-0.31880
O	0.89750	1.35880	2.90030
O	-2.90670	0.09040	-1.59500
C	0.63750	2.86890	0.67650
C	0.95300	3.99760	1.43760
C	1.28310	5.17900	0.77740
C	1.30440	5.21770	-0.62020
C	1.01070	4.07380	-1.36330
C	0.68030	2.87740	-0.72100
C	-1.59620	1.62920	1.89750
C	-2.43910	1.53890	0.63760
C	-2.92860	-1.24950	0.74870
C	-4.30140	-1.45290	0.58270
C	-4.94100	-2.39240	1.38840
C	-4.20970	-3.11850	2.33360
C	-2.83570	-2.91800	2.46920
C	-2.17380	-1.98310	1.66900
H	0.96370	3.93540	2.52200
H	1.53350	6.06600	1.35270
H	1.56720	6.14070	-1.13010
H	1.05070	4.09740	-2.44850
H	0.49530	1.97750	-1.30790
H	-1.85280	0.85290	2.62550
H	-1.68990	2.60970	2.37430
H	-2.20210	2.33350	-0.07720
H	-3.50880	1.57240	0.86560
H	-4.84300	-0.90300	-0.18140
H	-6.00710	-2.56600	1.27130
H	-4.71240	-3.85450	2.95540
H	-2.26330	-3.49880	3.18650
H	-1.09310	-1.86420	1.75110
O	2.13390	-0.75220	0.34840
O	1.23390	-1.98060	2.01170
O	0.13920	-1.70480	-1.56650
O	0.36200	0.16160	-2.81220
C	2.20500	-1.57060	1.37110
C	3.63120	-1.98280	1.70560
C	0.27280	-1.06390	-2.70330
C	0.31750	-1.99090	-3.90880
H	4.24520	-1.09570	1.89190
H	3.63210	-2.62770	2.58650
H	4.07200	-2.51490	0.85570
H	-0.54040	-2.67040	-3.89680
H	1.22410	-2.60460	-3.86700
H	0.31620	-1.40150	-4.82770



Allylbenzene

E_scf (B3LYP/LACVP*) = -348.94958989188 a.u.

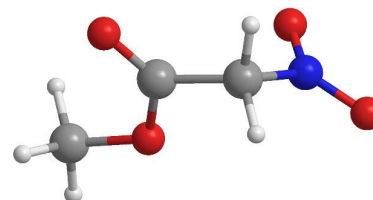


E_scf (M06/LACVP*) = -348.65922367329 a.u.
 E_Gibbs = -348.820936 a.u.
 E_Gibbs(318K) = -348.823816 a.u.
 E_solv = -348.95003655272 a.u.
 ZPE = 101.89 kcal/mol

C	0.00020	-0.00030	0.00110
C	0.00100	0.00060	1.40300
C	1.23210	0.00020	2.06850
C	2.43330	0.00670	1.35620
C	2.41950	0.01170	-0.03840
C	1.19710	0.00770	-0.71400
H	-0.94850	-0.01260	-0.53030
H	1.25180	-0.00320	3.15650
H	3.37910	0.00660	1.89250
H	3.35290	0.01620	-0.59550
H	1.17630	0.00900	-1.80110
C	-1.30980	0.01410	2.17810
H	-1.08280	-0.06950	3.25020
H	-1.81360	0.98110	2.04220
C	-2.25220	-1.09330	1.77680
H	-1.86270	-2.10590	1.88860
C	-3.48530	-0.91530	1.29980
H	-3.91000	0.07810	1.16760
H	-4.11970	-1.75470	1.02780

Methyl_Nitroacetate

E_scf (B3LYP/LACVP*) = -472.86630684448 a.u.
 E_scf (M06/LACVP*) = -472.62177095945 a.u.
 E_Gibbs = -472.807334 a.u.
 E_Gibbs(318K) = -472.810306 a.u.
 E_solv = -472.88312714186 a.u.
 ZPE = 58.922 kcal/mol

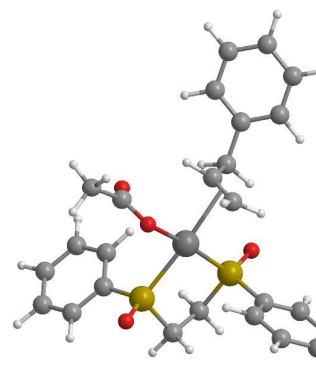


C	0.00000	0.00000	0.00000
H	0.00000	0.00000	1.09020
N	1.44460	0.00000	-0.43920
O	1.65900	-0.29460	-1.60970
O	2.28030	0.33100	0.39140
C	-0.75440	-1.19430	-0.57030
O	-0.13800	-2.33900	-0.24750
O	-1.78820	-1.10370	-1.18820
C	-0.74850	-3.53870	-0.76330
H	-0.12450	-4.35730	-0.40620
H	-1.77020	-3.63540	-0.38790
H	-0.76460	-3.51170	-1.85550
H	-0.44130	0.91940	-0.38660

Pd_Bisulfoxide_Allylbenzene_Acetate

E_scf (B3LYP/LACVP*) = -2192.62716360606 a.u.
 E_scf (M06/LACVP*) = -2191.68601439312 a.u.
 E_Gibbs = -2192.2284 a.u.
 E_Gibbs(318K) = -2192.2353 a.u.
 E_solv = -2192.70663936665 a.u.
 ZPE = 293.109 kcal/mol

O	0.51030	-0.10320	-0.82940
H	-0.56850	0.69510	2.30510
C	0.29650	0.50450	2.95030
C	1.48070	0.13100	2.11060
C	2.77460	0.58380	2.28080
C	-1.07000	-2.83710	4.22080

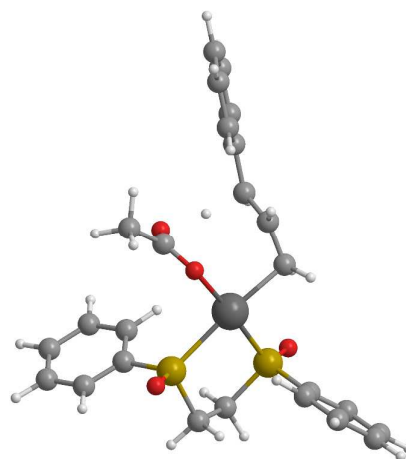


O	0.81410	3.77100	2.98410
C	-0.41640	-2.95820	5.44860
C	-0.84930	-1.71810	3.41830
C	0.02090	-0.70540	3.84150
C	0.68010	-0.83770	5.07070
C	0.45780	-1.95570	5.87320
H	-0.58930	-3.82980	6.07360
H	-1.75480	-3.61260	3.88890
H	-1.36420	-1.62520	2.46430
H	1.35940	-0.05730	5.40750
H	0.96490	-2.04370	6.83000
Pd	2.17760	1.86860	0.50280
S	1.78970	3.76850	1.85250
S	2.13810	3.35180	-1.41710
O	3.26300	3.33450	-2.39420
C	3.38570	4.38800	2.41030
C	3.40510	5.16860	3.56890
C	4.63130	5.65750	4.01970
C	5.80650	5.36340	3.32300
C	5.77030	4.56900	2.17400
C	4.55460	4.06560	1.71100
C	1.25550	5.05740	0.63230
C	2.09770	5.03610	-0.62890
C	0.55070	3.31690	-2.25920
C	0.49030	3.89440	-3.53230
C	-0.73800	3.91860	-4.18920
C	-1.87230	3.37100	-3.58090
C	-1.78530	2.77970	-2.31940
C	-0.56240	2.73770	-1.64710
H	2.48510	5.36830	4.10970
H	4.66830	6.26300	4.92010
H	6.75720	5.74530	3.68280
H	6.68650	4.32990	1.64270
H	4.52960	3.42220	0.83430
H	0.20190	4.82840	0.44080
H	1.32360	6.01930	1.15020
H	3.14820	5.27580	-0.43770
H	1.70610	5.72780	-1.38050
H	1.38600	4.29250	-3.99960
H	-0.80820	4.35800	-5.17970
H	-2.82510	3.39070	-4.10190
H	-2.66140	2.32830	-1.86410
H	-0.47360	2.21610	-0.69960
O	2.70450	0.36200	-0.74880
C	1.66600	-0.31720	-1.19680
C	2.03780	-1.38450	-2.20940
H	2.84610	-2.01450	-1.82660
H	2.40140	-0.90840	-3.12660
H	1.15990	-1.99190	-2.43560
H	1.30270	-0.66190	1.39050
H	0.48880	1.40150	3.54390
H	3.02210	1.25750	3.09850
H	3.60150	0.06650	1.80050

Pd_Bisulfoxide_Allylbenzene_Acetate_Abstraction_TS (**A**)

E_scf (B3LYP/LACVP*) = -2192.59758069590 a.u.
 E_scf (M06/LACVP*) = -2191.6529988305 a.u.
 E_Gibbs = -2192.205317 a.u.
 E_Gibbs(318K) = -2192.211977 a.u.
 E_solv = -2192.67481266168 a.u.
 ZPE = 289.469 kcal/mol

O	0.03180	-0.00940	0.01320
H	0.00920	0.01240	1.49290
C	0.19930	0.00430	2.79910
C	-0.26550	1.33240	2.88800
C	-1.63950	1.71300	2.96910
C	3.23750	-2.09900	3.59670
O	-4.83400	0.74560	2.82690
C	4.28870	-1.24690	3.25720
C	1.91810	-1.67780	3.43940
C	1.62980	-0.39410	2.95210
C	2.69500	0.45080	2.60070
C	4.01240	0.02890	2.75750
H	5.31760	-1.57450	3.37570
H	3.44320	-3.09310	3.98350
H	1.10300	-2.34710	3.70560
H	2.50430	1.44360	2.19890
H	4.82700	0.69470	2.48570
Pd	-2.33230	1.90660	0.94120
S	-4.48990	1.75910	1.78630
S	-3.35400	2.13420	-1.31630
O	-3.00920	3.24960	-2.24980
C	-5.03000	3.39400	2.31200
C	-6.02300	3.45770	3.29240
C	-6.45170	4.71360	3.72270
C	-5.89000	5.87320	3.18160
C	-4.88680	5.79010	2.21230
C	-4.43990	4.54320	1.77450
C	-5.54420	1.43260	0.29680
C	-5.15840	2.29170	-0.89350
C	-3.34350	0.56120	-2.19360
C	-3.60970	0.57990	-3.56600
C	-3.62190	-0.63020	-4.25760
C	-3.36440	-1.82830	-3.58390
C	-3.07730	-1.82550	-2.21740
C	-3.05490	-0.62240	-1.50980
H	-6.42920	2.54560	3.71890
H	-7.21890	4.78480	4.48780
H	-6.22610	6.84690	3.52570
H	-4.44090	6.69280	1.80560
H	-3.63520	4.47480	1.04610
H	-5.41170	0.36250	0.10600
H	-6.57750	1.61220	0.61010
H	-5.30220	3.35980	-0.70170
H	-5.72840	2.01300	-1.78460
H	-3.78060	1.52270	-4.07720
H	-3.82510	-0.63730	-5.32430
H	-3.37290	-2.76670	-4.13040
H	-2.85220	-2.75480	-1.70290
H	-2.78360	-0.60960	-0.45760
O	-0.46190	2.18430	0.04810
C	0.21310	1.17910	-0.38310
C	1.29200	1.46270	-1.40000
H	2.26980	1.23410	-0.96100
H	1.26490	2.50110	-1.73400
H	1.15960	0.79000	-2.25400
H	0.46270	2.13530	2.78010



H	-0.52240	-0.75880	3.10590
H	-2.31310	0.96660	3.39300
H	-1.80560	2.72390	3.34650

Pd_Bisulfoxide_Allylbenzene_Acetate_Abstraction_6Membered_TS (**B**)

E_scf (B3LYP/LACVP*) = -2192.60235869582 a.u.

E_scf (M06/LACVP*) = -2191.65784538658 a.u.

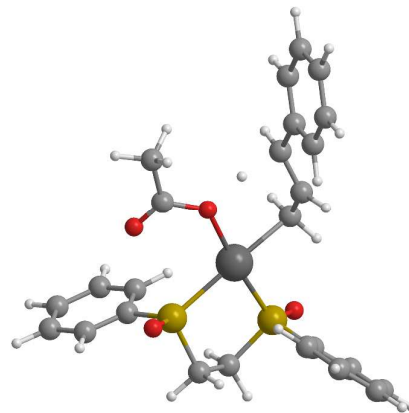
E_Gibbs = -2192.206611 a.u.

E_Gibbs(318K) = -2192.213173 a.u.

E_solv = -2192.68085368379 a.u.

ZPE = 289.490 kcal/mol

O	-0.00690	0.00240	-0.00370
H	-0.01670	0.00240	1.41630
C	0.54290	0.00230	2.67770
O	-0.53800	1.01470	-1.92780
C	1.46450	-0.96770	2.22500
C	2.52850	-0.71210	1.31830
C	-2.14260	-1.67680	4.79120
O	4.94030	-1.76270	-0.79550
C	-2.52690	-0.61820	5.61960
C	-1.15160	-1.49380	3.83050
C	-0.51850	-0.24760	3.68770
C	-0.92210	0.81240	4.51630
C	-1.91390	0.62770	5.47740
H	-3.30340	-0.76320	6.36500
H	-2.62320	-2.64610	4.89010
H	-0.88220	-2.32420	3.18140
H	-0.44510	1.78490	4.41470
H	-2.20690	1.45720	6.11480
Pd	1.67780	-0.98280	-0.69920
S	3.58810	-2.20090	-1.25340
S	0.96220	-1.22790	-3.05150
O	-0.38190	-1.74000	-3.45030
C	3.33300	-3.93280	-0.81390
C	4.47230	-4.69730	-0.55280
C	4.30930	-6.04030	-0.21130
C	3.03010	-6.59710	-0.13100
C	1.90060	-5.81260	-0.38060
C	2.04360	-4.46570	-0.71690
C	3.58760	-2.25400	-3.10410
C	2.19900	-2.47540	-3.67220
C	1.43740	0.19890	-4.04770
C	0.78230	0.37740	-5.26550
C	1.14780	1.46240	-6.06260
C	2.14300	2.34540	-5.63690
C	2.77430	2.15950	-4.40470
C	2.41520	1.08410	-3.59190
H	5.45670	-4.24160	-0.59910
H	5.18370	-6.64900	-0.00200
H	2.91170	-7.64240	0.13760
H	0.90690	-6.24370	-0.30520
H	1.16600	-3.84540	-0.88450
H	4.01530	-1.29100	-3.40170
H	4.27960	-3.05280	-3.38800
H	1.78020	-3.44510	-3.38630
H	2.20330	-2.40260	-4.76360
H	-0.00870	-0.30350	-5.56490
H	0.64660	1.62250	-7.01250
H	2.41790	3.19060	-6.26090
H	3.53220	2.86010	-4.06690
H	2.86960	0.96120	-2.61120
C	-0.77200	0.83230	-0.75240



C	-1.91050	1.47240	0.01020
H	-1.53020	2.05340	0.85800
H	-2.57740	0.70020	0.40920
H	-2.46810	2.12780	-0.66020
H	1.28570	-2.00810	2.49890
H	0.91150	1.03230	2.64760
H	2.85460	0.32540	1.21710
H	3.35170	-1.42200	1.35030

Pd_Bisulfoxide_Allylbenzene_Acetate_External_Acetate_Abstraction_TS (C)

E_scf (B3LYP/LACVP*) = -2421.28149348030 a.u.

E_scf (M06/LACVP*) = -2420.23362721681 a.u.

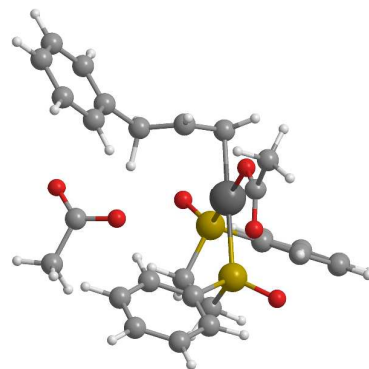
E_Gibbs = -2420.843169 a.u.

E_Gibbs(318K) = -2420.850349 a.u.

E_solv = -2421.3179288444 a.u.

ZPE = 321.551 kcal/mol

O	-0.01470	-0.01380	-0.00070
H	-0.00150	-0.01240	1.45920
C	0.33620	-0.00880	2.75600
O	-5.71850	-1.16100	1.63460
C	-0.96080	-0.20240	3.26570
C	-1.58350	-1.46660	3.50250
C	1.18780	3.70340	3.03470
O	0.16380	-3.30750	1.16800
C	2.55200	3.64610	3.33110
C	0.45630	2.53510	2.82830
C	1.07800	1.28030	2.92900
C	2.45200	1.23330	3.20870
C	3.18260	2.40400	3.41130
H	3.11880	4.56050	3.48670
H	0.69000	4.66620	2.94870
H	-0.58690	2.59630	2.53810
H	2.95010	0.26840	3.27620
H	4.24550	2.34330	3.63210
Pd	-2.73480	-1.64600	1.72450
S	-1.31230	-3.33300	0.90690
S	-3.87170	-1.89780	-0.46680
O	-4.73960	-3.08630	-0.79160
C	-1.95790	-4.92870	1.47920
C	-1.05060	-5.79770	2.08100
C	-1.52330	-7.00640	2.59630
C	-2.88020	-7.32260	2.51120
C	-3.77410	-6.43300	1.90790
C	-3.32250	-5.21990	1.38820
C	-1.64250	-3.41970	-0.93390
C	-2.30860	-2.14460	-1.43400
C	-4.54140	-0.42220	-1.25660
C	-5.60750	-0.59520	-2.13690
C	-6.14110	0.53230	-2.76120
C	-5.61120	1.79770	-2.49720
C	-4.55110	1.94630	-1.59980
C	-4.00850	0.83220	-0.95720
H	-0.00240	-5.52170	2.14200
H	-0.82920	-7.69670	3.06780
H	-3.24430	-8.26150	2.91920
H	-4.83060	-6.67750	1.84390
H	-4.02040	-4.53080	0.91730
H	-0.64700	-3.55450	-1.36710
H	-2.26050	-4.30110	-1.11940
H	-2.58250	-2.24210	-2.48950
H	-1.66800	-1.27220	-1.27330
H	-6.01040	-1.58890	-2.30630



H	-6.97480	0.42160	-3.44930
H	-6.03260	2.67220	-2.98600
H	-4.13910	2.92830	-1.38520
H	-3.19490	0.96390	-0.24800
O	-3.90130	-0.17710	2.54660
C	-5.18790	-0.28970	2.32400
C	-6.00200	0.80500	3.00610
H	-5.72230	1.78150	2.59530
H	-5.78560	0.82780	4.07930
H	-7.06720	0.63030	2.84360
H	-1.58520	0.67590	3.41920
H	0.97360	-0.89670	2.77130
H	-0.91630	-2.32490	3.57830
H	-2.39580	-1.45450	4.22810
C	-0.53130	1.11630	-0.39390
O	-1.25500	1.85400	0.28650
C	-0.21740	1.48530	-1.84540
H	-1.11570	1.33870	-2.45860
H	0.03980	2.54750	-1.89920
H	0.59730	0.88250	-2.25300

Pd_Bisulfoxide_Allylbenzene_Acetate_Sulfoxide_Abstraction_TS (D)

E_scf (B3LYP/LACVP*) = -2192.58810788547 a.u.

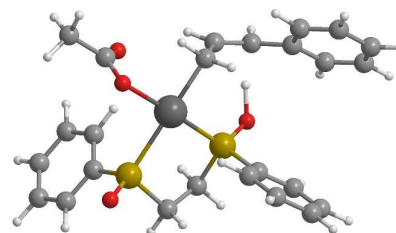
E_scf (M06/LACVP*) = -2191.64522399195 a.u.

E_Gibbs = -2192.194628 a.u.

E_Gibbs(318K) = -2192.201555 a.u.

E_solv = -2192.66925785363 a.u.

ZPE = 289.597 kcal/mol



C	-0.04780	0.07070	0.04430
C	-0.01580	0.12020	1.43910
C	1.07840	0.01350	-0.85270
H	0.90800	-0.52040	-1.78500
H	0.29390	1.52780	1.40450
O	0.70930	2.67520	1.53930
H	2.05690	-0.17990	-0.41570
H	-1.01190	0.20970	-0.44320
C	3.06510	-1.38910	4.04300
C	2.87200	-1.97280	2.79020
C	1.88850	-1.48480	1.93100
C	1.08260	-0.39720	2.31020
C	1.27520	0.16870	3.58300
C	2.25820	-0.32030	4.43970
H	3.82650	-1.77690	4.71400
H	3.47490	-2.82430	2.48690
H	1.71690	-1.98080	0.98040
H	0.64410	0.99410	3.90510
H	2.38900	0.12590	5.42180
H	-0.99300	0.17940	1.92580
Pd	1.12140	1.97730	-1.69970
S	1.57590	3.06950	0.29670
S	1.26760	4.27460	-2.77640
O	2.23530	4.61650	-3.86420
C	3.28260	3.11640	0.80490
C	3.61500	3.56670	2.08810
C	4.96080	3.59860	2.44480
C	5.94220	3.19180	1.53360
C	5.58950	2.73940	0.26070
C	4.24690	2.68890	-0.11470
C	1.11320	4.82840	-0.00730
C	1.73960	5.32190	-1.30130
C	-0.35850	4.96430	-3.13080
C	-0.41560	6.16640	-3.84420

C	-1.66450	6.72250	-4.11030
C	-2.82820	6.07870	-3.67450
C	-2.75140	4.86650	-2.98790
C	-1.50820	4.28790	-2.71700
H	2.84380	3.85670	2.79490
H	5.24470	3.93670	3.43660
H	6.98850	3.22080	1.82300
H	6.35370	2.41480	-0.43860
H	3.95650	2.32120	-1.09580
H	0.01880	4.81690	-0.04280
H	1.44620	5.40100	0.86350
H	2.83360	5.28670	-1.27470
H	1.42430	6.34840	-1.50790
H	0.49640	6.63560	-4.20250
H	-1.73150	7.65320	-4.66590
H	-3.79850	6.51670	-3.88980
H	-3.65700	4.35190	-2.68080
H	-1.45200	3.30640	-2.25220
O	0.66490	1.10500	-3.49570
O	-1.42020	1.07960	-2.65800
C	-0.61900	0.84540	-3.57130
C	-1.04160	0.21110	-4.88420
H	-0.60570	-0.79080	-4.96640
H	-0.66830	0.79760	-5.72890
H	-2.12960	0.13880	-4.92570

Pd_Bisulfoxide_Phenallyl_AcOH

E_scf (B3LYP/LACVP*) = -2192.64497188532 a.u.

E_scf (M06/LACVP*) = -2191.70998657237 a.u.

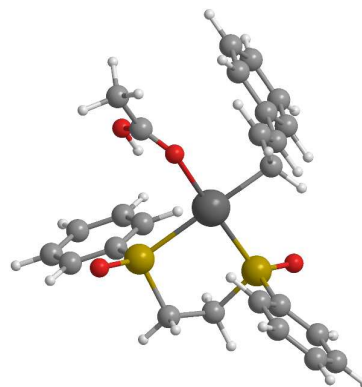
E_Gibbs = -2192.249059 a.u.

E_Gibbs(318K) = 2192.256157 a.u.

E_solv = -2192.71630543964 a.u.

ZPE = 292.843 kcal/mol

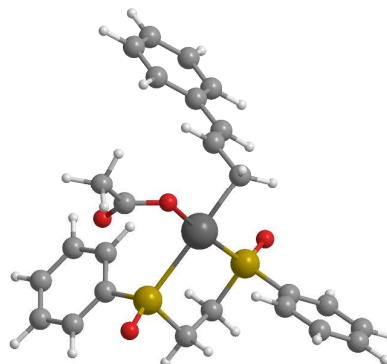
O	0.00000	0.00000	0.00000
H	0.00000	0.00000	0.97840
C	3.33970	0.00000	3.50520
C	3.11410	0.78840	2.40820
C	2.89280	2.21370	2.49580
C	4.00760	-3.41640	4.89190
O	1.77620	2.28570	5.77290
C	4.01110	-4.20920	3.74210
C	3.78520	-2.04670	4.79000
C	3.56080	-1.43590	3.53720
C	3.56850	-2.25400	2.38480
C	3.79120	-3.62120	2.49000
H	4.18870	-5.27820	3.81720
H	4.18270	-3.86520	5.86550
H	3.78850	-1.42920	5.68530
H	3.41070	-1.81740	1.40300
H	3.80000	-4.23630	1.59450
Pd	0.78230	2.19470	2.52900
S	0.79890	2.87300	4.79780
S	-1.74920	1.91930	2.77750
O	-2.79950	2.43570	1.83700
C	0.80680	4.67250	4.93850
C	1.22310	5.23280	6.14820
C	1.23420	6.62220	6.26780
C	0.83850	7.42450	5.19370
C	0.43990	6.84660	3.98610
C	0.43130	5.45820	3.84640
C	-0.90390	2.46380	5.41610
C	-1.99960	2.77570	4.40930



C	-2.15890	0.21710	3.24400
C	-3.47630	-0.20890	3.06500
C	-3.80680	-1.52300	3.39620
C	-2.82980	-2.39290	3.88940
C	-1.51140	-1.95920	4.04530
C	-1.16400	-0.64780	3.71310
H	1.55030	4.59340	6.96260
H	1.55950	7.07840	7.19790
H	0.85400	8.50570	5.29390
H	0.15000	7.47330	3.14810
H	0.15700	4.99860	2.89890
H	-0.85030	1.39770	5.65840
H	-1.04060	3.02750	6.34420
H	-2.03690	3.83940	4.15220
H	-2.98160	2.47210	4.78520
H	-4.21400	0.47510	2.65560
H	-4.82720	-1.87020	3.26300
H	-3.09490	-3.41530	4.14240
H	-0.75060	-2.63950	4.41650
H	-0.13370	-0.31180	3.82100
O	0.84910	2.06820	0.29650
C	0.46530	1.16430	-0.44870
C	0.46880	1.26720	-1.94390
H	0.90780	0.36690	-2.38360
H	1.01820	2.15550	-2.25470
H	-0.56680	1.33430	-2.29720
H	3.09310	0.34220	1.41600
H	3.36060	0.49990	4.47400
H	3.24150	2.69340	3.40960
H	3.08380	2.79900	1.59590

Pd_Bisulfoxide_Phenallyl_Acetate
 E_scf (B3LYP/LACVP*) = -2192.23937937424 a.u.
 E_scf (M06/LACVP*) = -2192.40522504855 a.u.
 E_Gibbs = -2191.854087 a.u.
 E_Gibbs(318K) = -2191.860639 a.u.
 E_solv = -2192.27515767617 a.u.
 ZPE = 285.096 kcal/mol

C	-0.24690	-0.37230	0.04290
O	1.47340	-2.62130	4.33380
C	0.32150	-1.55810	0.37890
C	-0.39150	-2.83150	0.39760
C	2.26530	2.45300	0.42050
O	-3.51750	-1.38110	1.26140
C	1.52110	3.53470	-0.06180
C	1.71210	1.17760	0.46540
C	0.39250	0.94360	0.02610
C	-0.34320	2.04650	-0.45130
C	0.21170	3.32340	-0.49670
H	1.95760	4.52950	-0.09470
H	3.28430	2.60700	0.76720
H	2.30710	0.35540	0.85200
H	-1.36460	1.88900	-0.79110
H	-0.38010	4.15480	-0.87200
Pd	-0.79480	-3.10040	2.43370
S	-3.03930	-2.50500	2.13670
S	-1.32850	-3.25140	4.97120
O	-1.04870	-4.37900	5.92550
C	-3.98630	-3.99220	1.71570
C	-5.22880	-3.82740	1.10170
C	-5.96580	-4.96180	0.76150
C	-5.45910	-6.23620	1.03040



C	-4.20700	-6.38510	1.63040
C	-3.45540	-5.25860	1.96890
C	-3.69320	-2.17200	3.83910
C	-3.19260	-3.16480	4.87450
C	-1.06750	-1.67030	5.81880
C	-1.19410	-1.64250	7.20720
C	-1.02440	-0.42960	7.87410
C	-0.72870	0.73080	7.15390
C	-0.58300	0.68130	5.76600
C	-0.73850	-0.52920	5.08760
H	-5.58980	-2.82820	0.87700
H	-6.93240	-4.85080	0.27810
H	-6.03580	-7.11600	0.75860
H	-3.80500	-7.37570	1.82290
H	-2.46120	-5.36040	2.39920
H	-3.36100	-1.15060	4.05240
H	-4.78470	-2.17770	3.75590
H	-3.49830	-4.18990	4.64120
H	-3.56040	-2.91090	5.87310
H	-1.39630	-2.56320	7.74740
H	-1.11590	-0.39130	8.95620
H	-0.59570	1.67330	7.67820
H	-0.32680	1.57880	5.21000
H	-0.57130	-0.59160	4.01610
O	1.13190	-3.87740	2.49180
C	1.85130	-3.43640	3.48500
C	3.25190	-4.03970	3.54330
H	3.88180	-3.45610	4.21810
H	3.70060	-4.08250	2.54630
H	3.18590	-5.06730	3.91970
H	1.36530	-1.59710	0.68450
H	-1.29780	-0.38160	-0.24670
H	-1.33170	-2.83390	-0.15760
H	0.22180	-3.71030	0.19980

Pd_Bisulfoxide_OH_Phenallyl_Acetate

E_scf (B3LYP/LACVP*) = -2192.59551642752 a.u.

E_scf (M06/LACVP*) = -2191.65696166141 a.u.

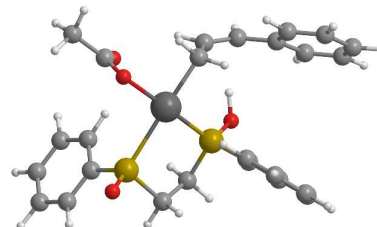
E_Gibbs = -2192.19613 a.u.

E_Gibbs(318K) = -2192.202887 a.u.

E_solv = -2192.68157277337 a.u.

ZPE = 292.323 kcal/mol

C	-0.01690	-0.01430	0.00890
C	-0.02060	-0.05670	1.37630
C	1.13770	0.00220	-0.89310
H	1.06890	-0.68180	-1.73770
H	0.23880	1.90360	1.09650
O	0.53070	2.87590	1.12580
H	2.12040	-0.02540	-0.42210
H	-0.98230	0.03390	-0.49180
C	3.18000	-0.61240	4.17660
C	3.23870	-1.24650	2.93390
C	2.21270	-1.07480	2.00670
C	1.10570	-0.25700	2.30600
C	1.04650	0.34830	3.57710
C	2.07640	0.18170	4.49820
H	3.97700	-0.75680	4.90070
H	4.07490	-1.89760	2.69450
H	2.24070	-1.62480	1.07150
H	0.18010	0.95120	3.84190
H	2.01220	0.65790	5.47280
H	-0.99900	0.00540	1.85610



Pd	1.05380	1.80780	-1.99270
S	1.43610	3.15030	-0.18770
S	1.05000	4.06770	-3.35850
O	2.01070	4.40110	-4.45830
C	3.09470	3.36560	0.41050
C	3.34220	4.07950	1.59080
C	4.66340	4.22740	2.00150
C	5.70450	3.67880	1.24230
C	5.43650	2.96730	0.07150
C	4.11930	2.79510	-0.35400
C	0.83900	4.83370	-0.63450
C	1.44880	5.25570	-1.96270
C	-0.60440	4.65950	-3.76720
C	-0.71270	5.80490	-4.56340
C	-1.98410	6.27750	-4.87930
C	-3.11970	5.60750	-4.40960
C	-2.99100	4.45140	-3.63960
C	-1.72380	3.95520	-3.31820
H	2.52820	4.48170	2.18590
H	4.88280	4.76850	2.91660
H	6.73150	3.80300	1.57240
H	6.24700	2.53740	-0.50850
H	3.89110	2.23230	-1.25560
H	-0.25020	4.73020	-0.68260
H	1.11050	5.50340	0.18710
H	2.54290	5.28580	-1.92880
H	1.07900	6.24520	-2.24540
H	0.17960	6.29190	-4.94670
H	-2.09070	7.16200	-5.50030
H	-4.10780	5.97980	-4.66410
H	-3.87380	3.91420	-3.30570
H	-1.62360	3.01100	-2.78660
O	0.65050	0.63750	-3.64410
O	-1.42730	0.81350	-2.81490
C	-0.63530	0.39740	-3.67390
C	-1.08770	-0.45980	-4.84170
H	-0.72720	-1.48540	-4.70200
H	-0.66100	-0.08960	-5.77830
H	-2.17740	-0.46630	-4.89760

Pd_Bisulfoxide_Phenallyl

E_scf (B3LYP/LACVP*) = -1963.56308152005 a.u.

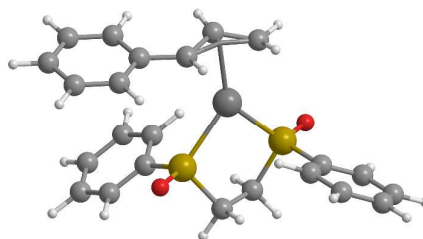
E_scf (M06/LACVP*) = -1962.74654372455 a.u.

E_Gibbs = -1963.219033 a.u.

E_Gibbs(318K) = -1963.224925 a.u.

E_solv = -1963.63786665601 a.u.

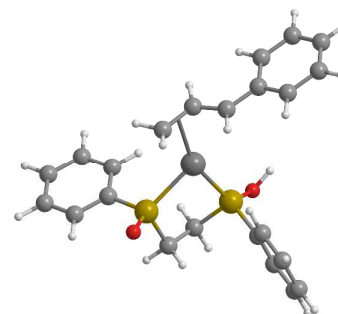
ZPE = 253.174 kcal/mol



C	0.01570	0.35520	-
0.01560			
S	2.53740	0.00500	-3.86360
S	-0.76280	0.63220	-3.55750
O	3.85910	0.71130	-3.90050
O	-2.09640	-0.04800	-3.54600
C	2.73800	-1.70180	-4.41170
C	3.84880	-2.00810	-5.19990
C	4.01190	-3.32120	-5.64130
C	3.07950	-4.30270	-5.29180
C	1.98290	-3.98270	-4.48790
C	1.80990	-2.67480	-4.03270
C	1.50150	0.68240	-5.25280
C	0.05180	0.22950	-5.18190
C	-0.98720	2.41510	-3.72690

C	-2.21650	2.87070	-4.20600
C	-2.40900	4.24530	-4.34730
C	-1.38860	5.13810	-4.00950
C	-0.16940	4.66620	-3.51600
C	0.03620	3.29460	-3.36180
H	4.57340	-1.23620	-5.44220
H	4.87020	-3.57900	-6.25480
H	3.21380	-5.32280	-5.63940
H	1.26700	-4.74960	-4.20760
H	0.97340	-2.42420	-3.38360
H	1.60240	1.76870	-5.16260
H	1.97300	0.36460	-6.18770
H	-0.04990	-0.85690	-5.26970
H	-0.55340	0.70170	-5.96180
H	-3.00350	2.16100	-4.44260
H	-3.35960	4.61880	-4.71620
H	-1.54740	6.20680	-4.12010
H	0.61700	5.36310	-3.24150
H	0.97380	2.92360	-2.95190
C	1.12190	-0.49670	0.18320
C	2.42430	-0.03210	-0.10370
H	0.12060	1.43190	0.11670
H	2.57270	1.04910	-0.08960
H	0.94830	-1.55670	0.34740
Pd	0.99970	0.15090	-1.95710
H	-0.99280	-0.03630	0.07930
C	6.07940	-2.22280	-0.23710
C	4.88370	-2.90710	0.01230
C	3.67840	-2.21780	0.05230
C	3.64910	-0.82120	-0.14860
C	4.86160	-0.14670	-0.39870
C	6.06520	-0.84180	-0.44400
H	7.01900	-2.76730	-0.26730
H	4.89690	-3.98020	0.17920
H	2.76280	-2.76640	0.25310
H	4.85110	0.92780	-0.56310
H	6.99070	-0.30820	-0.63820

Pd_Bisulfoxide_OH_Phenallyl
 E_scf (B3LYP/LACVP*) = -1963.81591428391 a.u.
 E_scf (M06/LACVP*) = -1962.99107432339 a.u.
 E_Gibbs = -1963.463286 a.u.
 E_Gibbs(318K) = -1963.469386 a.u.
 E_solv = -1964.04817028426 a.u.
 ZPE = 259.856 kcal/mol



S	13.53570	-0.16530	0.35560
S	16.81510	-0.01820	-0.30060
O	12.79700	-0.82060	1.47540
O	17.88590	0.26160	-1.53620
C	12.39370	0.38860	-0.91840
C	11.13140	-0.20960	-0.95650
C	10.23440	0.20340	-1.94230
C	10.60050	1.19270	-2.85960
C	11.86270	1.79000	-2.79530
C	12.77010	1.39700	-1.81070
C	14.47650	-1.50270	-0.54920
C	15.62970	-0.95350	-1.37200
C	17.56430	-1.25200	0.72690
C	18.23590	-2.34710	0.15820
C	18.78120	-3.29810	1.01410
C	18.64700	-3.15910	2.40230
C	17.98160	-2.05950	2.95100

C	17.44310	-1.08350	2.11470
H	10.85560	-0.96050	-0.22210
H	9.24610	-0.24360	-1.98780
H	9.89380	1.51060	-3.62020
H	12.13380	2.57070	-3.49960
H	13.74040	1.88450	-1.73020
H	14.80810	-2.18710	0.23830
H	13.77180	-2.02960	-1.20000
H	15.30970	-0.23480	-2.13290
H	16.20950	-1.74970	-1.84900
H	18.35460	-2.44310	-0.91690
H	19.31110	-4.15030	0.60020
H	19.07040	-3.91300	3.05920
H	17.88640	-1.95880	4.02750
H	16.92720	-0.22050	2.52790
C	15.15790	3.67010	1.70980
C	16.21100	3.56590	2.63950
C	13.97420	2.90290	1.85470
H	13.10450	3.14490	1.24880
H	16.04270	2.88240	3.47290
H	13.75670	2.40650	2.80090
H	15.21720	4.39780	0.90470
Pd	15.24440	1.52210	0.84960
C	19.76420	5.82410	3.00200
C	18.93130	6.04660	1.89230
C	17.77330	5.30640	1.73510
C	17.42150	4.31820	2.69580
C	18.28230	4.10610	3.80750
C	19.43930	4.85400	3.95820
H	20.66700	6.41650	3.12170
H	19.19340	6.80840	1.16480
H	17.12970	5.49710	0.88170
H	18.01390	3.36300	4.55460
H	20.08460	4.69610	4.81650
H	18.51420	0.96750	-1.27610

Pd_Bisulfoxide_Liniar_Product_TS

E_scf (B3LYP/LACVP*) = -2435.98718418631 a.u.

E_scf (M06/LACVP*) = -2434.93766112862 a.u.

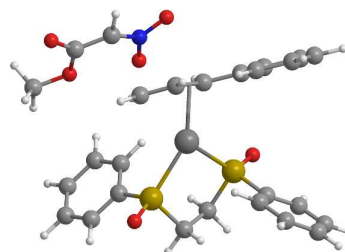
E_Gibbs = -2435.574805 a.u.

E_Gibbs(318K) = -2435.58222 a.u.

E_solv = -2436.03861411663 a.u.

ZPE = 304.374 kcal/mol

C	0.04440	0.07590	0.11060
C	-0.14590	-0.06370	2.69080
S	3.89750	0.02960	-2.79750
S	1.26270	2.22300	-2.88740
O	5.19330	-0.30950	-2.11270
O	0.12480	2.67920	-3.76250
C	3.89650	-0.75330	-4.43390
C	5.12470	-1.12230	-4.98370
C	5.14270	-1.73840	-6.23470
C	3.94630	-1.98530	-6.91380
C	2.72340	-1.62930	-6.34120
C	2.69120	-1.01740	-5.08670
C	4.04470	1.81110	-3.31340
C	2.78420	2.36670	-3.95410
C	1.63210	3.53530	-1.68660
C	1.23010	4.83410	-2.00520
C	1.48760	5.85990	-1.09660
C	2.12680	5.58250	0.11640
C	2.50090	4.27470	0.43190



C	2.24890	3.23740	-0.46950
H	6.03780	-0.94590	-4.42250
H	6.09050	-2.03380	-6.67640
H	3.96670	-2.46960	-7.88620
H	1.79340	-1.83880	-6.86210
H	1.74250	-0.76620	-4.61640
H	4.29880	2.33180	-2.38420
H	4.90070	1.87210	-3.99340
H	2.51500	1.82750	-4.86820
H	2.89490	3.42820	-4.19720
H	0.70480	5.02030	-2.93770
H	1.17940	6.87520	-1.33080
H	2.31880	6.38570	0.82250
H	2.96770	4.04560	1.38500
H	2.51100	2.21650	-0.20250
C	0.47990	-1.19870	-0.27900
C	1.84350	-1.57080	-0.19330
H	0.71130	0.79230	0.57490
H	2.47750	-0.96860	0.45610
H	-0.24790	-1.88450	-0.70560
Pd	1.65890	0.04350	-1.76360
H	-0.97500	0.38590	-0.09400
C	3.60500	-5.30330	-1.31800
C	2.25820	-5.06900	-1.61160
C	1.66230	-3.86310	-1.25710
C	2.40300	-2.86470	-0.59540
C	3.75770	-3.11560	-0.30590
C	4.35220	-4.32180	-0.66450
H	4.06690	-6.24720	-1.59620
H	1.67070	-5.83190	-2.11600
H	0.61410	-3.70210	-1.49350
H	4.33970	-2.35110	0.20080
H	5.39900	-4.49620	-0.43120
C	-0.77160	1.23310	2.86220
O	0.01370	2.31840	2.61000
O	-1.95360	1.34330	3.16580
H	-0.74820	-0.91890	2.95700
N	1.20320	-0.35500	2.59240
O	2.04690	0.53210	2.28280
O	1.53540	-1.56320	2.69840
C	-0.64980	3.57420	2.78410
H	0.08580	4.33190	2.50850
H	-1.53240	3.64780	2.14160
H	-0.96620	3.70920	3.82300

Pd_Bisulfoxide_Liniar_Product

E_scf (B3LYP/LACVP*) = -2436.04886058496 a.u.

E_scf (M06/LACVP*) = -2435.00696745586 a.u.

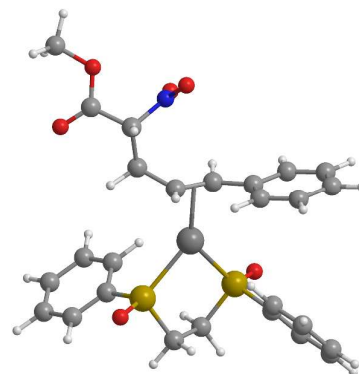
E_Gibbs = -2435.632045 a.u.

E_Gibbs(318K) = -2435.639506 a.u.

E_solv = -2436.08617857517 a.u.

ZPE = 305.822 kcal/mol

C	0.00000	0.00000	0.00000
C	0.00000	0.00000	1.54620
S	2.66530	0.00000	-4.86600
S	-0.38880	-1.29160	-4.25190
O	3.33880	1.31590	-5.22400
O	-1.17470	-2.58830	-4.12890
C	3.80790	-1.33380	-5.30310
C	4.64630	-1.17320	-6.41140
C	5.55920	-2.18270	-6.71740
C	5.63320	-3.33090	-5.91920



C	4.79840	-3.47220	-4.80780
C	3.88280	-2.46500	-4.48840
C	1.37880	-0.23950	-6.18300
C	0.45150	-1.40860	-5.90330
C	-1.58120	0.03040	-4.59120
C	-2.81110	-0.30700	-5.16140
C	-3.74700	0.70330	-5.38860
C	-3.45060	2.02740	-5.04480
C	-2.21950	2.34850	-4.46490
C	-1.27810	1.34400	-4.22410
H	4.59950	-0.26880	-7.01230
H	6.21800	-2.07070	-7.57510
H	6.35210	-4.11060	-6.15910
H	4.86730	-4.35680	-4.18000
H	3.24730	-2.55030	-3.61010
H	0.84120	0.71470	-6.21040
H	1.89880	-0.37090	-7.13740
H	0.98790	-2.36320	-5.86190
H	-0.33080	-1.47940	-6.66600
H	-3.03570	-1.34200	-5.40490
H	-4.71010	0.45490	-5.82790
H	-4.18550	2.80890	-5.22000
H	-1.99520	3.37510	-4.18640
H	-0.32940	1.57770	-3.74330
C	1.17770	-0.75910	-0.58060
C	2.42100	-0.14640	-0.82430
H	-0.01970	1.03880	-0.34310
H	2.46090	0.94060	-0.73910
H	1.15790	-1.82890	-0.37030
Pd	1.41770	-0.52580	-2.74250
H	-0.94720	-0.46080	-0.29300
C	6.35300	-1.89650	-0.58830
C	5.22550	-2.71630	-0.47610
C	3.94330	-2.17430	-0.55870
C	3.75430	-0.79380	-0.76170
C	4.90000	0.01770	-0.86890
C	6.18240	-0.52370	-0.78410
H	7.35170	-2.32180	-0.51760
H	5.34520	-3.78540	-0.31290
H	3.08390	-2.83300	-0.45710
H	4.77670	1.08980	-1.01520
H	7.04940	0.12880	-0.86550
C	-1.26320	0.64200	2.13390
O	-1.03230	1.24800	3.29490
O	-2.34790	0.54820	1.59800
H	0.08190	-1.01460	1.94690
N	1.22330	0.72700	2.05630
O	1.38810	1.88400	1.67260
O	1.98670	0.11770	2.79670
C	-2.17640	1.85510	3.95280
H	-1.78110	2.28060	4.87480
H	-2.60010	2.63430	3.31570
H	-2.92960	1.09390	4.16810

Linier_Product

E_scf (B3LYP/LACVP*) = -820.6294486975 a.u.

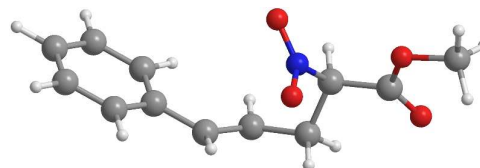
E_scf (M06/LACVP*) = -820.10962683226 a.u.

E_Gibbs = -820.424305 a.u.

E_Gibbs(318K) = -820.428596 a.u.

E_solv = -820.63883802547 a.u.

ZPE = 148.447 kcal/mol



C	0.00000	0.00000	0.00000
---	---------	---------	---------

C	0.00000	0.00000	1.54620
C	1.17770	-0.75910	-0.58060
C	2.42100	-0.14640	-0.82430
H	-0.01970	1.03880	-0.34310
H	2.46090	0.94060	-0.73910
H	1.15790	-1.82890	-0.37030
H	-0.94720	-0.46080	-0.29300
C	6.35300	-1.89650	-0.58830
C	5.22550	-2.71630	-0.47610
C	3.94330	-2.17430	-0.55870
C	3.75430	-0.79380	-0.76170
C	4.90000	0.01770	-0.86890
C	6.18240	-0.52370	-0.78410
H	7.35170	-2.32180	-0.51760
H	5.34520	-3.78540	-0.31290
H	3.08390	-2.83300	-0.45710
H	4.77670	1.08980	-1.01520
H	7.04940	0.12880	-0.86550
C	-1.26320	0.64200	2.13390
O	-1.03230	1.24800	3.29490
O	-2.34790	0.54820	1.59800
H	0.08190	-1.01460	1.94690
N	1.22330	0.72700	2.05630
O	1.38810	1.88400	1.67260
O	1.98670	0.11770	2.79670
C	-2.17640	1.85510	3.95280
H	-1.78110	2.28060	4.87480
H	-2.60010	2.63430	3.31570
H	-2.92960	1.09390	4.16810

Pd_Bisulfoxide_Quinone

E_scf (B3LYP/LACVP*) = -1996.87729924803 a.u.

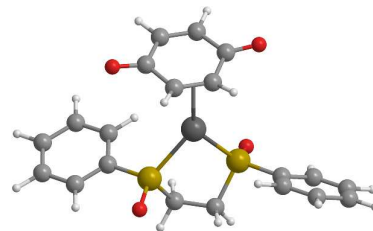
E_scf (M06/LACVP*) = -1996.10776602681 a.u.

E_Gibbs = -1996.600239 a.u.

E_Gibbs(318K) = -1996.606031 a.u.

E_solv = -1996.91290368969 a.u.

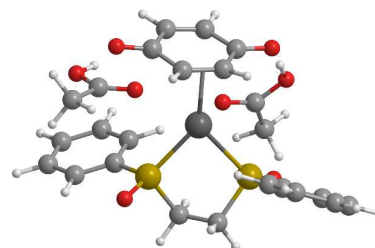
ZPE = 210.888 kcal/mol



O	0.04670	-0.17830	-0.11090
S	0.04780	-0.08450	1.39370
Pd	2.16850	-0.06260	2.64440
C	0.42400	2.63770	1.70230
C	-0.66200	1.57690	1.82490
S	1.71920	2.33140	2.99740
H	0.03040	3.64020	1.89960
H	-1.47590	1.74710	1.11260
H	-1.06530	1.53130	2.84140
H	0.90350	2.61710	0.71830
C	3.28340	-1.92780	2.75490
C	4.06880	-0.89600	3.31400
C	4.15510	-0.70980	4.78370
C	3.28250	-1.58170	5.62170
H	3.32390	-1.39440	6.69190
C	2.52970	-2.56080	5.09130
H	1.92150	-3.22180	5.70420
C	2.50650	-2.84520	3.62740
H	4.86980	-0.42990	2.74500
H	3.45870	-2.28780	1.74400
O	4.91940	0.09610	5.30590
O	1.88540	-3.81000	3.19080
C	4.92260	5.27990	1.57870
C	4.76410	4.06230	0.90970
C	3.79000	3.15600	1.32830

C	2.97430	3.49780	2.40810
C	3.12640	4.69930	3.09540
C	4.10960	5.59480	2.66960
H	5.69000	5.97760	1.25570
H	5.40660	3.81220	0.07010
H	3.67440	2.19470	0.83180
H	2.49180	4.90740	3.95170
H	4.24410	6.53580	3.19580
C	-3.29230	-2.78260	2.97740
C	-2.42940	-2.15650	3.88180
C	-1.40960	-1.32820	3.41550
C	-1.28540	-1.12670	2.03980
C	-2.12620	-1.75250	1.12340
C	-3.13890	-2.58500	1.60410
H	-4.07850	-3.43570	3.34600
H	-2.54240	-2.32300	4.94930
H	-0.71720	-0.85680	4.10990
H	-1.97090	-1.59090	0.06100
H	-3.80330	-3.08400	0.90420
O	1.15340	2.84660	4.29630

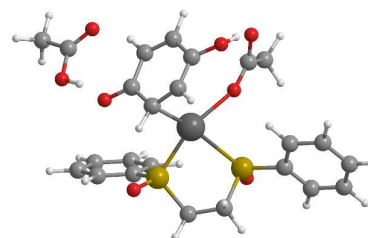
Pd_Bisulfoxide_Quinone_2AcOH (Poli_A)
 E_scf (B3LYP/LACVP*) = -2455.06242530146 a.u.
 E_scf (M06/LACVP*) = -2454.06065277806 a.u.
 E_Gibbs = -2454.671081 a.u.
 E_Gibbs(318K) = -2454.679407 a.u.
 E_solv = -2455.09599236341 a.u.
 ZPE = 290.546 kcal/mol



O	0.00000	0.00000	0.00000
S	0.00000	0.00000	1.50690
Pd	2.04260	0.00000	2.85990
C	0.22890	2.70310	2.03590
C	-0.80480	1.58590	2.04690
S	1.47640	2.38420	3.37480
O	0.90460	2.96830	4.64450
H	-0.21570	3.67460	2.27520
H	-1.60810	1.76820	1.32570
H	-1.23050	1.43640	3.04390
H	0.75060	2.76430	1.07580
C	3.52930	-1.61600	2.67930
C	3.93480	-0.68800	3.66380
C	3.59370	-0.89830	5.08180
C	2.84110	-2.13240	5.42710
H	2.59870	-2.26220	6.47760
C	2.50050	-3.04120	4.49850
H	1.97000	-3.95630	4.74750
C	2.83820	-2.85700	3.06470
H	4.70270	0.05380	3.46060
H	3.95210	-1.60780	1.67660
O	3.91340	-0.09110	5.96690
O	2.55430	-3.75280	2.25620
H	2.56850	0.44040	6.99030
O	1.81150	0.89350	7.44730
C	0.67980	0.68310	6.77390
O	0.59230	-0.06540	5.80870
C	-0.47060	1.48660	7.32740
H	-0.38720	1.60410	8.41080
H	-0.43150	2.47930	6.86520
H	-1.41820	1.01210	7.06440
O	4.42890	-2.18820	-0.38860
C	3.89760	-3.10880	-0.98760

O	3.05730	-3.98510	-0.43210
H	2.93450	-3.78190	0.53480
C	4.11240	-3.39500	-2.45780
H	4.46260	-4.42370	-2.59330
H	3.16250	-3.30120	-2.99550
H	4.84020	-2.69410	-2.86870
C	4.86800	5.14510	2.01880
C	4.61800	3.96640	1.30940
C	3.58240	3.11990	1.70320
C	2.79920	3.48290	2.80130
C	3.04020	4.64680	3.52600
C	4.08460	5.48220	3.12390
H	5.68300	5.79570	1.71400
H	5.23550	3.69920	0.45650
H	3.39680	2.18790	1.17310
H	2.42400	4.87330	4.39070
H	4.28920	6.39310	3.67960
C	-3.17650	-2.97810	2.90830
C	-2.39290	-2.31560	3.85780
C	-1.42010	-1.39970	3.45630
C	-1.26700	-1.15830	2.08840
C	-2.02810	-1.81480	1.12530
C	-2.99330	-2.73130	1.54680
H	-3.92630	-3.69530	3.23210
H	-2.53190	-2.51660	4.91670
H	-0.79870	-0.89640	4.19600
H	-1.85180	-1.60880	0.07370
H	-3.59660	-3.25520	0.81040

Pd_Bisulfoxide_Quinone_2AcOH (Poli_A-B)
 E_scf (B3LYP/LACVP*) = -2455.03418236377 a.u.
 E_scf (M06/LACVP*) = -2454.0309695028 a.u.
 E_Gibbs = -2454.646965 a.u.
 E_Gibbs(318K) = -2454.6544 a.u.
 E_solv = -2455.06910045294 a.u.
 ZPE = 288.711 kcal/mol

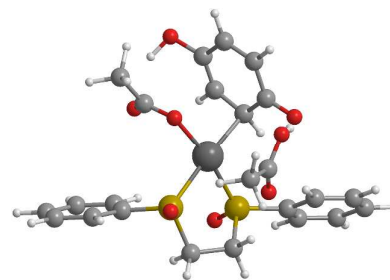


O	-0.52040	-0.18640	2.85750
H	-0.11310	-0.47850	4.14160
O	6.85970	-1.45570	3.65200
S	5.97850	-1.01360	2.51800
Pd	3.69280	-0.67780	2.92370
C	5.87860	1.78700	2.25800
C	6.68190	0.56880	1.82070
S	4.17330	1.61830	1.55260
O	4.36120	1.70060	0.05170
H	6.32150	2.70270	1.85240
H	7.70590	0.58540	2.20650
H	6.69760	0.48590	0.73160
H	5.79330	1.87010	3.34660
C	3.69270	-2.14180	4.46710
C	2.67200	-1.44960	5.16320
O	1.65900	0.08910	2.31720
C	0.01480	0.35500	0.60340
C	1.34910	-1.58630	4.72490
H	-1.06740	0.46930	0.51700
C	1.00080	-2.68070	3.81460
H	0.52080	1.25750	0.24700
H	-0.05210	-2.79710	3.57540
H	0.34200	-0.47350	-0.03600
C	1.93010	-3.50770	3.28680
H	4.05430	-5.42940	2.12050
H	1.67850	-4.34190	2.63640

O	4.15930	-6.20860	1.50480
C	3.34900	-3.31970	3.61630
C	2.97060	-6.59150	1.02970
H	2.91030	-0.68670	5.89720
O	1.91220	-6.05370	1.30740
H	4.70970	-2.14530	4.85600
C	3.08820	-7.78340	0.10420
O	0.42710	-0.72980	5.06800
H	3.78140	-7.56020	-0.71360
O	4.23330	-4.09840	3.21850
H	3.49520	-8.64110	0.65100
H	2.10580	-8.03520	-0.29740
C	0.42280	0.06390	2.04100
C	2.28370	5.50440	3.04900
C	3.10900	5.52500	1.92040
C	3.69420	4.34630	1.46090
C	3.44720	3.16010	2.15450
C	2.60020	3.11480	3.26370
C	2.02640	4.30490	3.71540
H	1.82580	6.42550	3.39920
H	3.28970	6.45840	1.39440
H	4.31150	4.33190	0.56700
H	2.36590	2.16640	3.73810
H	1.36230	4.28800	4.57490
C	6.36990	-3.99270	-0.91780
C	6.96810	-4.25630	0.31420
C	6.88170	-3.32400	1.35050
C	6.19300	-2.13600	1.12100
C	5.57050	-1.85840	-0.10110
C	5.67550	-2.79640	-1.12580
H	6.43350	-4.72530	-1.71770
H	7.48990	-5.19420	0.47880
H	7.32120	-3.51100	2.32450
H	5.01170	-0.93730	-0.25390
H	5.20410	-2.59660	-2.08390

Pd_Bisulfoxide_Quinone_2AcOH (Poli_B)
 E_scf (B3LYP/LACVP*) = -2455.04534352613 a.u.
 E_scf (M06/LACVP*) = -2454.0456716079 a.u.
 E_Gibbs = -2454.653343 a.u.
 E_Gibbs(318K) = -2454.66087 a.u.
 E_solv = -2455.0798515766 a.u.
 ZPE = 291.586 kcal/mol

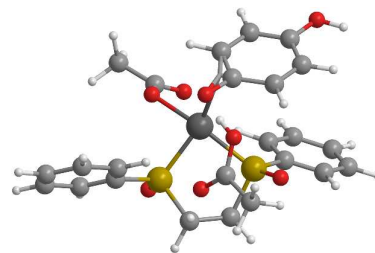
O	1.36190	1.12940	2.97480
S	1.63910	0.13520	1.87670
Pd	-0.01050	0.17820	0.19220
C	2.83760	2.06990	0.36520
C	3.14070	0.70350	0.96600
S	1.40410	1.96780	-0.82430
O	1.94100	1.88640	-2.22530
H	3.67470	2.43990	-0.23480
H	3.94760	0.76770	1.70290
H	3.35730	-0.05180	0.20460
H	2.58170	2.79840	1.13850
C	-1.15070	-1.46920	1.13550
C	-2.42710	-0.83880	1.32830
C	-3.42730	-1.02150	0.41330
C	-3.28530	-2.00740	-0.62110
H	-4.13400	-2.15490	-1.28480
C	-2.12980	-2.71030	-0.79460
H	-2.03050	-3.45610	-1.57760
C	-0.99370	-2.50410	0.08200



H	-2.55340	-0.12770	2.13850
H	-0.54130	-1.65060	2.01830
O	-4.55290	-0.24590	0.38420
O	0.06870	-3.17090	-0.02650
H	-4.24570	0.67390	0.53210
O	-2.64200	1.84260	-0.03870
C	-2.33350	1.35540	-1.13800
O	-1.33350	0.55650	-1.36490
C	-3.19220	1.63120	-2.36920
H	-3.49680	2.68180	-2.37460
H	-2.67230	1.38060	-3.29580
H	-4.10020	1.02050	-2.29410
H	0.80800	-3.02740	-1.51860
O	1.15080	-2.77430	-2.43000
C	2.07640	-1.83090	-2.32560
O	2.52070	-1.42940	-1.25200
C	2.52590	-1.27650	-3.65550
H	2.10050	-1.83570	-4.49050
H	2.20960	-0.22880	-3.70500
H	3.61880	-1.29200	-3.70670
C	2.98680	-3.75820	3.84240
C	2.92320	-3.68480	2.44860
C	2.52310	-2.50670	1.81760
C	2.20830	-1.40780	2.62030
C	2.25560	-1.45790	4.01420
C	2.65290	-2.64890	4.62330
H	3.28750	-4.68600	4.32170
H	3.16720	-4.55350	1.84380
H	2.43700	-2.44770	0.73770
H	1.98090	-0.58270	4.59460
H	2.69380	-2.71020	5.70730
C	-0.50630	6.07730	-0.26250
C	-0.58070	5.13660	0.76900
C	0.00470	3.88120	0.61600
C	0.67570	3.60240	-0.57650
C	0.74980	4.51870	-1.62210
C	0.15310	5.76970	-1.45410
H	-0.97510	7.04990	-0.13960
H	-1.11070	5.37180	1.68730
H	-0.06910	3.13410	1.40220
H	1.25710	4.24320	-2.54150
H	0.19950	6.50040	-2.25680

Pd_Bisulfoxide_Quinone_2AcOH (Poli_B-C)
 E_scf (B3LYP/LACVP*) = -2455.01951872337 a.u.
 E_scf (M06/LACVP*) = -2454.02215163565 a.u.
 E_Gibbs = -2454.627321 a.u.
 E_Gibbs(318K) = -2454.634625 a.u.
 E_solv = -2455.06233453529 a.u.
 ZPE = 290.555 kcal/mol

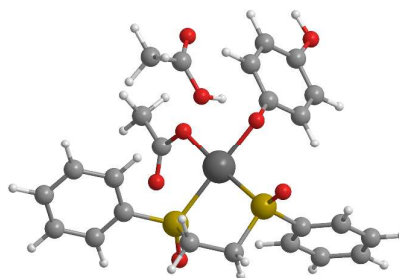
Pd	0.19070	-0.07250	0.06780
S	0.16360	0.05340	2.39350
O	1.30470	0.09920	3.37220
S	-0.82870	1.98250	-0.02710
O	-2.31010	2.05840	-0.22690
C	-0.82820	1.62000	2.64750
C	-0.42490	2.67620	1.62960
H	-0.58700	1.92640	3.66970
H	-1.88880	1.36390	2.57190
H	-1.02200	3.58770	1.73570
H	0.64870	2.89380	1.65780
O	-0.36860	-0.17920	-1.90090



C	-1.48810	-0.84260	-2.07530
O	-2.12590	-1.40480	-1.18520
C	1.32980	-2.32270	-0.74820
C	2.17450	-1.53850	0.12380
C	0.84470	-3.58030	-0.37290
C	2.47500	-2.12900	1.41310
C	1.17100	-4.10890	0.86810
H	0.21840	-4.15520	-1.04720
C	1.99440	-3.36990	1.75500
H	1.16890	-1.94790	-1.75180
H	3.50190	0.29940	0.70100
O	4.12720	0.69900	1.45200
C	3.63870	1.80880	1.95850
O	2.67520	2.42180	1.49640
C	4.36930	2.26830	3.20180
H	4.00380	1.67830	4.05080
H	5.44470	2.09290	3.10980
H	4.16750	3.32430	3.39090
H	3.10460	-1.56960	2.09560
H	2.24280	-3.79430	2.72750
C	-1.94070	-0.86940	-3.53120
H	-2.86510	-1.44370	-3.62110
H	-2.10340	0.15240	-3.89010
H	-1.16350	-1.31570	-4.16110
O	0.68280	-5.33900	1.19860
H	0.98770	-5.57550	2.08860
O	2.64740	-0.39060	-0.24950
C	1.13450	4.67670	-3.12750
C	-0.25700	4.59570	-3.04180
C	-0.85150	3.78580	-2.07450
C	-0.02300	3.07850	-1.20380
C	1.37010	3.13190	-1.27540
C	1.94220	3.95160	-2.24750
H	1.59270	5.30210	-3.88930
H	-0.88150	5.15590	-3.73220
H	-1.92920	3.68800	-1.99270
H	1.98710	2.56520	-0.58330
H	3.02420	4.01310	-2.32110
C	-2.89120	-2.96070	3.97690
C	-1.87980	-2.47810	4.81160
C	-0.94220	-1.56970	4.32030
C	-1.04680	-1.15420	2.99260
C	-2.03590	-1.63880	2.13260
C	-2.96460	-2.54730	2.64490
H	-3.61850	-3.66980	4.36370
H	-1.81950	-2.80930	5.84490
H	-0.13670	-1.18920	4.94120
H	-2.07530	-1.34610	1.08290
H	-3.73890	-2.93750	1.99070

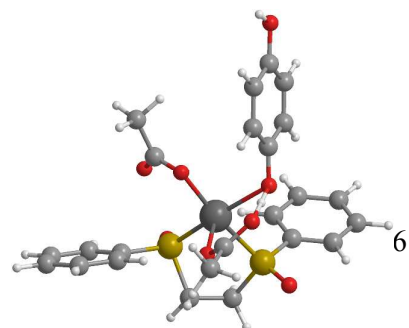
Pd_Bisulfoxide_Quinone_2AcOH (Poli_C)
 E_scf (B3LYP/LACVP*) = -2455.04725149665 a.u.
 E_scf (M06/LACVP*) = -2454.03291604582 a.u.
 E_Gibbs = -2454.657418 a.u.
 E_Gibbs(318K) = -2454.665029 a.u.
 E_solv = -2455.09371181012 a.u.
 ZPE = 291.316 kcal/mol

Pd	-0.53930	0.09530	
0.16470			
S	-1.71020	0.65770	2.17330
O	-1.69540	2.02170	2.79020
S	-2.58850	-0.90200	-0.56140



O	-3.06490	-2.26740	-0.16220
C	-3.47460	0.17420	1.79920
C	-3.74040	0.27250	0.30400
H	-4.07740	0.88150	2.37660
H	-3.64170	-0.84230	2.16380
H	-4.75940	-0.04640	0.06320
H	-3.56380	1.27890	-0.08900
O	0.57520	-0.32560	-1.49970
C	0.34600	-1.44200	-2.13640
O	-0.59710	-2.20970	-1.92470
C	1.38690	-1.73330	-3.21230
H	1.42520	-0.90760	-3.93070
H	2.37740	-1.81010	-2.75170
H	1.14100	-2.66440	-3.72660
H	2.94570	1.46110	-0.96950
O	1.05900	1.16240	0.84830
O	6.01680	-1.25990	1.06870
H	6.55920	-1.00150	0.30710
H	4.14110	-1.55820	2.79910
C	2.25830	0.55070	0.86010
H	0.66890	2.45320	-0.31810
C	3.20570	0.79180	-0.15610
O	0.14080	3.08270	-0.88570
C	2.61100	-0.30440	1.92700
C	0.88660	3.53880	-1.90480
C	4.46420	0.20350	-0.09330
O	2.04380	3.22470	-2.10280
C	3.86250	-0.90060	1.98090
C	0.11240	4.50650	-2.77570
H	1.88240	-0.48530	2.71270
H	-0.78810	4.02510	-3.17300
C	4.79650	-0.64890	0.96670
H	-0.21440	5.36650	-2.18110
H	5.19600	0.40580	-0.87490
H	0.74430	4.84390	-3.59840
C	-0.48960	-2.38800	5.35890
C	-0.52960	-1.02600	5.66320
C	-0.92410	-0.10030	4.69420
C	-1.27930	-0.56850	3.43170
C	-1.23470	-1.92670	3.10040
C	-0.84170	-2.83630	4.08150
H	-0.17650	-3.10290	6.11450
H	-0.24850	-0.67920	6.65350
H	-0.95220	0.96570	4.89780
H	-1.49800	-2.27320	2.10320
H	-0.80320	-3.89520	3.84260
C	-3.73040	0.00980	-4.88360
C	-2.95460	0.92690	-4.16950
C	-2.59050	0.66090	-2.84940
C	-3.03050	-0.52960	-2.26870
C	-3.79250	-1.46340	-2.96420
C	-4.14520	-1.18150	-4.28460
H	-4.00250	0.22050	-5.91430
H	-2.62030	1.84520	-4.64420
H	-1.95720	1.35130	-2.29690
H	-4.07680	-2.39210	-2.47970
H	-4.73560	-1.89860	-4.84810

Pd_Bisulfoxide_Quinone_2AcOH (Poli_C-D)
 E_scf (B3LYP/LACVP*) = -2455.03646923901 a.u.
 E_scf (M06/LACVP*) = -2454.02291978903 a.u.
 E_Gibbs = -2454.649746 a.u.
 E_Gibbs(318K) = -2454.657231 a.u.

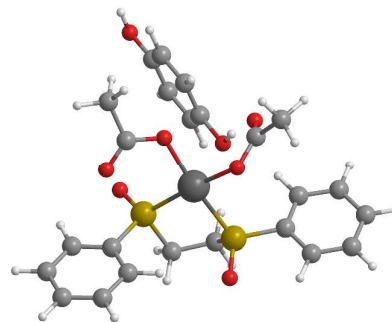


E_solv = -2455.07938681832 a.u.
 ZPE = 288.577 kcal/mol

Pd	0.58780	-0.05140	-0.01110
S	1.89480	1.20610	-1.50550
O	2.12190	2.67260	-1.32870
S	2.32050	-1.66930	-0.05640
O	2.56300	-2.67490	-1.14340
C	3.53770	0.31890	-1.50390
C	3.70900	-0.44240	-0.19830
H	4.27700	1.11720	-1.61860
H	3.57250	-0.35070	-2.36680
H	4.64180	-1.01480	-0.19280
H	3.65760	0.21370	0.67630
O	-0.71630	-1.04200	1.14520
C	-0.78740	-2.34810	1.08800
O	-0.46260	2.72770	1.61100
O	0.01120	-3.08550	0.51060
C	0.59090	2.26250	2.17340
O	1.28470	1.30910	1.73750
C	0.97530	2.92830	3.48480
H	1.99610	2.66700	3.77140
H	0.86990	4.01360	3.40100
H	0.28720	2.59130	4.26980
C	-1.99050	-2.89780	1.84620
H	-1.87730	-2.69070	2.91620
H	-2.90440	-2.39970	1.50870
C	-2.39440	1.10410	-0.38990
H	-2.06430	-3.97540	1.68790
C	-3.19550	1.16350	0.76220
H	-2.77380	1.55280	1.68280
C	-2.94650	0.59430	-1.57770
O	-1.11120	1.54610	-0.39120
C	-4.52200	0.73930	0.71720
O	-6.35690	-0.20590	-0.56940
C	-4.26490	0.15980	-1.61890
H	-6.78910	-0.10100	0.29230
H	-2.32130	0.54650	-2.46490
H	-4.69850	-0.23550	-2.53270
C	-5.06180	0.23280	-0.46930
H	-0.80890	2.13290	0.57580
H	-5.13970	0.79860	1.61270
C	0.30990	0.48840	-5.73210
C	0.58270	1.78460	-5.29090
C	1.08500	1.99850	-4.00540
C	1.30920	0.89470	-3.18660
C	1.03010	-0.41150	-3.60120
C	0.53390	-0.60550	-4.88990
H	-0.08490	0.32830	-6.73140
H	0.39980	2.63350	-5.94340
H	1.29310	2.99550	-3.63010
H	1.19210	-1.26010	-2.94020
H	0.31460	-1.61320	-5.23060
C	3.30780	-3.47610	4.01220
C	2.85760	-2.15550	3.93900
C	2.54370	-1.58290	2.70570
C	2.70520	-2.36600	1.56100
C	3.14060	-3.68740	1.60770
C	3.44710	-4.24000	2.85140
H	3.54150	-3.91440	4.97890
H	2.73990	-1.56800	4.84550
H	2.16900	-0.56280	2.63980
H	3.21530	-4.26250	0.69050

H 3.78530 -5.27060 2.91300
 Pd_Bisulfoxide_Quinone_2AcOH (Poli_D)
 E_scf (B3LYP/LACVP*) = -2455.05169440408 a.u.
 E_scf (M06/LACVP*) = -2454.03155458226 a.u.
 E_Gibbs = -2454.66157 a.u.
 E_Gibbs(318K) = -2454.672228 a.u.
 E_solv = -2455.09685599479 a.u.
 ZPE = 291.906 kcal/mol

Pd	-1.14580	0.25390	0.48950
S	-1.43580	-1.39790	2.14050
O	-2.82680	-1.89950	2.41270
S	-1.23070	-1.54130	-1.09110
O	-0.21810	-1.70440	-2.17300
C	-0.45340	-2.74160	1.35490
C	-1.11160	-3.05850	0.02060
H	-0.47250	-3.60440	2.02840
H	0.56330	-2.35200	1.23970
H	-0.51140	-3.76120	-0.56430
H	-2.11980	-3.45520	0.15810
O	-0.68390	1.72230	-0.82270
C	-1.22370	1.69330	-2.01320
O	-1.96590	0.80850	-2.43680
C	2.74900	-0.19230	-0.28960
C	2.83230	1.02900	-0.96360
C	3.54940	-1.26330	-0.69670
C	3.71330	1.17150	-2.03630
C	4.42320	-1.12120	-1.77140
H	3.48260	-2.20710	-0.16230
C	4.50870	0.09920	-2.44660
H	3.77500	2.12560	-2.55850
C	-0.83050	2.89550	-2.86410
H	-1.07450	3.82580	-2.34120
H	0.25150	2.88500	-3.03400
H	-1.35340	2.85610	-3.82150
H	2.19840	1.85530	-0.65620
O	1.89120	-0.37930	0.76920
O	5.39080	0.18430	-3.49810
H	5.32800	1.07230	-3.88190
H	5.04670	-1.94810	-2.09810
H	1.54110	0.49920	1.08150
O	1.02430	1.90490	1.91070
C	-0.11040	2.29920	2.21930
O	-1.23260	1.72780	1.90100
C	-0.28880	3.57080	3.03500
H	0.63690	3.80840	3.56320
H	-0.52740	4.39370	2.35080
H	-1.11910	3.47200	3.73910
C	0.54820	-0.72050	6.20090
C	-0.81530	-1.00530	6.10530
C	-1.40130	-1.21820	4.85690
C	-0.59660	-1.14120	3.72030
C	0.76870	-0.85130	3.78890
C	1.33430	-0.64720	5.04790
H	0.99930	-0.55080	7.17460
H	-1.42780	-1.05940	7.00090
H	-2.45830	-1.44110	4.75480
H	1.38300	-0.77390	2.89760
H	2.39380	-0.41950	5.12120
C	-5.42410	-2.19140	-2.78300
C	-4.30530	-2.28790	-3.61230
C	-3.02550	-2.08930	-3.09240



C	-2.89840	-1.80850	-1.73700
C	-4.00010	-1.70000	-0.88820
C	-5.27180	-1.89700	-1.42570
H	-6.41870	-2.33750	-3.19580
H	-4.42700	-2.50560	-4.66980
H	-2.14050	-2.12600	-3.71920
H	-3.87900	-1.46910	0.16730
H	-6.14290	-1.81460	-0.78170

Calculation of theoretical KIE

The calculation of the theoretical KIE was performed using the Hessian calculated in gas phase for the complexes **A**, **B** and **C**. To allow comparison with experimental results the Gibbs free energy was evaluated at $T=318\text{ K}$ ($=45\text{ }^{\circ}\text{C}$).

This gave the following results:

(A) Pd_Bisulfoxide_Allylbenzene_Acetate_Abstraction_TS	-2192.212272 a.u.
Pd_Bisulfoxide_Allylbenzene_Acetate_Abstraction_TS_D	-2192.21407 a.u.
(B) Pd_Bisulfoxide_Allylbenzene_Acetate_Abstraction_6Membered_TS	-2192.213173 a.u.
Pd_Bisulfoxide_Allylbenzene_Acetate_Abstraction_6Membered_TS_D	-2192.214928 a.u.
(C) Pd_Bisulfoxide_Allylbenzene_Acetate_External_Acetate_Abstraction_TS	-2420.850806 a.u.
Pd_Bisulfoxide_Allylbenzene_Acetate_External_Acetate_Abstraction_TS_D	-2420.852142 a.u.

From this we find the activation energies and differences in energy.

A: Activation energy of 60.5 kJ/mol (proton) and 64.9 kJ/mol (deuterium). Energy difference of 4.4 kJ/mol.

B: Activation energy of 58.1 kJ/mol (proton) and 62.6 kJ/mol (deuterium). Energy difference of 4.5 kJ/mol.

C: Activation energy of 15.6 kJ/mol (proton) and 20.3 kJ/mol (deuterium). Energy difference of 4.7 kJ/mol.

These results corresponds to the following differences in reaction rates, when using a Boltzmann expression evaluated at $T=318\text{ K}$.

A: Reaction rate of 5.3

B: Reaction rate of 5.5

C: Reaction rate of 5.9

Performing a similar analysis using differences in zero-point energies (ZPE) resulted in the following calculated KIE values:

A: KIE value of 5.7

B: KIE value of 5.9

C: KIE value of 3.7

Thus documenting that the observed effect arises mainly from the differences in bond strengths, as expected.

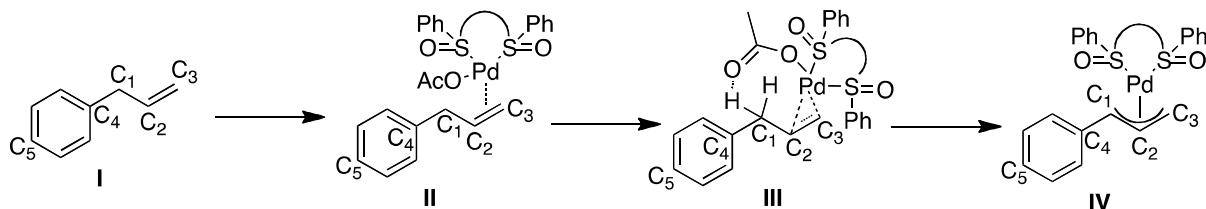
The results for **C** are obtained with a starting point structure where the O-H distance between substrate and acetate is set to $1.6\text{ \AA}$. If a starting point with $1.5\text{ \AA}$ is used the following results are obtained:

Reaction rate from Gibbs free energy ($T=318\text{ K}$) = 3.9

KIE value from ZPE = 2.5

Calculated charges

The electrostatic potential (esp) charges for the structures shown here were calculated. The various carbons in the allylbenzene substrate are marked.



Scheme S6 Numbering of the carbon atoms used for the calculation of charges.

In the figure the absolute charges of the marked carbons in the structures are plotted.

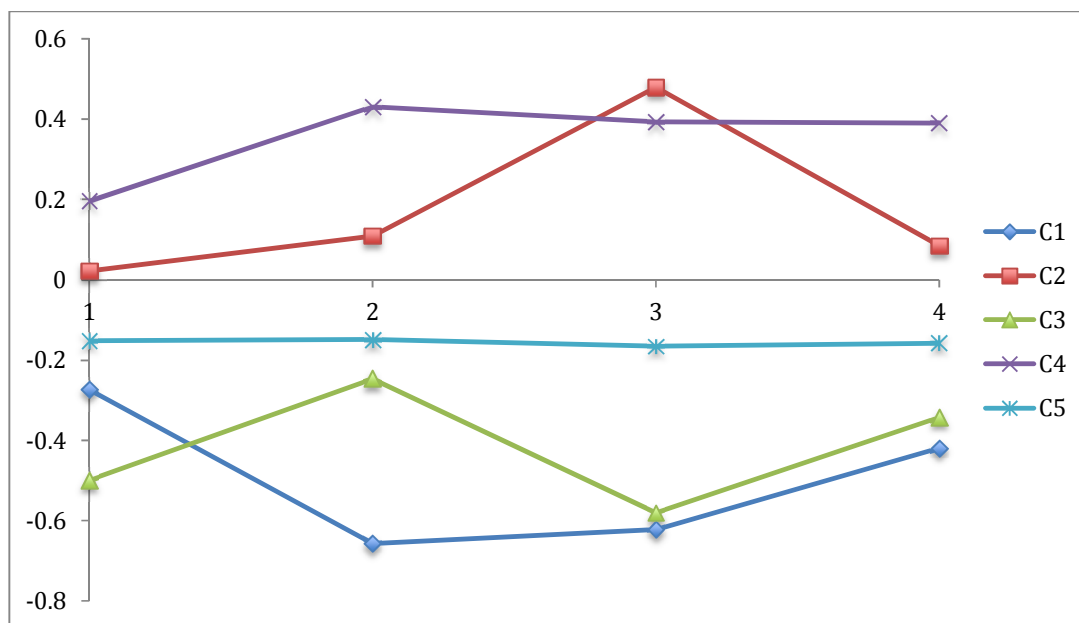


Figure S19 Calculated charges for the carbon atoms along the C-H activation pathway.

- (1) Biellmann, J.; d'Orchymont, H. *J. Org. Chem.* **1982**, *47*, 2882-2886.
- (2) Bialecki, J.; Ruzicka, J.; Attygalle, A. *J. Mass Spectrom.* **2006**, *41*, 1195-1204.
- (3) López-Pérez, A.; Adrio, J.; Carretero, J. C. *Org. Lett.* **2009**, *11*, 5514-5517.
- (4) Young, A. J.; White, M. C. *J. Am. Chem. Soc.* **2008**, *130*, 14090-14091.