

Supporting Information:

**Photochemically-induced Changes in Dissolved Organic Matter
Identified by Ultrahigh Resolution Fourier Transform - Ion Cyclotron
Resonance - Mass Spectrometry**

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Dissolved Organic Carbon Analysis. Small aliquots from all four water samples (dark controls as well as irradiated samples) were acidified and their DOC content measured using a Shimadzu TOC 5000 total organic carbon analyzer.

UV/Vis-Absorbance. A Cary Varian 100 dual beam UV/Vis spectrophotometer and a 5 cm path length quartz cell were used to determine the UV/Vis absorbance between 280-700 nm. The spectra were baseline corrected with Milli-Q water as a blank. The raw optical density A (absorbance) was normalized to the pathlength (L) of the cuvette and converted into a CDOM absorption coefficient ($a_{\text{CDOM}}(\lambda)$):

$$a_{\text{CDOM}}(\lambda) = 2.303 \times A(\lambda) \times L^{-1} \quad (1)$$

The spectral slope S was determined across the wavelength range 280 - 700 nm as described elsewhere (1). The extent of photodegradation across the entire wavelength range (280-700 nm) was quantified for specific wavelengths (λ) using the absolute decrease in the absorption coefficient ($\Delta a_{\text{CDOM}}(\lambda)$). Additionally, the value of $a_{\text{CDOM}}(\lambda)$ at 355 nm ($a_{\text{CDOM}}(355)$) was normalized to the DOC content.

Excitation emission matrix fluorescence (EEM) spectroscopy. EEM fluorescence was measured using a Jobin Yvon SPEX FluoroMax-3 fluorescence spectrometer. The emission was recorded between 280 and 600 nm for excitation wavelengths at 5 nm intervals ranging from 250-500 nm. The EEM spectra collected in the present study were calibrated to quinine sulfate units (QSU) and corrected for Raleigh and Raman scattering peaks using a collection of software tools incorporated into a MATLAB

toolbox "FLToolbox 1.91" (2). The total EEM fluorescence (EEM_{TOTAL}) was calculated using the entire integral of the spectrum and does not have a unit.

Additional Method Description: ESI-FT-ICR-MS Analysis.

The sample solution was injected at a flow rate of $0.5 \mu\text{L min}^{-1}$ into a single spray source to which a needle voltage of -2.3 kV was applied to generate negative molecular ions. Octopole frequencies were set to 1.6 MHz to maximize the data set distribution ($<1000 \text{ m/z}$) for each experiment. After ion accumulation in the octopoles, the ions were transferred into the cylindrical ion cell (ICR cell), where they were separated by their unique cyclotron frequencies for analysis. A Fourier Transformation is then applied to the data to give a spectrum of the relative abundance of each ion as a function of its m/z ratio. Mass spectra were calibrated using the exact masses of internal calibrant ions that showed high relative abundances. These calibrant ions were also part of Kendrick series and have been observed in previous DOM mass spectra. The mass accuracy achieved using the above calibration procedure was $< 0.2 \text{ ppm}$.

A spectrum of the m/z ions present in one DOM sample can contain several thousand peaks and so data analysis begins with setting a signal to noise threshold (in this study: 5) and generating corresponding peak lists. To sort through thousands of peaks, each m/z ratio or peak location is assigned a new Kendrick Mass, which rescales each peak from the IUPAC atomic mass of $^{12}\text{C} = 12.00000 \text{ Dalton (Da)}$ to the Kendrick mass scale where $\text{CH}_2 = 14.00000 \text{ Da}$ (3-5). The deviation of the Kendrick mass (KM) from the

nominal mass (NM) is referred to as the Kendrick mass defect (KMD). All members of a homologous series of molecules differing only in the number of CH₂ groups can be identified because they will all share the same KMD value.

References:

(1) Gonsior, M.; Peake, B. M.; Jaffé, R.; Cooper, W. J.; Kahn, A.; Young, H.; Kowalczyk, P. Spectral characterization of chromophoric dissolved organic matter (CDOM) in a fjord (Doubtful Sound, New Zealand). *Aqua. Sci.* **2008**, xxx, xxx.

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(5) Kujawinski, E. B.; Hatcher, P. G.; Freitas, M. A. High-Resolution Fourier Transform Ion Cyclotron Resonance Mass Spectrometry of Humic and Fulvic Acids: Improvements and Comparisons. *Anal. Chem.* **2002**, 74, 413-419.

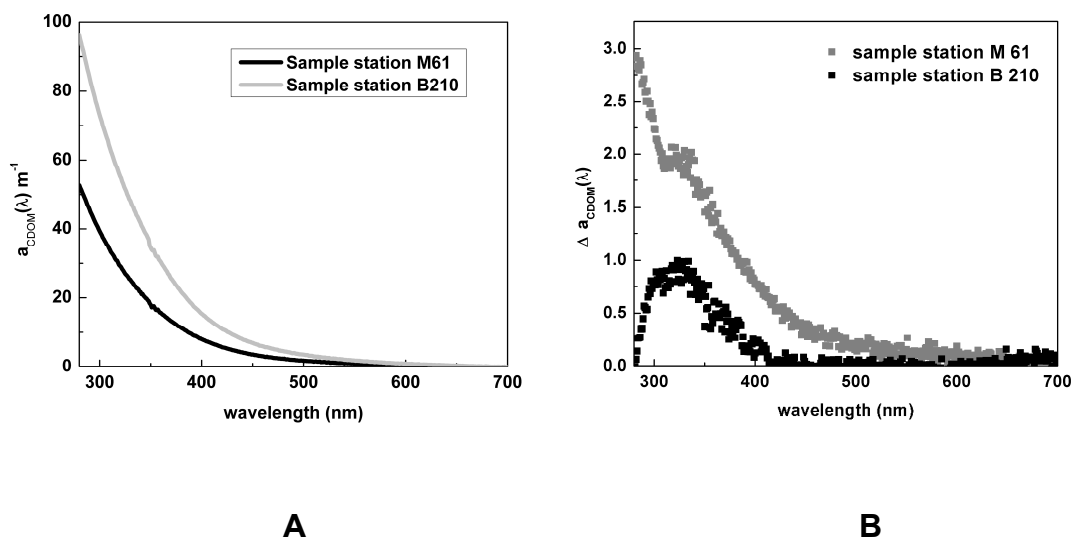


Figure S1. $a_{\text{CDOM}}(\lambda)$ values before irradiation (A) and difference plot of the absolute changes in $a_{\text{CDOM}}(\lambda)$ after simulated sunlight exposure (B) in sample M61 and B210.

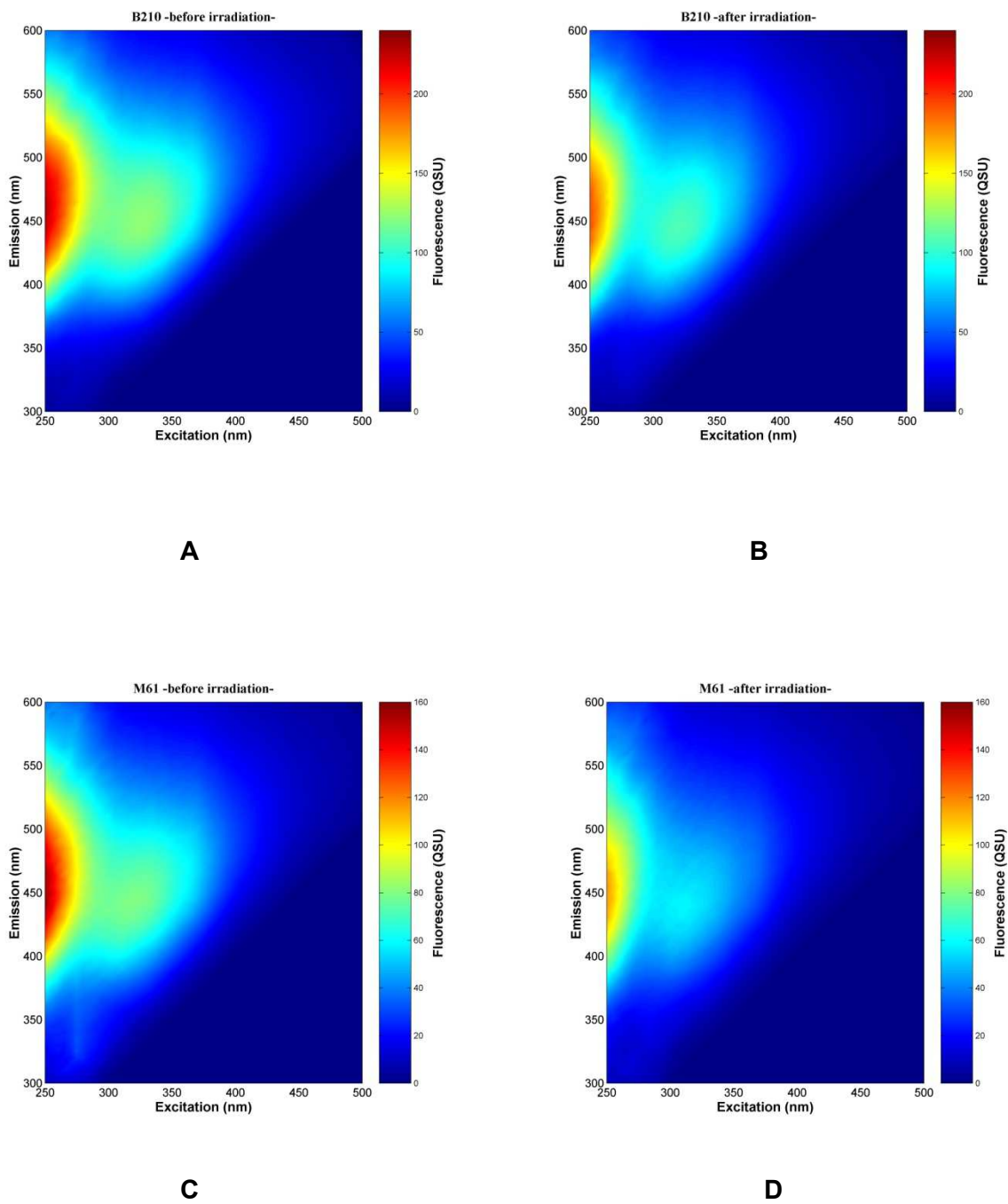


Figure S2: Excitation emission matrix (EEM) fluorescence spectra of sample B210 before (A) and after (B) irradiation and sample M61 before (C) and after (D) irradiation.

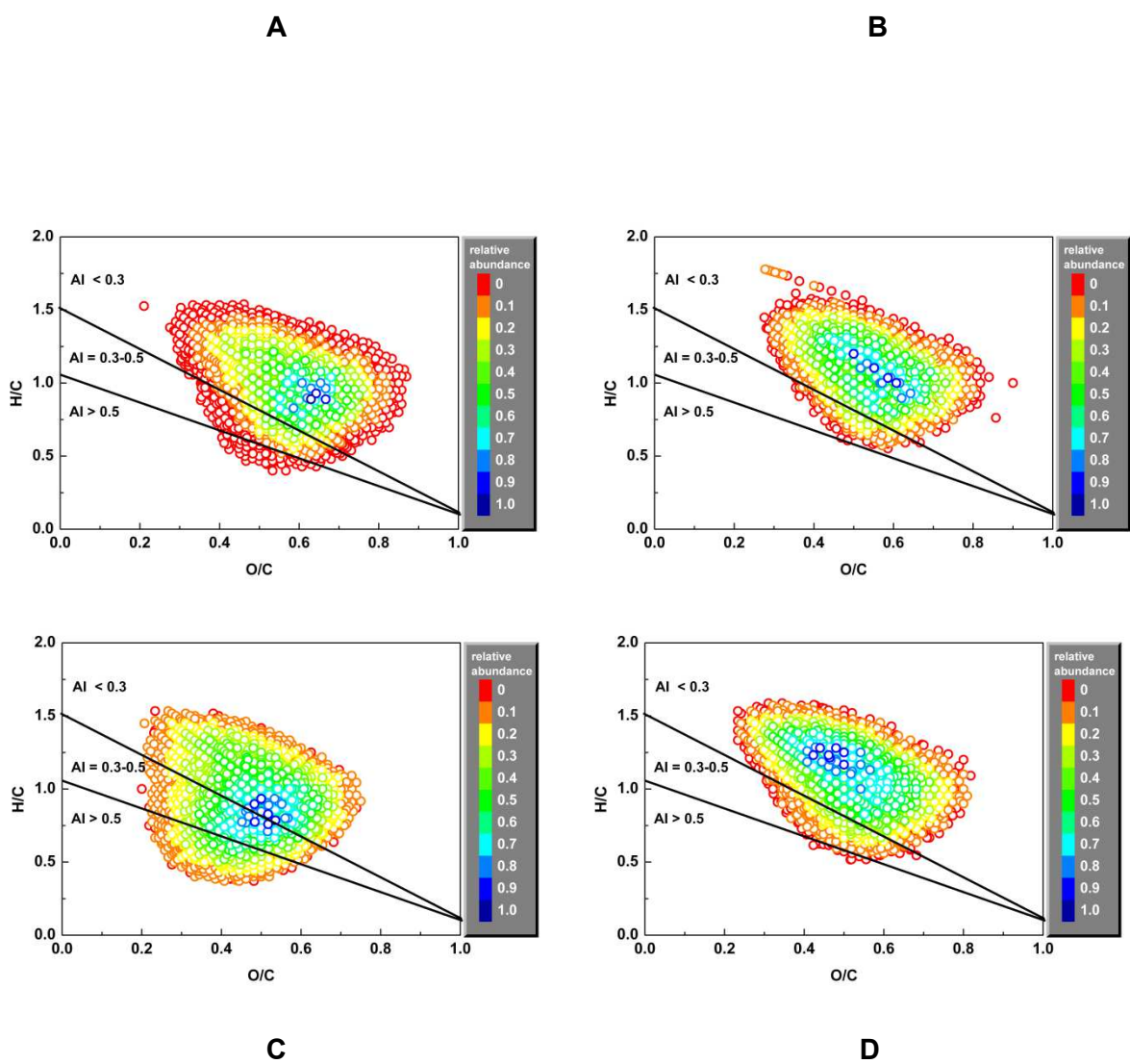


Figure S3. Van Krevelen plots before (A) and after (B) irradiation of sample B210 and before (C) and after (D) irradiation of sample M61 (lines separate areas with aromaticity indices (AI) of AI > 0.5; AI = 0.3-0.5 and AI < 0.3).

Table S1: One hundred most abundant photo-degraded compounds in sample M61

100 most abundant, completely photo-degraded compounds in sample M61					
Formulas sorted from highest to lowest relative abundance	carbon	hydrogen	oxygen	DBE-O	molecular mass (Da)
1	27	18	11	8	518.08492
2	28	20	11	8	532.10056
3	27	20	10	8	504.10565
4	27	16	12	8	532.06418
5	30	20	12	9	572.09548
6	28	18	12	8	546.07983
7	29	20	11	9	544.10056
8	29	22	11	8	546.11622
9	32	22	12	10	598.11113
10	26	18	11	7	506.08492
11	26	18	10	8	490.0900
12	28	22	10	8	518.1213
13	28	16	12	9	544.06418
14	31	18	13	10	598.07475
15	29	18	11	10	542.08492
16	25	18	11	6	494.08492
17	28	18	11	9	530.08492
18	31	20	13	9	600.09039
19	24	16	10	7	464.07435
20	29	24	10	8	532.13695
21	30	18	12	10	570.07983
22	32	20	13	10	612.09039
23	27	18	10	9	502.0900
24	29	18	12	9	558.0798
25	28	18	10	10	514.0900
26	25	18	10	7	478.0900
27	26	16	10	9	488.07435
28	24	18	9	7	450.09508
29	32	24	12	9	600.12678
30	33	22	14	9	642.10096
31	33	20	15	9	656.08023
32	25	16	10	8	476.07435
33	29	16	13	9	572.05909
34	27	16	10	10	500.07435
35	25	14	11	8	490.05361
36	25	18	9	8	462.09508
37	28	16	11	10	528.06926
38	31	22	11	10	570.11622
39	26	14	12	8	518.04853
40	23	16	10	6	452.07435

41	27	22	10	7	506.1213
42	31	24	11	9	572.13186
43	34	28	12	9	628.15808
44	31	22	12	9	586.11113
45	33	24	12	10	612.12678
46	30	20	11	10	556.10056
47	29	22	10	9	530.1213
48	28	20	10	9	516.10565
49	30	22	10	10	542.1213
50	23	14	10	7	450.0587
51	27	16	11	9	516.06926
52	22	16	10	5	440.07435
53	32	20	12	11	596.09548
54	30	18	13	9	586.07475
55	23	14	11	6	466.05361
56	29	18	10	11	526.0900
57	29	16	12	10	556.06418
58	29	20	10	10	528.10565
59	27	14	11	10	514.05362
60	32	18	13	11	610.07475
61	33	22	12	11	610.11113
62	25	20	9	7	464.11073
63	31	20	12	10	584.09548
64	31	20	11	11	568.10056
65	25	16	9	9	460.07943
66	27	22	9	8	490.12639
67	34	26	13	9	642.13734
68	30	22	11	9	558.11622
69	30	26	10	8	546.1526
70	33	26	11	10	598.14752
71	31	16	15	9	628.04893
72	25	14	12	7	506.04853
73	25	22	9	6	466.12639
74	22	14	11	5	454.05361
75	30	24	10	9	544.13695
76	28	24	9	8	504.14204
77	30	18	14	8	602.06966
78	26	14	11	9	502.05361
79	28	16	13	8	560.05909
80	35	28	13	9	656.15300
81	24	14	10	8	462.05870
82	27	24	9	7	492.14204
83	27	20	9	9	488.11073
84	34	20	13	12	636.09039
85	26	20	9	8	476.11073
86	31	16	13	11	596.05909
87	33	26	12	9	614.14243

88	28	14	12	10	542.04853
89	27	14	12	9	530.04853
90	34	30	12	8	630.17373
91	28	14	11	11	526.05362
92	24	14	11	7	478.05361
93	30	16	12	11	568.06418
94	26	14	13	7	534.04345
95	28	20	9	10	500.11073
96	30	16	14	9	600.05401
97	31	18	14	9	614.06966
98	28	22	9	9	502.12639
99	24	14	12	6	494.04853
100	26	22	9	7	478.12639

Table S2: One hundred most abundant photo-produced compounds in sample M61

100 most abundant, newly formed, photo-produced compounds in sample M61					
Formulas sorted from highest to lowest relative abundance	carbon	hydrogen	oxygen	DBE-O	molecular mass (Da)
1	23	30	14	-5	530.16356
2	22	26	15	-5	530.12717
3	24	32	14	-5	544.17921
4	22	28	14	-5	516.14791
5	27	28	19	-5	656.12248
6	33	28	18	2	712.12757
7	25	36	12	-4	528.22068
8	27	32	18	-6	644.15887
9	26	36	14	-5	572.21051
10	20	26	12	-4	458.14243
11	27	38	14	-5	586.22616
12	25	34	14	-5	558.19486
13	21	24	15	-5	516.11153
14	20	20	15	-4	500.08022
15	21	26	14	-5	502.13226
16	26	30	18	-6	630.14322
17	27	34	17	-6	630.17961
18	28	34	16	-4	626.18469
19	28	36	17	-6	644.19525
20	21	20	16	-4	528.07514
21	24	34	13	-5	530.19994
22	25	24	19	-5	628.09119
23	19	20	14	-4	472.08531
24	26	26	19	-5	642.10684
25	28	40	14	-5	600.24181
26	19	22	13	-4	458.10604
27	19	24	12	-4	444.12678
28	27	40	12	-4	556.25198
29	31	44	15	-5	656.26802
30	23	32	13	-5	516.1843
31	21	22	16	-5	530.09079
32	32	44	15	-4	668.26802
33	29	42	14	-5	614.25746
34	30	38	13	-1	606.23124
35	20	24	14	-5	488.11661
36	28	32	19	-6	672.15378
37	28	38	16	-6	630.21599
38	24	30	16	-6	574.15339
39	21	28	13	-5	488.15299
40	28	42	12	-4	570.26763

41	26	28	19	-6	644.12248
42	21	18	16	-3	526.05949
43	29	38	15	-4	626.22108
44	30	42	15	-5	642.25238
45	24	20	18	-3	596.06497
46	20	22	15	-5	502.09587
47	31	44	11	-1	592.28837
48	20	18	15	-3	498.06457
49	36	48	16	-3	736.29424
50	23	26	17	-6	574.117
51	22	28	15	-6	532.14283
52	31	42	15	-4	654.25238
53	30	44	14	-5	628.27311
54	23	30	15	-6	546.15847
55	33	46	15	-4	682.28368
56	25	30	17	-6	602.14831
57	22	30	13	-5	502.16865
58	25	36	13	-5	544.2156
59	30	46	10	-2	566.3091
60	29	40	16	-6	644.23164
61	23	34	10	-3	470.2152
62	28	42	10	-2	538.2778
63	19	18	14	-3	470.06966
64	27	40	13	-5	572.24689
65	24	26	18	-6	602.11192
66	29	36	18	-6	672.19017
67	22	26	16	-6	546.12209
68	31	28	18	0	688.12757
69	24	32	15	-6	560.17413
70	23	34	11	-4	486.21011
71	26	34	16	-6	602.18469
72	23	28	16	-6	560.13774
73	24	24	18	-5	600.09627
74	21	24	16	-6	532.10644
75	35	48	15	-3	708.29932
76	22	18	16	-2	538.05949
77	34	42	15	-1	690.25238
78	25	34	15	-6	574.18978
79	25	26	19	-6	630.10684
80	21	26	15	-6	518.12717
81	29	44	10	-2	552.29345
82	22	22	17	-5	558.0857
83	26	38	13	-5	558.23124
84	29	44	12	-4	584.28328
85	35	48	16	-4	724.29424

86	20	28	11	-4	444.16316
87	33	40	18	-4	724.22147
88	21	28	14	-6	504.14791
89	31	46	11	-2	594.30402
90	18	22	11	-3	414.11622
91	22	32	11	-4	472.19446
92	33	38	20	-5	754.19565
93	22	20	17	-4	556.07006
94	20	18	16	-4	514.05949
95	31	28	20	-2	720.1174
96	20	24	15	-6	504.11152
97	22	30	14	-6	518.16356
98	18	20	13	-4	444.0904
99	26	20	19	-2	636.05989
100	25	38	11	-4	514.24141