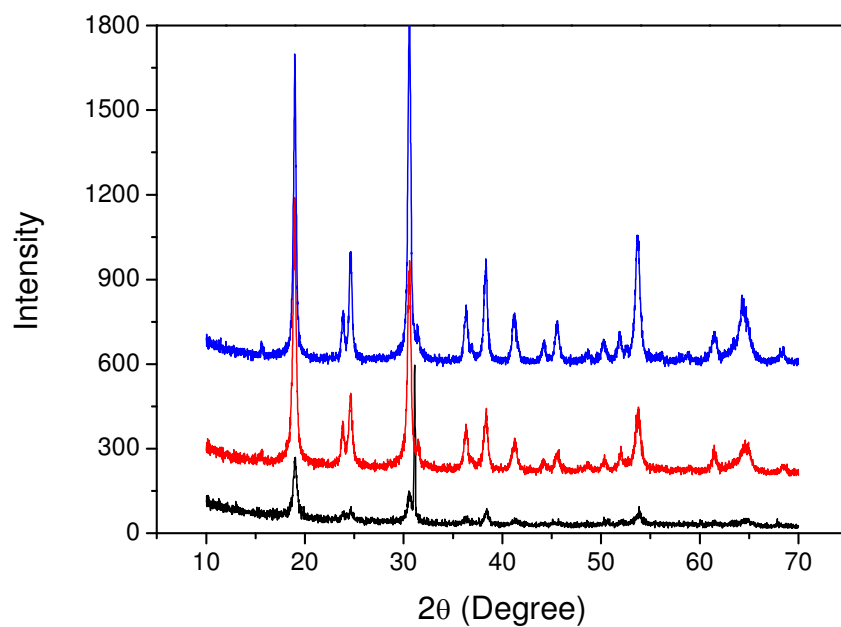


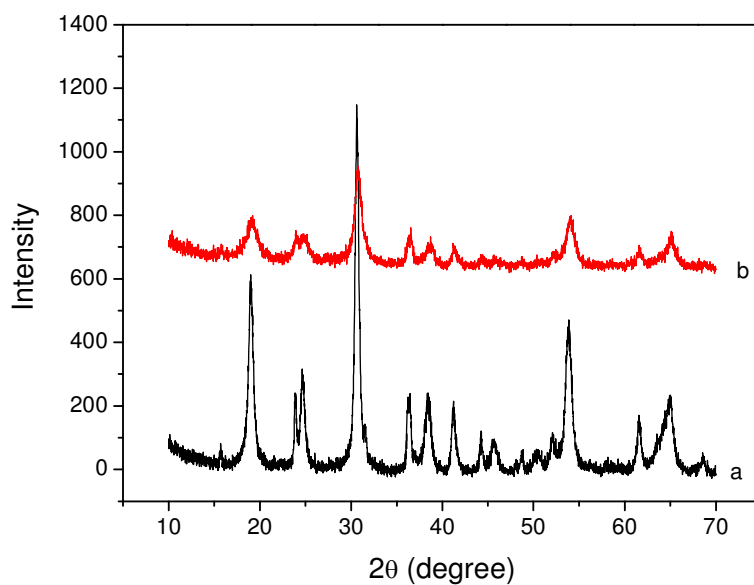
## Supporting Information

### Hierarchical FeWO<sub>4</sub> Microcrystals: Solvothermal Synthesis, and Their Photocatalytic and Magnetic Properties

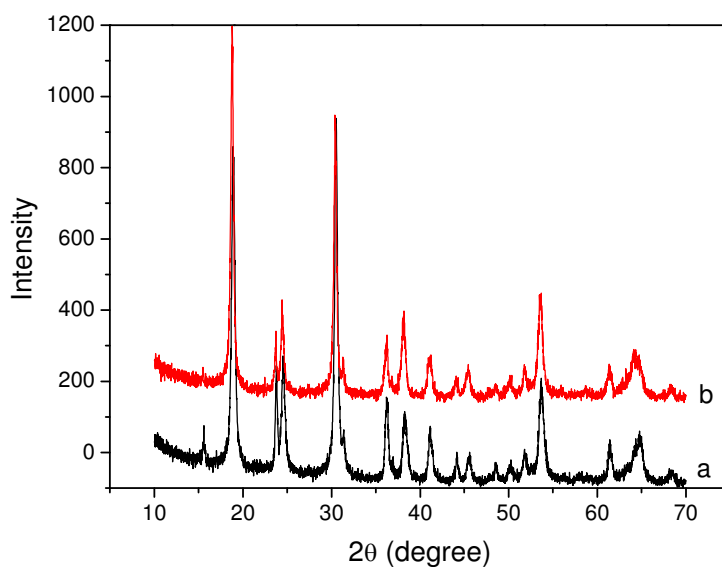
*Yu-Xue Zhou, Hong-Bin Yao, Qiao Zhang, Jun-Yan Gong, Shu-Juan Liu, Shu-Hong Yu\**



*Figure S1.* XRD patterns of FeWO<sub>4</sub> obtained at 200°C for different reaction time, [FeCl<sub>3</sub>] = [Na<sub>2</sub>WO<sub>4</sub>] = 0.02 mol L<sup>-1</sup>, [NaAc] = 0.2 mol L<sup>-1</sup>, V<sub>H<sub>2</sub>O</sub>: V<sub>EG</sub> = 1:9. (a) 4 h, (b) 5 h, (c) 12 h.



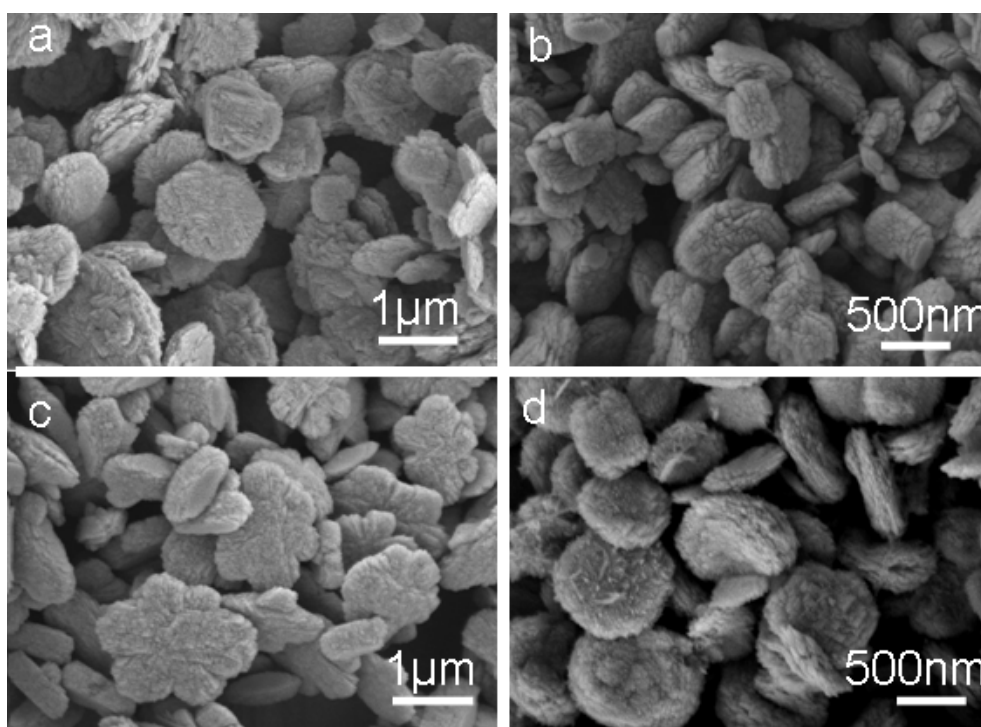
**Figure S2.** XRD patterns of  $\text{FeWO}_4$  microcrystals obtained at  $200\text{ }^\circ\text{C}$  for 12 h in media with different volume ratio of  $\text{H}_2\text{O}$  and EG ( $V_{\text{H}_2\text{O}}: V_{\text{EG}}$ ) (a) 2:8 , (b) 3:7 .  $[\text{FeCl}_3] = [\text{Na}_2\text{WO}_4] = 0.02\text{ mol L}^{-1}$ ,  $[\text{NaAc}] = 0.2\text{ mol L}^{-1}$ .



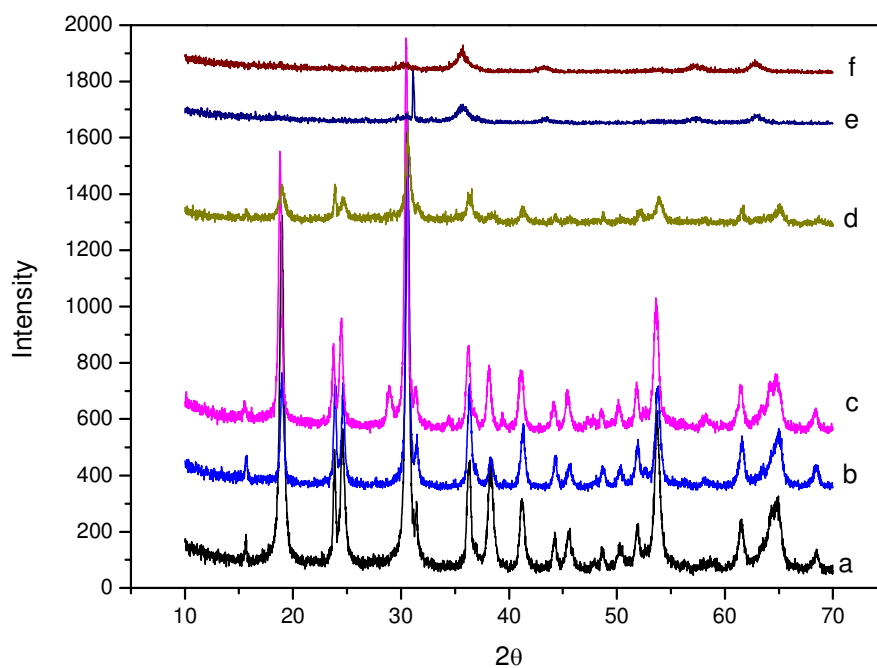
**Figure S3.** XRD patterns of  $\text{FeWO}_4$  microcrystals prepared at  $200\text{ }^\circ\text{C}$  for 12 h,  $[\text{FeCl}_3] = [\text{Na}_2\text{WO}_4] = 0.02\text{ mol L}^{-1}$ . (a)  $[\text{NaAc}] = 0.04\text{ mol}\cdot\text{L}^{-1}$ , (b)  $[\text{NaAc}] = 0.4\text{ mol}\cdot\text{L}^{-1}$ .

**Table S1.** Products obtained by using different alkaline source in reaction solution.

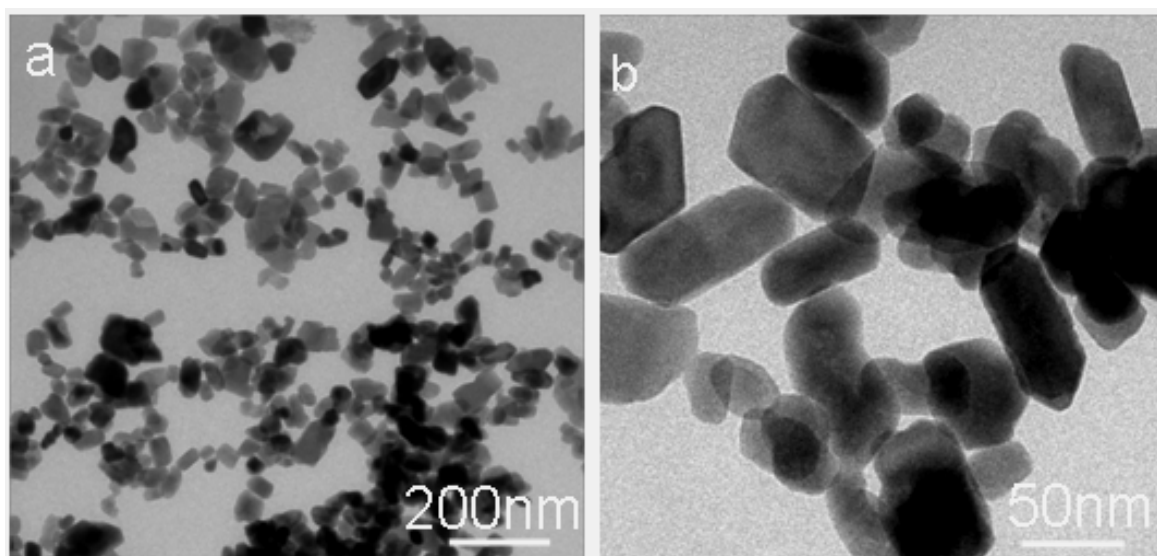
Dosage of alkaline source (10 mmol)	Composition of the products
HCOONa (a)	disk-like FeWO <sub>4</sub> microcrystals
Na <sub>2</sub> C <sub>2</sub> O <sub>4</sub> (b)	irregular FeWO <sub>4</sub> disk-like FeWO <sub>4</sub> microcrystals
CH <sub>3</sub> CH <sub>2</sub> COONa (c)	non-uniform hexangular FeWO <sub>4</sub> microcrystals
NH <sub>3</sub> .H <sub>2</sub> O (d)	disk-like FeWO <sub>4</sub> microcrystals
NaOH (e)	maghemite
Na <sub>2</sub> CO <sub>3</sub> (f)	maghemite



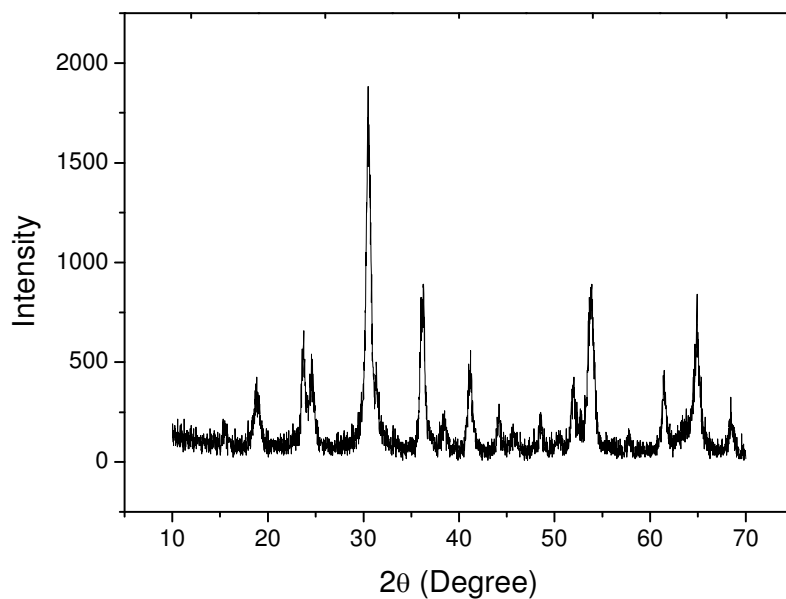
**Figure S4.** SEM images of FeWO<sub>4</sub> microcrystals prepared at 200 °C for 12 h by adding 10 mmol different alkaline source into reaction solution. (a) HCOONa, (b) Na<sub>2</sub>C<sub>2</sub>O<sub>4</sub>, (c) CH<sub>3</sub>CH<sub>2</sub>COONa and (d) NH<sub>3</sub>.H<sub>2</sub>O into reaction solution as alkaline source, [FeCl<sub>3</sub>] = [Na<sub>2</sub>WO<sub>4</sub>] = 0.02 mol L<sup>-1</sup>.



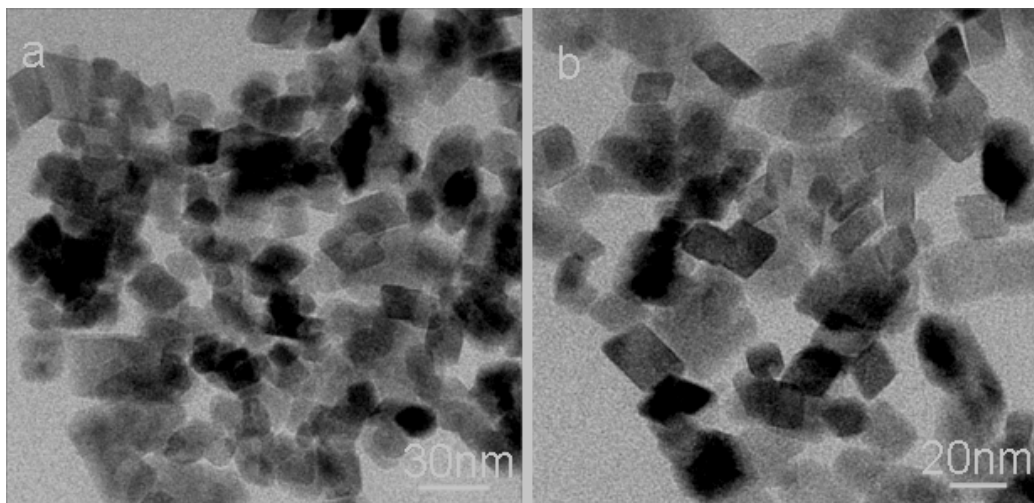
*Figure S5.* XRD patterns of the samples prepared at 200°C for 12 h by adding 10 mmol different alkaline source into reaction solution. (a) HCOONa, (b) Na<sub>2</sub>C<sub>2</sub>O<sub>4</sub>, (c) CH<sub>3</sub>CH<sub>2</sub>COONa, (d) NH<sub>3</sub>·H<sub>2</sub>O, (e) NaOH and (f) Na<sub>2</sub>CO<sub>3</sub>. [FeCl<sub>3</sub>] = [Na<sub>2</sub>WO<sub>4</sub>] = 0.02 mol L<sup>-1</sup>.



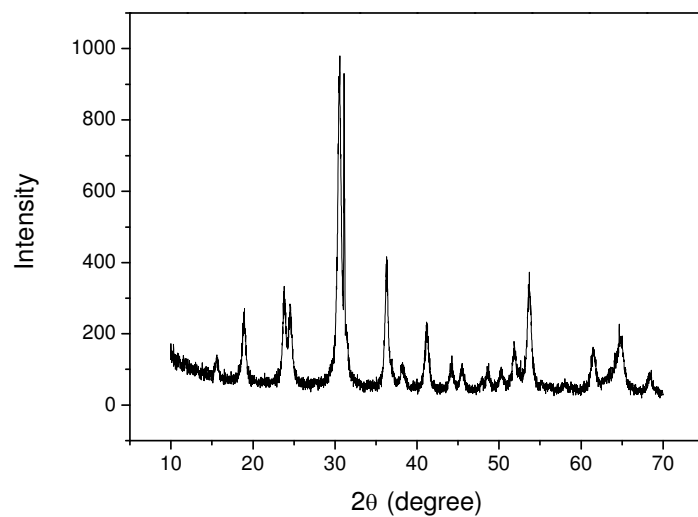
*Figure S6.* TEM images of the  $\text{FeWO}_4$  nanocrystals prepared at  $200^\circ\text{C}$  for 12 h without using EG.  $[\text{FeCl}_2] = [\text{Na}_2\text{WO}_4] = 0.02 \text{ mol}\cdot\text{L}^{-1}$ ,  $[\text{NaAc}] = 0.2 \text{ mol}\cdot\text{L}^{-1}$ ,  $V_{\text{H}_2\text{O}} = 50 \text{ ml}$ .



*Figure S7.* XRD pattern of the  $\text{FeWO}_4$  nanocrystals prepared at  $200^\circ\text{C}$  for 12 h without using EG.  $[\text{FeCl}_2] = [\text{Na}_2\text{WO}_4] = 0.02 \text{ mol}\cdot\text{L}^{-1}$ ,  $[\text{NaAc}] = 0.2 \text{ mol}\cdot\text{L}^{-1}$ ,  $V_{\text{H}_2\text{O}} = 50 \text{ ml}$ .



*Figure S8.* TEM images of  $\text{FeWO}_4$  nanocrystals prepared at  $200^\circ\text{C}$  for 12 h,  $V_{\text{H}_2\text{O}}: V_{\text{EG}} = 1: 9$ ,  $[\text{FeCl}_2] = [\text{Na}_2\text{WO}_4] = 0.02 \text{ mol}\cdot\text{L}^{-1}$ ,  $[\text{NaAc}] = 0.2 \text{ mol}\cdot\text{L}^{-1}$ .



*Figure S9.* XRD pattern of the  $\text{FeWO}_4$  nanocrystals prepared at 200 °C for 12 h in a mixed solvent.  $V_{\text{H}_2\text{O}}: V_{\text{EG}} = 1: 9$ ,  $[\text{FeCl}_2] = [\text{Na}_2\text{WO}_4] = 0.02 \text{ mol}\cdot\text{L}^{-1}$ ,  $[\text{NaAc}] = 0.2 \text{ mol}\cdot\text{L}^{-1}$ .