Supporting Information

Core/Shell Structured Noble Metal (Alloy)/Cadmium Selenide Nanocrystals

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Synthesis of FePt nanoparticles

FePt nanoparticles were prepared by a procedure modified from those reported previously.^{1, 2} For a typical reaction, platinum (II) acetylacetonate (100 mg), iron (III) acetylacetonate (180 mg), 1,2-hexadecandiol (260 mg) and oleylamine (10mL) were mixed in a 50 mL three-necked flask, and the mixture was heated to 120 °C under argon flow and held at 120 °C for 20 min, followed by keeping at 240 °C for 1 h and subsequently at 298 °C for 2 h before being cooled to RT. The resulting FePt nanoparticles were purified by washing with ethanol and subsequent centrifuging and dispersed in n-hexane.

Synthesis of NiPt nanoparticles

NiPt nanoparticles were prepared by a procedure modified from those reported previously.³ For a typical reaction, 78 mg of nickel (II) acetylacetonate, 0.2 mL of oleylamine, 0.2 mL of oleic acid and 10mL of benzyl ether were mixed in a 50 mL three-necked flask, and the mixture was heated to 120 °C and held at 120 °C for 30 min under argon flow before 390 mg of 1,2-hexadecandiol was added. The mixture was then heated to 200 °C and 40 mg of platinum (II) acetylacetonate was added. The reaction mixture was stirred for 1 h at 200 °C. At RT, the resulting NiPt nanoparticles were purified by washing with ethanol and subsequent centrifuging and dispersed in n-hexane.

Synthesis of CoPt nanoparticles

CoPt nanoparticles were synthesized via co-reduction of metal salts. 1,4 The mixture of

platinum (II) acetylacetonate (50 mg), cobalt (II) acetylacetonate (65 mg), 1,2-hexadecandiol (130 mg) and OAm(5mL) was heated to 140 °C under argon flow and held at 140 °C for 20 min. After the hold, the mixture was heated to 210 °C and held at 210 °C for 40 min. At RT, the resulting CoPt nanoparticles were purified by washing with ethanol and subsequent centrifuging and dispersed in n-hexane.

Synthesis of Au nanoparticles

2 nm Au nanoparticles were synthesized by a procedure modified from reported methods.⁵ In the modified procedure, only tetraoctylammonium bromide was substituted by didodecyldimethylammonium bromide,⁶ the resulting Au nanoparticles were 2 nm in size. A seed-mediated growth method was used to synthesize 6 nm Au nanoparticles, with 2 nm Au nanoparticles as seeds and hydrazine hydrate as reducing agent. The as-prepared 6 nm Au nanoparticles were dispersed in toluene.

Synthesis of noble metal (alloy)/cadmium chalcogenide nanocrystals

The mixture of CdO (40 mg) and stearic acid (SA, 350 mg)⁷ was heated to 160 °C and held at 160 °C for 20 min, and added with 15 mL of solvent and 10 mL of n-hexane or toluene solution containing appropriate amount of noble metal (alloy) nanoparticles at RT. Purging with Ar, the above mixture was heated to 85-110 °C to remove n-hexane or toluene and subsequently heated to 290 - 300 °C. At this temperature, TOPSe (79 mg of Se powder in 2.5 mL of TOP) or DOAS (20 mg of S powder in 2.5 mL of dioctylamine) was injected swiftly, and products with different shell sizes could be produced by modifying reaction time. After being cooled to RT, the products were purified by size-selective precipitation/washing steps, finally the products were dispersed in n-hexane.

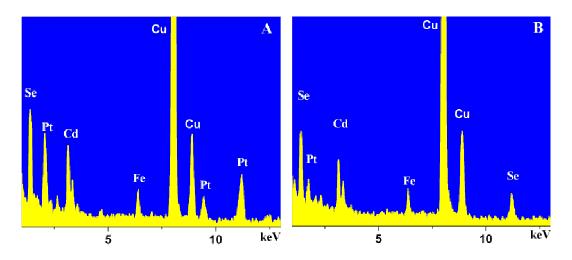


Figure S1. EDX Spectra of the as-prepared FePt/CdSe nanocrystals. (A) the core and (B) the shell.

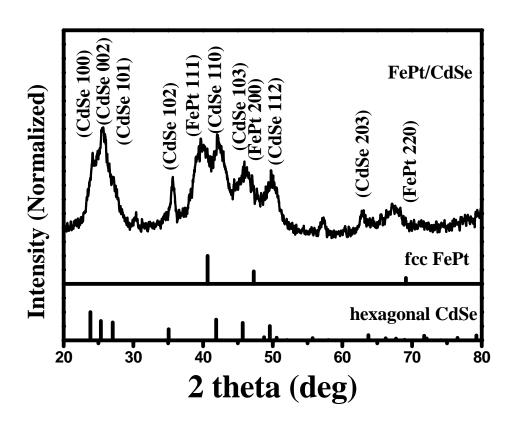


Figure S2 XRD pattern of the as-prepared FePt/CdSe nanocrystals.

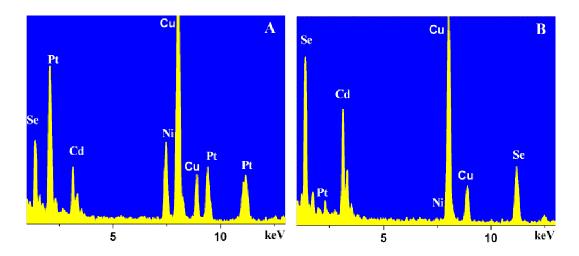


Figure S3 EDX Spectra of the as-prepared NiPt/CdSe nanocrystals. (A) the core and (B) the shell.

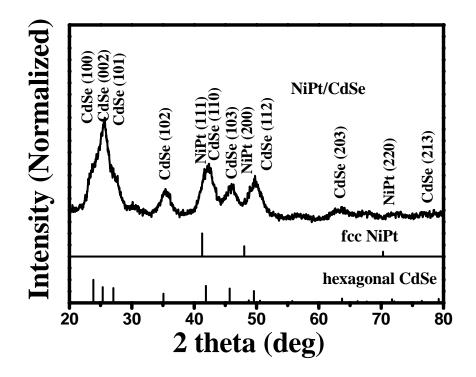


Figure S4 XRD pattern of the as-prepared NiPt/CdSe nanocrystals.

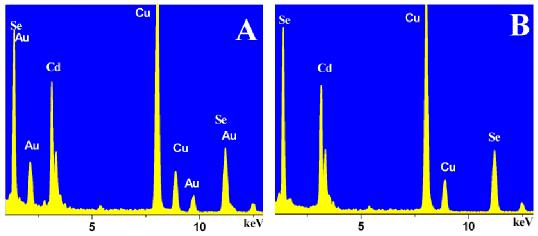


Figure S5 EDX Spectra of the as-prepared Au/CdSe nanocrystals. (A) the core and (B) the shell.

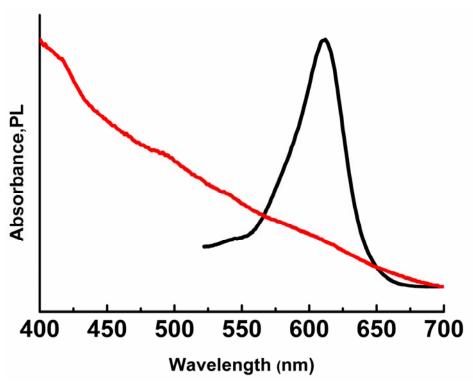


Figure S6 Absorption and emission spectra of one of the as-prepared FePt/CdSe nanocrystals samples.

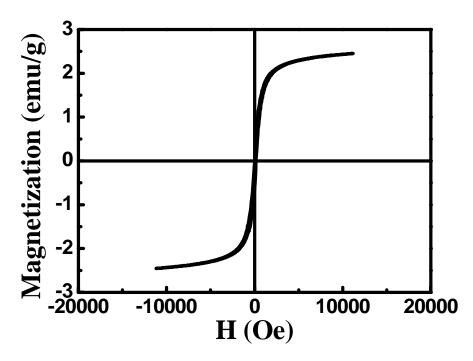


Figure S7 RT hysteresis loop of one of the as-prepared FePt/CdSe nanocrystals samples.

⁽¹⁾Elkins, K. E.; Vedantam, T. S.; Liu, J. P.; Zeng, H.; Sun, S.; Ding, Y.; Wang, Z. L. *Nano Lett.* **2003**, *3*, 1647.

⁽²⁾ Chen, M.; Liu, J. P.; Sun, S. J. Am. Chem. Soc. 2004, 126, 8394.

⁽³⁾Ahrenstorf, K.; Albrecht, O.; Heller, H.; Kornowski, A.; Görlitz, D.; Weller, H. Small 2007, 3, 271.

⁽⁴⁾ Chinnasamy, C. N.; Jeyadevan, B.; Shinoda, K.; Tohji, K. J. Appl. Phys. 2003, 93, 7583.

⁽⁵⁾Shi, W.; Zeng, H.; Sahoo, Y.; Ohulchanskyy, T. Y.; Ding, Y.; Wang, Z. L.; Swihart, M.; Prasad, P. N. *Nano Lett.* **2006**, *6*, 875.

⁽⁶⁾Jana, N. R.; Peng, X. J. Am. Chem. Soc. 2003, 125, 14280.

⁽⁷⁾Peng, Z. A.; Peng, X. J. Am. Chem. Soc. 2001, 123, 183.