

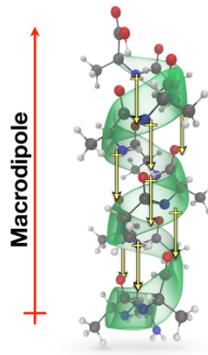
Origins of Negative and Positive Electromechanical Response of Oligopeptide Piezoelectrics

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ABSTRACT In response to an applied electric field, materials and even individual molecules can exhibit electromechanical response. Previous work has demonstrated the piezoelectric distortion of polar organic crystals, biomaterials, and even single molecular monolayers, increasing length in response to the change in potential energy of interaction between the polarization or dipole moment and the applied electric field. In this work, we demonstrate through density functional calculations that helical oligopeptides, depending on sequence can exhibit both positive and surprisingly also negative piezoresponse. While the overall molecular dipole moment may favor positive piezoresponse, individual hydrogen bonding interactions along the backbone have the opposite direction, enabling tailored sequences to yield either net expansion or contraction in response to an electric field. While negative piezoresponse has largely been considered in polyvinylidene difluoride (PVDF) and related systems, understanding the atomic and molecular basis for both electromechanical effects will yield interesting new applications.

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Introduction

Piezoelectric materials generate electric charge in response to physical stresses (the direct piezoelectric effect) or through deformation under an electric field (the inverse piezoelectric effect). Due to this property of piezoelectric materials, they have caught the attention of researchers who are interested in improving energy efficiency through a different method of energy harvesting. For example, piezoelectric materials find wide uses ranging from consumers to medical applications such as force sensors¹, actuators¹, and pacemakers². On the other hand, a piezoelectric actuator directly converts an electrical signal into a precisely controlled physical displacement. For example, medical implant devices use piezoelectric responses to monitor temperature or blood pressure.³ Piezoelectric motors can be powered by converting the vibrations of piezoelectric materials⁴ for applications such navigating through the bloodstream.

The geometric change of a material in the presence of electric fields determines the piezoresponse of a material. To date, the most highly responsive piezo materials are ceramics (e.g., lead zirconium titanate, PZT) and semi-crystalline polymers such as polyvinylidene difluoride (PVDF).⁵ While a piezoresponse is found in many polar inorganic materials, the piezoelectric effect is also found in a number of biomaterials such as peptides and amino acids. The assembly of proteins show adaptation due to environment changes (for example redox, pH, and chemical gradient), and these characteristics extend to applied electric fields.⁵ For example, collagen,⁶ viruses,⁷ aligned peptide crystals,⁸ horse muscle,⁹ fish scales,¹⁰ and peptide nanotubes,¹¹ exhibit

piezoresponse in prior studies. Despite both experimental and computational studies, full understanding of what molecular properties yield large piezo response does not yet exist.

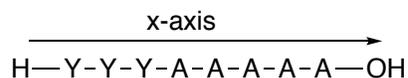
Previous computational studies have demonstrated the piezoelectric effect in hydrogen-bonded molecules^{7, 12-13} and individual molecules.⁷ Recent experimental studies have measured piezo response in individual molecular monolayers using piezo force microscopy.⁵ While such experiments have demonstrated the effect of molecular flexibility on piezoresponse in oligopeptides, computational studies can guide further experimental investigation, including the effect of amino acid side chains on piezo response.

In this work, we particularly focus on substitutions with two particular amino acids: tyrosine and phenylalanine in helical oligoalanine backbones. Both add polarizable aromatic phenyl rings, potentially increasing the electrostatic molecular response to an applied electric field. The sole difference is in the additional hydrogen bonding -OH group in tyrosine. Additionally, the effects of oligopeptide length and repeated phenylalanine or tyrosine substitutions will be considered.

Computational Methods

Adopting methods from our previous work on π -conjugated helicenes,⁷ all molecular geometries were optimized using the B3LYP density functional¹⁴⁻¹⁵ and the 6-31G(d) basis set using Gaussian 09.¹⁶ While little experimental data is yet available for comparison with computed piezoelectric response, the B3LYP functional has generally been the method of choice for accurate molecular geometries, and focus mainly on the structure/property trends with the computed geometric deformations.¹⁷⁻¹⁸ When our group compared computed to experimental piezoelectric coefficients of crystalline 2-methyl-4-nitroaniline, high agreement was found between computational and experimental piezoelectric responses (in hydrogen bonding interactions) when using the B3LYP functional.¹⁹ Since DFT methods are known to be asymptotically incorrect and ignore dispersion,²⁰⁻
²⁶ we also tested results with the ω -B97X-D3 functional.²⁷

All oligopeptides are indicated using standard three-letter abbreviations, for example Tyr₃-Ala₅ (Scheme 1), represents the neutral peptide aligned along the x-axis, with the origin at the terminal NH₂ group, and the tyrosine residues at the bottom of the peptide structure. Each sequence investigated is compiled in Table S1. For the present study, initial geometries were generated using the Avogadro molecular editor,¹⁸ assuming a helical secondary structure, followed by density functional optimization as described above.



Scheme 1. Amino acid notation of Tyr₃-Ala₅ sequence which was optimized on x-direction.

As in our previous work, the molecular geometry is oriented to a specific frame of reference (i.e., aligned along the Cartesian x-axis) using Avogadro¹⁸ or a custom Python script (see Supporting Information), and the geometry was optimized under different applied electric fields along the molecular (x) axis. The molecular length after optimization under each electric field was used to calculate piezocoefficient (d_{33}).

The piezoelectric coefficient is a tensor derived from the direction of the applied field and the resulting molecular geometric distortion.²⁸ We restrict our discussion to the largest such contribution, and the piezoelectric coefficient, d_{33} , was calculated from the molecular length at maximum extension (l_{\max}) and at zero applied field (l_{zero}) using Equation 1:

$$d_{33} \left[\frac{\text{pm}}{\text{V}} \right] = \frac{(l_{\max} - l_{\text{zero}})}{l_{\text{zero}}} \times \frac{1}{\text{field}} \left[\frac{\text{nm}}{\text{V}} \right] \times 1000 \left[\frac{\text{pm}}{\text{nm}} \right] \quad (1)$$

Note that while this formula suggests a linear response of the piezoelectric distortion to the applied electric field, there is no guarantee that all molecules will exhibit linear electromechanical responses.

For example, Figure 1 displays the deformation of the Phe₃-Ala₅ oligopeptide sequence under different applied electric fields. As we increase the field strength, the Phe₃-Ala₅ oligopeptide increases in length, yielding a computed piezo response of 15.23 pm/V. Meanwhile, Figure 2 displays the deformation of the Ala₉-Phe₃ oligopeptide sequence under applied electric fields. As

we increase the field strength, this oligopeptide contracts, yielding a computed piezo response of -0.75 pm/V .

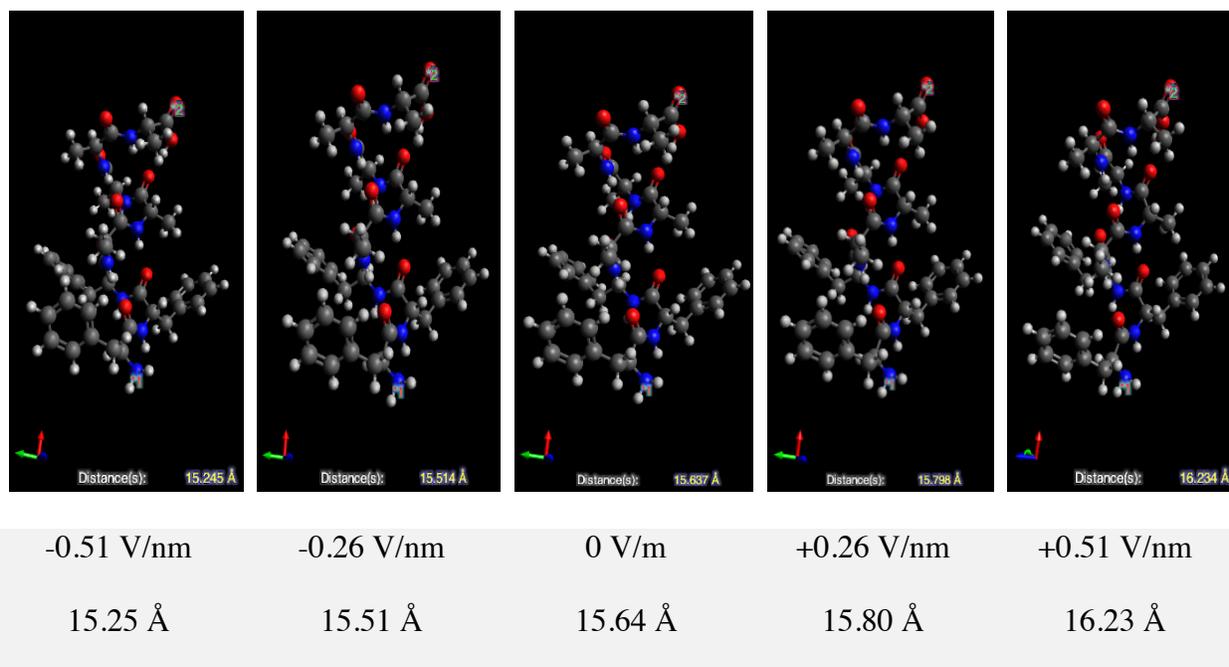


Figure 1. B3LYP-optimized geometries of Phe₃-Ala₅ with varied applied electric fields, with the N to O distance indicated. Note that the molecule extends with increasing electric field, reflecting a positive d_{33} piezo response (15.23 pm/V).

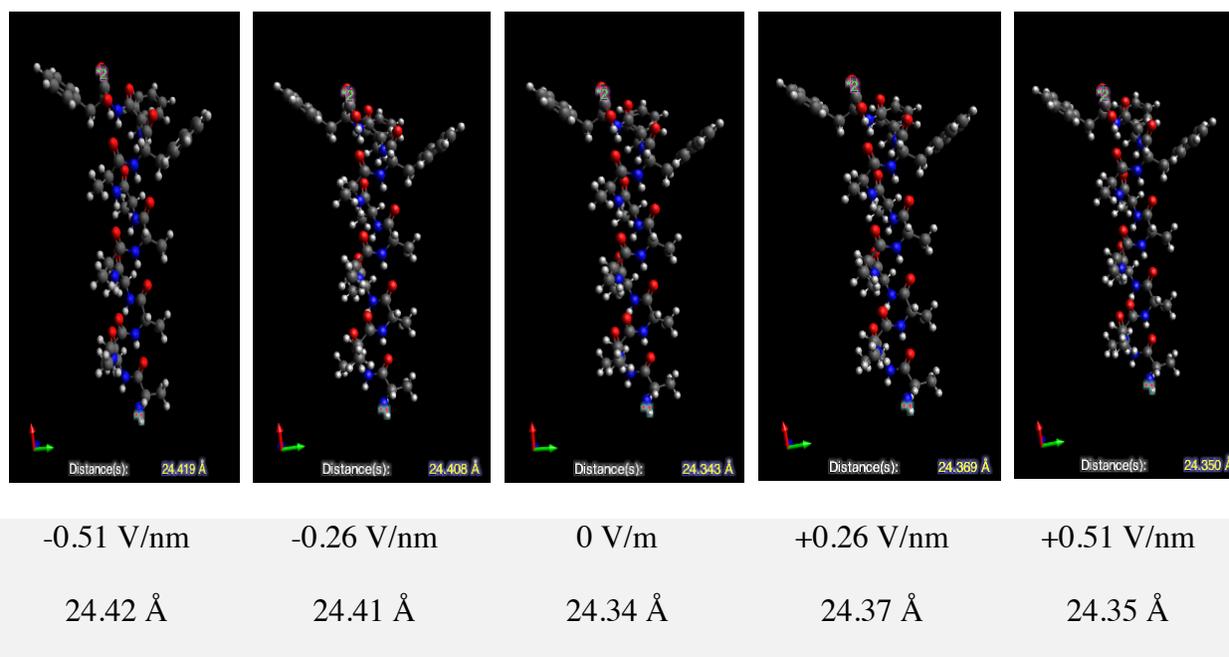


Figure 2. B3LYP-optimized geometries of Ala₉-Phe₃ with varied applied electric fields, with the N to O distance indicated. Note that with the same alignment, the molecule contracts with increasing electric field, reflecting a negative piezo response (-0.75 pm/V).

Multiple helical oligopeptide sequences were studied, including both tyrosine (Tyr) and phenylalanine (Phe) substitution of oligoalanines to consider the effect of polarizable side chains. The number of amino acids along the helix is also varied from 8 to 14 residues. All sequences and naming conventions are compiled in Table S1.

Results and Discussion

The overall purpose of this study is to understand the molecular properties of peptides that give rise to a large piezoresponse by simulating the deformation of helical oligopeptide structures. In particular, the use of tyrosine and phenylalanine side chains should introduce polarizable side chains, presumably increasing the induced dipole moment under applied field and thus the electrostatic interaction involved in piezo response of the peptides. Such Tyr and Phe substitutions will be considered at the top (C-terminus), at the bottom (N-terminus) as well as in the middle of the sequence of alanine residues.

Based on the optimized geometries under an applied electric field (e.g., Figures 1 and 2), one can compute a piezo response using Equation 1 (i.e., from the slope of the distance-field plots in Figure 3). Surprisingly, different sequences yield completely different *signs* of the piezo response: from a large positive response (expansion) to a small negative response (contraction).

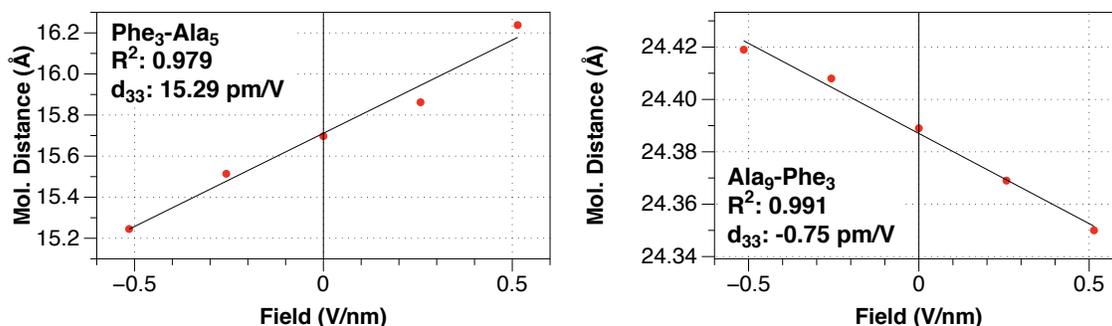


Figure 3. Plots of the length of the optimized geometries from peptide sequence (left) Phe₃-Ala₅ with positive piezocoefficient and (right) Ala₉-Phe₃ with negative piezo coefficient calculated using B3LYP.

To understand the effect of substitution on piezo response, first three Phe or Tyr residues were substituted at the bottom (N-terminus) of the sequences. As illustrated in Figure 4, the B3LYP-computed d_{33} are compared as a function of sequence length and side-chain substitution. In general, both Phe and Tyr-substituted sequences show similar computed piezo response. The largest piezo response of +15.29 pm/V is found in Phe₃-Ala₅ structures (at eight residues), followed by Tyr₃-Ala₅ (at eight residues) with +14.68 pm/V. In one case, Phe₃-Ala₉ yields negative piezo response of -0.61 pm/V.

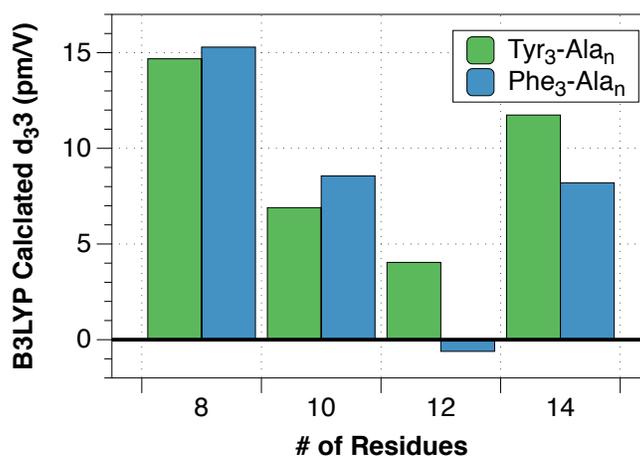


Figure 4. The resulting B3LYP-computed d_{33} obtained from phenylalanine and tyrosine substitutions at the bottom of the peptide structure (N terminus) e.g., Phe₃-Ala_n or Tyr₃-Ala_n.

Similarly, the B3LYP-computed piezo response was considered for substitution at the top (C-terminus) of the sequences (Figure 5). The largest positive piezo-coefficient of +7.7 pm/V is found in Ala₁₁-Phe₃ with a similarly large positive response in Ala₅-Tyr₃ (+7.46 pm/V). On the other hand, both types of substitutions show very small positive and negative piezo response at 8 residues, 10 residues, and 12 residues. Interestingly, both Phe and Tyr substitutions exhibit negative piezo-coefficient of -0.75 pm/V is found with Ala₉-Phe₃ (at 12 residues) and -0.73 pm/V is found with Ala₉-Tyr₃. Given only two outliers yielding strong positive piezo response, the C-terminal substitutions evidently strongly decrease the computed response. Even in the case of

Ala₁₁-Phe₃ and Ala₅-Tyr₃, the computed response is much smaller than the corresponding N-terminal substitutions (Figure 4).

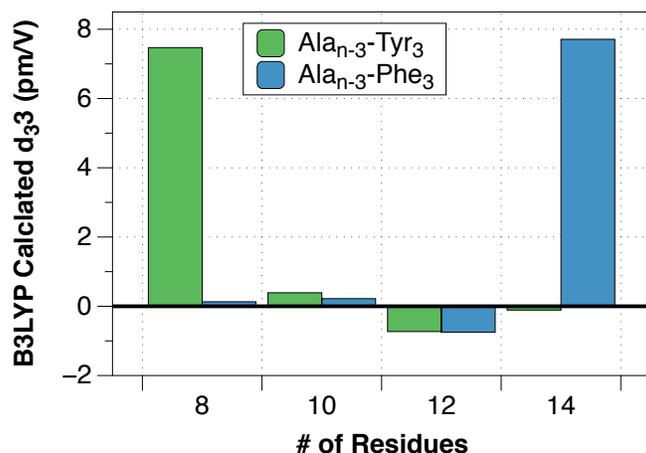


Figure 5. The resulting d_{33} obtained from phenylalanine and tyrosine with added residues. Both phenylalanine and tyrosine are located at the top of the peptide structure, (C terminus) e.g., Ala_n-Phe₃ or Ala_n-Tyr₃.

Given the small electromechanical deformations with C-terminal substitutions, clearly the overall molecular length does not show significant distortion. Smaller local effects may occur, for example between the C=O and NH hydrogen-bonding interactions along the helix, since those have been previously found to exhibit piezo response.²⁹⁻³¹ In Figure 6, the average distance between O and N hydrogen-bonding neighbors for Phe₃-Ala₅ and Ala₉-Phe₃ are compared as a function of electric field strength. The plot shows a *negative* linear trend (i.e., contraction) for both sequences. Despite the different overall piezo response from the two molecules, the hydrogen bonds contract, consistent with recent analysis of negative piezo response in hydrogen bond pairs.²⁹

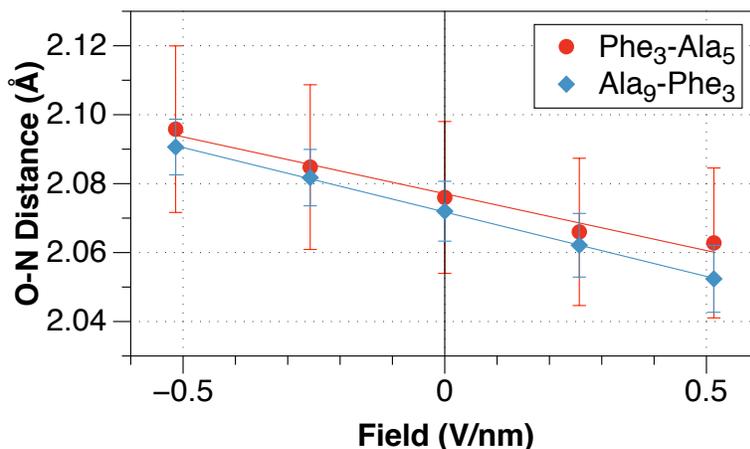


Figure 6. Average O-N hydrogen bond distances obtained from along the oligopeptides Phe₃-Ala₅ and Ala₉-Phe₃.

Since the hydrogen bonds contract with similar trends regardless of the sequence, one might wonder how the overall piezo response still favor positive expansion. The macrodipole along the helix strongly interacts with the applied electric field, producing potential energy driving towards an expanded length. These effects can be revealed through the backbone torsion angles ϕ and φ . Using a Python script to extract the Ramachandran angles, Figure 7 illustrates that the ϕ backbone angles from B3LYP-computed optimized geometries change significantly as a function of applied field strength in both oligopeptides with positive and negative computed piezo response. In the case of a ϕ backbone angle shift from $\sim 61^\circ$ to $\sim 59^\circ$, the helix elongates with increasing applied electric field (i.e., a positive piezo response from the helix macrodipole coupling to the field).

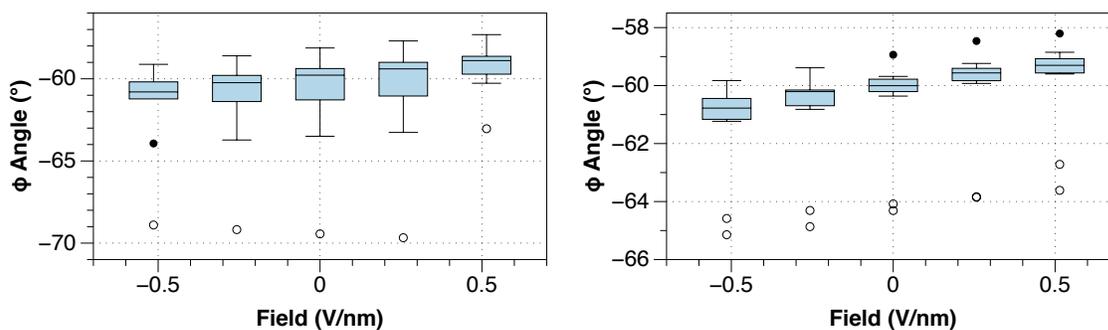


Figure 7. Ramachandran angle ϕ for B3LYP-optimized geometries of (left) Tyr₃-Ala₁₁ and (right) Ala₁₁-Tyr₃ as a function of applied electric field.

The geometric comparisons in Figures 6 and 7 illustrate that regardless of the overall computed piezo response (i.e., positive or negative), the peptide sequences respond in similar ways to the applied electric fields. The hydrogen bonds contract slightly with positive electric fields, consistent with negative piezo response.²⁹ At the same time, the ϕ dihedral angles along the chain twist towards less negative values (i.e. a longer helix). In the case of top-substituted C-terminal sequences, steric hindrance evidently results in a trend towards negative overall piezo response (e.g., Ala₉-Phe₃ in Figures 2 and 3).

Consequently, sequences with “middle” substitutions were also considered, for example, Ala-Tyr-Ala and Ala-Phe-Ala, in which the two types of side chains are buried inside the oligoalanine helix (Figure 8), nearer to the bottom of the sequence (N-terminus) or nearer to the C-terminus. Note that, both phenylalanine and tyrosine have similar trends. The largest piezo-coefficient is found in Ala-Phe₃-Ala₄ (at eight residues), $d_{33} \approx 15.40$ pm/V, with response typically decreasing with sequence length. In all cases, the computed piezo response is positive, although substitution near the C-terminus with eight- and ten-residue sequences yields smaller computed response than the corresponding N-terminus.

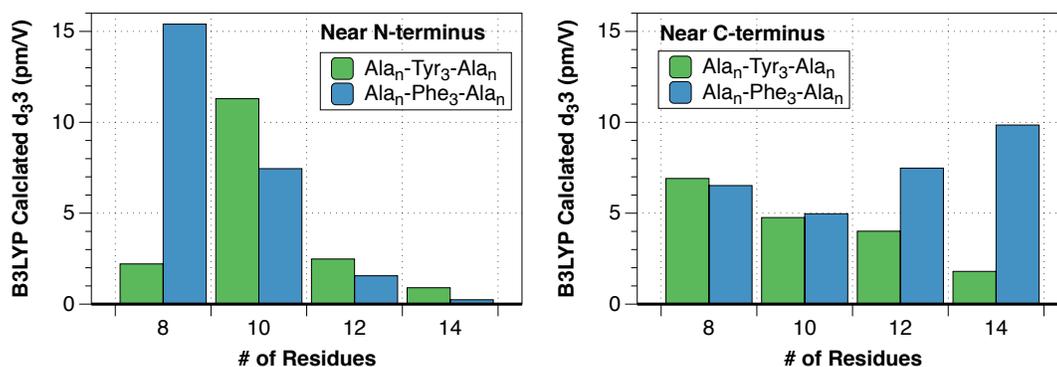


Figure 8. Computed B3LYP d_{33} piezo response as a function of oligopeptide length and “internal” side chain substitution closer to the (left) N-terminus e.g., Ala-Tyr₃-Ala₄, Ala-Phe₃-Ala₄, Ala₂-Tyr₃-Ala₅, Ala₂-Phe₃-Ala₄, etc., or (right) C-terminus, e.g., Ala₄-Tyr₃-Ala, Ala₄-Phe₃-Ala, Ala₅-Tyr₃-Ala₂, Ala₅-Phe₃-Ala₂ etc.

Conclusions

Using density functional optimized geometries of oligopeptides in response to applied electric fields along the peptide helix, the deformation of sequences containing alanine, phenylalanine, and tyrosine were compared. The length of the oligopeptides and the nature of the sequences yielded both positive and negative piezoresponse. Some variability between particular sequences likely occurs from the large number of geometry optimization calculations with large molecules – with many local minima. Although every effort was made to ensure all geometries are the global minima for the sequence and applied field, some errors may occur.

Overall, we note very similar computed piezo coefficients from both Tyr and Phe substitutions and no clear effect of sequence length on piezo response. On the other hand, we find that two measures of structural distortion – the distance between O and NH hydrogen bond atoms along the helix, and the ϕ backbone angles show opposing trends. That is, the local effects of hydrogen bonds act to *contract* the oligopeptide sequence (i.e., a *negative* piezo response) matching recent research on the electromechanical response of h-bonds.²⁹ At the same time, the macrodipole associated with the helical secondary structure push the backbone angles to elongate, a *positive* piezo response. In cases with Tyr or Phe substitution near the C-terminus, this latter effect is diminished, reflecting an overall negative piezo contraction (e.g., Figures 2, 3, 5). In remaining cases with substitution near the N-terminus or internal substitution, the overall piezo response is positive, with d_{33} piezo coefficients only slightly smaller than PVDF.

There is an increasing interest in the electromechanical response of organic and biomaterials,^{5, 32-33} and computational design of peptides with increased response will guide efforts to produce new active materials for applications including touch and force sensors, and electroactive actuators.

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ASSOCIATED CONTENT

Table S1. Naming convention of peptide sequences which were considered in this work.

Peptide Sequence	Amino Acid Notation
Tyr3-Ala5	H ₂ N-Y-Y-Y-A-A-A-A-A-CO ₂ H
Tyr3-Ala7	H ₂ N-Y-Y-Y-A-A-A-A-A-A-A-CO ₂ H
Tyr3-Ala9	H ₂ N-Y-Y-Y-A-A-A-A-A-A-A-A-A-CO ₂ H
Tyr3-Ala11	H ₂ N-Y-Y-Y-A-A-A-A-A-A-A-A-A-A-A-CO ₂ H
Ala5-Tyr3	H ₂ N-A-A-A-A-A-Y-Y-Y-CO ₂ H
Ala7-Tyr3	H ₂ N-A-A-A-A-A-A-A-Y-Y-Y-CO ₂ H
Ala9-Tyr3	H ₂ N-A-A-A-A-A-A-A-A-A-Y-Y-Y-CO ₂ H
Ala11-Tyr3	H ₂ N-A-A-A-A-A-A-A-A-A-A-A-Y-Y-Y-CO ₂ H
Phe3-Ala-5	H ₂ N-F-F-F-A-A-A-A-A-CO ₂ H
Phe3-Ala7	H ₂ N-F-F-F-A-A-A-A-A-A-A-CO ₂ H
Phe3-Ala9	H ₂ N-F-F-F-A-A-A-A-A-A-A-A-A-CO ₂ H
Phe3-Ala11	H ₂ N-F-F-F-A-A-A-A-A-A-A-A-A-A-A-CO ₂ H
Ala5-Phe3	H ₂ N-A-A-A-A-A-F-F-F-CO ₂ H
Ala7-Phe3	H ₂ N-A-A-A-A-A-A-A-F-F-F-CO ₂ H
Ala9-Phe3	H ₂ N-A-A-A-A-A-A-A-A-A-F-F-F-CO ₂ H
Ala11-Phe3	H ₂ N-A-A-A-A-A-A-A-A-A-A-A-F-F-F-CO ₂ H
Ala1-Tyr3-Ala4	H ₂ N-A-Y-Y-Y-A-A-A-A-CO ₂ H
Ala2-Tyr3-Ala5	H ₂ N-A-A-Y-Y-Y-A-A-A-A-A-CO ₂ H
Ala3-Tyr3-Ala6	H ₂ N-A-A-A-Y-Y-Y-A-A-A-A-A-A-CO ₂ H
Ala4-Tyr3-Ala7	H ₂ N-A-A-A-A-Y-Y-Y-A-A-A-A-A-A-A-CO ₂ H
Ala4-Tyr3-Ala1	H ₂ N-A-A-A-A-Y-Y-Y-A-CO ₂ H
Ala5-Tyr3-Ala2	H ₂ N-A-A-A-A-A-Y-Y-Y-A-A-CO ₂ H
Ala6-Tyr3-Ala3	H ₂ N-A-A-A-A-A-A-Y-Y-Y-A-A-A-CO ₂ H
Ala7-Tyr3-Ala4	H ₂ N-A-A-A-A-A-A-A-Y-Y-Y-A-A-A-A-CO ₂ H
Ala1-Phe3-Ala4	H ₂ N-A-F-F-F-A-A-A-A-CO ₂ H
Ala2-Phe3-Ala5	H ₂ N-A-A-F-F-F-A-A-A-A-A-CO ₂ H
Ala3-Phe3-Ala6	H ₂ N-A-A-A-F-F-F-A-A-A-A-A-A-CO ₂ H
Ala4-Phe3-Ala7	H ₂ N-A-A-A-A-F-F-F-A-A-A-A-A-A-A-CO ₂ H
Ala4-Phe3-Ala1	H ₂ N-A-A-A-A-F-F-F-A-CO ₂ H
Ala5-Phe3-Ala2	H ₂ N-A-A-A-A-A-F-F-F-A-A-CO ₂ H
Ala6-Phe3-Ala3	H ₂ N-A-A-A-A-A-A-F-F-F-A-A-A-CO ₂ H
Ala7-Phe3-Ala4	H ₂ N-A-A-A-A-A-A-A-F-F-F-A-A-A-A-CO ₂ H