

Supporting Information For:

Stark Effects after Excited-State Interfacial Electron Transfer at Sensitized TiO₂ Nanocrystallites

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Figure S1 – (A) UV-Vis absorption spectra of $[\text{Ru}(\text{dtb})_2(\text{dcb})]^{2+}$ anchored to TiO_2 thin films in neat CH_3CN and after introduction of 100 mM LiClO_4 highlighting the differences in the spectra, λ_{max} , and isosbestic points for the Li^+ -induced spectral changes as a function of the indicated surface coverage. (B) Fraction growth of the spectroscopic changes monitored at 505 nm as a function of the concentration of LiClO_4 in solution. Overlaid on the data are fits to the Langmuir isotherm model: a best fit (green) and a range of fits that bracket the spectroscopic data (orange) that result in the adduct formation constants shown.

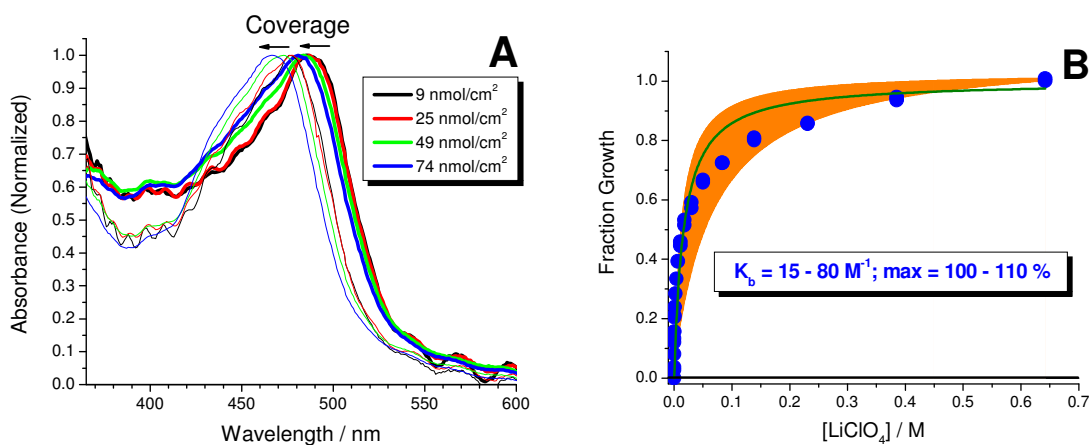


Figure S2 – (A) UV-Vis absorption spectra of $[\text{Ru}(\text{dtb})_2(\text{dcb})]^{2+}$ anchored to a TiO_2 thin film in 500 mM $\text{LiClO}_4/\text{CH}_3\text{CN}$ (black) and after four (orange; 31 mM), eight (violet; 2.0 mM), and ten (cyan; 0.49 mM LiClO_4) 1:2 dilutions of the solution electrolyte. The spectra were converted from wavelength to wavenumber space by a well-known correction based on the change of variables theorem. The most dilute spectrum depicted was initially modeled as a sum of three Gaussians in energy in the $18835 - 25582 \text{ cm}^{-1}$ ($390.9 - 530.9 \text{ nm}$) range. The collective Gaussian fit is overlaid in pink and individual Gaussian components are shown in light green. (B) Difference absorption spectra with the same color scheme as in Panel A referenced to the most dilute spectrum (*i.e.*, cyan spectrum in panel A). Overlaid on the spectra at 500 mM LiClO_4 (black) are spectral simulations using the difference between (a) the most dilute spectrum from Panel A and the same spectrum shifted 220 cm^{-1} (pink) and (b) the original and scaled & shifted (210 cm^{-1}) versions of the Gaussians from panel A (light green).

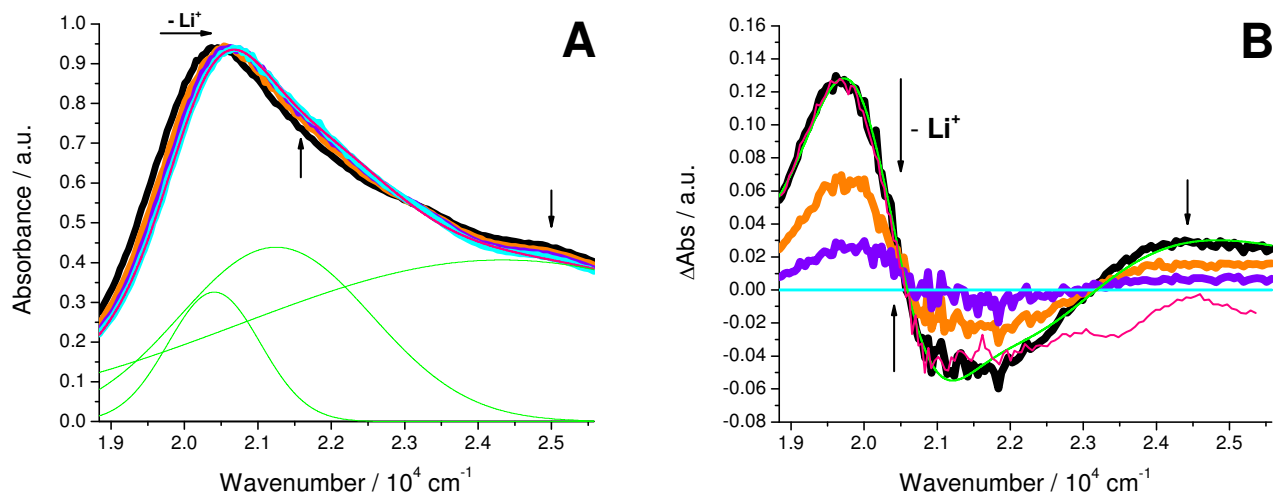


Figure S3 – Extinction coefficient spectra of $\text{TiO}_2(e^-)$ s and the sensitizers and oxidized donors required for spectral modeling in this study. Spectroelectrochemical reduction, chemical oxidation with Br_2 , and Li^+ titration were all employed in order to obtain these difference spectra.

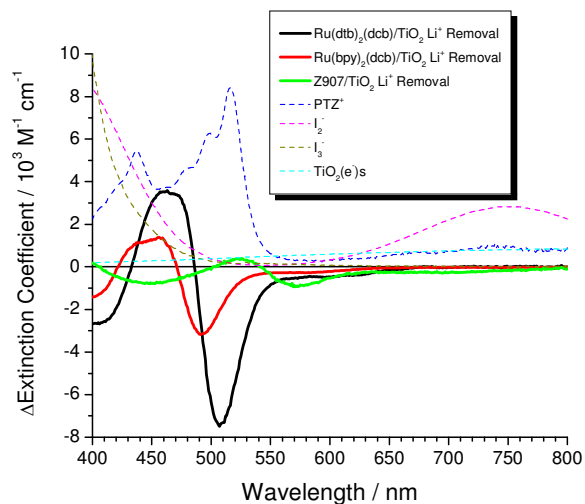
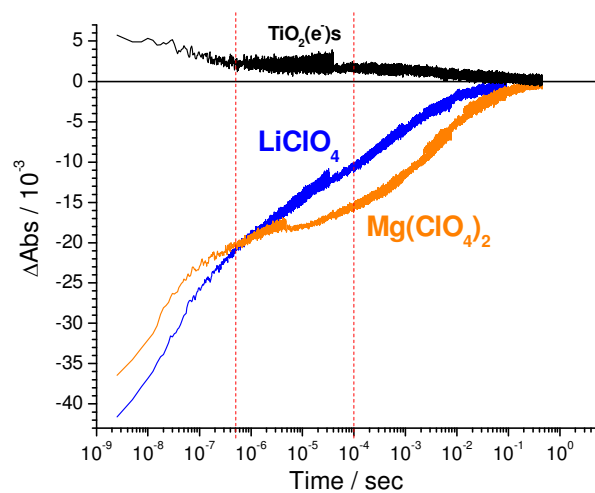


Figure S4 – Absorption difference spectra at 510 nm after pulsed 532 nm excitation of a $\text{Ru}(\text{dtb})_2(\text{dcb})/\text{TiO}_2$ thin film in 500 mM TBAI/ CH_3CN with either 500 mM LiClO_4 or 500 mM $\text{Mg}(\text{ClO}_4)_2$. Also shown is a representative $\text{TiO}_2(e^-)$ spectrum at 750 nm, as both were within error the same.



Complete reference 7: Kroon, J. M.; Bakker, N. J.; Smit, H. J. P.; Liska, P.; Thampi, K. R.; Wang, P.; Zakeeruddin, S. M.; Grätzel, M.; Hirsch, A.; Hore, S.; Würfel, U.; Sastrawan, R.; Durrant, J. R.; Palomares, E.; Pettersson, H.; Gruszecki, T.; Walter, J.; Skupien, K.; Tulloch, G. E. *Prog. Photovolt. Res. Appl.* **2007**, *15*, 1-18.