Supporting Information for:

Development, Mechanism, and Scope of the Palladium-Catalyzed Enantioselective Allene Diboration

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A. General.

¹H NMR spectra were recorded on Bruker DRX 400 and 500 MHz spectrometer as well as a Varian Unity Inova 500 spectrometer. Chemical shifts are reported in ppm with the solvent resonance as the internal standard (CDCl₃: 7.26 ppm). Data are reported as follows: chemical shift, integration, multiplicity (s = singlet, d = doublet, t = triplet, q = quartet, br = broad, m = multiplet), coupling constants (Hz) and assignment. ¹³C{¹H}NMR spectra were recorded on a Bruker DRX 400 (100 MHz), 500 (125 MHz) MHz, or Varian Gemini-500 MHz (125 MHz) spectrometers with chemical shifts are reported in ppm with the solvent resonance as the internal standard (CDCl₃: 77.16 ppm). ³¹P{¹H}NMR (161 MHz or 202 MHz) were recorded on a Bruker DRX 400 or Varian Unity Inova 500 spectrometer. Chemical shifts are reported for ³¹P NMR spectra using phosphoric acid as an external standard. ¹⁹F NMR (376 MHz) and ²H spectra (61 MHz) were recorded on a Bruker DRX 400 MHz spectrometer. Low-resolution mass spectrometry was performed at The University of North Carolina at Chapel Hill Mass Spectrometry Facility. High-resolution CI-GC/MS, EI, and MALDI-TOF were recorded at Indiana University, Bloomington, IN.

Liquid Chromatography was performed using forced flow (flash chromatography) on silica gel (SiO_2 , 230 x 450 Mesh) purchased from Silicycle. Thin Layer Chromatography was performed on 25 μ m silica gel glass backed plates from EMD Chemicals, Inc. Visualization was performed using, ultraviolet light, phosphomolybdic acid (PMA), and potassium permanganate (KMNO₄).

Analytical gas-liquid chromatography (GLC) was performed on a Hewlett-Packard 6890 Series chromatography equipped with a CTC Analysis Combi Pal autosampler by Leap Technologies (Carrboro, North Carolina), a split mode capillary injection system, a flame ionization detector and a Supleco β -Dex 120 column with helium as a carrier gas.

Tris(dibenzylideneacetone)dipalladium(0) (Pd_2dba_3), mercury(II) chloride ($HgCl_2$), and chlorodicarbonylrhodium(I) dimer were purchased from Strem Chemicals, Inc. Bis(pinacoloto)diboron ($B_2(pin)_2$) was obtained from Allychem Co., Ltd. and recrystallized from pentanes prior to use. Toluene- d_8 and acetone- d_6 were purchased from Cambridge Isotope Laboratories and dried and degassed prior to use. Methyl allene (1,2-butadiene) was purchased from Chem. Samp. Co. and used as a solution in toluene. All reactions were conducted in oven or flame-dried glassware under an inert atmosphere of nitrogen or argon. Toluene was distilled over calcium hydride and degassed by freeze-pump cycles prior to use. Tetrahydrofuran was distilled from sodium-benzophenone. Triethylamine was distilled from calcium hydride. Allenes were synthesized according to literature procedures. Phosphoramidite ligands were prepared as described in the general procedure below. All other reagents were purchased from Aldrich and used without further purification.

B. Diboration Procedures.

Representative Procedure for Diboration of Allenes. A 20 mL vial with magnetic stir bar was charged with Pd_2dba_3 (7.62 mg, 0.0083 mmol), (R,R)-xylylTADDOL-PNMe₂ ((R,R)-6) (13.0 mg, 0.0199 mmol), and toluene (2.2 mL) in an inert atmosphere dry box. After stirring for 1 hour, $B_2(pin)_2$ (101.3 mg, 0.399 mmol) was added to the mixture followed by tridec-1,2-diene (60 mg, 0.3327 mmol). The vial was sealed with a polypropylene cap, removed from dry box, and stirred at room temperature for 12 hours. The resulting solution was concentrated via rotary evaporation and chromatographed on

¹ Myers, A.G.; Zheng, B. *J. Am. Chem. Soc.* **1996**, *118*, 4492. For aromatic allenes: Huang, C.- W.; Shanmugasundaram, M.; Chang, H.- M.; Cheng, C.- H. *Tetrahedron*, **2003**, *59*, 3635. For OTBS protected allene: Trost, B.M.; Pinkerton, A.B.; Seidel, M. *J. Am. Chem. Soc.* **2001**, *123*, 12466.

silica gel (2% ethyl acetate/hexanes) to afford the 1,2-bis(boronate)ester product in 68% yield (99.7 mg).

Modified Procedure for Diboration of 1,2-Butadiene. Tris(dibenzylideneacetone)dipalladium(0) (16.3 mg, 0.0178 mmol) and (R,R)-6 (27.7 mg, 0.0426 mmol) were complexed for 1 h in toluene (4.23 mL). The reaction was charged with B₂(pin)₂ (216.3 mg, 0.852 mmol) and 0.5 mL of a 1.42 M solution of 1,2-butadiene in toluene (as determined by 1 H NMR). The vial was sealed with a polypropylene cap, removed from dry box, and stirred at room temperature for 14 hours. The resulting solution was concentrated via rotary evaporation and chromatographed (4% ethyl acetate/hexanes) on silica gel to afford 169.2 mg (77% yield) of the methyl allene diboration product.

Representative Procedure for Ligand Screening - Conversion by 1 H NMR. In the dry box, a 6 dram vial with magnetic stir bar was charged with Pd_2dba_3 (3.6 mg, 3.9 μ mol) and (R,R)-TADDOLPNMe₂ (R,R)-1 (5.3 mg, 9.9 μ mol) in toluene- d_8 (1.10 mL, 0.15 M). The metal and ligand were complexed for 1 hour at which time $B_2(pin)_2$ (50.6 mg, 0.199 mmol) was added following by tridec-1,2-diene (30 mg, 0.166 mmol). The reaction was allowed to stir for 2 minutes and then transferred to an oven dried NMR tube. The NMR tube was sealed with a cap and removed from the glove box. Single pulse 1 H NMR was used to determine the extent of reaction at 20 minutes.

C. Phosphoramidite Ligand Synthesis.

General Procedure for (*R*,*R*)-TADDOL Phosphoramidite Ligand Synthesis. To an oven-dried flask with magnetic stir bar under nitrogen, was added flame-dried 4Å molecular sieves, the respective TADDOL derivative (1 equiv), and tetrahydrofuran (0.25 M). The flask was cooled to 0 °C (ice-water bath) and charged with triethylamine (3.4 equiv) and phosphorus trichloride (1.2 equiv). The mixture was warmed to ambient temperature and stirred for 30 minutes when it was cooled to 0 °C (ice-water bath) and charged with dimethylamine in tetrahydrofuran (10 equiv) or the respective amine (5 equiv, neat). The reaction was stirred overnight at ambient temperature, was diluted with diethyl ether, and filtered over celite. The solvent was removed via rotary evaporator and the crude mixture was chromatograhed on silica gel.

Procedure for the Preparation of 3,5-(CF₃)₂TADDOLPNMe₂. To a 100 mL round-bottom flask equipped with an oven-dried magnetic stir bar was added 4Å molecular sieves, followed by 3,5-(CF₃)₂C₆H₃TADDOL² (2.43 g, 2.40 mmol, 1 equiv) and tetrahydrofuran (9.6 mL, 0.25 M). The flask was cooled to 0 °C (ice-water bath) and charged with triethylamine (1.13 mL, 8.16 mmol) and phosphorus trichloride (251 μL, 2.88 mmol). The mixture was allowed to warm to ambient temperature and stirred for 30 minutes, at which time it was cooled to 0 °C (ice-water bath) and charged with dimethylamine (12 mL, 2.0 M in tetrahydrofuran). The mixture warmed to ambient temperature, stirred overnight, then diluted with diethyl ether and filtered over celite. The solvent was removed via rotary evaporator and the crude reaction mixture was chromatographed on silica gel (10% dichloromethane/hexanes) then recrystallized from dichloromethane to afford the desired phosphoramidite **8** (597 mg, 22% yield).

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² 3,5-(CF₃)₂C₆H₃TADDOL was prepared from L-tartaric acid and 1,3-bis(trifluoromethyl)-5-bromobenzene. See: Seebach, D.; Beck, A.K.; Keckel, A. *Angew. Chem. Int. Ed.* **2001**, *40*, 92.

(3aR,8aR)-4,4,8,8-tetrakis(3,5-bis(trifluoromethyl)phenyl)-N,N,2,2-tetramethyl-tetrahydro-[1,3]dioxolo94,5-e][1,3,2]dioxaphosphenpin-6-amine.

¹H NMR (500 MHz, CDCl₃) δ 0.35 (3H, s, CH₃CCH₃), 1.51 (3H, s, CH₃CCH₃), 2.85 (6H, d, J = 11.0 Hz, (CH₃)₂N), 4.33 (1H, d, J = 8.5 Hz, OCHC), 4.94 (1H, dd, J = 9.0, 3.5 Hz, OCHC), 7.83-7.89 (8H, m, Ar-H), 8.03 (2H, s, Ar-H), 8.30 (2H, s, Ar-H).

¹³C NMR (125 MHz, CDCl₃) δ 25.4, 27.2, 35.2 (2C, d, ${}^2J_{CP} = 20.2$ Hz), 79.7, 80.2 (1C, d, ${}^2J_{CP} = 8.2$ Hz), 81.9 (1C, d, ${}^3J_{CP} = 22.2$ Hz), 82.9 (1C, d, ${}^3J_{CP} = 3.1$ Hz), 113.4, 122.5 (1C, q, ${}^3J_{CF} = 3.5$ Hz), 122.6 (1C, q, ${}^3J_{CF} = 3.6$ Hz), 122.9 (1C, q, ${}^3J_{CF} = 3.7$ Hz, 123.0 (1C, q, ${}^3J_{CF} = 3.5$ Hz), 123.24 (4C, q, 2.5 (2C, q, ${}^1J_{CF} = 272.2$ Hz), 126.8 (2C), 127.0

 1 G_{F3} F₃C (1C, d, 3 G_{CF} = 3.1 Hz), 113.4, 122.5 (1C, d, 3 G_{CF} = 3.5 Hz), 122.6 (1C, d, 3 G_{CF} = 3.6 Hz), 122.9 (1C, q, 3 J_{CF} = 3.7 Hz, 123.0 (1C, q, 3 J_{CF} = 3.5 Hz), 123.24 (4C, q, 1 J_{CF} = 272.6 Hz), 123.25 (2C, q, 1 J_{CF} = 272.7 Hz), 123.3 (2C, q, 1 J_{CF} = 272.2 Hz), 126.8 (2C), 127.0 (2C), 128.7 (2C), 128.8 (2C), 131.6 (2C, q, 2 J_{CF} = 33.5 Hz), 132.12 (2C, q, 2 J_{CF} = 33.5 Hz), 132.15 (2C, q, 2 J_{CF} = 33.2 Hz), 132.6 (2C, q, 2 J_{CF} = 34.2 Hz), 141.8, 142.7, 146.5, 147.4. ³¹P NMR (202 MHz, CDCl₃) δ 140.7. ¹⁹F (376 MHz, CDCl₃) δ -63.8 (6F, s), -63.7 (6F, s), -63.6 (6F, s), -63.5 (6F, s). Purification Technique: very sensitive to silica gel, will decompose on the column, 10% dichloromethane/hexanes, silica gel, recrystallized from dichloromethane. R_f = 0.51 (10% dichloromethane/hexanes). LRMS – (ESI+): for C₄₁H₂₇F₂₄NO₄P calc'd: 1084.12 (M+H)⁺, observed: 1084.2 (M+H)⁺.

D. Preparation of Rhodium Complexes for analysis by IR:3

Chlorocarbonyl-bis(triphenylphosphine)rhodium and chlorocarbonylbis(tricyclohexylphosphine)-rhodium were prepared according to literature procedure and spectral data and analysis of the complexes by MALDI are in accordance with the literature.

Preparation of Chlorocarbonylbis((*R***,***R***)-1)rhodium:** In the dry box, a 25 mL round-bottom flask was charged with chlorodicarbonylrhodium(I) dimer (80 mg, 0.2057 mmol). To a separate 25-mL pear shaped flask was added (R,R)-TADDOLPNMe₂ (R,R)-1 (444.1 mg, 0.8231 mmol). Both flasks were sealed with septa, removed from the glove box. Chlorodicarbonylrhodium(I) dimer was dissolved in benzene (1 mL) and (R,R)-TADDOLPNMe₂ was dissolved in benzene (5 mL) after mild heating. The solution of (R,R)-TADDOLPNMe₂ was cannulated into the chlorodicarbonylrhodium(I) dimer solution and CO evolution was immediate. The reaction mixture stirred at room temperature for 30 minutes when solvent was removed on the rotary evaporator. Residual solvent was removed overnight in vacuo to afford 473.6 mg (93% yield) of a yellow solid. ¹H NMR (500 MHz, CDCl₃) δ 0.49 (6H, s, (CH₃)₂), $0.58 (6H, s, (CH_3)_2), 2.23 (12H, t, J = 5.2 Hz, 2 x (N(CH_3)_2)), 5.42 (2H, d, J = 8.0 Hz, OCH), 5.49 (2H, d, J = 8.0 Hz, OCH), 5.40 (2H, d, J = 8.0 Hz$ d, J = 8.0 Hz, OCH), 7.16-7.36 (28H, m, Ar-H), 7.45 (4H, t, J = 7.7 Hz, Ar-H), 7.54-7.56 (4H, m, Ar-H) **H**), 7.70-7.72 (4H, m, Ar-**H**). ¹³C NMR (125 MHz, CDCl₃) δ 26.6 (2C), 26.9 (2C), 38.5 (4C, t, J_{CP} = 5.6 Hz), 79.3 (2C), 79.5 (2C), 87.2 (2C), 88.1 (2C), 115.1 (2C), 126.8 (2C), 126.9 (4C), 127.2 (4C), 127.4 (4C), 127.7 (2C), 17.9 (2C), 128.0 (4C), 128.3 (4C), 128.4 (2C), 128.7 (4C), 129.0 (4C), 129.3 (4C). ³¹P NMR (202 MHz, CDCl₃) δ 117.6, 118.6. IR (CH₂Cl₂): 1982 (s), 1607 (s), 1491 (m) cm⁻¹. MALDI-TOF with CAA overlay: calc'd for $C_{66}H_{68}N_2O_8P_2Rh$ 1181.35 (M-CO-Cl)⁺, observed 1181.488 $(M-CO-Cl)^+$.

Preparation of chlorocarbonylbis((R,R)-6)**rhodium:** In the dry box, a 25 mL round-bottom flask was loaded with chlorodicarbonylrhodium(I) dimer (30 mg, 0.0771 mmol). To a separate 25-mL pear shaped flask was added (R,R)-xylylTADDOLPNMe₂ (R,R)-6 (201.1 mg, 0.3086 mmol). Both flasks were sealed with septa and removed from the glove box. Chlorodicarbonylrhodium(I) dimer and (R,R)-xylylTADDOLPNMe₂ were dissolved in benzene (1 mL). The solution of (R,R)-xylylTADDOLPNMe₂

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³ Wilkinson, G.; McClevery, J.A. *Inorganic Syntheses*, **1966**, *8*, 214-217.

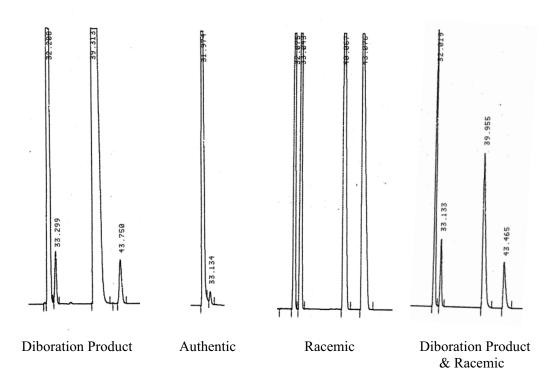
in benzene was cannulated into the chlorodicarbonylrhodium(I) dimer solution and CO evolution was immediate. The reaction mixture was stirred at room temperature for 5 minutes when solvent was removed on the rotary evaporator. Residual solvent was removed overnight *in vacuo* and the crude reaction mixture was recrystallized form pentanes to afford 114 mg (50% yield) of a yellow solid. ¹H NMR (500 MHz, CDCl₃) δ 0.44 (6H, s, (CH₃)₂), 0.54 (6H, s, (CH₃)₂), 2.04 (12H, t, J = 5.2 Hz, 2 x (N(CH₃)₂), 2.21 (24H, s, Ar-CH₃), 2.25 (12H, s, Ar-CH₃), 2.42 (12H, s, Ar-CH₃), 5.26 (2H, d, J = 7.5 Hz, OCH), 5.64 (2H, d, J = 8.0 Hz, OCH), 6.80 (2H, s, Ar-H), 6.82 (2H, s, Ar-H), 6.85 (2H, s, Ar-H), 6.90 (4H, s, Ar-H), 6.94 (2H, s, Ar-H), 7.07 (4H, s, Ar-H), 7.10 (4H, s, Ar-H), 7.18 (4H, s, Ar-H). ¹³C NMR (125 MHz, CDCl₃) δ 21.5 (4C), 21.6 (4C), 21.7 (4C), 21.8 (4C), 26.5 (2C), 27.0 (2C), 38.2 (4C, t, J = 6.3 Hz), 78.2 (2C), 79.7 (2C), 88.4 (2C), 86.7 (2C), 115.1 (2C), 125.3 (2C), 126.9 (4C), 127.2 (6C), 128.5 (2C), 128.6 (2C), 129.3 (4C), 129.7 (4C), 135.5 (4C), 135.9 (4C), 137.0 (4C), 137.1 (4C), 141.7 (2C), 142.3 (2C), 143.0 (2C), 144.9 (2C). ³¹P NMR (202 MHz, CDCl₃) δ 116.5, 117.5. MALDI-TOF with CAA overlay: calc'd for C₈₂H₁₀₀N₂O₈P₂Rh 1405.601 (M-CO-Cl)[†], observed 1405.778 (M-CO-Cl)[†].

E. Characterization of Substrates – Table 2, entries 6,8,9,10,11

 1 . LRMS-(ESI+): for $C_{22}H_{35}B_{2}O_{4}$ calc'd: 385.2 (M+H)⁺, observed 385.2 (M+H)⁺. Purification Technique: silica gel, 2% ethyl acetate/hexanes. $R_{\rm f} = 0.24$ (4% ethyl acetate/hexanes).

Proof of Stereochemistry. Stereochemistry was determined by diimide reduction (2-nitrobenzenesulfonylhydrazide (3 equiv), triethylamine (6 equiv), dioxanes (0.1 M), 80 °C, overnight) of the diboron followed by basic hydrogen peroxide oxidation (3M NaOH, H₂O₂, quench with Na₂S₂O₃). The resultant diol was protected as the acetonide (*p*-toluenesulfonic acid and 2,2-dimethoxypropane). Racemic material was prepared from the diboration of 1-methyl-4-(propa-1,2-dienyl)benzene as described in the general procedure using tricyclohexylphosphine. Absolute stereochemistry was established in comparison to authentic material prepared by Sharpless asymmetric oxidation of 4-methyl-*trans*-β-methylstyrene (Norrby, P-O.; Becker, H.; Sharpless, K.B. *J. Am. Chem. Soc.* **1996**, *118*, 35.). The resultant diol was protected as the acetonide (*p*-toluenesulfonic acid and 2,2-dimethoxypropane).

Chiral GLC (β -dex, Supelco, 115 °C) – analysis of acetonide derived from reduction and oxidation of reaction product as described above

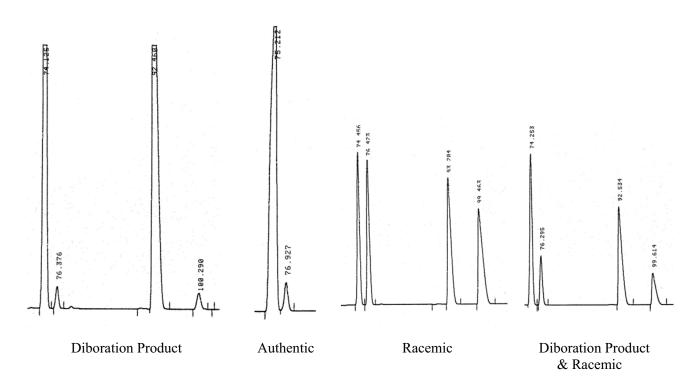


(*S*)-2-(1-(4-methoxyphenyl)-1-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)prop-2-en-2-yl)-4,4,5,5-tetramethyl-1,3,2-dioxaborolane. ¹H NMR (400 MHz, CDCl₃) δ 1.22 (6H, s, OCCH₃), 1.25 (12H, s, OCCH₃), 1.26 (6H, s, OCCH₃), 3.31 (1H, s, ArCHB), 5.27 (1H, s, CBCH), 5.80 (1H, br, q, *J* = 1.2 Hz, CBCH), 6.81 (2H, d, *J* = 8.8 Hz, Ar-H), 7.11 (2H, d, *J* = 8.4 Hz, Ar-H). ¹³C NMR (100 MHz, CDCl₃) δ 24.6, 24.7, 24.9, 25.0, 55.1, 83.4 (2C), 83.5 (2C), 113.7,

127.7, 130.9, 132.5, 157.6. IR (neat): 3069 (m), 2965 (s), 2922 (s), 2720 (w), 2589 (w), 2546 (w), 2421 (w), 2050 (w), 1876 (w), 1658 (w), 1609 (s), 1576 (s) cm $^{-1}$. LRMS-(ESI+): for $C_{22}H_{35}B_2O_5$ calc'd 401.2 (M+H) $^+$, observed: 401.2 (M+H) $^+$. Purification Technique: silica gel, 5% ethyl acetate/hexanes. $R_f = 0.09$ (4% ethyl acetate/hexanes).

Proof of Stereochemistry. Stereochemistry was determined by diimide reduction (2-nitrobenzenesulfonylhydrazide (3 equiv), triethylamine (6 equiv), dioxanes (0.1 M), 80 °C, overnight) of the diboron followed by basic hydrogen peroxide oxidation (3M NaOH, H₂O₂, quench with Na₂S₂O₃). The resultant diol was protected as the acetonide (*p*-toluenesulfonic acid and 2,2-dimethoxypropane). Racemic material was prepared from the diboration of 1-methoxy-4-(propa-1,2-dienyl)benzene as described in the general procedure using tricyclohexylphosphine. Absolute stereochemistry was established in comparison to authentic material prepared by Sharpless asymmetric oxidation of anethole (Norrby, P-O.; Becker, H.; Sharpless, K.B. *J. Am. Chem. Soc.* **1996**, *118*, 35) The resultant diol was protected as the acetonide (*p*-toluenesulfonic acid and 2,2-dimethoxypropane).

Chiral GLC (β -dex, Supelco, 120 °C) – analysis of acetonide derived from reduction and oxidation of reaction product as described above.

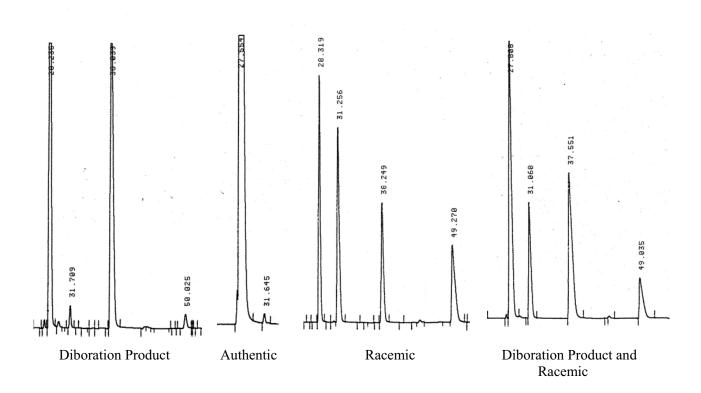


(*S*)-4,4,5,5-tetramethyl-2-(1-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)-1-(4-(trifluoromethyl)phenyl)prop-2-en-yl)-1,3,2-dioxaborolane.
¹H NMR (400 MHz, CDCl₃) δ 1.24 (12H, s, OCCH₃), 1.25 (12H, s, OCCH₃), 3.44 (1H, s, ArCHB), 5.34 (1H, s, CBCH), 5.86 (1H, s, CBCH), 7.33 (2H, d, J = 8.0 Hz, Ar-H), 7.51 (2H, d, J = 7.5 Hz, Ar-H).
¹³C NMR (100 MHz, CDCl₃) δ 24.7 (2C), 24.8 (2C), 24.9 (2C), 25.0 (2C), 83.8 (2C), 83.9 (2C), 124.6 (1C, q, ${}^{1}J_{CF}$ = 271 Hz), 125.1 (2C, q, ${}^{3}J_{CF}$ = 3.7 Hz), 127.8 (1C, q, ${}^{2}J_{CF}$ = 32.2 Hz), 128.9, 130.3 (2C), 145.6.
¹⁹F NMR

 $(376.4 \text{ MHz}, \text{CDCl}_3) \delta$ -62.8 (s). IR (neat): 2984 (s), 2930 (s), 2859 (w), 2354 (w), 1627 (m), 1366 (s), 1323 (s), 1133 (s) cm⁻¹. LRMS-(ESI+): for $C_{22}H_{31}B_2F_3O_4$ calc'd: 439.2 (M+H)⁺, observed: 439.2 (M+H)⁺. Purification Technique: silica gel, 4% ethyl acetate/hexanes. $R_f = 0.23$ (4% ethyl acetate/hexanes).

Proof of Stereochemistry. Stereochemistry was determined by diimide reduction (2-nitrobenzenesulfonylhydrazide (3 equiv), triethylamine (6 equiv), dioxanes (0.1 M), 80 °C, overnight) of the diboron followed by basic hydrogen peroxide oxidation (3M NaOH, H₂O₂, quench with Na₂S₂O₃). The resultant diol was protected as the acetonide (*p*-toluenesulfonic acid and 2,2-dimethoxypropane). Racemic material was prepared from the diboration of 1-(propa-1,2-dienyl)-4-(trifluoromethyl)benzene as described in the general procedure using tricyclohexylphosphine. Absolute stereochemistry was established in comparison to authentic material prepared by Sharpless asymmetric oxidation of 4-trifluoromethyl-*trans*-β-methylstyrene (Norrby, P-O.; Becker, H.; Sharpless, K.B. *J. Am. Chem. Soc.* **1996**, *118*, 35) The resultant diol was protected as the acetonide (*p*-toluenesulfonic acid and 2,2-dimethoxypropane).

Chiral GLC (β -dex, Supelco, 105 °C) – analysis of acetonide derived from reduction and oxidation of reaction product as described above.

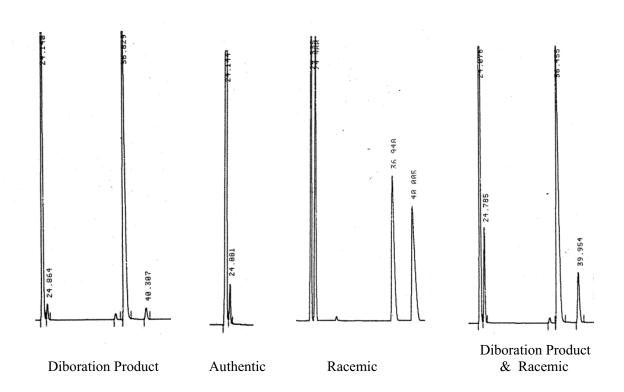


(*S*)-4,4,5,5-tetramethyl-2-(1-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)-1-(3-trifluoromethyl)phenyl)prop-2-en-2-yl)-1,3,2-dioxaborolane. ¹H NMR (400 MHz, CDCl₃) δ 1.24 (6H, s, OCCH₃), 1.24 (6H, s, OCCH₃), 1.25 (6H, s, OCCH₃), 1.26 (6H, s, OCCH₃), 3.42 (1H, s, ArCHB), 5.33 (1H, s, CBCH), 5.85 (1H, s, CBCH), 7.36-7.42 (3H, m, Ar-H), 7.53 (1H, s, ortho-ArH). ¹³C NMR (100 MHz, CDCl₃) δ 24.6 (2C), 24.7 (2C), 24.9 (2C), 25.0 (2C), 83.8 (2C), 83.9 (2C), 122.4 (1C, q, ${}^3J_{CF}$ = 3.8 Hz), 124.6 (1C, q, ${}^1J_{CF}$ = 272.3 Hz), 126.9 (1C, q, ${}^3J_{CF}$ = 3.8 Hz), 128.6, 128.9, 130.4 (1C, q, ${}^2J_{CF}$ = 32.2 Hz), 133.5, 142.2. ¹⁹F NMR (376 MHz, CDCl₃) δ -63.1. IR (neat): 3064 (w), 2978 (s),

2932 (s), 2357 (w), 1603 (m), 1445 (m), 1366 (s), 1320 (s) cm $^{-1}$. LRMS-ESI+: For $C_{22}H_{31}B_2F_3O_4$ calc'd: 439.2 (M+H) $^+$, observed: 439.2 (M+H) $^+$. Purification Technique: silica gel, 4% ethyl acetate/hexanes. $R_f = 0.18$ (4% ethyl acetate/hexanes).

Proof of Stereochemistry. Stereochemistry was determined by diimide reduction (2-nitrobenzenesulfonylhydrazide (3 equiv), triethylamine (6 equiv), dioxanes (0.1 M), 80 °C, overnight) of the diboron followed by basic hydrogen peroxide oxidation (3M NaOH, H₂O₂, quench with Na₂S₂O₃). The resultant diol was protected as the acetonide (*p*-toluenesulfonic acid and 2,2-dimethoxypropane). Racemic material was prepared from the diboration of 1-(propa-1,2-dienyl)-3-(trifluoromethyl)benzene as described in the general procedure using tricyclohexylphosphine. Absolute stereochemistry was established in comparison to authentic material prepared by Sharpless asymmetric oxidation of 3-trifluoromethyl-*trans*-β-methylstyrene (Norrby, P-O.; Becker, H.; Sharpless, K.B. *J. Am. Chem. Soc.* **1996**, *118*, 35) The resultant diol was protected as the acetonide (*p*-toluenesulfonic acid and 2,2-dimethoxypropane).

Chiral GLC (β -dex, Supelco, 100 °C) – analysis of acetonide derived from reduction and oxidation of reaction product as described above

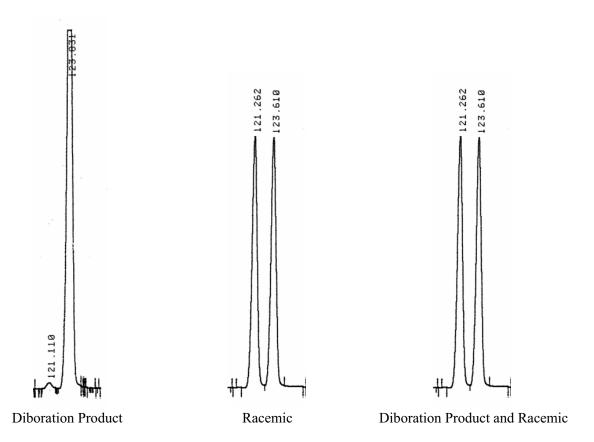


(*R*)-(3,4-bis(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)pent-4-enyloxy)(tert-butyl)dimethylsilane. 1 H NMR (400 MHz, CDCl₃) δ 0.003 (6H, s, (CH₃)₂Si), 0.89 (9H, s, (CH₃)₃CSi), 1.19 (6H, s, OCCH₃), 1.20 (6H, s, OCCH₃), 1.22 (12H, s, OCCH₃), 1.74 (1H, m, CH₂CH_AH_BCHB), 1.85 (1H, m, CH₂CH_AH_BCHB), 1.98 (1H, t, J = 7.4 Hz, CH₂CHBCB), 3.55 (2H, m, OCH₂CH₂), 5.55 (1H, d, J = 2.4

Hz, CHCB), 5.75 (1H, d, J = 3.2 Hz, CHCB). ¹³C NMR (100 MHz, CDCl₃) δ -5.1, -5.07, 18.5, 24.7 (2C), 24.8 (2C), 24.9 (2C), 25.0 (2C), 26.2 (3C), 32.7, 63.0, 83.1 (2C), 83.4 (2C), 128.0. IR (neat): 3055 (w), 2973 (s), 2848 (s), 2734 (w), 2289 (w), 1610 (s), 1475 (s), 1361 (s), 1274 (s), 1241 (s) cm⁻¹. LRMS-(ESI+): for $C_{23}H_{46}B_2NaO_5Si$ calc'd 475.3 (M+Na)⁺, observed 475.3 (M+Na)⁺. Purification Technique: silica gel, 5% ethyl acetate/hexanes. R_f = 0.33 (5% ethyl acetate/hexanes).

Proof of Stereochemistry. Stereochemistry was determined by diimide reduction (2-nitrobenzenesulfonylhydrazide (3 equiv), triethylamine (6 equiv), dioxanes (0.1 M), 80 °C, overnight) of the diboron followed by basic hydrogen peroxide oxidation (3M NaOH, H₂O₂, quench with Na₂S₂O₃). The resultant diol was protected as the diacetate (acetic anhydride (1.2 equiv), triethylamine (3 equiv), and 4-dimethylaminopyridine (cat) in dichloromthane (0.25 M). Racemic material was prepared from the diboration of *tert*-butyldimethyl(penta-3,4-dienyloxy)silane as described in the general procedure using tricyclohexylphosphine.

Chiral GLC (β -dex, Supelco, 100 °C for 25 minutes then 0.5 deg/min to 120 °C) – analysis of acetate derived from reduction and oxidation of reaction product as described above



F. Preparation of pinacol- d_{12} and $B_2(pin-d_{12})_2$, and subsequent Crossover Experiment.

Pinacol- d_{12} . To a 1 L two-neck flask equipped with a magnetic stir bar was added magnesium turnings (3.08 g, 127 mmol), which were flame-dried under vacuum. Under nitrogen, HgCl₂ (3.45 g, 12.71 mmol) was added, followed by benzene (500 mL, 0.025 M). The flask was charged with acetone- d_6 (65.5 mL, 890 mmol) and the resultant mixture stirred for 24 hours at 60 °C under nitrogen. The mixture was cooled to ambient temperature, quenched with distilled water (200 mL) and stirred at 60 °C for an additional three hours. The suspension was extracted with diethyl ether and the organic layers combined and filtered on a Büchner funnel. The solvent was removed by distillation to afford a yellow oil which was purified on silica gel (30% ethyl acetate/hexanes) to afford 2.27 g (13% yield) of a clear oil. Spectral data are in accordance with those in the literature.

 $\mathbf{B_2(pin-}d_{12})_2$ was prepared according to literature procedure without deviation.⁵

Crossover experiment. In the dry box, a 6 dram vial with a magnetic stir bar was charged with Pd₂dba₃ (7.6 mg, 8.3 µmol), (R,R)-xylylTADDOLPNMe₂ ((R,R)-6, 13.2 mg, 19.9 µmol), and toluene (2.21 mL, 0.15 M). The metal and ligand were complexed for 1 hour, at which time B₂(pin)₂ (43.2 mg, 0.170 mmol) and B₂(pin- d_{12})₂ (46.3 mg, 0.166 mmol) were added simultaneously to the reaction mixture. Tridec-1,2-diene (60.3 mg, 0.332 mmol) was added to the reaction mixture, the vial was sealed with a polyethylene cap and electrical tape then removed from the dry box. The reaction stirred at room temperature for 15 hours when it was concentrated to dryness. Residual solvent was removed *in vacuo* and the material was purified with by passage through a short column of silica gel (5% ethyl acetate/hexanes). All fractions were pooled, concentrated, and submitted for LRMS-ESI⁺ analysis. For C₂₅H₄₈B₂O₄ (M+H)⁺ calc'd: 435.3, observed: 435.3 (M+H)⁺. For C₂₅H₂₅D₂₄B₂O₄ (M+H)⁺ calc'd: 459.5, observed: 459.5. C₂₅H₃₆D₁₂B₂O₄ was not observed.

⁵ Ishiyama, T.; Murata, M.; Ahiko, T- A.; Miyaura, N. Organic Syntheses, 2000, 77, 176-185.

⁴ Boller, T.M.; Murphy, J.M.; Hapke, M.; Ishiyama, T.; Miyaura, N.; Hartwig, J.F. J. Am. Chem. Soc. 2005, 127, 14263.

G. Synthesis of (R)-9- d_1

$$\begin{array}{c} \text{O} \\ \text{C}_{10}\text{H}_{21} \\ \text{H} \end{array} \xrightarrow{\begin{array}{c} \text{1.} \\ \text{MBuLi (1 equiv)} \\ \text{TMS} & = \\ \text{(1 equiv)} \\ \text{THF (1.6 M)} \end{array}} \begin{array}{c} \text{OH} \\ \text{C}_{10}\text{H}_{21} \\ \text{TMS} \end{array} \xrightarrow{\begin{array}{c} \text{PDC (1.5 equiv)} \\ \text{CH}_{2}\text{Cl}_{2} (1.07 \text{ M}) \end{array}} \begin{array}{c} \text{O} \\ \text{C}_{10}\text{H}_{21} \\ \text{TMS} \end{array}$$

1-(Trimethylsilyl)-tridec-1-yn-3-one. To a 25 mL round-bottom flask equipped with magnetic stir bar and sealed with a septum, was added trimethylsilylacetylene (1 mL, 7.07 mmol) and THF (4.4 mL). The flask was cooled to 0 °C (ice-water bath) and charged with 1.54 M "BuLi (4.59 mL). After stirring for 30 min at 0 °C, undecanal was added slowly over 30 min and the reaction allowed to stir at 0 °C for an additional 60 min. The reaction was quenched with water (10 mL) and extracted with ethyl acetate (3 times). The organic layers were combined, washed with 0.5 M HCl, saturated NaHCO₃, and brine. Organic extracts were dried (MgSO₄) and concentrated on a rotary evaporator to afford 1-trimethylsilanyl-tri-dec-1-yl-ol as an oil.

PDC Oxidation: To a septum-sealed 100 mL round-bottom flask with a magnetic stir bar and 4Å MS (1.27 g) was added CH₂Cl₂ (10 mL, 1.07 M) and pyridinium dichromate (3.98 g). In a separate flask, 1-trimethylsilanyl-tri-dec-1-yl-ol was dissolved in CH₂Cl₂ (4.4 mL) and transferred by cannula into the reaction flask. After the reaction stirred for 16 hours, it was diluted with diethyl ether and filtered over celite. The filtrated was concentrated to afford a clear oil, which was purified by silica gel chromatography (2% ethyl acetate/hexanes) to afford 1.17 g (62% yield) of a clear oil.

¹H NMR (400 MHz, CDCl₃): δ 0.24 (9H, s, (CH₃)₃Si), 0.85 (3H, t, J = 6.8 Hz, CH₃CH₂), 1.24-1.27 (14H, m, br, (CH₂)₇), 1.63 (2H, p, J = 7.2 Hz, CH₂CH₂CO), 2.52 (2H, t, J = 7.4 Hz, CH₂CH₂CO). ¹³C NMR (100 MHz, CDCl₃): δ -0.67 (3C), 14.2, 22.8, 24.0, 29.0, 29.4, 29.5, 29.6, 31.9, 97.5, 102.2, 188.0. IR (neat): 2942 (s), 2843 (s), 2148 (m), 2089 (w), 1681 (s), 1459 (s), 1404 (w), 1241 (s) cm⁻¹. LRMS-(ESI+): for C₁₆H₃₀NaOSi calc'd 289.2 (M+Na)⁺, observed 289.3 (M+Na)⁺. Purification: silica gel, 2% ethyl acetate/hexanes, $R_f = 0.45$ (10% ethyl acetate/hexanes), stains in PMA.

$$\begin{array}{c|c} & & & & \\ & & & & \\ & & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & \\ & & & \\ &$$

(S)-1-(trimethylsilyl)tridec-1-yn-3-ol. To a 100 mL round-bottom flask with magnetic stir bar was added 1-(trimethylsilyl)-tridec-1-yn-3-one (1 g, 3.76 mmol) in isopropanol (37.6 mL). Next, Ru[(1S,2S)-p-TsNCH(C₆H₅)CH(C₆H₅)NH](η^6 -p-cymene) (112.6 mg, 0.187) was added. The reaction was stirred for 16 hours after which time the crude reaction mixture was concentrated, and purified on silica gel to afford 1.00 g (91% yield, 99% ee) of a clear oil.

¹H NMR (400 MHz, CDCl₃): δ 0.15 (9H, s, Si(CH₃)₃), 0.86 (3H, t, J = 6.7 Hz, CH₃CH₂CH₂), 1.24 (14 H, s, br, (CH₂)₇), 1.38-1.44 (2H, m, CH₂CH₂CHOH), 1.63-1.69 (2H, m, CH₂CH₂CHOH), 2.20 (1H, s br, OH), 4.32 (1H, t, J = 6.6 Hz, CHOH). ¹³C NMR (100 MHz, CDCl₃): δ -0.008 (3C), 14.2, 22.8, 25.2, 29.3, 29.4, 29.6, 29.7, 29.8, 32.0, 37.8, 62.9, 89.2, 107.2. IR (neat): 3315 (s), 2935 (s), 2859 (s), 2164 (s), 1453 (s), 1399 (m), 1252 (s) cm⁻¹. LRMS-(ESI+) for C₁₆H₃₁NaOSi calc'd 290.2042

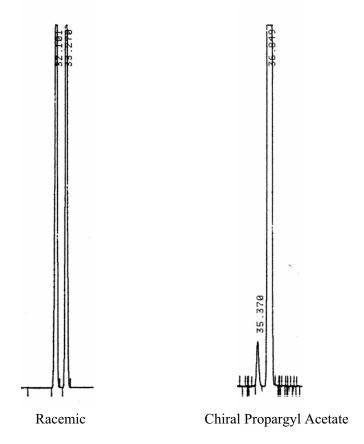
⁶ Matsumara, K.; Hashiguchi, S.; Ikariya, T.; Noyori, R. J. Am. Chem. Soc. 1997, 119, 8738-8739.

⁷ Catalyst Preparation: Haack, K-J.; Hashiguchi, S.; Fujii, A.; Ikariya, T.; Noyori, R. *Angew. Chem. Int. Ed. Eng.* **1997**, *36*, 285-287. Transfer Hydrogenation: Matsumara, K.; Hashiguchi, S.; Ikariya, T.; Noyori, R. *J. Am. Chem. Soc.* **1997**, *119*, 8738-8739.

 $(M+Na)^+$, observed 291.3 $(M+Na)^+$. Purification Technique: silica gel, 5% ethyl acetate/hexanes, $R_f = 0.45 (10\% \text{ ethyl acetate/hexanes})$, stains in PMA.

Determination of Enantioselectivity. The trimethylsilyl group of (*S*)-1-(trimethylsilyl)tridec-1-yn-3-ol was deprotected with potassium carbonate (1 equiv) in methanol (1.0 M). The hydroxyl group was then protected using acetic anhydride (1.2 equiv), triethylamine (3 equiv), and 4-dimethylaminopyridine (cat) in dichloromethane (0.25 M). The enantiomerically enriched material was compared to racemic product.

Chiral GLC (β -dex, Supelco, 140 °C) – analysis of acetate derived from TMS deprotection and alcohol protection of the reaction product as described above



$$\begin{array}{c} OH \\ C_{10}H_{21} \end{array} \xrightarrow{\text{TMS}} \begin{array}{c} K_2CO_3 \text{ (1 equiv)} \\ \hline MeOD \text{ (1 M)} \end{array} \begin{array}{c} OH \\ C_{10}H_{21} \end{array}$$

(S)-1-Deuteriotridec-1-yn-3-ol. To a 50 mL round-bottom flask with magnetic stir bar, was added (S)-1-(trimethylsilyl)tridec-1-yn-3-ol (2.23 mg, 8.31 mmol) in methanol-OD (8.3 mL). Potassium carbonate (anhydrous) (1.132 g, 8.314 mmol) was added under nitrogen. The reaction was stirred for 3 hours when it was quenced with water and diluted with ethyl acetate. The aqueous layer was washed twice with ethyl acetate, then the organic extracts were combined and washed with brine and dried (MgSO₄). The organic layer was filtered over celite and concentrated on a rotary evaporator. The reaction mixture was purified on silica gel (10% ethyl acetate/hexanes) to afford 1.548 g (94% yield) of a white solid.

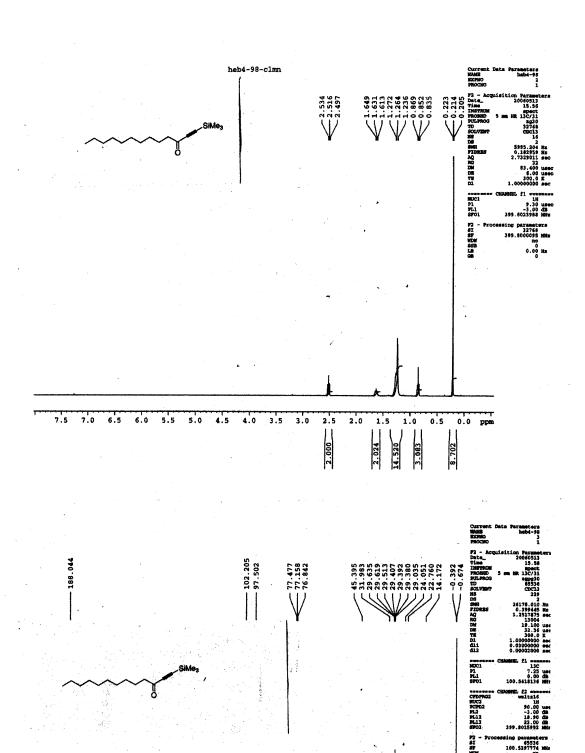
¹H NMR (400 MHz, CDCl₃) δ 0.85 (3H, t, J = 6.8 Hz, CH₃CH₂), 1.24 (14H, m, (CH₂)₇), 1.42 (2H, m, CH₂CH₂CHOH), 1.68 (2H, m, CH₂CH₂CHOH), 2.53 (1H, s, OH), 4.34 (1H, t, J = 6.8 Hz, CH₂CHOH). ¹³C NMR (100 MHz, CDCl₃) δ 14.2, 22.8, 25.1, 29.3, 29.4, 29.6, 29.7, 29.8, 31.9, 37.7, 62.3 (1C, $^3J_{CD} = 2.3$ Hz), 72.6 (1C, $^IJ_{CD} = 38.1$ Hz), 84.8 (1C, $^2J_{CD} = 7.2$ Hz). IR (CH₂Cl₂): 3364 (s), 2952 (s), 2843 (s), 2593 (s), 1980 (s), 1632 (w), 1464 (s) cm⁻¹. LRMS-(ESI+) for C₁₃H₂₃DNaO calc'd 220.1 (M+Na)⁺, observed 220.1 (M+Na)⁺. Purification Technique: silica gel, 10% ethyl acetate/hexanes, R_f = 0.21 (10% ethyl acetate/hexanes), developed with PMA.

(*R*)-1-Deuteriotridec-1,2-diene. ⁸ To a two neck 100 mL round-bottom flask with a magnetic stir bar, was added triphenylphosphine (1.01 g, 3.86 mmol) in THF (12 mL). This was cooled to -40 °C (ethylene glycol:ethanol, dry ice) and diethyl azodicarboxylate (607 μL, 3.86 mmol) was added slowly. The reaction stirred for 15 minutes at -40 °C when the labeled alkyne (586 mg, 2.662 mmol) was transferred by cannula as a solution in THF (9.1 mL). After an additional 15 minutes at -40 °C, *o*-nitrobenzenesulfonylhydrazine (NBSH, 838.2 mg, 3.860 mmol) was added as a solution in THF (12 mL). The reaction was stirred at -40 °C for an additional 2 hours, then was allowed to gradually warm to ambient temperature and stir overnight. The crude reaction mixture was concentrated, dissolved in methylene chloride and purified on silica gel (pentanes) to afford 263 mg (50% yield) of a clear oil.

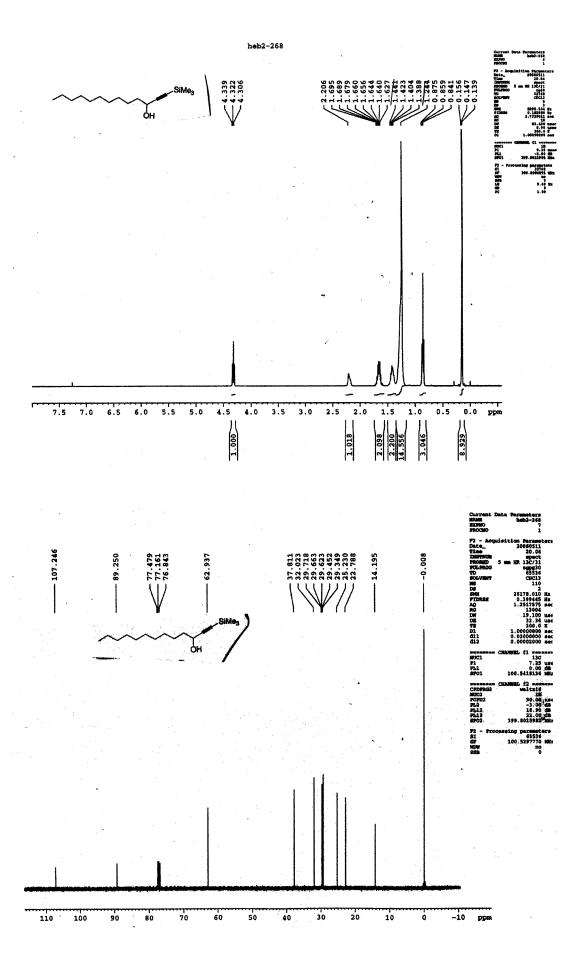
¹H NMR (400 MHz, CDCl₃): δ 0.90 (3H, t, J = 6.8 Hz, CH₃CH₂), 1.29 (14H, s, br, (CH₂)₇), 1.38-1.45 (2H, m, CH₂CH₂CH), 2.00 (2H, qd, J = 7.2, 7.2, 7.1, 3.0 Hz, CH₂CH₂CH), 4.65-4.62 (1H, m, CHD), 5.09 (1H, q, J = 7.2, 6.8 Hz, CH₂CHC). ¹³C NMR (100 MHz, CDCl₃) δ 14.2, 22.9, 28.5, 29.3, 29.4, 29.5, 29.6, 28.9, 32.1, 74.4 (1C, t, ${}^{I}J_{CD} = 25.6$ Hz), 90.3, 208.73. ²H NMR (61.4 MHz, CHCl₃): δ 4.68. IR (neat) 2962 (s), 2924 (s), 2848(s), 1958 (s), 1458 (s), 1372 (m) cm⁻¹. HRMS-CI GC/MS: for C₁₃H₂₃D calc'd 181.194 (M⁺), observed 181.193 (M⁺). Purification: silica gel, pentanes, R_f = 0.92 (pentanes), developed with KMnO₄.

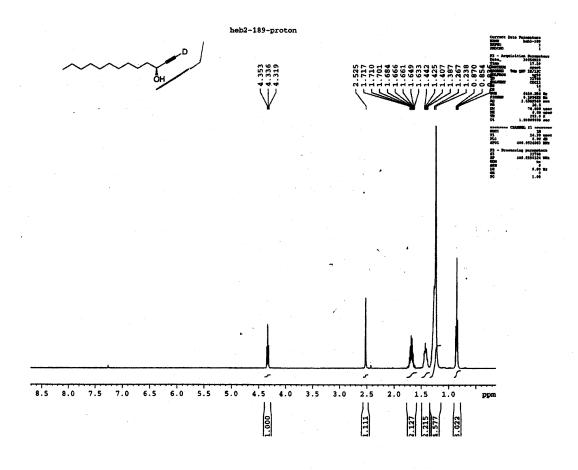
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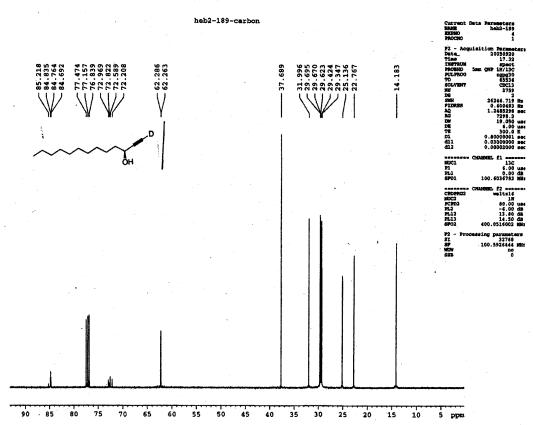
⁸ Myers, A.G.; Zheng, B. J. Am. Chem. Soc. 1996, 118, 4492-4493.

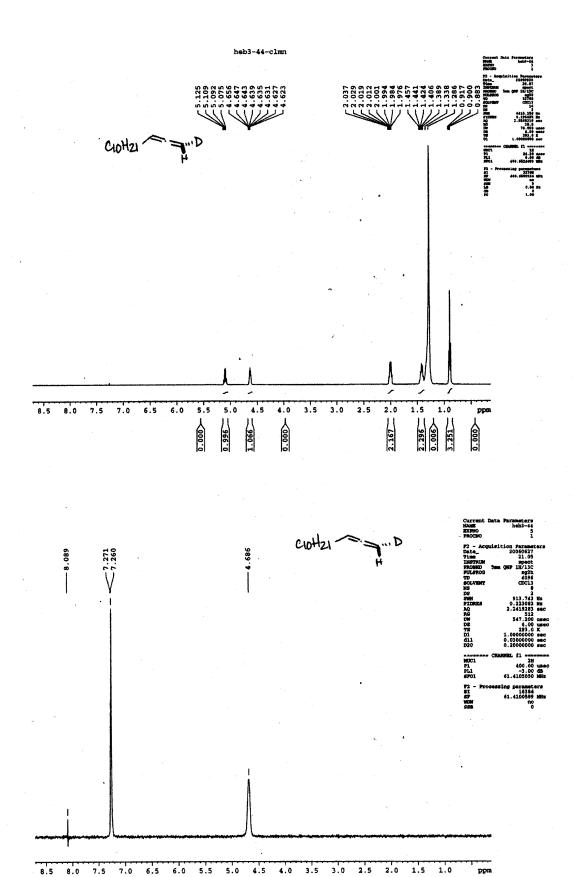


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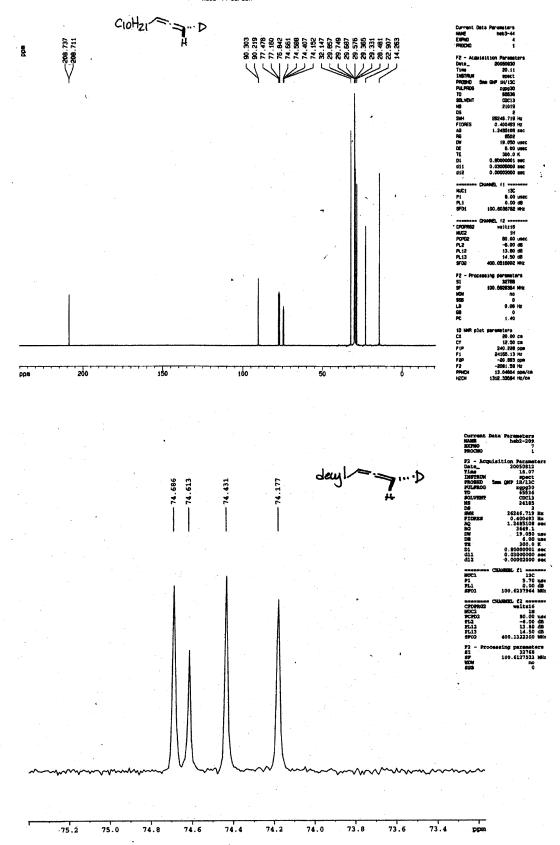






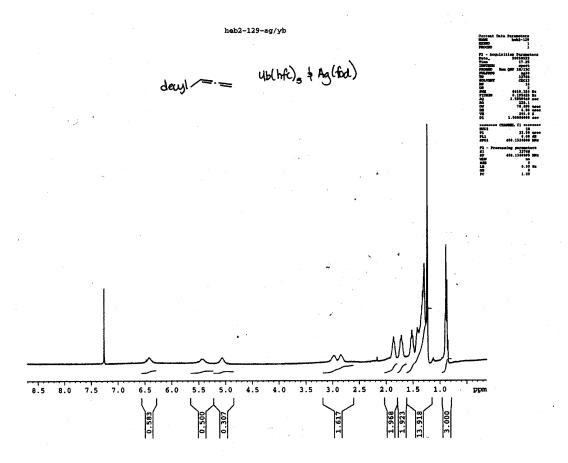


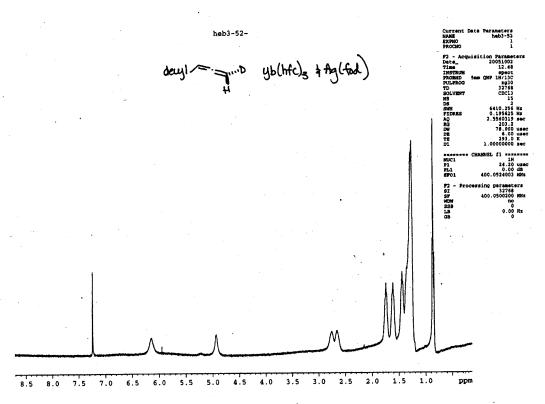


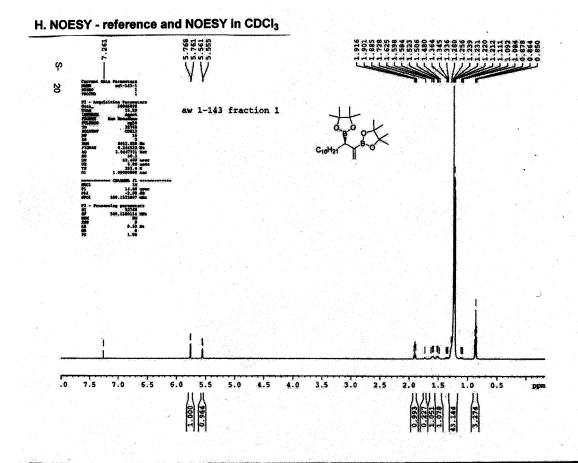


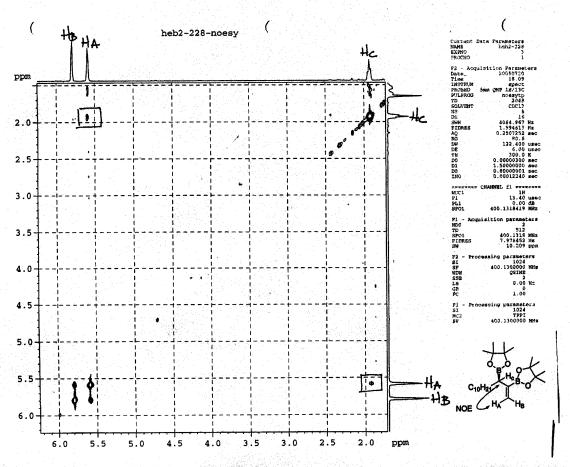
Enatiomeric purity determination of (*R*)-9- d_1 . To a 1 dram vial was added ytterbium tris[3-(heptafluoropropylhydroxymethylene)-(+)-camphorate] (Yb(hfc)₃) (23.4 mg, 0.019 mmol) and (6,6,7,7,8,8-Heptafluoro-2,2-dimethyl-3,5-octandionato)silver (Ag(fod)) (8.0 mg, 0.019 mmol). To this was added 750 μ L of CDCl₃ and decyl allene (5.8 mg, 0.0321 mmol). ¹H NMR of the reference unlabeled decyl allene is below. The signals at δ 5.0 and 5.5 ppm correspond to the two terminal protons on the allene. A small amount of protonation was observed in the labeled decyl allene (δ 5.2 ppm).

⁹ Chirality of Allene. Reference: Offermann, W.; Mannschreck, A. *Organic Magnetic Resonance*. **1984**, *22*, 355-363. Mannschreck, A.; Munninger, W.; Burgemeister, T.; Gore, J.; Cazes, B. *Tetrahedron*, **1986**, *42*, 399-408.





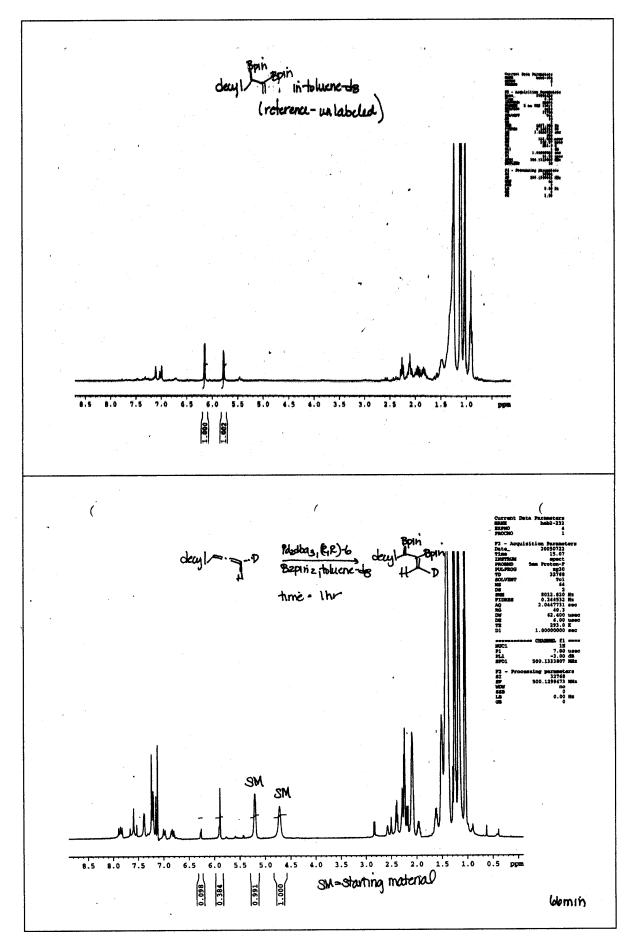


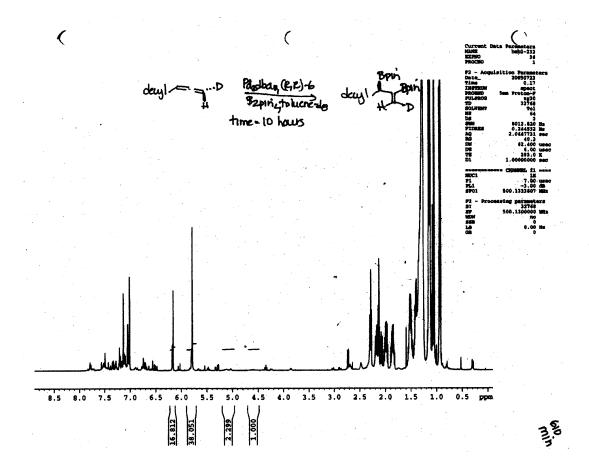


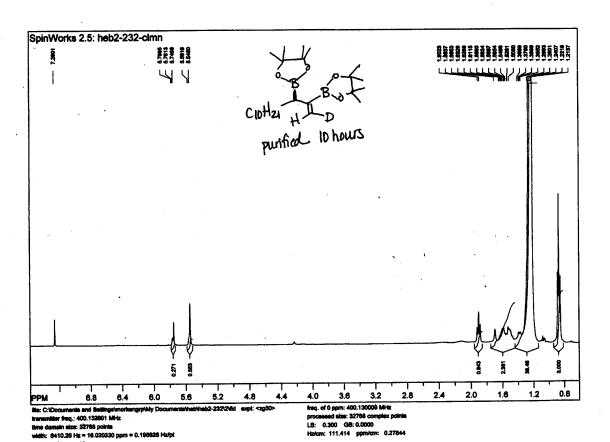
Supporting Information - Page S22

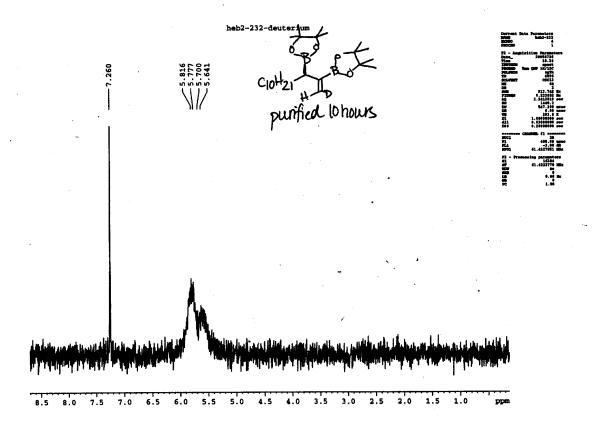
I. Diboration of (R)-9- d_1

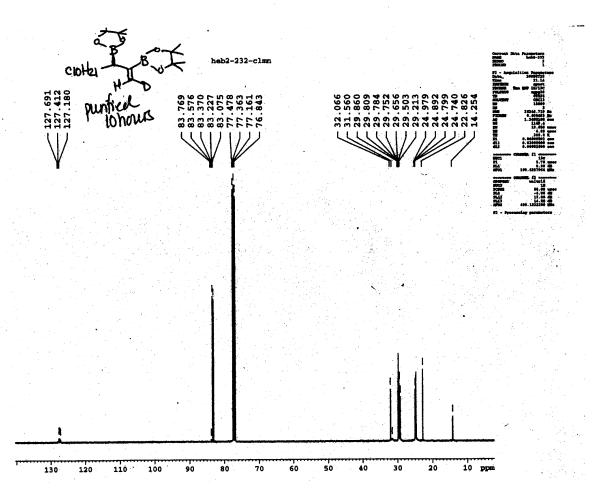
Diboration with(R)-9- d_1 In the dry box, a 1 dram vial equipped with a stir bar was loaded with Pd₂dba₃ (4.0 mg, 0.0043 mmol) and (R,R)-6 (6.6 mg, 0.010 mmol). To this was added 300 μL of toluene- d_8 . The metal/ligand mixture stirred for 1 hour at which time B₂(pin)₂ (46.3 mg, 0.1824 mmol) was added followed by the labeled allene (27.5 mg, 0.152 mmol). The reaction mixture was transferred to an ovendried J-Young tube and diluted with the remaining 700 μL of toluene- d_8 . The reaction progress was monitored by ¹H NMR every 17 minutes. After completion of the reaction, solvent was removed *in vacuo* and the crude material was purified on silica gel (2% ethyl acetate/hexanes) to afford 45.5 mg (68% yield) of the labeled product.



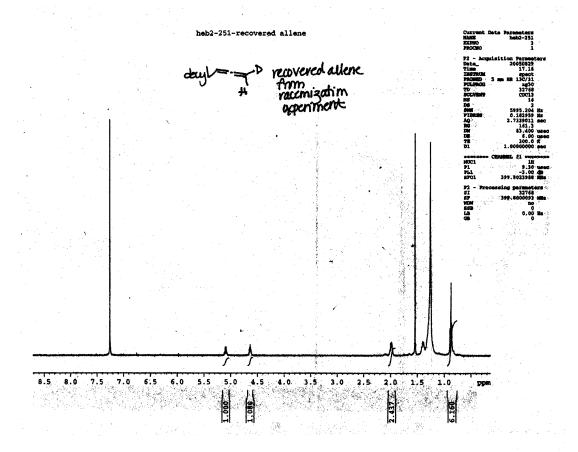


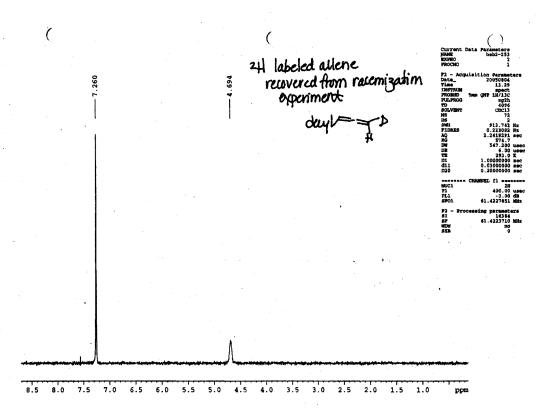


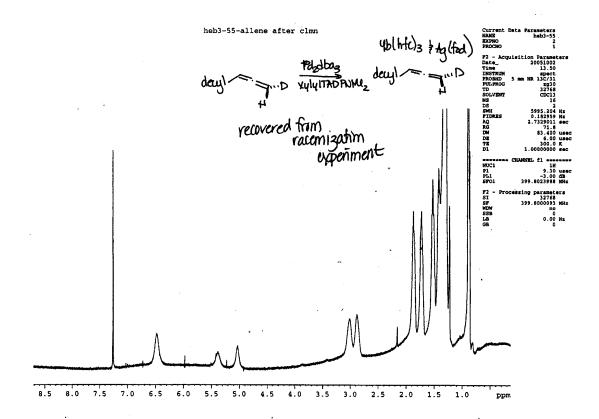


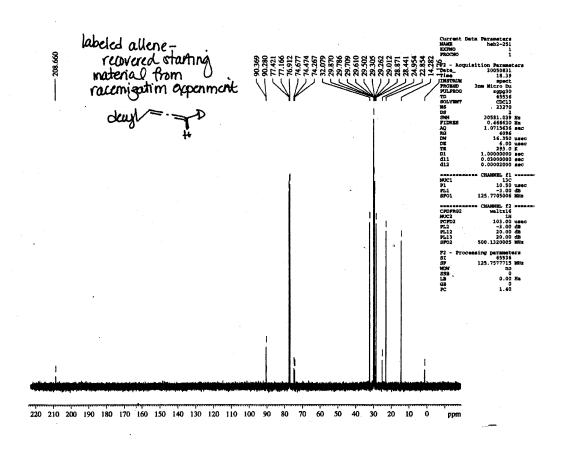


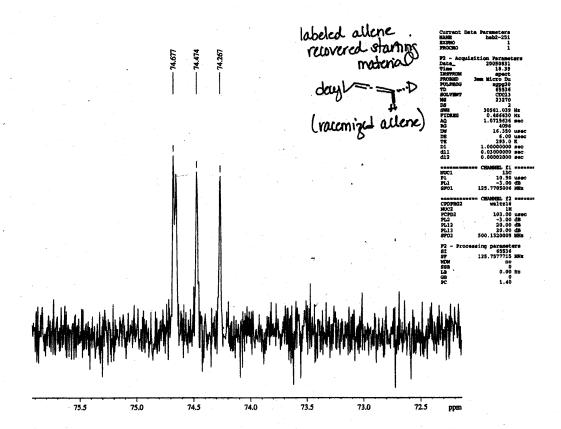
J. Racemization experiment with labeled allene. In the dry box, a 2 dram vial equipped with a stir bar was charged Pd_2dba_3 (3.5 mg, 0.0038 mmol) and (R,R)-6 (6 mg, 0.0091 mmol). To this was added 1 mL of toluene and the metal/ligand mixture complexed for 1 hour at which time the labeled allene (27.5 mg, 0.152 mmol) was added. The vial was sealed with a polypropylene cap and removed from the glove box. After a reaction time of 1 hour, solvent was removed in vacuo and the crude reaction mixture was purified on silica gel (pentanes) to afford 7.4 mg (27%) of the recovered allene. Analysis by Yb(hfc)₃ and Ag(fod) indicates that the allene has racemized under the reaction conditions in the absence of $B_2(pin)_2$.











K. Kinetics Studies for Allene Diboration

A. Preparation of capillary tubes containing 1,3-dimethoxybenzene. A stock solution of 1,3-dimethoxybenzene (100 μ L, 0.7635 mmol) in toluene- d_8 (5 mL) was prepared. A 6-inch 22 gauge needle attached to a 100 μ L gas-tight syringe was used to transfer 70 μ L of the stock solution to a capillary tube. Each capillary was flame-sealed and its accuracy checked by ¹H NMR against a stock solution of tridec-1,2-diene.

B. Initial Rate Studies with varied $[B_2(pin)_2]$:

Preparation of stock solutions: All operations were performed in the dry box, all glassware was ovendried overnight prior to use, and gas-tight microliter syringes were used for all operations. Stock solutions were prepared immediately before use.

<u>Tridec-1,2-diene solution:</u> To a 6 dram vial was added 405.3 mg (2.247 mmol) of tridec-1,2-diene and this was diluted with toluene- d_8 (1.5 mL).

<u>Catalyst solution:</u> To a 6 dram vial was added Pd_2dba_3 (11.8 mg, 0.0128 mmol), (*R*,*R*)-xylylTADDOLPNMe₂ (18.8 mg, 0.0288 mmol), and toluene- d_8 (4 mL). This mixture was allowed to incubate for 1 hour prior to use.

 $\underline{B_2(pin)_2}$ solution: To a 6 dram vial was added bis(pinacolato)diboron (457.7 mg, 1.802 mmol) and toluene- d_8 (3 mL).

Kinetic Runs:

 $[\mathbf{B_2(pin)_2}] = 0.45 \,\mathrm{M}$: To a J-Young tube was added 200 $\mu\mathrm{L}$ of the catalyst solution. To this was added 100 $\mu\mathrm{L}$ of toluene- d_8 and 1.5 mL of $\mathrm{B_2(pin)_2}$ solution. The J-Young tube was sealed with a Teflon cap and inverted several times. The Teflon cap was removed, the flame-sealed capillary inserted, and 200 $\mu\mathrm{L}$ of tridec-1,2-diene solution was added. The J-Young tube was sealed with the Teflon cap and inverted several more times.

 $[B_2(pin)_2] = 0.30 \text{ M}$: This experiment was conduct as above except that 1 mL of $B_2(pin)_2$ solution and 600 µL of toluene were employed.

 $[B_2(pin)_2] = 0.15 \text{ M}$: This experiment was conduct as above except that 0.5 mL of $B_2(pin)_2$ solution and 1.1 mL of toluene were employed.

Results. Single-pulse ¹H NMR was utilized to measure the formation of product (in comparison to the internal standard) versus time.

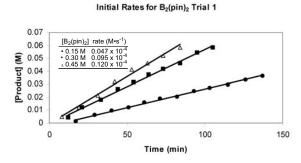


Figure S1. Initial Rates for B₂(pin)₂ Trial 1: 0.2 mol % Pd₂dba₃ and 0.48 mol % (R,R)-6

Initial Rates for B2(pin)2 Trial 2

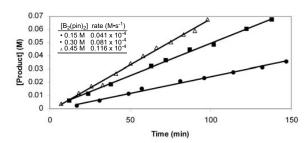


Figure S2. Initial Rates for B₂(pin)₂ Trial 2: Catalyst loading: 0.2 mol % Pd₂dba₃ and 0.48 mol % (*R*,*R*)-6

Initial Rates for B2(pin)2 Trial 3

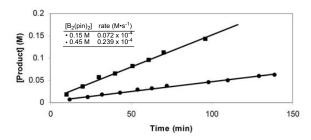


Figure S3. Initial Rates for B₂(pin)₂ Trial 3: Catalyst loading: Catalyst loading: 0.5 mol % Pd₂dba₃ and 1.2 mol % (R,R)-6

C. Initial Rate Studies with varied [Allene]:

Preparation of stock solutions: All operations were performed in the dry box, all glassware was ovendried overnight prior to use, and gas-tight microliter syringes were used for all operations. Stock solutions were prepared immediately before each kinetic run.

<u>Tridec-1,2-diene solution:</u> To a 6 dram vial was added 324.4 mg (1.799 mmol) of tridec-1,2-diene and this was diluted with toluene- d_8 (3 mL).

<u>Catalyst solution:</u> To a 6 dram vial was added Pd_2dba_3 (12.0 mg, 0.0131 mmol), (*R,R*)-xylylTADDOLPNMe₂ (19.8 mg, 0.0303 mmol), and toluene- d_8 (4 mL). The metal and ligand were incubated for 1 hour in the dry box prior to use.

 $\underline{B_2(pin)_2}$ solution: To a 6 dram vial was added bis(pinacolato)diboron (571.0 mg, 2.248 mmol) and toluene- d_8 (1.5 mL).

Kinetic Runs:

[tridec-1,2-diene] = 0.45 M: To a J-Young tube was added 200 μ L of the catalyst solution. To this was added 100 μ L of toluene- d_8 and 200 μ L of B₂(pin)₂ solution. The J-Young tube was sealed with a Teflon cap and inverted several times. The Teflon cap was removed, the flame-sealed capillary inserted, and 1.5 mL of tridec-1,2-diene solution was added. The J-Young tube was sealed with the Teflon cap and the NMR tube inverted several more times.

[tridec-1,2-diene] = 0.30 M: This was carried out as above except that 1 mL of the tridec-1,2-diene solution and 600 μ L of toluene were used.

[tridec-1,2-diene] = 0.15 M: This was carried out as above except that 0.5 mL of the tridec-1,2-diene solution and 1.1 mL of toluene were used.

Results. Single-pulse ¹H NMR was utilized to measure the formation of product (in comparison to the internal standard) versus time.

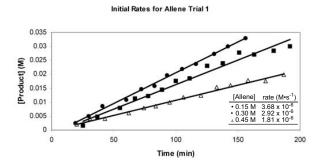


Figure S4. Initial Rates for Allene – Trial 1: Catalyst loading: 0.2 mol % Pd₂dba₃ and 0.48 mol % (*R*,*R*)-6
Initial Rates for Allene Trial 2

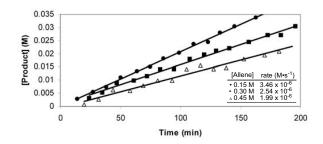


Figure S5. Initial Rates for Allene – Trial 2: Catalyst loading: 0.2 mol % Pd₂dba₃ and 0.48 mol % (*R*,*R*)-6

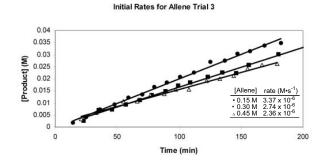
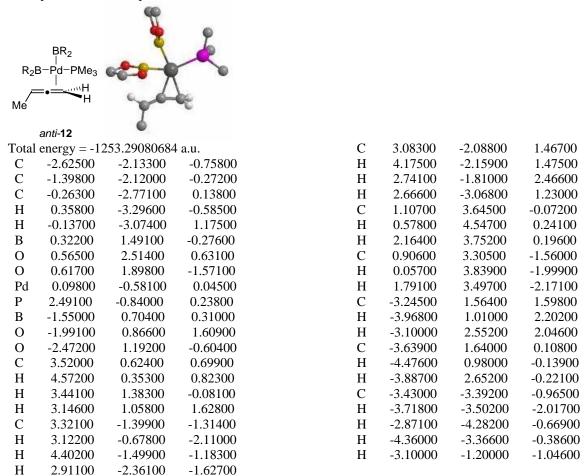


Figure S6. Initial Rates for Allene – Trial 3: Catalyst loading: 0.2 mol % Pd₂dba₃ and 0.48 mol % (*R*,*R*)-6

L. DFT Calculations.

Computational Methods: DFT calculations were performed using B3LYP^{10,11} with the basis set of Stuttgart RSC 1997 ECP¹² for Pd, 6-311+G*¹³ for others. All calculations were conducted using Gaussian 03 C02 package¹⁴ with tight SCF convergence and ultra fine integration grids on the 128-CPU SGI Altix 3700 SMP machine¹⁵ at University of North Carolina at Chapel Hill. In search for the transition state structure, a single-point frequency calculation has been performed to ensure that the final structure obtained (i) has only one imaginary frequency and (ii) the vibration mode of the negative frequency corresponds to the anticipated bond formation.



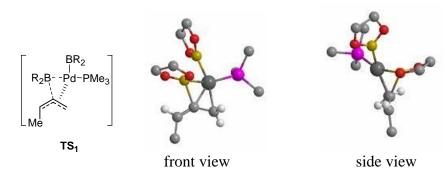
- (10) Becke, A.D. J. Chem. Phys. 1993, 98, 1372.
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- (15) http://its.unc.edu/hpc/hardware/#Cedar/Cypress/outline

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 H

syn-**12**

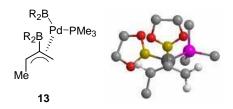
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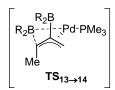
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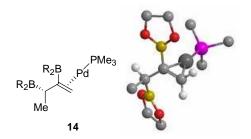
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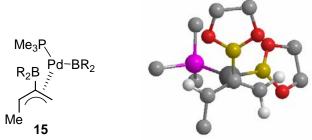
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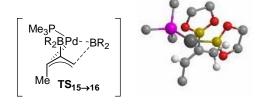
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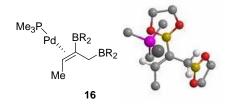
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C	-1.11800	1.56400	2.94700
Н	-0.71800	2.57700	2.84500
Η	-0.31000	0.92100	3.30200
Н	-1.87400	1.60500	3.74400
C	-1.74100	1.08700	1.66000
Н	-2.38400	1.81000	1.16300
C	-1.84500	-0.23500	1.23000
C	-0.92100	-1.26200	1.64800
Н	-1.07800	-2.26200	1.26400
Н	-0.49000	-1.24300	2.64600
В	-2.80700	-0.54200	0.03900
O	-3.07900	-1.81500	-0.40300
O	-3.50600	0.42000	-0.65900
Pd	0.22900	0.14600	0.48600
P	1.31500	1.81000	-0.71200
В	1.69900	-1.24000	0.12500
O	2.11300	-1.62500	-1.15000
O	2.43200	-1.90100	1.09800
C	1.35600	1.47600	-2.52300
Η	1.93000	2.23500	-3.06300
Η	1.79600	0.49100	-2.68600
Η	0.33600	1.45800	-2.91300
C	3.09700	2.09200	-0.32000
Η	3.65400	1.16900	-0.48600
Η	3.52600	2.88500	-0.94000
Η	3.20500	2.36400	0.73100
C	0.64600	3.53300	-0.64300
Η	0.66900	3.89600	0.38700
Η	1.22200	4.21900	-1.27100
Η	-0.39400	3.53400	-0.97400
C	3.43800	-2.71400	0.46600
Η	4.42100	-2.26900	0.65300
Η	3.41900	-3.71200	0.90800
C	3.06000	-2.70500	-1.02600
Η	2.57500	-3.63500	-1.34000
Η	3.90900	-2.51400	-1.68600
C	-3.92500	-1.71900	-1.55900
Н	-3.33800	-1.98800	-2.44300
Η	-4.75100	-2.42600	-1.45800
C	-4.38500	-0.24500	-1.58000
Η	-5.41500	-0.12300	-1.23200
Η	-4.29100	0.22000	-2.56400



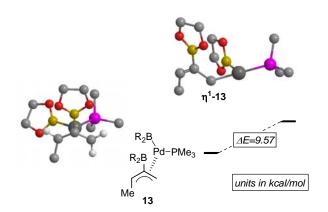
Total energy = -1253.32837014 a.u.

C	1.04900	-0.58500	3.44300
Н	0.73200	-1.59100	3.73400
Н	0.16000	0.04500	3.39100
Н	1.67500	-0.20900	4.26500
C	1.82800	-0.61500	2.16000
Н	2.57900	-1.39800	2.08500
C	1.77200	0.29000	1.13400
C	0.79300	1.41600	1.05900
Н	1.16700	2.22400	0.43300
Η	0.45900	1.79600	2.02100
В	2.79600	0.12500	-0.02700
O	3.03100	1.10200	-0.96700
O	3.59500	-0.98500	-0.21100
Pd	-0.35700	-0.29700	0.32900
P	-1.86700	-1.79600	-0.52800
В	-1.01600	1.62600	0.06800
O	-0.97900	2.25500	-1.17400
O	-1.92300	2.25700	0.91700
C	-1.99500	-1.79400	-2.37000
Н	-2.76400	-2.48700	-2.72400
Н	-2.23000	-0.78600	-2.71600
Н	-1.03400	-2.07700	-2.80500
C	-3.62400	-1.52700	-0.02500
Н	-3.93300	-0.52000	-0.30900
Н	-4.29600	-2.25400	-0.49000
Н	-3.71100	-1.60400	1.06100
C	-1.66400	-3.60300	-0.17200
Н	-1.69300	-3.76800	0.90700
Н	-2.44800	-4.20500	-0.64300
Н	-0.69100	-3.94200	-0.53400
C	-2.61700	3.27000	0.17400
Н	-3.63600	2.92500	-0.03500
Н	-2.67600	4.18100	0.77500
C	-1.78500	3.44100	-1.11000
Η	-1.12700	4.31600	-1.06500
Н	-2.39500	3.51300	-2.01300
C	3.96500	0.58800	-1.92700
Н	3.43300	0.39700	-2.86400
Н	4.73900	1.33600	-2.11100
C	4.51000	-0.70400	-1.28100
Н	5.51200	-0.57000	-0.86000
Н	4.53000	-1.55300	-1.96800

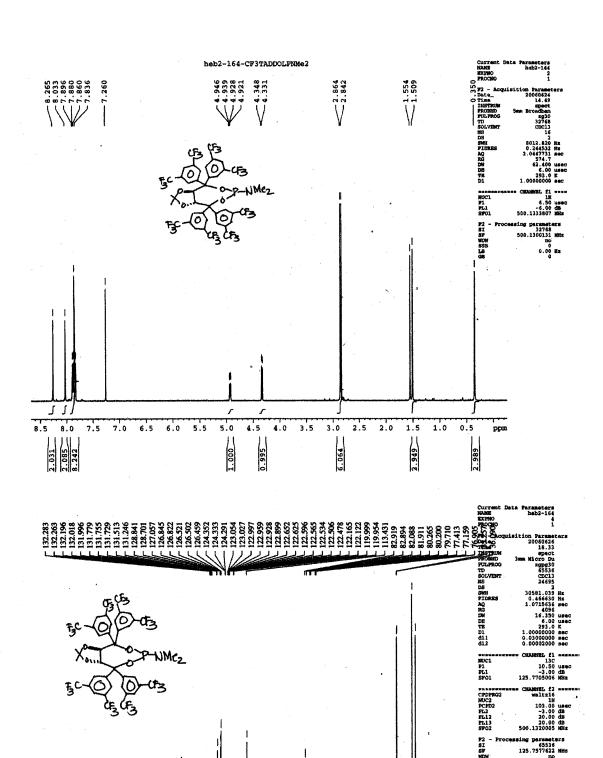


Total energy = -1253.36450257 a.u.

C	0.34200	1.50100	2.96700
Η	-0.23000	0.86700	3.65100
Η	-0.34600	2.20600	2.50100
Η	1.04900	2.07400	3.58400
C	1.09400	0.66600	1.96700
Η	1.60800	-0.19900	2.38400
C	1.46300	1.05400	0.68900
C	1.09800	2.42500	0.10100
Η	1.75700	2.62100	-0.74900
Η	1.30800	3.20100	0.84800
В	2.48700	0.17300	-0.07400
O	2.85700	0.40400	-1.38600
O	3.21300	-0.86200	0.49400
Pd	-0.15500	-0.50200	0.48800
P	-1.71000	-2.08600	-0.13900
В	-0.38600	2.58700	-0.39000
O	-0.78700	2.41000	-1.69400
O	-1.41500	3.01600	0.42200
C	-1.90100	-2.35900	-1.96000
Н	-2.64700	-3.13100	-2.17900
Н	-2.19900	-1.42800	-2.44400
Н	-0.94300	-2.66000	-2.38700
C	-3.47200	-1.84100	0.37400
Н	-3.84200	-0.89400	-0.02300
Η	-4.11600	-2.65100	0.01900
Η	-3.53300	-1.79100	1.46300
C	-1.41100	-3.81400	0.45100
Η	-1.40300	-3.83300	1.54300
Η	-2.17700	-4.50700	0.08800
Н	-0.43200	-4.15300	0.10700
C	-2.63100	3.01000	-0.34000
Н	-3.28600	2.22700	0.05200
Н	-3.13000	3.97400	-0.22200
C	-2.18100	2.73500	-1.79500
Н	-2.29000	3.60900	-2.44200
Н	-2.71100	1.89600	-2.25200
C	3.93400	-0.48400	-1.71300
Н	3.73500	-0.95200	-2.67900
Н	4.85900	0.09700	-1.79000
C	3.97000	-1.49200	-0.54600
H	4.97900	-1.69700	-0.18300
Н	3.49100	-2.44300	-0.80500



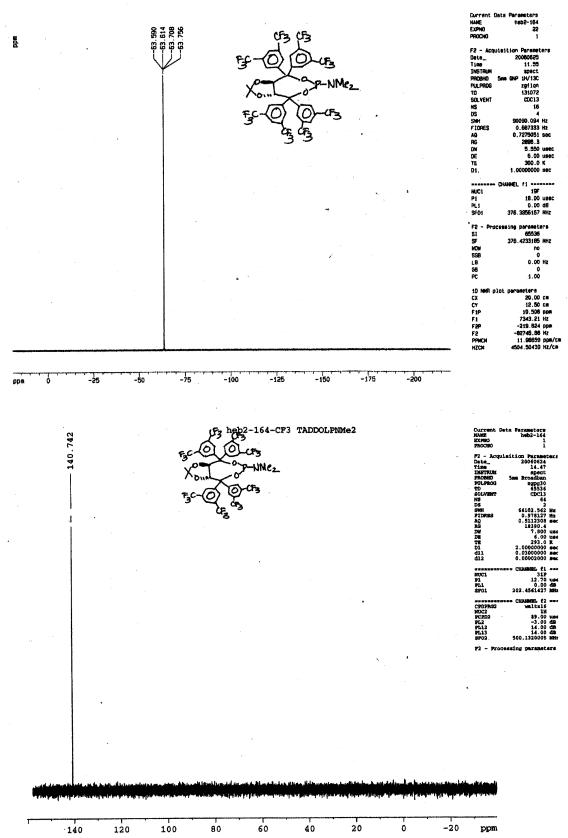
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C	2.62000	2.69200	-1.76800
Н	3.30700	3.50200	-1.50700
Н	1.60100	3.03000	-1.55500
Η	2.68500	2.55300	-2.85400
C	2.95800	1.42700	-1.02400
Η	3.88500	1.43300	-0.45600
C	2.21900	0.29300	-1.01300
C	0.92900	0.14300	-1.74800
Н	0.89700	-0.83600	-2.25400
Η	0.77400	0.92900	-2.49100
В	2.75500	-0.91200	-0.19900
O	2.25700	-2.19500	-0.31200
O	3.79800	-0.84000	0.70700
Pd	-0.94600	-0.18200	-0.85300
P	-3.06300	-0.61200	0.11000
В	-0.23400	1.11800	0.47400
O	0.26100	0.71900	1.70100
O	-0.24000	2.49100	0.33300
C	-2.98800	-1.37300	1.78800
Н	-2.38000	-0.74700	2.44300
Η	-3.98300	-1.49800	2.22500
Η	-2.50200	-2.34800	1.72300
C	-4.21900	-1.75400	-0.77700
Н	-3.74600	-2.72900	-0.91100
Н	-5.15500	-1.89000	-0.22800
Н	-4.44900	-1.35400	-1.76700
C	-4.13800	0.86200	0.39200
Н	-4.35600	1.34500	-0.56200
Н	-5.07900	0.59200	0.88000
Н	-3.60900	1.58700	1.01400
C	4.10300	-2.16900	1.14900
Н	5.05000	-2.48300	0.69900
Н	4.21500	-2.17100	2.23500
C	2.91100	-3.02000	0.66100
Η	3.21700	-3.95800	0.19300
Н	2.20200	-3.24300	1.46500
C	0.81700	1.87900	2.35400
Н	0.51200	1.88000	3.40300
Η	1.90600	1.81600	2.29700
C	0.25600	3.07300	1.55600
Н	1.01500	3.81700	1.31000
Н	-0.57300	3.57100	2.06900

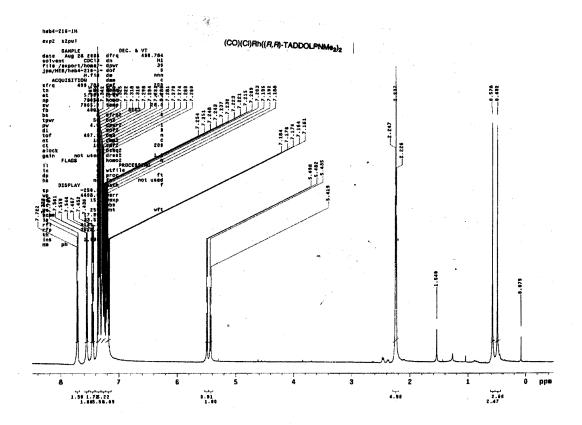


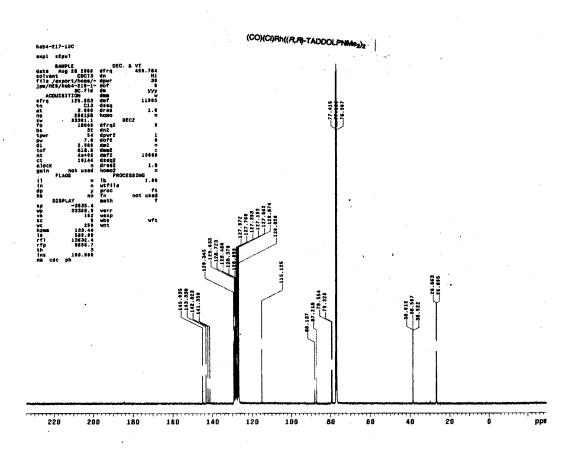
70

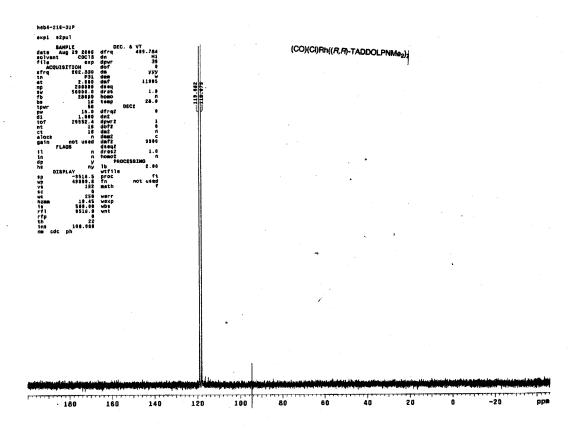
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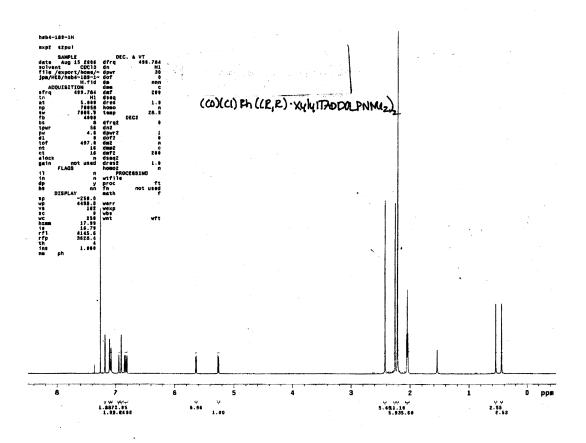


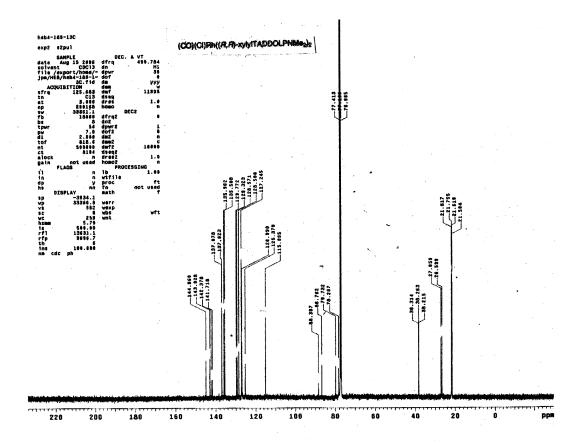


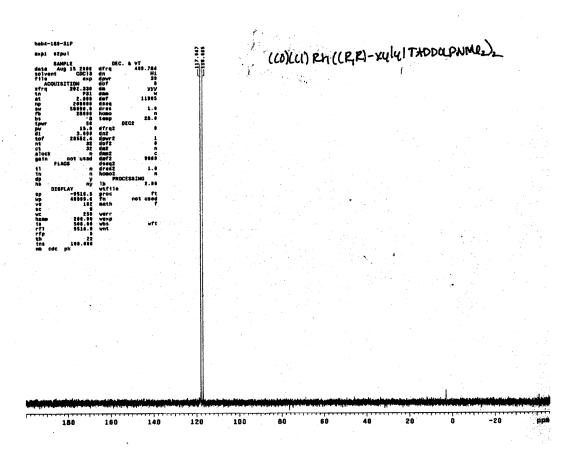


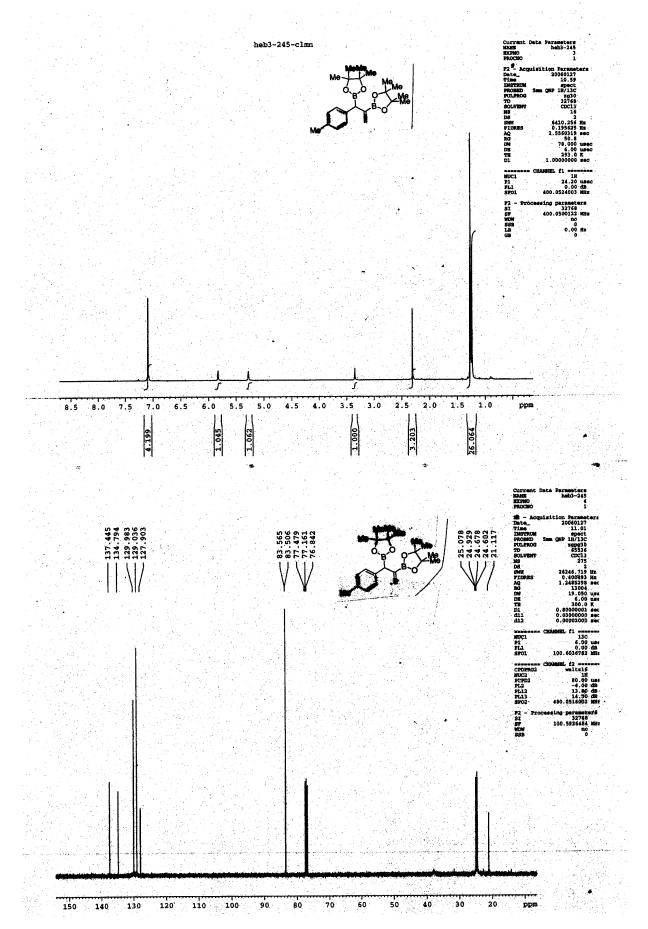




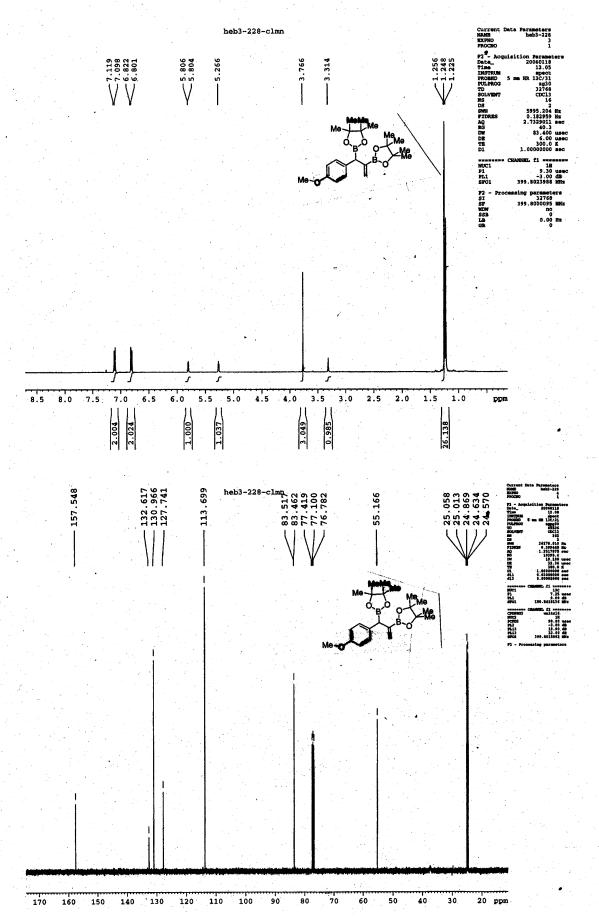








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