

## Supplementary Information.

Table S1. NPA Charges on Atoms in Substituted Staggered 1,1,1-Trifluoroethanes

X	F <sub>ap</sub>	F <sub>g</sub>	C of CF <sub>3</sub>	H of CH <sub>2</sub>	C of CH <sub>2</sub>	X
F	-0.414	-0.407	1.199	0.212	0.026	-0.420
SO <sub>2</sub> F	-0.405	-0.398	1.243	0.295	-0.775	0.142
OH	-0.419	-0.409	1.223	0.209	-0.110	-0.291
CF <sub>3</sub>	-0.412	-0.406	1.246	0.274	-0.594	0.022
H	-0.419	-0.419	1.233	0.247	-0.715	0.247
NH <sub>2</sub>	-0.416	-0.422	1.220	0.238	-0.302	-0.135
NF <sub>2</sub>	-0.414	-0.402	1.232	0.267	-0.344	-0.206
SiH <sub>3</sub>	-0.417	-0.419	1.241	0.268	-0.988	0.468
SiF <sub>3</sub>	-0.412	-0.412	1.246	0.283	-1.097	0.521
CH <sub>3</sub>	-0.418	-0.421	1.240	0.246	-0.518	0.044
CH <sub>2</sub> Li	-0.433	-0.427	1.232	0.223	-0.517	0.125
CH <sub>2</sub> F <sup>a</sup>	-0.416	-0.408 <sup>b</sup>	1.242	0.260 <sup>b</sup>	-0.558	0.048
		-0.421		0.253		
CH <sub>2</sub> F <sup>c</sup>	-0.411	-0.418	1.240	0.257	-0.555	0.050
CH <sub>2</sub> <sup>-</sup>	-0.459	-0.441	1.218	0.208	-0.530	-0.764
CF <sub>2</sub> <sup>-</sup>	-0.456	-0.425	1.236	0.228	-0.612	-0.744
BH <sub>3</sub> <sup>-</sup>	-0.460	-0.436	1.237	0.212	-0.756	-0.573
BF <sub>3</sub> <sup>-</sup>	-0.455	-0.430	1.241	0.224	-0.870	-0.508
B(OH) <sub>3</sub> <sup>-</sup>	-0.456	-0.440 <sup>d</sup>	1.243	0.217 <sup>d</sup>	-0.896	-0.513
		-0.434 <sup>d</sup>		0.229 <sup>d</sup>		
NH <sub>3</sub> <sup>+</sup>	-0.371	-0.401	1.233	0.287	-0.301	0.668
NF <sub>3</sub> <sup>+</sup>	-0.367	-0.380	1.241	0.313	-0.288	0.550
OH <sub>2</sub> <sup>+</sup>	-0.367	-0.397	1.214	0.286	-0.106	0.482

<sup>a</sup>The fluorine atom is gauche to the C-C bond of the trifluoroethane group. <sup>b</sup>Same side of the plane defined by the three carbons of the molecule as is the fluorine atom of the CH<sub>2</sub>F group. <sup>c</sup>The fluorine atom is antiperiplanar to the C-C bond of the trifluoroethane group. <sup>d</sup>The orientation of the hydrogens on the oxygens remove the plane of symmetry from this species.

**Table S2.** NPA Charges on Atoms in Substituted Eclipsed 1,1,1-Trifluoroethanes

X	F <sub>syn</sub>	F <sub>g</sub>	C of CF <sub>3</sub>	H of CH <sub>2</sub>	C of CH <sub>2</sub>	X
F	-0.398	-0.415	1.194	0.212	0.027	-0.418
SO <sub>2</sub> F	-0.400	-0.404	1.244	0.299	-0.775	0.083
OH	-0.398	-0.420	1.218	0.209	-0.111	-0.289
CF <sub>3</sub>	-0.403	-0.411	1.246	0.276	-0.596	0.023
H	-0.419	-0.419	1.230	0.251	-0.724	0.251
NH <sub>2</sub>	-0.420	-0.420	1.214	0.242	-0.306	-0.132
NF <sub>2</sub>	-0.395	-0.412	1.230	0.269	-0.342	-0.206
SiH <sub>3</sub>	-0.421	-0.417	1.241	0.273	-1.001	0.468
SiF <sub>3</sub>	-0.417	-0.411	1.246	0.288	-1.107	0.522
CH <sub>3</sub>	-0.419	-0.420	1.238	0.249	-0.524	0.046
CH <sub>2</sub> Li	-0.421	-0.433	1.227	0.227	-0.523	0.128
CH <sub>2</sub> <sup>-</sup>	-0.432	-0.455	1.210	0.213	-0.539	-0.757
CF <sub>2</sub> <sup>-</sup>	-0.411	-0.448	1.234	0.232	-0.616	-0.774
BH <sub>3</sub> <sup>-</sup>	-0.427	-0.453	1.235	0.218	-0.770	-0.569
BF <sub>3</sub> <sup>-</sup>	-0.421	-0.446	1.241	0.230	-0.882	-0.506
B(OH) <sub>3</sub> <sup>-</sup>	-0.434	-0.450 <sup>a</sup>	1.242	0.224 <sup>a</sup>	-0.850	-0.517
		-0.446 <sup>a</sup>		0.231 <sup>a</sup>		
NH <sub>3</sub> <sup>+</sup>	-0.412	-0.382	1.234	0.288	-0.305	0.670
NF <sub>3</sub> <sup>+</sup>	-0.388	-0.373	1.243	0.313	-0.285	0.551
OH <sub>2</sub> <sup>+</sup>	-0.405	-0.378	1.214	0.287	-0.111	0.486

<sup>a</sup>The orientation of the hydrogens on the oxygens remove the plane of symmetry from this species.

**Table S3.** NPA Charges on Atoms in Eclipsed CH<sub>3</sub>CH<sub>2</sub>X

X	H <sub>syn</sub>	H <sub>g</sub>	C of CH <sub>3</sub>	H of CH <sub>2</sub>	C of CH <sub>2</sub>	X
F	0.246	0.225	-0.697	0.186	0.083	-0.452
SO <sub>2</sub> F	0.241	0.244	-0.655	0.276	-0.697	0.071
OH	0.243	0.220	-0.671	0.180	-0.048	-0.325
CF <sub>3</sub>	0.238	0.232	-0.657	0.248	-0.523	-0.018
H	0.219	0.219	-0.658	0.219	-0.658	0.219
NH <sub>2</sub>	0.216	0.220	-0.648	0.213	-0.241	-0.171
NF <sub>2</sub>	0.250	0.231	-0.679	0.241	-0.268	-0.246
SiH <sub>3</sub>	0.212	0.226	-0.641	0.245	-0.948	0.428
SiF <sub>3</sub>	0.224	0.232	-0.647	0.263	-1.059	0.490
CH <sub>3</sub>	0.215	0.220	-0.648	0.217	-0.454	0.011
CH <sub>2</sub> Li	0.214	0.205	-0.648	0.194	-0.455	0.091
CH <sub>2</sub> <sup>-</sup>	0.207	0.177	-0.658	0.181	-0.471	-0.793
CF <sub>2</sub> <sup>-</sup>	0.233	0.187	-0.654	0.198	-0.540	-0.807
BH <sub>3</sub> <sup>-</sup>	0.206	0.181	-0.628	0.180	-0.702	-0.597
BF <sub>3</sub> <sup>-</sup>	0.217	0.187	-0.632	0.194	-0.809	-0.538
B(OH) <sub>3</sub> <sup>-</sup>	0.213	0.181 <sup>a</sup>	-0.626	0.184 <sup>a</sup>	-0.774	-0.555
		0.183 <sup>a</sup>		0.194 <sup>a</sup>		
NH <sub>3</sub> <sup>+</sup>	0.235	0.268	-0.690	0.261	-0.233	0.629
NF <sub>3</sub> <sup>+</sup>	0.266	0.280	-0.705	0.288	-0.199	0.502
OH <sub>2</sub> <sup>+</sup>	0.251	0.274	-0.718	0.262	-0.028	0.424

<sup>a</sup>The orientation of the hydrogens on the oxygens remove the plane of symmetry from this species.

**Table S4.** Numerical Values of the Potential Gradient for CH<sub>3</sub>CH<sub>2</sub>X<sup>a</sup>

X	Position	$\gamma$ (AIM)	$\gamma$ (NPA)	X	Position	$\gamma$ (AIM)	$\gamma$ (NPA)
F	ap	0.0045	-0.0531	e-CF <sub>3</sub>	syn	0.1108	0.0208
	g	0.0742	-0.0109		g	0.0617	-0.0141
OH	ap	-0.0140	-0.0519	H	ap	0.0052	
	g	0.0649	0.0096		g	0.0052	
CH <sub>3</sub>	ap	0.0123	-0.0573	e-CH <sub>2</sub> Li	syn	0.0155	-0.0363
	g	0.0116	-0.0447		g	-0.0275	-0.0870
CF <sub>3</sub>	ap	0.0479	-0.0233	NF <sub>2</sub>	ap	0.0005	-0.0424
	g	0.0965	0.0106		g	0.0971	0.0236
e-F	syn	0.1109	0.0614	CH <sub>2</sub> Fin	ap	0.014	-0.0523
	g	0.0224	-0.0087		gs	0.0859	0.0134
e-OH	syn	0.1121	0.0454		go	0.007	-0.0518
	g	0.0062	-0.0363	CH <sub>2</sub> Fout	ap	0.0456	-0.0260
NH <sub>2</sub>	ap	0.0234	-0.0367		g	0.0222	-0.0375
	g	0.0408	-0.0260	CF <sub>2</sub> H	ap	0.0073	-0.0505
CH <sub>2</sub> Li	ap	-0.0345	-0.0964		g	0.0785	0.0050
	g	-0.0048	-0.0584	e-CF <sub>2</sub> H <sup>b</sup>	ap	0.0637	
SiH <sub>3</sub>	ap	0.0195	-0.0549		g	0.0466	
	g	-0.0444	-0.0725	e-CH <sub>3</sub>	syn	-0.0011	-0.0567
SiF <sub>3</sub>	ap	0.0272	-0.0224		g	0.0012	-0.0453
	g	-0.0288	-0.0310	NH <sub>3</sub> <sup>+</sup>	ap	0.0812	0.0708
e-SiF <sub>3</sub>	syn	-0.0659	-0.0567		g	0.0534	0.0220
	g	0.0136	-0.0268	BH <sub>3</sub> <sup>-</sup>	ap	-0.0755	-0.1774
SO <sub>2</sub> F	ap	0.0638	-0.0021		g	-0.052	-0.0993
	g	0.1143	0.0221	NF <sub>3</sub> <sup>+</sup>	ap	0.1064	0.0962
e-SO <sub>2</sub> F	syn	0.061			g	0.1129	0.0812
	g	0.0992		CH <sub>2</sub> <sup>-</sup>	ap	-0.0930	-0.1709
e-SiH <sub>3</sub>	syn	-0.0822	-0.0856		g	-0.0329	-0.0923
	g	0.0042	-0.0601	CF <sub>2</sub> <sup>-</sup>	ap	-0.1134	-0.1786
Cl	ap	-0.0110	-0.0492		g	0.0266	-0.0424
	g	0.0256	-0.0196	e-NH <sub>3</sub> <sup>+</sup>	syn	0.0421	-0.0030
e-NF <sub>3</sub> <sup>+</sup>	syn	0.111	0.0654		g	0.0842	0.0575
	g	0.1114	0.0931	BF <sub>3</sub> <sup>-</sup>	ap	-0.0672	-0.1427
e-CH <sub>2</sub> <sup>-</sup>	syn	-0.0062	-0.0567		g	-0.034	-0.0613
	g	-0.0775	-0.1487	OH <sub>2</sub> <sup>+</sup>	ap	0.0927	0.0928
e-CF <sub>2</sub> <sup>-</sup>	syn	0.0597	0.0222		g	0.0522	0.0480
	g	-0.0784	-0.1435	e-BF <sub>3</sub> <sup>-</sup>	syn	-0.0308	-0.0333
O <sup>-</sup>	ap	-0.0677	-0.1451		g	-0.057	-0.1206
	g	0.0435	-0.0432	e-BH <sub>3</sub> <sup>-</sup>	syn	-0.0493	-0.0665
PH <sub>3</sub> <sup>+</sup>	ap	0.0694	0.0551		g	-0.0683	-0.1566
	g	-0.0082	-0.0221				

<sup>a</sup>An "e" in front of the X group designation indicates the eclipsed conformer. <sup>b</sup>The hydrogen atom in X is eclipsed with the C-C bond of the ethyl moiety.

**Table S5.** The Potential Gradient Values for CF<sub>3</sub>CH<sub>2</sub>X<sup>a</sup>

X	Position	$\gamma$ (AIM)	$\gamma$ (NPA)	X	Position	$\gamma$ (AIM)	$\gamma$ (NPA)
F	ap	0.0159	-0.0391	e-CF <sub>3</sub>	syn	0.1132	0.0581
	g	0.1143	0.0355		g	0.0794	0.0157
OH	ap	-0.0045	-0.0692	H	ap	0.0374	-0.0487
	g	0.1080	0.0152		g	0.0374	-0.0487
CH <sub>3</sub>	ap	0.0432	-0.0680	e-CH <sub>2</sub> Li	syn	0.0441	-0.0283
	g	0.0357	-0.0443		g	0.0025	-0.0931
CF <sub>3</sub>	ap	0.0760	-0.0305	NF <sub>2</sub>	ap	0.0158	-0.0563
	g	0.1077	-0.0007		g	0.1283	0.0191
e-F	syn	0.1535	0.0665	CH <sub>2</sub> Fin	ap	0.0420	-0.0597
	g	0.0447	-0.0170		gs	0.1027	0.0080
e-OH	syn	0.1562	0.0530		go	0.0192	-0.0571
	g	0.0273	-0.0446	CH <sub>2</sub> Fout	ap	0.0803	-0.0282
NH <sub>2</sub>	ap	0.0449	-0.0478		g	0.0380	-0.0392
	g	0.0722	-0.0257	e-NH <sub>3</sub> <sup>+</sup>	syn	0.0631	-0.0160
CH <sub>2</sub> Li	ap	-0.0050	-0.1106		g	0.1019	0.0468
	g	0.0281	-0.0543	e-CH <sub>3</sub>	syn	0.0309	-0.0331
SiH <sub>3</sub>	ap	0.0711	-0.0570		g	0.0379	-0.0605
	g	-0.0260	-0.0805	NH <sub>3</sub> <sup>+</sup>	ap	0.1148	0.0653
SiF <sub>3</sub>	ap	0.0746	-0.0164		g	0.0760	0.0042
	g	-0.0158	-0.0559	BH <sub>3</sub> <sup>-</sup>	ap	-0.0240	-0.1870
e-SiF <sub>3</sub>	syn	0.0243	-0.0586		g	-0.0177	-0.1026
	g	0.0530	-0.0486	e-SiH <sub>3</sub>	syn	-0.0738	-0.0942
SO <sub>2</sub> F	ap	0.0939	-0.0040		g	0.0439	-0.0637
	g	0.1092	-0.0015				

<sup>a</sup>See note *a* of Table S4.

**Table S6.** The Potential Gradient (AIM Method) for Fluorine Atoms in Compounds of the Type  $\text{CF}_3(\text{CH}_2)_n\text{NH}_3^{+a}$

Compound	Position	$\gamma$ (AIM)
$\text{CF}_3\text{CH}_2\text{NH}_3^+$	ap	0.0978
	g	0.0454
e- $\text{CF}_3\text{CH}_2\text{NH}_3^+$	syn	0.0261
	g	0.0808
$\text{CF}_3(\text{CH}_2)_2\text{NH}_3^+$	ap	0.0382
	g	0.0094
$\text{CF}_3(\text{CH}_2)_4\text{NH}_3^+$	ap	0.0520
	g	0.0041
$\text{CF}_3(\text{CH}_2)_4\text{NH}_3^{+b}$	a	-0.0248
	b	0.0829
	c	0.0195

<sup>a</sup>An e before the compound indicates an eclipsed conformer. <sup>b</sup>Twisted conformer as illustrated in Figure 2.