

Supporting Information for

Regioselective synthesis of
7-(trimethylsilylethynyl)pyrazolo[1,5-*a*]pyrimidines
***via* reaction of pyrazolamines with enynones**

Pavel Golubev, Ekaterina A. Karpova, Alena S. Pankova,* Mariia Sorokina, Mikhail A. Kuznetsov

*Institute of Chemistry, Saint Petersburg State University,
Universitetsky pr. 26, 198504, Saint Petersburg, Russia*

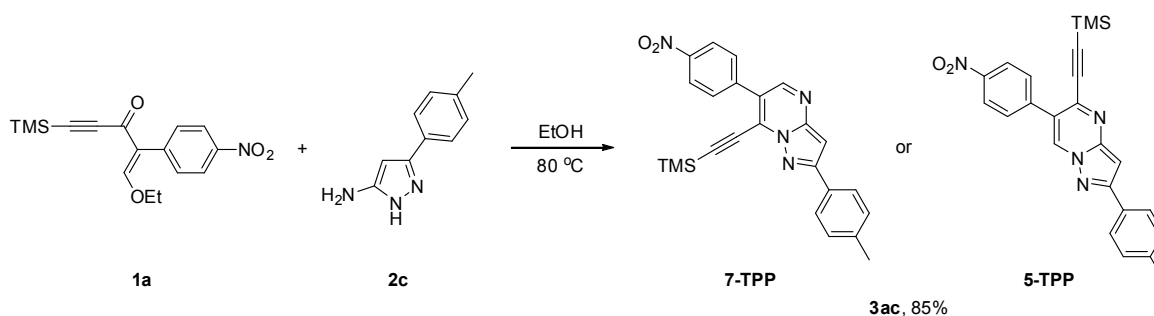
a.pankova@spbu.ru

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NMR structure evaluation for compound **3ac**

Initially the reaction was tested using ketone **1a** and pyrazolamine **2c**, and a single product was isolated in 85% yield. According to its ^1H and ^{13}C NMR spectra and HRMS data, this product was identified as a (trimethylsilylethynyl)pyrazolo[1,5-*a*]pyrimidine. However, position of the alkynyl substituent could not be established unambiguously, and the product could either be a 7-(trimethylsilylethynyl)pyrazolopyrimidine (**7-TPP**) or a 5-(trimethylsilylethynyl)pyrazolopyrimidine (**5-TPP**). It is necessary to mention that signals of H^3 ($\delta = 7.08$ ppm) and $\text{H}^{5(7)}$ ($\delta = 8.50$ ppm) atoms are easy to distinguish in its ^1H NMR spectrum.



First, ^1H - ^{15}N HMBC spectrum of the product was recorded (Figure 1). For both **5-TPP** and **7-TPP** one could expect to observe cross-peaks of H^3 proton with all three cyclic nitrogen atoms. But H^5 atom in **7-TPP** was expected to correlate with N^4 nitrogen atom (and probably with N^8), while H^7 proton in **5-TPP** would correlate with N^1 and N^8 nitrogen atoms (and probably with N^4).

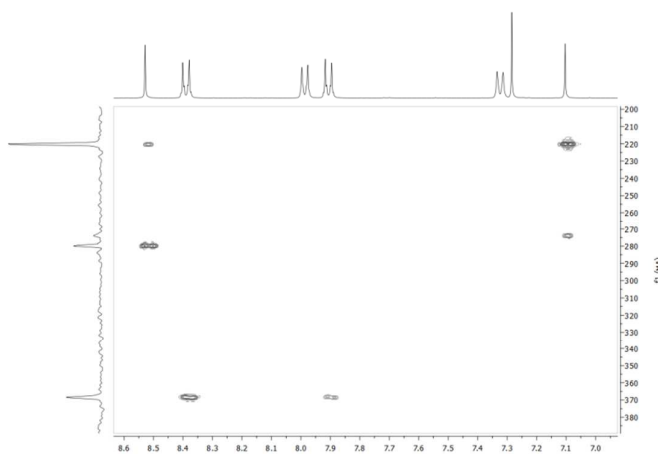


Figure 1. ^1H - ^{15}N HMBC (400 MHz) spectrum of compound **3ac**.

Projections of all four nitrogen atoms were visible in the spectrum at $\delta \approx 220$ (N^8), 270, 280 (N^1 and N^4) and 365 (NO_2) ppm. Unfortunately, not all cross-peaks were visible in the spectrum. Interaction between NO_2 and protons of the adjacent aromatic ring was clearly seen. Nodal N^8 atom correlates

with both H³ and H⁵⁽⁷⁾. Each of these protons correlates with one of N¹ or N⁴, but this is possible for **7-TPP** (H³-N¹⁽⁴⁾, H⁵-N⁴), as well as for **5-TPP** (H³-N¹⁽⁴⁾, H⁷-N¹). We then proposed that choice between the two structures could be made on the basis of ¹H-¹⁵N HSQMBC spectrum (Figure 2). Relying on the literature data¹ for ¹H-¹⁵N spin-spin coupling constants in pyrrole and pyridine, we predicted ¹H-¹⁵N constants sets for structures **7-TPP** and **5-TPP**. The key difference is a large N⁴-H⁵ constant (²J ≈ 10 Hz) in **7-TPP**, while all constants in **5-TPP** were expected to be lower.

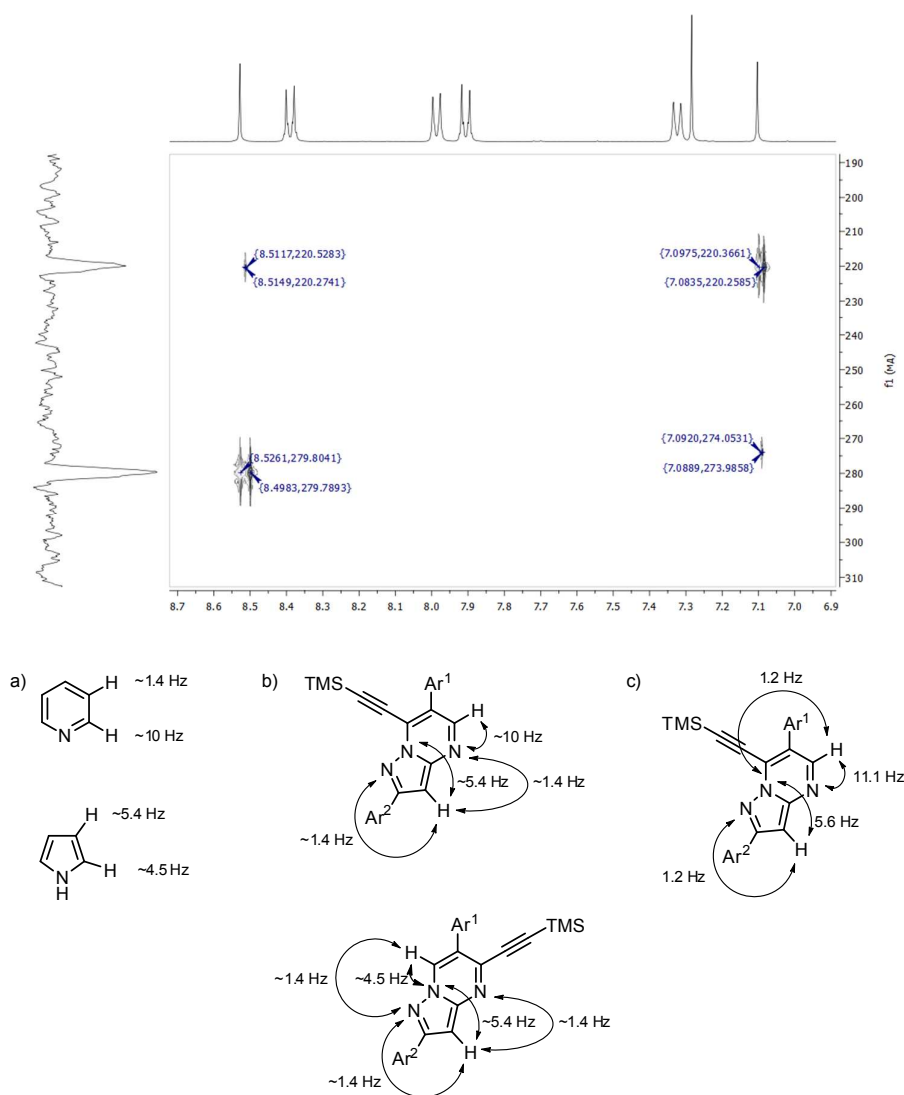


Figure 2. Top: ¹H-¹⁵N HSQMBC (400 MHz) spectrum of compound **3ac**. Bottom: N-H spin-spin coupling constants: a) literature data for pyridine and pyrrole;¹ b) predicted values for **7-TPP** and **5-TPP**; c) experimental data for the obtained product.

Experimental values of N-H spin-spin coupling constants were in agreement with those proposed for **7-TPP**, therefore the structure of 7-(trimethylsilylethynyl)pyrazolo[1,5-*a*]pyrimidine was ascribed to compound **3ac** and all obtained products.

X-Ray analysis data for compound 3aa

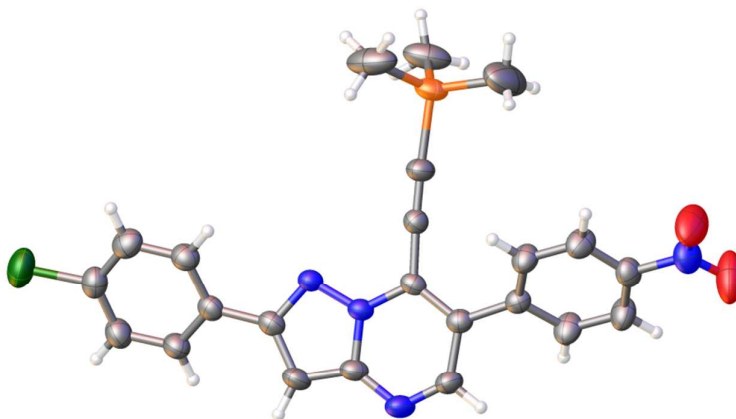


Figure 3. Olex2 representation of compound 3aa crystal structure.

Thermal ellipsoids are given at 50% probability level.

Single crystals of $C_{23}H_{19}ClN_4O_2Si$ were obtained from chloroform solution. A suitable crystal was selected and put on a diffractometer. The crystal was kept at 100(2) K during data collection. Using Olex2,² the structure was solved with the ShelXS³ structure solution program using Direct Methods and refined with the XL³ refinement package using CGLS minimization.

Crystal structure determination of 3aa

Crystal Data for $C_{23}H_{19}ClN_4O_2Si$ ($M = 446.96$ g/mol): monoclinic, space group Cc (no. 9), $a = 6.9864(6)$ Å, $b = 30.542(2)$ Å, $c = 11.0261(8)$ Å, $\beta = 106.329(8)^\circ$, $V = 2257.8(3)$ Å³, $Z = 4$, $T = 100(2)$ K, $\mu(\text{MoK}\alpha) = 0.249$ mm⁻¹, $D_{\text{calc}} = 1.315$ g/cm³, 10094 reflections measured ($6.222^\circ \leq 2\theta \leq 52.99^\circ$), 4196 unique ($R_{\text{int}} = 0.0175$, $R_{\text{sigma}} = 0.0223$) which were used in all calculations. The final R_1 was 0.0339 ($I > 2\sigma(I)$) and wR_2 was 0.1095 (all data).

Refinement model description

Number of restraints - 2, number of constraints - unknown.

Details:

1. Fixed Uiso

At 1.2 times of:

All C(H) groups

At 1.5 times of:

All C(H,H,H) groups

2.a Aromatic/amide H refined with riding coordinates:

C12(H12), C14(H14), C15(H15), C16(H16), C17(H17), C18(H18), C21(H21),
C22(H22), C24(H24), C27(H27)

2.b Idealised Me refined as rotating group:

C26(H26A,H26B,H26C), C29(H29A,H29B,H29C), C36(H36A,H36B,H36C)

Table 1. Crystal data and structure refinement for 3aa

Empirical formula	C ₂₃ H ₁₉ ClN ₄ O ₂ Si
Formula weight	446.96
Temperature/K	100(2)
Crystal system	monoclinic
Space group	Cc
a/Å	6.9864(6)
b/Å	30.542(2)
c/Å	11.0261(8)
α/°	90
β/°	106.329(8)
γ/°	90
Volume/Å ³	2257.8(3)
Z	4
ρ _{calc} /g/cm ³	1.315
μ/mm ⁻¹	0.249
F(000)	928.0
Crystal size/mm ³	0.22 × 0.20 × 0.15
Radiation	MoKα (λ = 0.71073)
2θ range for data collection/°	6.222 to 52.99
Index ranges	-8 ≤ h ≤ 8, -35 ≤ k ≤ 38, -13 ≤ l ≤ 13
Reflections collected	10094
Independent reflections	4196 [R _{int} = 0.0175, R _{sigma} = 0.0223]
Data/restraints/parameters	4196/2/283
Goodness-of-fit on F ²	0.884
Final R indexes [I ≥ 2σ (I)]	R ₁ = 0.0339, wR ₂ = 0.1052
Final R indexes [all data]	R ₁ = 0.0362, wR ₂ = 0.1095
Largest diff. peak/hole / e Å ⁻³	0.21/-0.18
Flack parameter	0.034(19)

Table 2. Fractional Atomic Coordinates (×10⁴) and Equivalent Isotropic Displacement Parameters (Å²×10³) for 3aa. U_{eq} is defined as 1/3 of the trace of the orthogonalised U_{ij} tensor.

Atom	x	y	z	U(eq)
Si1	519.0(13)	1198.7(3)	6402.9(8)	48.7(2)
Cl1	-10664.7(14)	2960.5(4)	-76.3(12)	77.4(3)
N2	-937(3)	1445.1(7)	1443.0(19)	33.6(4)
N3	-2422(3)	1681.9(7)	1687(2)	35.5(5)
C4	468(4)	1209.1(7)	2328(2)	32.1(5)

C5	1852(4)	982.9(8)	1899(3)	35.7(5)
C6	412(4)	1229.2(8)	3610(3)	36.1(6)
C7	-3393(4)	1852.8(9)	549(2)	37.9(6)
C8	3398(4)	697.2(8)	2696(2)	34.9(5)
C9	-5151(4)	2133.9(9)	432(3)	37.4(5)
C10	430(4)	1232.5(10)	4697(3)	43.5(7)
C11	-962(5)	1462.8(9)	179(3)	40.2(6)
C12	-6312(5)	2093(1)	1260(3)	44.0(6)
C13	-8516(5)	2646.5(10)	142(3)	49.6(7)
C14	2981(4)	407.1(9)	3566(3)	45.3(7)
C15	-5700(5)	2443.2(10)	-531(3)	43.8(6)
C16	1728(4)	1021.5(10)	586(3)	42.7(6)
C17	5306(5)	696.9(11)	2545(3)	50.5(7)
C18	-2540(5)	1729.1(10)	-395(3)	44.7(6)
N19	419(4)	1243.9(9)	-242(2)	48.1(6)
C20	6284(4)	125.7(9)	4045(3)	45.1(6)
C21	-7996(5)	2348.5(11)	1128(3)	51.3(7)
C22	4426(5)	118.8(10)	4248(3)	50.6(7)
N23	7822(5)	-177.7(10)	4767(3)	61.3(8)
C24	-7376(5)	2700.2(10)	-680(3)	49.4(7)
O25	9526(5)	-129.5(13)	4683(5)	102.6(13)
C26	1283(9)	1736.6(15)	7134(4)	78.0(13)
C27	6754(4)	409.8(11)	3215(3)	51.9(7)
O28	7336(5)	-467.6(12)	5361(4)	96.1(12)
C29	-2024(8)	1052(2)	6463(4)	94.8(18)
C36	2383(11)	776(2)	7099(5)	113(2)

Table 3. Anisotropic Displacement Parameters ($\text{\AA}^2 \times 10^3$) for 3aa. The Anisotropic displacement factor exponent takes the form: $-\pi^2 [h^2 a^{*2} U_{11} + 2hka^*b^* U_{12} + \dots]$

Atom	U_{11}	U_{22}	U_{33}	U_{23}	U_{13}	U_{12}
Si1	66.5(5)	52.2(5)	28.1(4)	3.5(3)	14.2(4)	0.5(4)
Cl1	53.5(5)	85.1(7)	94.9(8)	3.8(5)	23.0(5)	22.3(5)
N2	39.6(11)	35.4(10)	26.5(10)	0.5(8)	10.3(9)	-0.1(9)
N3	40.1(11)	38.9(10)	27.9(10)	2.1(8)	10.3(9)	4.6(9)
C4	37.6(12)	30.2(11)	27.1(12)	1.2(8)	6.7(10)	-4.4(9)
C5	38.8(12)	35.7(12)	34.3(13)	0.1(10)	12.9(11)	-3.4(11)
C6	37.4(12)	36.6(13)	34.0(14)	2.0(9)	9.6(11)	2.1(10)
C7	40.3(13)	39.1(12)	31.9(13)	0.2(10)	6.3(11)	-3.2(10)
C8	37.3(12)	32.8(11)	35.7(13)	-1.2(9)	12.4(11)	-1.6(10)
C9	36.2(12)	39.9(13)	33.2(13)	-1.1(10)	5.2(11)	-3.3(10)
C10	52.1(16)	48.6(15)	31.1(15)	1.8(10)	14.2(13)	5.4(12)
C11	50.7(15)	43.4(13)	27.9(12)	1.9(10)	13.2(12)	2.4(12)
C12	45.4(15)	45.9(14)	40.9(15)	4.9(11)	12.2(13)	-2.9(12)
C13	40.0(14)	49.0(15)	57.0(19)	-1.1(13)	9.1(14)	3.3(12)
C14	41.5(14)	45.0(15)	54.6(17)	9.9(12)	21.9(13)	2.3(11)
C15	43.1(14)	51.8(15)	35.9(14)	6.2(11)	10.1(12)	1.2(12)

C16	48.9(15)	47.4(15)	35.6(14)	0.2(11)	18.2(12)	7.7(12)
C17	45.5(15)	47.7(16)	64.4(19)	14.5(13)	25.4(15)	1.2(12)
C18	52.7(15)	53.2(15)	26.1(12)	3.0(11)	7.4(12)	5.5(13)
N19	60.6(16)	55.2(14)	31.5(13)	0.7(10)	17.7(12)	10.7(12)
C20	45.9(15)	38.3(14)	49.1(16)	2.6(12)	10.0(13)	7.5(11)
C21	45.6(16)	56.4(17)	56.0(19)	-2.6(14)	20.7(14)	-6.3(13)
C22	58.1(18)	41.8(15)	58.7(19)	13.8(13)	27.6(15)	7.4(13)
N23	60.5(18)	57.6(16)	64.0(18)	11.3(14)	14.3(15)	19.1(13)
C24	46.0(15)	51.8(16)	47.8(17)	9.1(13)	8.8(13)	4.9(13)
O25	49.9(16)	99(2)	155(4)	50(2)	22.6(19)	21.1(16)
C26	126(4)	70(2)	38.7(17)	-2.8(15)	25(2)	-21(2)
C27	38.5(14)	54.6(17)	66(2)	12.8(15)	20.2(14)	6.3(13)
O28	94(2)	100(2)	106(3)	60(2)	48(2)	46(2)
C29	106(4)	134(5)	56(3)	-24(3)	42(3)	-45(4)
C36	163(6)	116(4)	57(3)	28(3)	25(3)	65(4)

Table 4. Bond Lengths for 3aa

Atom	Atom	Length/Å	Atom	Atom	Length/Å
Si1	C10	1.867(3)	C8	C17	1.389(4)
Si1	C26	1.842(4)	C9	C12	1.386(4)
Si1	C29	1.852(5)	C9	C15	1.392(4)
Si1	C36	1.840(5)	C11	C18	1.373(4)
Cl1	C13	1.740(3)	C11	N19	1.358(4)
N2	N3	1.352(3)	C12	C21	1.385(5)
N2	C4	1.377(3)	C13	C21	1.386(5)
N2	C11	1.389(3)	C13	C24	1.374(5)
N3	C7	1.352(3)	C14	C22	1.391(4)
C4	C5	1.375(4)	C15	C24	1.380(5)
C4	C6	1.427(4)	C16	N19	1.289(4)
C5	C8	1.472(4)	C17	C27	1.385(4)
C5	C16	1.431(4)	C20	C22	1.378(5)
C6	C10	1.195(4)	C20	N23	1.472(4)
C7	C9	1.474(4)	C20	C27	1.367(4)
C7	C18	1.389(4)	N23	O25	1.229(5)
C8	C14	1.395(4)	N23	O28	1.205(4)

Table 5. Bond Angles for 3aa

Atom	Atom	Atom	Angle/°	Atom	Atom	Atom	Angle/°
C26	Si1	C10	108.38(16)	C15	C9	C7	120.4(3)
C26	Si1	C29	111.2(3)	C6	C10	Si1	176.1(3)
C29	Si1	C10	106.80(17)	C18	C11	N2	105.5(2)
C36	Si1	C10	106.2(2)	N19	C11	N2	120.9(2)
C36	Si1	C26	111.0(3)	N19	C11	C18	133.6(2)
C36	Si1	C29	112.9(3)	C21	C12	C9	121.2(3)
N3	N2	C4	124.95(19)	C21	C13	C11	119.7(3)
N3	N2	C11	112.6(2)	C24	C13	C11	118.7(3)

C4	N2	C11	122.4(2)	C24	C13	C21	121.5(3)
C7	N3	N2	103.6(2)	C22	C14	C8	120.9(3)
N2	C4	C6	117.8(2)	C24	C15	C9	121.3(3)
C5	C4	N2	116.7(2)	N19	C16	C5	126.1(3)
C5	C4	C6	125.4(2)	C27	C17	C8	121.2(3)
C4	C5	C8	124.2(2)	C11	C18	C7	105.8(2)
C4	C5	C16	117.2(2)	C16	N19	C11	116.6(2)
C16	C5	C8	118.6(2)	C22	C20	N23	118.7(3)
C10	C6	C4	177.1(3)	C27	C20	C22	122.2(3)
N3	C7	C9	119.6(2)	C27	C20	N23	119.1(3)
N3	C7	C18	112.4(3)	C12	C21	C13	118.7(3)
C18	C7	C9	127.9(3)	C20	C22	C14	118.4(3)
C14	C8	C5	121.8(2)	O25	N23	C20	117.3(3)
C17	C8	C5	119.8(2)	O28	N23	C20	118.8(3)
C17	C8	C14	118.4(3)	O28	N23	O25	123.8(3)
C12	C9	C7	121.1(3)	C13	C24	C15	118.9(3)
C12	C9	C15	118.4(3)	C20	C27	C17	118.8(3)

Table 6. Hydrogen Atom Coordinates ($\text{\AA}\times 10^4$) and Isotropic Displacement Parameters ($\text{\AA}^2\times 10^3$) for 3aa

Atom	<i>x</i>	<i>y</i>	<i>z</i>	U(eq)
H12	-5954	1890	1914	53
H14	1719	407	3691	54
H15	-4923	2477	-1084	53
H16	2680	872	303	51
H17	5616	893	1984	61
H18	-2953	1810	-1242	54
H21	-8763	2321	1690	62
H22	4146	-74	4828	61
H24	-7727	2906	-1325	59
H26A	2695	1772	7283	117
H26B	957	1754	7923	117
H26C	594	1964	6579	117
H27	8024	410	3102	62
H29A	-2839	985	5627	142
H29B	-2595	1295	6794	142
H29C	-1957	802	7000	142
H36A	2147	522	6567	170
H36B	2278	699	7922	170
H36C	3695	888	7172	170

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Copies of the NMR spectra

