

SUPPORTING INFORMATION

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Innovative cavitand-based sol-gel coatings for the environmental monitoring of benzene and chlorobenzenes via solid-phase microextraction

Analytical and spectral characterizations of the developed coating like ¹H-NMR, ESI, ²⁹Si and ¹³C solid state NMR spectra and scanning electron micrographies are provided in this section in order to better support the characterization of the developed materials. A detailed description of the GC-MS experimental conditions used in this work are provided in Table S-1.

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Table S-1. GC-MS analysis	
BTEX	
GC oven temperature program	40°C for 5 min, 10°C min ⁻¹ to 170°C, 170°C for 2 min.
Time scheduled selected-ion monitoring ^a	<p>Delay time: 1 min</p> <p>hexane: <i>m/z</i> 43, 57 and 86 from 1.0 to 4.3 min; heptane and cyclohexane: <i>m/z</i> 41, 56, 71, 84 and 100 from 4.3 to 5.1 min; octane: <i>m/z</i> 43, 85 and 114 from 5.1 to 6.0 min; nonane: <i>m/z</i> 43, 57 and 99 from 6.0 to 7.1 min; benzene: <i>m/z</i> 77 and 78 from 7.1 to 9.0 min; toluene: <i>m/z</i> 91 and 92 for from 9.0 to 11.50 min; ethylbenzene, <i>m</i>-, <i>p</i>-xylene and <i>p</i>-xylene-d₁₀: <i>m/z</i> 91, 98, 105, 106 and 116 from 11.50 to 15.0 min <i>o</i>-xylene: <i>m/z</i> 91, 105 and 106 from 15.0 to 20.0 min.</p> <p>Dwell time: 100 ms</p>
Chlorobenzenes	
GC oven temperature program	65°C, 10°C min ⁻¹ to 200°C
Time scheduled selected-ion monitoring ^a	<p>Delay time: 1min</p> <p>benzene and toluene: <i>m/z</i> 77, 78, 91 and 92 from 1 to 2.5 min; chlorobenzene: <i>m/z</i> 77, 112 and 114 from 2.5 to 3.1 min; 2-chlorotoluene: <i>m/z</i> 91 and 126 from 3.1 to 4.3 min; 1,2-dichlorobenzene: <i>m/z</i> 111, 146 and 148 from 4.3 to 5.4 min; for 1,2,4-trichlorobenzene: <i>m/z</i> 145, 180 and 182 from 5.4 to 13.5 min.</p> <p>Dwell time: 100 ms</p>

^aFor all the analyses, the corresponding ion ratios were used to confirm the identification of the analytes.

Table S-2. $^1\text{H-NMR}$ and ESI-MS characterization of Cavitands 1-3		
	$^1\text{H-NMR}$ (300 MHz, CDCl_3):	ESI-MS
Cavitand 1	$\delta = 8.16$ (s, 4H, ArH), 7.78 (m, 8H, AA' part of an AA'BB' system), 7.47 (m, 8H, BB' part of an AA'BB' system), 7.21 (s, 4H, ArH), 5.83 (m, 4H, CH=CH ₂) 5.57 (t, 4H, CHAr ₂ , J=7.7 Hz), 4.98 (m, 8H, CH=CH ₂), 2.27 (m, 8H, Ar ₂ CHCH ₂), 1.34 (m, 48H, CH ₂)	m/z (%): 1547 [M+H ⁺ , 100].
Cavitand 2 and Cavitand 3	$\delta = 8.12$ (s, 4H, ArH), 7.77 (m, 8H, AA' part of an AA'BB' system), 7.45 (m, 8H, BB' part of an AA'BB' system), 7.18 (s, 4H, ArH), 5.53 (t, 4H, ArCH, J=7.7 Hz), 5.41 (bm, 2H, CH=CH ₂), 3.80 (m, 18H, Si-O-CH ₂), 2.24 (m, 8H, -CH ₂ CHAr ₂), 1.99 (m, 8H), 1.40 (m, 52H, CH ₂), 1.22 (t, 27H, Si-O-CH ₂ CH ₃ + 3H, CH ₃)	m/z (%): (3) 1875 [M+H ⁺ , 50], 1898 [M+Na ⁺ , 30], (2) 2040 [M+H ⁺ , 85], 2062 [M+Na ⁺ , 50]

Table S-3. LOD values obtained with the QxCav fiber in soil

	LOD (ngKg⁻¹)*
	QxCav
Benzene	13.93
Toluene	8.68
Chlorobenzene	2.34
2-Chlorotoluene	2.72
1,2-Dichlorobenzene	0.95
1,2,4-Trichlorobenzene	0.53

*Incubation temperature: 45°C, incubation time: 5 min, stirring: 500 rpm.
Extraction temperature: 45°C, extraction time: 20 min, stirring: 250 rpm,
soil sample: 3g in 12 ml of water

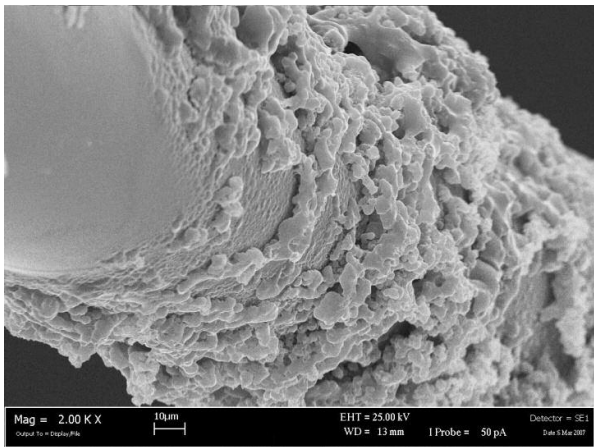


Figure S-1. Scanning electron micrography of QxCav-based coating using acetone as solvent

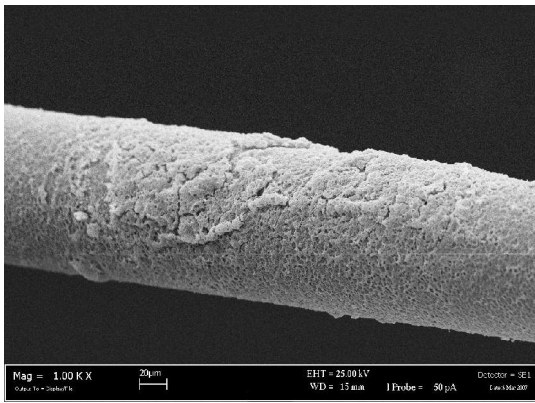


Figure S-2. Scanning electron micrography of QxCav-based coating using CH_2Cl_2 as solvent

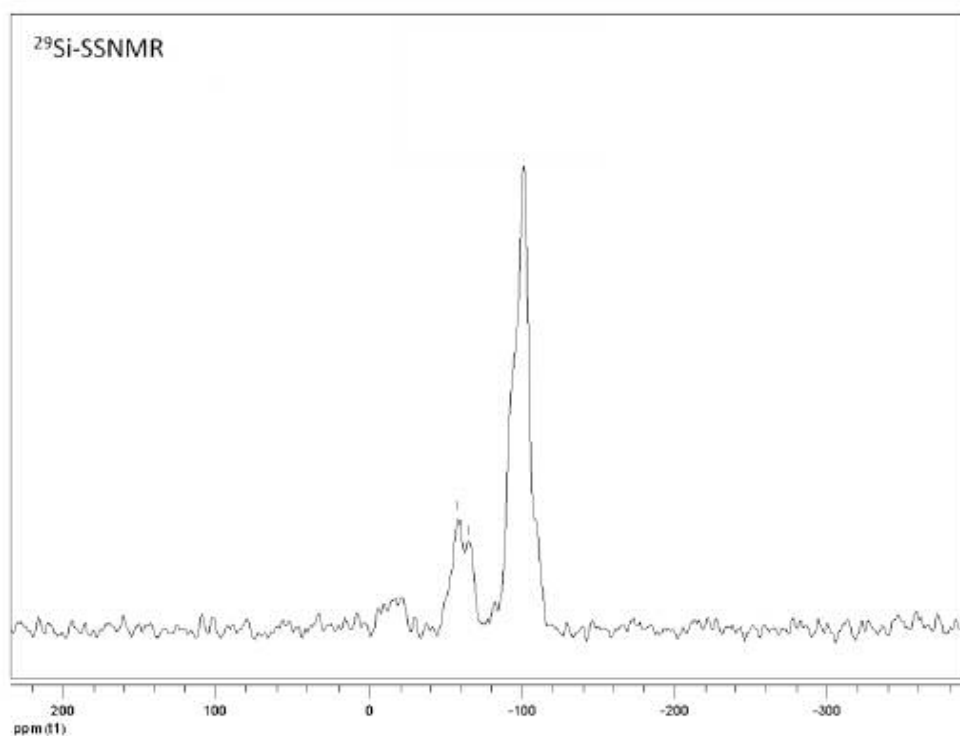


Figure S-3. ^{29}Si solid state NMR of the gel without plasticizer

The ^{29}Si spectrum in cross polarization conditions of the dried gel is constituted by three partially superimposed peaks: Q^2 (≈ -92 ppm), assigned to silicon atoms bonded to two hydroxyl groups, Q^3 (≈ -101 ppm), assigned to silicon atoms with one hydroxyl group and Q^4 (≈ -111 ppm), assigned to silicon atoms without hydroxyl groups. Additional signals appeared in the region between -70 and -55 ppm (T^2 and T^3) belonging to silicon atoms bonded to the organic receptor.

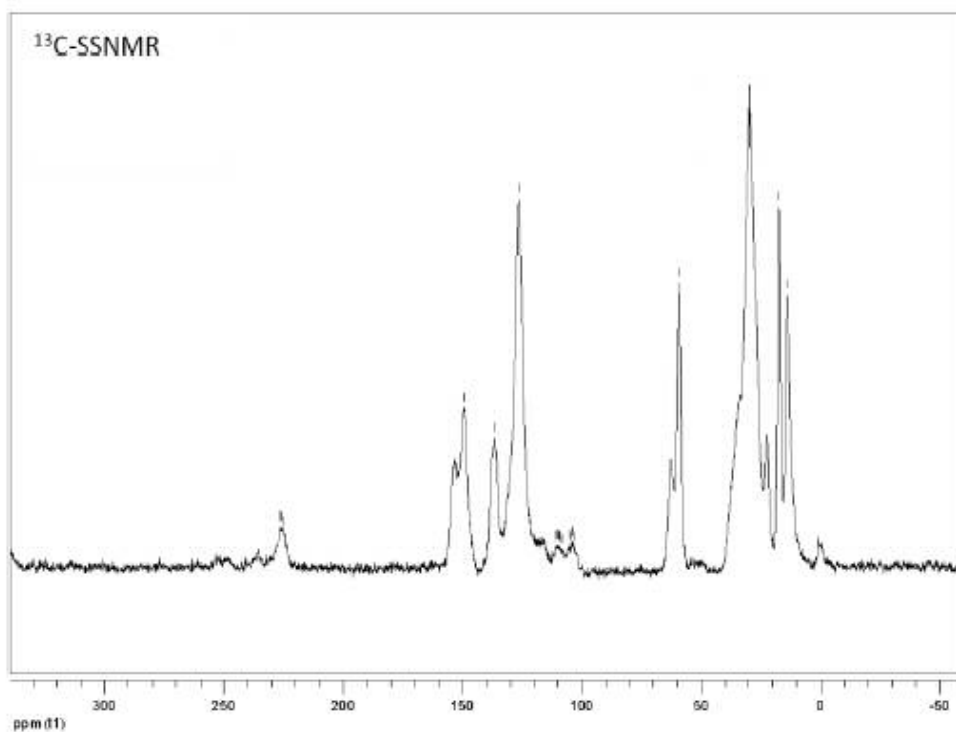


Figure S-4. ^{13}C solid state NMR of the gel without plasticizer

^{13}C -SSNMR: δ = cluster at ~150 ppm (resorcinarene aromatic carbons); multiplet centered at ~ 137 ppm (quinoxaline aromatic carbons); multiplet centered at ~ 125 ppm (resorcinarene aromatic carbons); multiplet centered at ~ 60 ppm (Si-O-CH₂); multiplet at 20-40 ppm (lower rim aliphatic chains carbons); 17.4 ppm (Si-O-CH₂-CH₃), 12.7 ppm (terminal CH₃ of alkyl chains).