

Supporting Information for “Applying Polymer Solubility Theory to Single Walled Carbon Nanotube Solutions”

Table S1 presents the molar volume values for the aryl carboxylic acid and aryl hydroxyl groups calculated from group additivity theory¹ as well as estimates for the bare carbon nanotube species under investigation².

Table S1. Summary of Molar Volume Calculations

CNT	d (nm)	#C/nm	V (nm ³)
(6,5)	0.76	89.5	953
(11,3)	1.01	119.8	1448
(10,10)	1.38	162.5	2321
R Groups			
C6H4-COOH	-	-	0.131
C6H4-OH	-	-	0.107

Table S2 shows the group additivity values used in the Refined Solubility Parameter and Fedors polymer solubility theories.

Table S2. Summary of Group Additivity Values

Data from Van Krevelen, D.W., *Properties of Polymers*, Elsevier, New York, 1976

Refined Solubility Parameter						
	>C=	>C<	=CH-	-COOH	-OH	Ring
F _{di}	70	-70	200	530	210	1270
F _{pi}	0	0	0	420	500	110
E _{hi}	0	0	0	10000	20000	0

Fedors Method						
Cohesive Energy (J/cm ³)						
>C=	>C<	=CH-	-COOH	-OH	Ring	
89518	0	0	0	0	0	
93889	4370	17481	4370	0	4370	
93889	4370	17481	0	4370	4370	
119808	0	0	0	0	0	
125657	5849	23396	5849	0	5849	
125657	5849	23396	0	5849	5849	
162537	0	0	0	0	0	
170472	7935	31740	7935	0	7935	
170472	7935	31740	0	7935	7935	

Table S3 summarizes the solubility parameters (δ , SI units) and Flory-Huggins interaction parameters (χ) calculated for the OH-SWNT reaction series.

Table S3. Calculated Solubility Parameters and Flory-Huggins Interaction Parameter for OH-SWNT

Sample	# FG / 100 C's	Refined Solubility Parameter			X - water			X - DMF		
		(6,5)	(11,3)	(10,10)	(6,5)	(11,3)	(10,10)	(6,5)	(11,3)	(10,10)
P2	0.0	10.9	9.6	8.1	10.05	10.77	11.62	6.05	7.23	8.71
P2OH1	1.5	14.1	12.7	11.2	8.42	9.09	9.91	3.62	4.57	5.82
P2OH3	5.1	18.2	16.9	15.4	6.50	7.06	7.80	1.38	1.95	2.80
P2OH4	7.9	20.1	18.9	17.4	5.71	6.19	6.85	0.71	1.10	1.72
P2OH2	8.5	20.4	19.2	17.7	5.59	6.06	6.70	0.62	0.98	1.57

Sample	# FG / 100 C's	Fedors			X - water			X - DMF		
		(6,5)	(11,3)	(10,10)	(6,5)	(11,3)	(10,10)	(6,5)	(11,3)	(10,10)
P2	0.0	31.4	29.5	27.1	2.02	2.51	3.19	1.36	0.68	0.17
P2OH1	1.5	31.4	29.7	27.6	2.03	2.46	3.07	1.34	0.73	0.23
P2OH3	5.1	31.3	30.0	28.3	2.05	2.38	2.86	1.30	0.83	0.37
P2OH4	7.9	31.3	30.1	28.6	2.06	2.34	2.75	1.29	0.88	0.45
P2OH2	8.5	31.2	30.2	28.7	2.06	2.34	2.73	1.28	0.89	0.47

Sample	# FG / 100 C's	Combined Maiti - Fedors			X - water			X - DMF		
		(6,5)	(11,3)	(10,10)	(6,5)	(11,3)	(10,10)	(6,5)	(11,3)	(10,10)
P2	0.0	26.0	22.5	19.4	3.54	4.75	6.00	0.04	0.17	0.93
P2OH1	1.5	26.7	23.7	20.9	3.32	4.33	5.39	0.11	0.04	0.49
P2OH3	5.1	27.8	25.5	23.2	2.99	3.72	4.52	0.28	0.01	0.09
P2OH4	7.9	28.3	26.4	24.4	2.84	3.44	4.10	0.38	0.07	0.01
P2OH2	8.5	28.4	26.5	24.6	2.81	3.39	4.03	0.40	0.09	0.00

Sample	# FG / 100 C's	Combined Maiti - RSP			X - water			X - DMF		
		(6,5)	(11,3)	(10,10)	(6,5)	(11,3)	(10,10)	(6,5)	(11,3)	(10,10)
P2	0.0	26.0	22.5	19.4	3.54	4.75	6.00	0.04	0.17	0.93
P2OH1	1.5	29.8	26.0	22.4	2.44	3.56	4.79	0.76	0.04	0.18
P2OH3	5.1	35.3	31.3	27.4	1.18	2.06	3.13	3.44	1.29	0.20
P2OH4	7.9	38.1	34.0	29.9	0.73	1.45	2.39	5.48	2.60	0.82
P2OH2	8.5	38.5	34.4	30.4	0.66	1.36	2.28	5.86	2.86	0.96

Table S4 summarizes the solubility parameters (δ , SI units) and Flory-Huggins interaction parameters (χ) calculated for the OH-SWNT reaction series.

Table S4. Calculated Solubility Parameters and Flory-Huggins Interaction Parameter for COOH-SWNT

Sample	# FG / 100 C's	Refined Solubility Parameter			X - water			X - DMF		
		(6,5)	(11,3)	(10,10)	(6,5)	(11,3)	(10,10)	(6,5)	(11,3)	(10,10)
P2	0.0	10.9	9.6	8.1	10.05	10.77	11.62	6.05	7.23	8.71
P2COOH3	0.7	12.3	11.0	9.5	9.31	10.01	10.86	4.89	5.98	7.37
P2COOH2	1.0	12.8	11.4	9.9	9.09	9.78	10.62	4.56	5.61	6.98
P2COOH1	1.2	13.1	11.8	10.3	8.90	9.58	10.42	4.29	5.30	6.64
P2COOH4	1.6	13.6	12.3	10.8	8.64	9.31	10.14	3.91	4.88	6.18
P2COOH5	2.2	14.4	13.1	11.6	8.24	8.89	9.70	3.37	4.27	5.49
P2COOH6	3.0	15.3	14.0	12.4	7.84	8.46	9.25	2.86	3.67	4.80

Sample	# FG / 100 C's	Fedors			X - water			X - DMF		
		(6,5)	(11,3)	(10,10)	(6,5)	(11,3)	(10,10)	(6,5)	(11,3)	(10,10)
P2	0.0	31.4	29.5	27.1	2.02	2.51	3.19	1.36	0.68	0.17
P2COOH3	0.7	31.1	29.4	27.2	2.09	2.55	3.19	1.24	0.64	0.17
P2COOH2	1.0	31.0	29.3	27.2	2.12	2.56	3.19	1.20	0.63	0.17
P2COOH1	1.2	30.9	29.3	27.2	2.14	2.57	3.18	1.17	0.62	0.17
P2COOH4	1.6	30.8	29.2	27.2	2.17	2.59	3.18	1.12	0.60	0.17
P2COOH5	2.2	30.6	29.1	27.2	2.22	2.62	3.18	1.05	0.57	0.17
P2COOH6	3.0	30.4	29.0	27.2	2.27	2.64	3.17	0.98	0.55	0.18

Sample	# FG / 100 C's	Combined Maiti - Fedors			X - water			X - DMF		
		(6,5)	(11,3)	(10,10)	(6,5)	(11,3)	(10,10)	(6,5)	(11,3)	(10,10)
P2	0.0	26.0	22.5	19.4	3.54	4.75	6.00	0.04	0.17	0.93
P2COOH3	0.7	28.1	22.9	21.0	2.92	4.61	5.35	0.33	0.11	0.46
P2COOH2	1.0	28.7	23.1	21.5	2.73	4.56	5.14	0.47	0.10	0.35
P2COOH1	1.2	29.3	23.2	22.0	2.57	4.52	4.96	0.62	0.09	0.26
P2COOH4	1.6	26.3	23.3	20.6	3.47	4.46	5.51	0.06	0.07	0.56
P2COOH5	2.2	26.3	23.6	21.0	3.44	4.37	5.35	0.07	0.05	0.47
P2COOH6	3.0	26.4	23.8	21.4	3.41	4.28	5.19	0.08	0.03	0.37

Sample	# FG / 100 C's	Combined Maiti - RSP			X - water			X - DMF		
		(6,5)	(11,3)	(10,10)	(6,5)	(11,3)	(10,10)	(6,5)	(11,3)	(10,10)
P2	0.0	26.0	22.5	19.4	3.54	4.75	6.00	0.04	0.17	0.93
P2COOH3	0.7	26.1	24.4	20.0	3.50	4.09	5.75	0.05	0.01	0.74
P2COOH2	1.0	26.2	25.0	20.2	3.49	3.89	5.67	0.06	0.00	0.68
P2COOH1	1.2	26.2	25.5	20.3	3.48	3.71	5.60	0.06	0.01	0.63
P2COOH4	1.6	30.1	26.2	22.6	2.36	3.47	4.72	0.86	0.06	0.15
P2COOH5	2.2	31.3	27.4	23.7	2.04	3.12	4.33	1.32	0.20	0.04
P2COOH6	3.0	32.7	28.7	24.8	1.73	2.75	3.93	1.92	0.46	0.00

References:

- (1) Ammon, H. L., Mitchell, S. *Propellants, Explosives, Pyrotechnics* **1998**, 23, 260-265.
- (2) Nair, N.; Kim, W. J.; Braatz, R. D.; Strano, M. S. *Langmuir* **2008**, 24, 1790-1795.