

Supporting Information (ES0353964)

Appendix A: Velocity Gradient Calculations.

Table S-1. Volatilization data of phenol from its binary-solute solution with *m*-cresol at 25 °C.

Table S-2. Experimental K_L values of benzene, toluene, ethylbenzene, and *o*-xylene from their multi-solute solutions at 25 °C

Appendix A: Velocity Gradient Calculations

The velocity gradient G (s^{-1}) is defined as

$$G = (P/V\mu)^{0.5}$$

where P is the stirring power (W) for the solution, V is the solution volume in the reactor vessel (m^3), and μ is the solution dynamic viscosity (N.m/s). For water at 25 °C, $\mu = 0.000900$ N.m/s.

The stirring power is determined from

$$P = (C_D A \rho v^3)/2$$

where C_D is the drag coefficient (approximately 1.8 for a blade moving perpendicular to the water solution), A is the blade area (m^2) and ρ is the water-solution density (approximately 1000 kg/m^3), and v is the velocity of the blade relative to water (m/s). The value of v is related to the stirring rpm and blade diameter approximately as

$$v = (rpm/60)\pi D(0.75)$$

where D is the blade diameter (m) and 0.75 is an empirical experimental constant. With the solution $V = 0.0001$ m^3 and the blade $D = 0.06$ m and $A = (0.06$ m) (0.015 m) = 0.0009 m^2 in our experiments, the G values calculated for stirring rates at 0, 20, 50, and 100 rpm stirring are 0, 31, 121, and 344 s^{-1} , respectively.

Table S-1. K_L (cm/min), k_L (cm/min), $k_G H$ (cm/min), and α values of phenol from its binary-solute solutions with *m*-cresol at 25 °C under various system settings^a.

Air Speed = 0 m/s	K_L	α	k_L	$k_G H$
Stirring: 0 rpm	1.24E-4	1.02	-6.20E-3	1.22E-4
Stirring: 20 rpm	1.52E-4	1.03	-5.07E-3	1.47E-4
Stirring: 50 rpm	1.72E-4	1.05	-3.44E-3	1.64E-4
Stirring: 100 rpm	2.29E-4	1.09	-2.54E-3	2.10E-4
Air Speed = 0.2 m/s				
Stirring: 0 rpm	1.24E-3	0.961	0.0318	1.29E-3
Stirring: 20 rpm	1.31E-3	0.963	0.0354	1.36E-3
Stirring: 50 rpm	1.46E-3	0.973	0.0541	1.50E-3
Stirring: 100 rpm	1.55E-3	0.975	0.0620	1.59E-3
Air Speed = 2.0 m/s				
Stirring: 0 rpm	2.90E-3	0.884	0.0250	3.28E-3
Stirring: 20 rpm	3.29E-3	0.889	0.0296	3.70E-3
Stirring: 50 rpm	3.47E-3	0.904	0.0361	3.84E-3
Stirring: 100 rpm	3.64E-3	0.919	0.0449	3.96E-3
Air Speed = 4.0 m/s				
Stirring: 0 rpm	4.53E-3	0.866	0.0338	5.23E-3
Stirring: 20 rpm	4.64E-3	0.879	0.0383	5.28E-3
Stirring: 50 rpm	4.85E-3	0.887	0.0429	5.47E-3
Stirring: 100 rpm	5.04E-3	0.895	0.0480	5.63E-3
Air Speed = 6.0 m/s				
Stirring: 0 rpm	5.49E-3	0.817	0.0300	6.72E-3
Stirring: 20 rpm	5.89E-3	0.846	0.0382	6.96E-3
Stirring: 50 rpm	6.02E-3	0.828	0.0350	7.27E-3
Stirring: 100 rpm	6.37E-3	0.839	0.0396	7.59E-3

^a See footnotes in Table 3 for calculations of k_L , $k_G H$, and α values. See also the pure-liquid $\bar{\beta}$ values in Table 3 for calculations of the k_G values.

Table S-2. K_L (cm/min) values of benzene, toluene, ethylbenzene, and *o*-xylene from their multi-solute solutions at 25 °C under various system settings^a

Air Speed = 0 m/s	Benzene	Toluene	Ethylbenzene	<i>o</i> -Xylene
Stirring: 0 rpm	9.6E-3	9.0 E-3	8.3E-3	7.9E-3
Stirring: 20 rpm	0.0506	0.0485	0.0483	0.0452
Stirring: 50 rpm	0.0802	0.0778	0.0725	0.0662
Stirring: 100 rpm	0.114	0.112	0.108	0.0899
Air Speed = 0.2 m/s				
Stirring: 0 rpm	0.0138	0.0129	0.0116	0.0114
Stirring: 20 rpm	0.0515	0.0500	0.0476	0.0432
Stirring: 50 rpm	0.0817	0.0804	0.0775	0.0702
Stirring: 100 rpm	0.119	0.119	0.109	0.0920
Air Speed = 2.0 m/s				
Stirring: 0 rpm	0.0246	0.0233	0.0219	0.0212
Stirring: 20 rpm	0.0536	0.0519	0.0494	0.0489
Stirring: 50 rpm	0.0842	0.0836	0.0801	0.0719
Stirring: 100 rpm	0.121	0.123	0.111	0.0989
Air Speed = 4.0 m/s				
Stirring: 0 rpm	0.0314	0.0307	0.0301	0.0279
Stirring: 20 rpm	0.0633	0.0565	0.0526	0.0510
Stirring: 50 rpm	0.0969	0.0953	0.0881	0.0756
Stirring: 100 rpm	0.126	0.124	0.118	0.101
Air Speed = 6.0 m/s				
Stirring: 0 rpm	0.0398	0.0385	0.0375	0.0349
Stirring: 20 rpm	0.0640	0.0582	0.0560	0.0535
Stirring: 50 rpm	0.0981	0.0971	0.0906	0.0798
Stirring: 100 rpm	0.130	0.127	0.121	0.107

^a See footnotes in Table 3 for the velocity gradients (G) at given stirring rates.